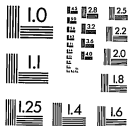


0 10 20 30 40 50 60 70 80 90 100 110 120
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Thomas A Edison Papers

A SELECTIVE MICROFILM EDITION

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(1887-1898)*

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START

100

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Notebook, N-90-01-04.2

This notebook was begun during January 1890. Most of the entries are by Edison. There are also some entries by an unidentified assistant. The book contains lists of mines, mainly in North Carolina, but also in Georgia, the New England states, and the Great Lakes region. Among the ores noted are gold, copper, and iron. Also included are notes by Edison to surveyors Samuel G. Burn, Arthur C. Payne, and Charles J. Reed regarding mining regions in the northeastern United States. The front cover is labeled "#585." The inside front cover is inscribed "#585 T A Edison Jan 4. 90." The pages are unnumbered, and the book has been used in both directions. Approximately 70 pages have been used.

585

XE-172
N-90-01-09.2

St. Edin
Jan 4.90

Rudisal Mine - Gold & Cop -
Saled pyrite of Cu with Au at water line
3 @ 4 ft 1/2 mile - Double vein which
if united at 600 ft make mineral vein

Northampton Mining Dist
Mass - Vein 3 miles long
Cu pyrite Blender head with Au
6 @ 12 ft thick - Enough ore for
ages ~~###~~ It is Massachusetts

Mining Mag Vol 2 634 p.
Lordsville Mine big mine same Dist

North Carolina Cop mine - excerpt
in 1834 says best in US -
Mining Mag Vol 2 = 77

Chatham Nickel + Cobalt
Mine Chatham Conn
Unlimited quantity or
cheaply mined 2 @ 3 pct.
run of mine - been wkd -
Munseyllaq Vol 2 p 113

Middletown Conn Lead Mine
+ at depth Copper mine
3/4 mile long 5 ft 4
inches from Chatham
mine. - Contains Nickel
as well - 7503 Silver ton
ton less

Vanderburgh Mine NC
immense deposit gold
Mining Map Val 2 310-11-
p. 10

Lancaster Goldmine
Georgia Federal Union
Co - 3 miles long 2 1/2 miles
side walks rich gold
Mining Map Val 2 309

Plymouth Copper Mine
Township of Plymouth
Connecticut - Run 3/4
mile long
Mining Map Val 2
316

Steele Mine - Montgomery Co NC near Uwharrie
River

Henderson Mine Montgomery Co NC near
Uwharrie River

Davis Mine Union Co 6 miles south
of Matthews on Carolina Central Railway
Dyers, Eliffer & Lewis mines are close together.
Bedded veins - not yet shown to be in commercial value
exists quantity - gold on 400 fine - Ven matter
Galena with Silver - Gold -

Hemby Mine - just north of Lewis Mine - Galena
in gold deposits which are numerous.

Smart Mine - Union Co NC - 4 miles nearly
Southeast of Matthews + about 1 mile south of
C C R R - Galena in relatively large quantities,
resources mine unknown by reason want home market
for Galena - Mine was worked as gold mine + Galena
was objectionable - abandoned - Ore body proved
depth 110 ft 3 levels. 35 60 95 ft - each 60 ft
length J C Bates Esq of Monroe has information about
underground work - Vein 3 @ 4 feet thick - ore
seam considerably less - Ore Galena - pyrites in
Quantity Matrix - Assay No 1 $\frac{9}{10}$ oz gold
 $\frac{3}{10}$ Silver No 2 - $2\frac{1}{4}$ oz gold 4.5% Silver -
parties claim average value over 30 parts $\frac{1}{2}$ Ag $\frac{1}{2}$ Au -

Lemmons Mine - Union County 2 miles NE of
Smart, sometimes called Marion Mine. gold silver
galena -

Stewart Mine - Union Co. NC.

Moors Mine. Union Co. $1\frac{1}{2}$ miles NE of
Stewart,

Long or Crowell mine same neighborhood as
Moors mine.

Rocky River Mine near Rocky River in
Cabarrus Co 10 miles nearly SE from Concord

Allen Furr Mine. or (Silver Valley) 11 miles.
Southeast. Concord + same neighborhood as Rocky
River Mine.

McMakin Mine. in SW of Gold Hill Dist. in
Cabarrus Co.

Trautman Mine Cabarrus Co. N.E. 1 mile.
SE of Gold Hill

Silver Hill Mine. Davidson Co. N.E.
10 miles SE from Lexington + Silver Valley
5 miles NE from Silver Hill. + near border
of Randolph Co.

Silver Valley Mine 5 m NE Silver Hill

Wellborn Mine. Davidson Co. 2 m W of
Silver Hill.

Steep Mine. - Montgomery Co. 10 @ 12 m
E of above belt. It is on east side of
Wharris River 2 m SE of Eldorado Village

Henderson Mine near Eldon's Village
Montgomery Co NC

Baryte Mine along east flank of King
Mountain Range from Crocoders Mountain
to Yorkville SC.

Flint Knob Mine at Flint Knob, Wilkes
Co NC West part of County 6 mi E of
Deep Gap of the Blue Ridge not far from
both Watanga & Abbe Cos

Queen Cedar Cove & Dobson Mines
near the east foot of Mountain
in McCowan Co -

Daker Mine Caldwell Co NC 12
miles nearly N of Morgantown E side
Johns River -

Peach Bottom Mine (or Maxwell) in
Alleghany Co NC

Copper Mine CM N Carolina -

Hodges Mine. Guilford Co. near Fisher Hill.

North Carolina Mine or Fortress Mine.

Gardner Hill Mine. Guilford Co NC
3 m nearly E of Jamestown

Twin Mine. Guilford Co 6 m SE of
Greensboro.

cm

North State Mine. or McCulloch Mine.
Seielford Co 2 m S slightly E from
Gambelton —

Jack's Hill Mine. $\frac{1}{3}$ mile beyond
Northstate.

Phoenix Mine. SE part Cabarrus Co NC

Pioneer Mills Mine SE part Cabarrus Co

CM

Morris Mine 1 mile from Pioneer Mine.

Crosby (of Poplar) Mine same vicinity

Rogers Mine same vicinity

The Ray mine.

Ferris Mine

McGuire Mine
5 mi W Charlotte

} Mecklenburg Co

Kerns, or (Hopewell)

11 miles NE Charlotte

By C Cathey Mine's

Mo Leary Mine's

Cathey Mine's

5 miles SE Charlotte

Mocklenburg Co

Dunn Mine: 90 miles N of W. from Charlotte

Conrad Hill Mine. Davidson Co 7 miles a
little S of E from Lexington.

Gold Hill Mine. Gold Hill Rowan Co.

Spencer Copper Mine. Randolph Co

Spain Mine. at Fork of Deep & Rocky Rivers
1 mile from each Randolph Co

Clegg Mine - Chatham Co.

Emmons or (Davidson Mine) Davidson Co.
7 miles E of Lexington.

Cid Mine. $1\frac{1}{4}$ miles NE of Emmons Mine

Foust Mine: Alamance Co. South foot of
Bass Mountain.

Chick Mine, Chatham Co - 1 mile
N of Deep River

Phillips Tract Mine; tool hole - $2\frac{1}{2}$ m NE
of Chick. Chatham Co -

Big American Mine (Registers)
Granville Co - $\frac{1}{2}$ mile E of Blue Wing

Luck mine near Big American

Silver Nugget Mine. near Big American

Halloway Mine. $2\frac{1}{2}$ m a little W of
D from Big American Mine -

Maestron Mine (alcalantia) is near
Granville - Person County Line.

2) Poole Mine. Person Co $\frac{1}{4}$ mile nearly
W of Maestron Mine.

Buckeye Mine. near Poole Mine.

Gillis Mine. 1 mile S of Buckeye

Copper World Mine. Person Co. $1\frac{1}{2}$
m a little W of S from Gillis Mine. or Gillis -

Yancy Mine. Person Co. $1\frac{1}{2}$ miles nearly
South of above.

Burrell Wells Copper vein - Gaston Co
2 m S of Tuckasee Ford on Catawba
River

Graham Copper vein - Lincoln Co - nc
on farm Maj W A Graham - 4 miles NE
from Iron.

Waryhnt Mine. - Jackson Co on
Waryhnt Creek. 6 m SE of Webster.

Cullowhee Mine. on Cullowhee
Mountain SE of farm.

Savannah Mine 9 miles SW of
Webster on Savannah Creek - Jackson Co.

A Copper lode - Northern part Haywood
Co near Wilkins Creek

Copper Outcrop - near head Big Gray Creek.
not far from Madison + Buncombe Co line.
see C. D. Smith Review - write Essay on
Geology NC - seems to know lot - might say
himself

Elk Knob Copper Mine -
NE Slope of Elk Knob Mountain.

Mittell mine Southern Base Mountain

Copper Knob mine (Gap mine) SE
part of Sta. Co. 3 m W of Summit of Blue
Ridge 1 1/2 miles E of New River -

Ore Knob Copper Mine - most remarkable vein
in state - situated at Ore Knob in S.E. corner
of Ashe Co near Top Blue Ridge Mt. + 2 miles from
New River -

gold mines

~~Mine - NE New working - P~~

Portis Mine: NE Cor Franklin Co. near to
Nash & Warren - on hill 108 ft above Shocco
Creek -

Arrington Mine Nash Co. 1 mile SE.
of Portis

Thomas Mine - Kearney - Taylor -
Mann - & Davis Mines in Halifax Co

Nick Arrington Mine 12 miles E of
Portis -

Bell Mine Moore Co - 10 m NW of
Cathage. + $\frac{1}{4}$ mile W of Red Sandstone

Franksville Mine Moore Co
3 m SW of Bell Mine -

Brown Mine Moore Co on NW Edge
of the Gold District - on the Road from Elhoffetts
to Richardson's Mill -

Bat Roost Mine

Shields Mine

} same vicinity

Cagle Mine Moore Co. 2 or 3 mi
S E of Brown & Bat Roost.

Clegg Mine Moore Co. $\frac{1}{4}$ mi W of
Cagle on opposite side of Cabin Creek

Burns Mine - Moore Co. $\frac{1}{2}$ mi. S of
Cagle.

Moore Mine - North East of Reynolds Troy
Montgomery Co NE

The Reynolds Mine. 6 mi NE of Troy

The Carter Mine a little nearer Troy

San Christian Mtn. 5 W of Troy
Allegheny Co.

Russell Mtn. Montgomery Co.

Morris Mountain Mtn. or (Diasis)
or Ditten - NW part Montgomery Co.
1 m nearly W of Coopers

Appalachian Mtn. or (Coopers)
NW part Montgomery Co. 1 1/2 NW of
Steel

Steel Mine & Scumdera E side of the
Wharris Riv. & 2 m SE of Eldorado
Village

Beaver Dam Mine. Flagtown P.O.
2 m NE junction Beaver Dam Creek &
the Yodkin River

Griffin Mine. $1\frac{1}{2}$ m NE of Russell &
Nall, near Stokes Ferry

The Wharris Mine short distance
north east of the Russell Mine just
over the line in Randolph County

Wilmington and Slack Mines
Randolph County two and a half
miles south of Ashboro

Dani Mountain Mine, Randolph
County four miles southwest of
Ashboro

Sanger Mine, Randolph County
five miles northwest of Ashboro.

Irishboro Mine, Randolph County
five miles south west of Ashboro.

The Jones or Keystone Mine, Randolph
County, twelve miles a little East of
South from Thomasville and near
Davidson County line.

The Parish Mine, Randolph County,
Same place as above.

The Hooper Hill Mine, Randolph
County, seventeen miles a little East
of South from High Point.

The Wilson Hurdley Mines Randolph
County, half a mile southwest from
Homer Hill

Hearn's Mine, Stanley County, two
miles north west of Albemarle.

Haircock Mines - Same place as
above.

Parker Mine, Stanley County, at
Bilesville, ten miles southeast
of Gold Hill -

Flint Spring mine, same place as
above

Crowell Mines - same place as
above -

Barringer Mines, Stanley
County, four miles Southeast
of Gold Hill -

Howie Mine, Union County
1/4 miles N. W. of Washington

Washington Mine, Union Co.
eight miles distant a little
S. of W. from Monroe.

Danis Mine, Phiffer Mine,
Lewis Mine, Herby Mine,
Union County, near Washington
Mine -

Woodhill Mine situated
1 m. S. E. of the Danis mine -

Folger Hill Mine half mile
NW of the Danis

Harkness Mine $\frac{1}{2}$ m., E. of
the Lewis Mine -

Crouse Mine, Union County
four m. S. from Stout's
Station on Carolina Central
Railroad

Stewart Mine towards the
northern part of Union
County 15 m. nearly S. E.
of Charlotte -

Reed Mine, Cabarrus County
10 m. S. E. of Concord.

Rocky River Mine, Union
County, ten m. nearly South
E. of Concord not far distant
from the Reed Mine, it
includes the Jake Skin and
Tom Skin mine

Gold Hill Mines 14 m.
South E. from Salisbury in
the South E. part of Rowan
County and the North East
part of Cabarrus County.

Randolph Mine, Gold Hill
district,

The Barnett, Gold Hill district

Old Field Mine - Gold Hill
district

Krautman Mine, Gold Hill
district

Mc Mackin Mine, Gold Hill
district -

Hammicut Mine, Gold Hill
district

Conrad Hill Mines, Davidson
County about 7 M. nearly E.
from Lexington -

Hard Mine, Davidson County
1 M., W. of Delk -

The Hamilton or Bailey Mine
Davidson County 2 M., S. E. of
Hadesboro -

Jessie Cox Mine, Anson County
1 M., S. E. of Wadesboro.

Lalor Mine, Davidson County,
two M., S. E. of Thomasville.

The Loftin Mine, Davidson
County - $1\frac{1}{2}$ M., S. E. of Thomas
ville.

Eureka Mine - $\frac{1}{2}$ M. W. of the
Lalor Mine.

Jadwin Mine Rowan County -
 $\frac{1}{2}$ M. S. of Salisbury.

The Dunns Mountain Mine
4 M. S. E. Salisbury.

P.

Rimmer Mine, Rowan County,
5 M. S. E. Salisbury -

Bullion Mine - $\frac{1}{2}$ M., E.
of the Rimmer Mines -

Gold Knot Mine, 9 M., S. E.
Salisbury -

Haynes Mine - Same Place

Holtshausen Mine - Same place

Dutch Creek Mines - Stokes
Ferry Road - 10 M S. E. of
Salisbury and just to the S. E.
of Dutch Second Creek

Atlas and Rome Mines - near
the above.

Quaker City Mine - Cabarrus
County 5 M. a little S. of E.
from Concord.

Tucker Mines 8 M., S. E. of
Concord.

Phoenix Mine. Same place.

Dandson Hill Mine. Mecklenburg
County 1 M. west of Charlotte.

The Point Mine, Rowan Co.
Same Place.

Rudisil Mine: 1 M. from
Charlotte.

St. Catherine Mine $\frac{1}{2}$ M.,
S. of Charlotte.

Blake Mine, Parkes Mine,
Clark Mine, Smith & Palmer
Mine, McDonald Mine, Wilson
Mine, Horrell Mine, Trotter Mine,
Carson Mine, Taylor Mine, John
Lynch Mine, all in the
vicinity of Charlotte.

Hayes Mine - Wallace Mine,
Bradley Mine, Fraser Mine,
Ship Mine, Campbell Mine,
Todd Mine, Arlington Mine
Capps Mine, Mc Levin Mine,
Stephen Wilson Mine, Frost
Frautman Mine, Peun Mine,
Abernathy Mine, Alexander
Mine, Dunn Mine, Sloan
Mine, Mc Coubert Mine, Cathy
Mine, all within 5 to 10 M.
W. and N.W. of Charlotte.

Ferris Mine, 6 M., N. of
Charlotte.

Hunter Mine, Fredrick Mine
Ray Mine, at Providence
Worship about Sadies Church
Some 5 to 10 M., S. E. and E.
of Charlotte.

Johnson Mine, Stinson Mine
Maxwell Mine, Black Mine,
Harris Mine at Ponce de Villa
Cabarrus County extending
into S. E. part of Mecklenburg
County.

Baltimore and South Carolina
Mining Co's. Mine - Mecklenburg
County, 9 M., S. E., of Charlotte
and 1 M. from Matthews.

The Oliver Mine, Gaston
County, 12 M. N. W. of
Charlotte on W side of Cata-
wamba River.

Farrar Mine $\frac{1}{2}$ M beyond
Oliver Mine.

Rhyme Mine, 17 M. W. of
Charlotte.

Derr Mine - Same as Rhym

Rhodes Mine, Gaston County,
18 M., S. W. of Charlotte

Duffie Mine - Gaston County
16 M. from W. from Charlotte
on the Tuckasegee Road.

Robinson Mine, adjacent to
Duffie -

Smith Mine - 13 M. W. of
Charlotte:

~~Saw Betty's~~

Saw Beattie Mine, just south
of the Smith Mine.

Long Creek Mine, near the
the Smith Mine.

Burrell-Gills Mine $3\frac{1}{2}$ M
S. of the Duffy

M^cLeans Mine, Gaston County
15 M., S. W. from Charlotte.

Catawba on Kings Mountain
Mine, Gaston County, $1\frac{1}{2}$ M
nearly South of Kings Mountain
Station on the Atlanta and
Charlotte Airline railroad.

Coopers Mountain Mine, 4 M
East of Catawba Mine on the
E. side of Kings Mountain.

Patterson Mine, Easton Co.
1/2 N.E. of Crowder Mine.

Hoke Mine, Lincoln Co. 4
miles from Lincolnton.

Shuford Mine, Catawba Co.,
4 1/2 M. slightly S. of E. from
Catawba Station on the W.N.
C. Railroad and 6 M. W. of
Sherrille Ford.

Butter Mine, Davie County
8 M., S. W. of Mocksville.

The Isaac Allen Mine 1 m.
N. W. of Mocksville.

Clarksville Township Mine
7/2^{mi} N. W. of Mocksville.

Fulton Township Mine
7 M. N. E. of Mocksville

Barnes Mine, Alexander Co.,
8 M. N. of Taylorsville -

Wicks Mine, Caldwell Co. 2 M
E. of John's river - Same place
Michaux Mine, Page Hill
Mine, Corpey^{ing} Mine.

Mine worked in NC

[THIS BOOK WAS USED IN BOTH DIRECTIONS.
THE FOLLOWING PAGES WERE FILMED FROM
THE BACK END FORWARD.]

THOS. A. EDISON,
ORANGE, N. J.

Iron Ore Hunters Island - extension
Vermilion Range into Canada -

Longest of the Plate Islands -
Lake Superior -

Iron Cap - near Uchiapikatan -

North of Balchawana Bay outlet -
Bell location - larger granites
at head of Pancake River

Township of Dalhousie, Lanark Co.
and adjacent regions -

Rail Portage E side of Winnipeg River
bet Lake Woods + English River -

Blairton Iron Mines Lot 8
1st Concession of Belcourt in
Co of Pelly - map 130

Glendonville Mine Range
20 miles long. Township of
Bedford Co of Frontenac
4 m E of Kingston & Pembroke
R.R. crossed by a S.W. rd.

Calabogie Mine. Mine No 4.
Lot 16 in 8th Concan of
Bagot 1 1/2 m direct line
E. of Kingston & Pembroke R.R.

Haliburton Mines -

Howland Mine. Lot 26,
Concan 4 of Snowden -

p. 131

Pine Lake Mine - p 132

old Geology 1863 -

p 33 34 p 35 - p 37

p 245 important p 247 246 -

248 249 251 :

Chrono 266 - do 270 464

472 - 489

important 501 502

505 506 - nickel - Elzibitum -

515 - 609 613 615 -

616 = ~~673~~ 673 674

675 676 677 678

737 738 - nickel 749 - chrono

750 Cobalt = 754 -

Barris p 798 - 799
802 - 803

See N-90-07-12
Page 127. →
Fried

Notes for Paine -
Haversham Mining Co - p 220
Mining Map Val 2 750 acres
West bank of Hudson 5 mi from Rye
formerly named Horsehoe River
mine - workable, mined for
25c - magnetic

= Burns = p 552 553

Mining Map Val 1

Paine = p 715 of US geol. S.
Rep 1887

Burns = p 739 USG Rep 1887

p 741 chromite - do p 742

Paine = p 745 = 759 -

Reid - p 762 - Geol US 1887
p 762

Paine 768 - US Geol Rep 1887

Burns 779 - Chromium

Paine Connecticut.

Geology. by Percival - get
Shepherds = p 16 - 97

p 98 114 118 122

124 135 137 161

170 183 197 203

235 280 473

Paine Mass - Geology of Mass -

52 - p 53 54 55

399 296 347 348

359 360 361 363

369 - done

Burns Tysens 2nd Rep
Geology Maryland

p 55 56 - 66 Chrom.

67 68

50 -

50,

75° -
24.
8.

250,

107.

60

75

25

25

292.

160

12.

12.

476.

71.

8

568

476

92.

71

710

476

234.

Notebook, N-90-01-04.3

This notebook covers the period January-December 1890. All entries are by Edison. At the beginning of the book is a list of iron mines in New York, New Jersey, and Pennsylvania. The book also contains drawings of the phonograph and the electric railway; notes regarding the technical and commercial development of the electric light, including notes on the contractual arrangement between the laboratory and the manufacturing company (possibly the Edison General Electric Co.); lists and drawings for an ore milling operation; and an enumeration of salaries and other operating expenses at the West Orange laboratory. There are also hypotheses on conductivity and resistance and notes on filament experiments. The front cover is labeled "#581." The inside front cover is inscribed "#581 T A Edison Jan 4, 90." The pages are unnumbered, and the book has been used in both directions. Several pages have been removed from the middle of the book. Approximately 125 pages have been used.

581

XE-172

N-90-01-04.3

La Edison
Jan 4, 90

Iron Mines etc

Potter Mine $4\frac{2}{3}$ miles N 40°
W from Fort Ann near Lake George
Washington Co. N.Y.
Operated John T. Harris 1880
who raised 12 172 tons sent to Canal
8 miles at Fort Ann & thence to Pa
from description in Vol 15 - US Census Report
on mining ~~to~~ p 105 - immense body
lean ore on side walls
Mt Hope Mine $\frac{1}{2}$ mile NW of Potter
worked 1881 by W.P. Ostrander & Co

p 108 says Immense masses titaniferous
ore near old Village of Adirondack
and also near the lake at in Westport
Township & elsewhere - See Comments
in Geology 2nd District p 244

O.V. 21

also properly Reprinted from 32nd Annual
Rep State Museum Natural History
Albany March 1880 by Chas. E. Hall.
describes these deposits. He says
Juraniferous *ore* belong to upper
Lamentian (Liberadite) series while
Magnetite *ore* belong to lower
Lamentian.

New Deep Mine Italian Island
Serpentine = Limonite, contains:
39 pct. Fe - 1.91 Seignol Chromium
0.55 Nickel Sulphide 0.17 Cobalt
Sulphide

Northampton Mine Pa - (Limonite)
has 0.14 Seignol Cobalt
most all ^{is} from 0.1 to 0.5

Beyond Cornwallis Mine in Laurent
is Dillbury Mine. near Cornwall
is French Creek - St. Marys ^{mine} PO - Jones
mine, Wheatfield Mine, Fritz Island
mine - all magnetic.

NY Mine owners or lessees.

Crane Iron Co

Randall Hill -

Dulyrville -

Kispaugh
Hill.

Glendon Iron Co

Hibernia

Leabo

Cooper, Hewitt & Co.

Union Church

Sanson

Charlotte Singh

Darien post

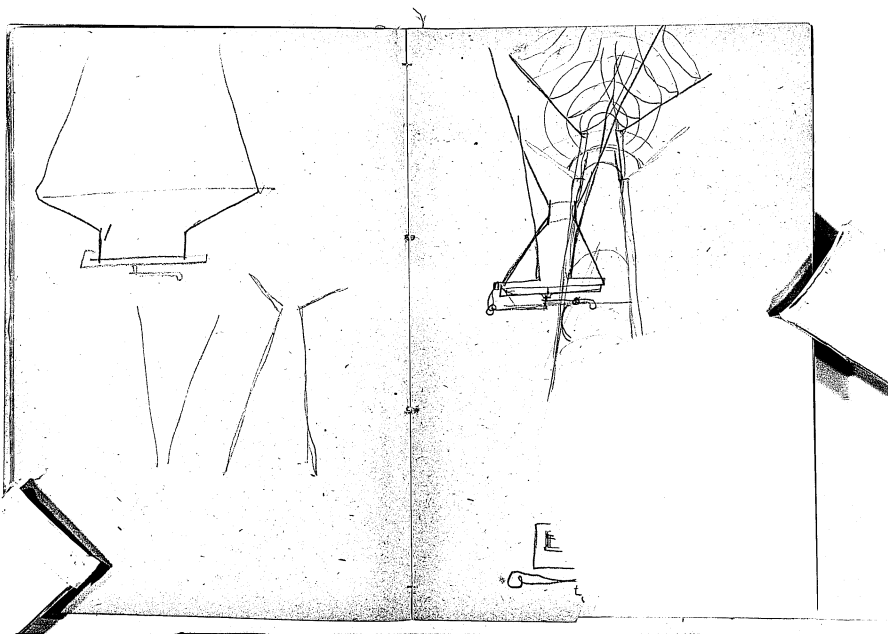
Pitney

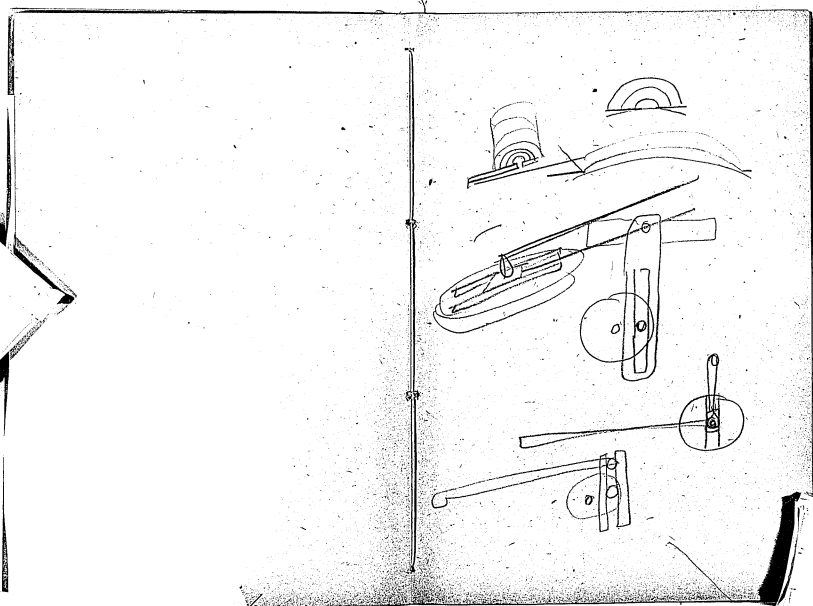
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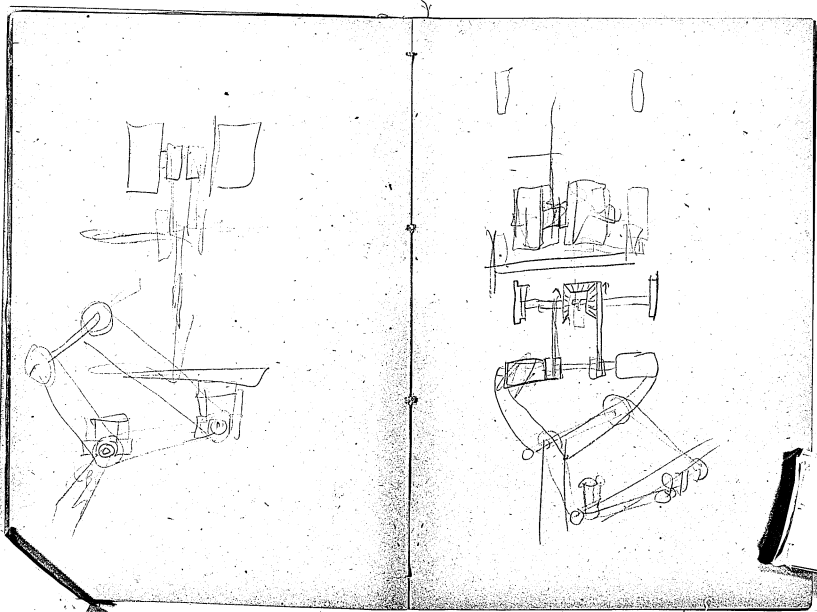
William Carpenter Minn.

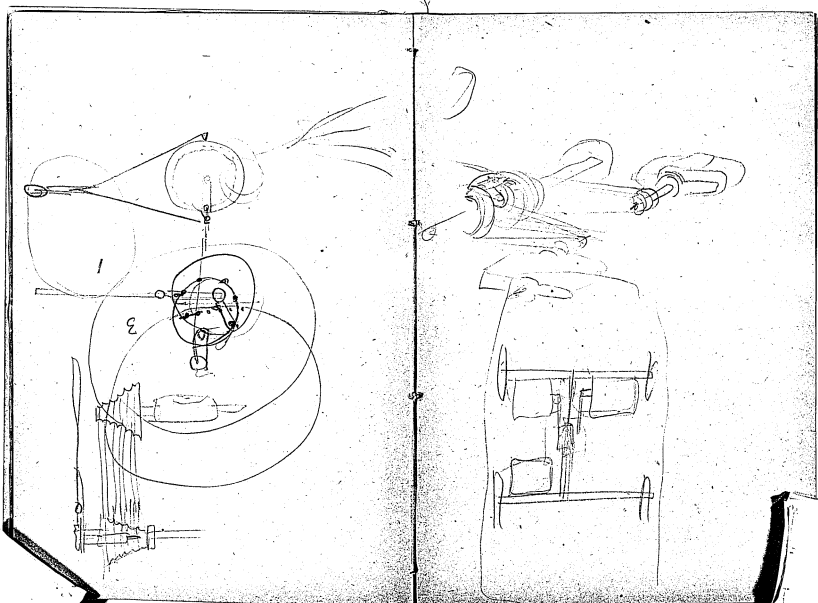
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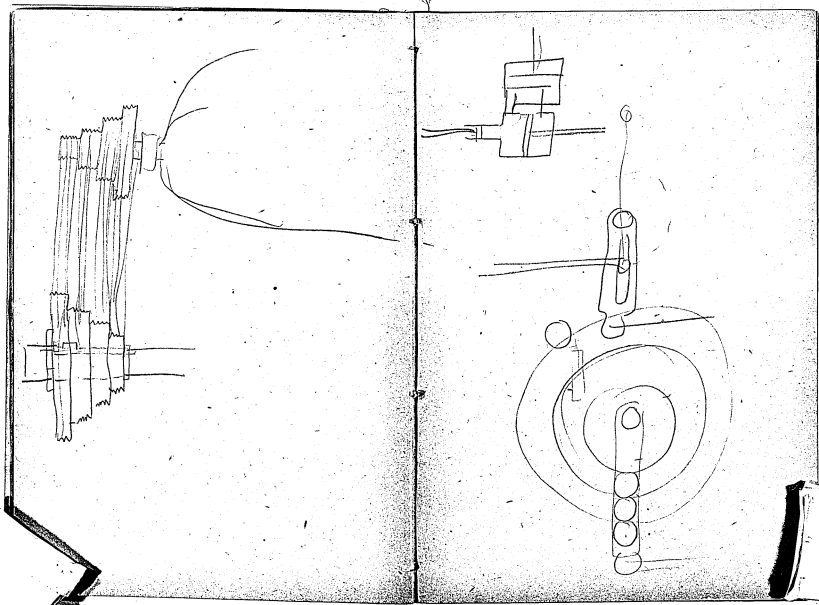
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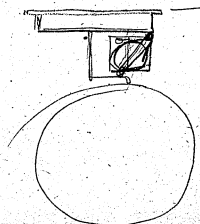
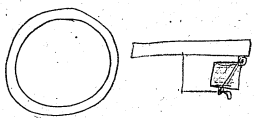
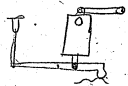
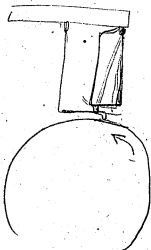
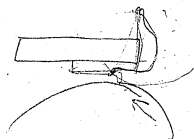


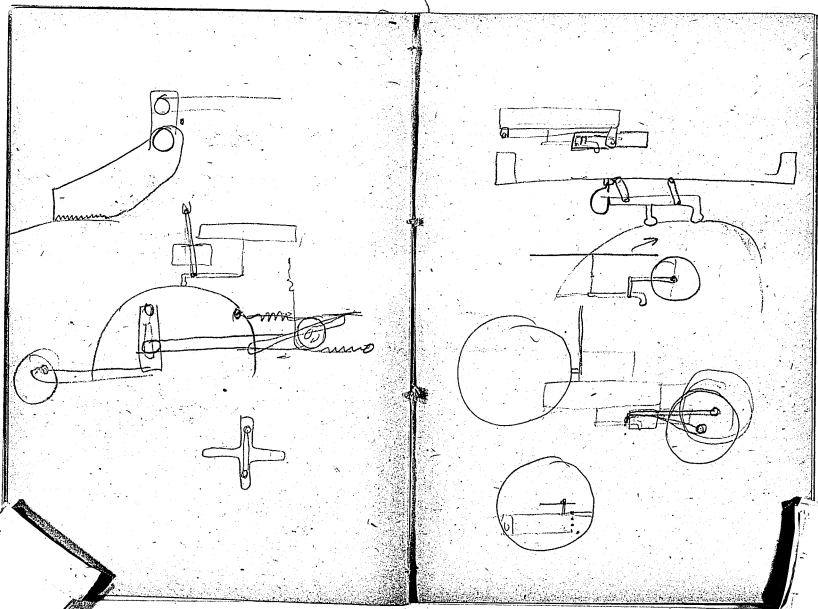


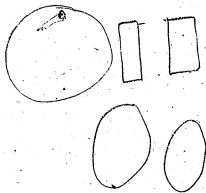
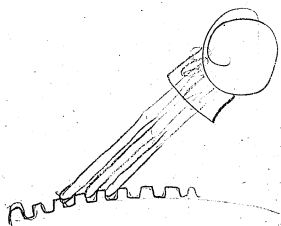


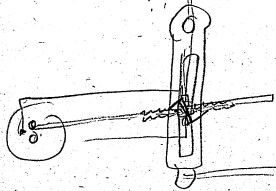
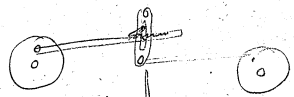
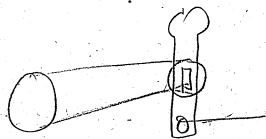


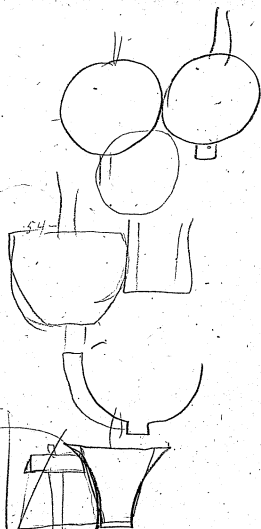
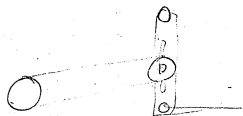


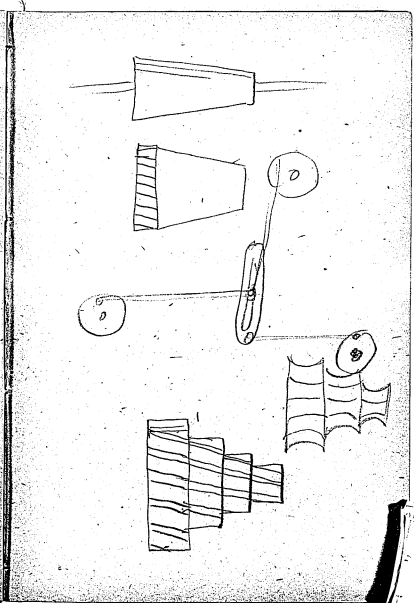
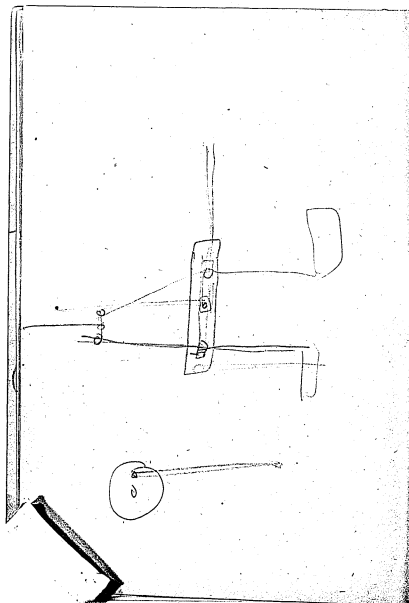


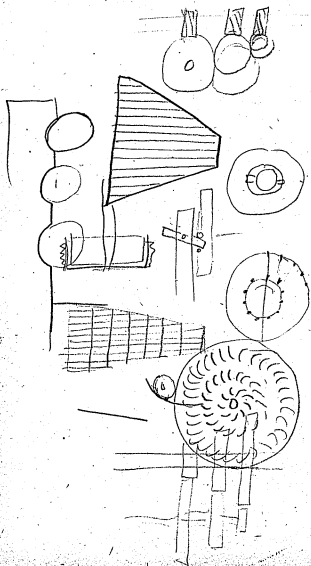
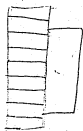
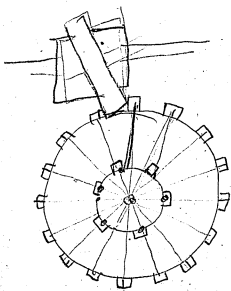


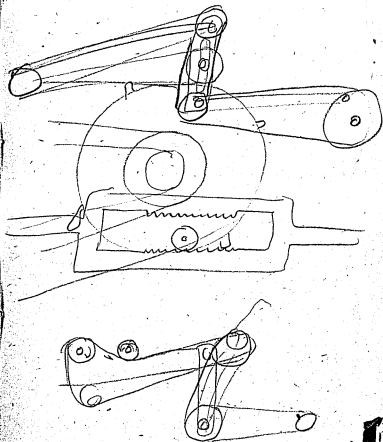
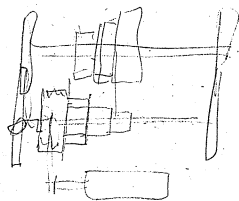


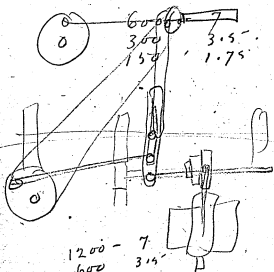












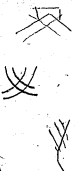
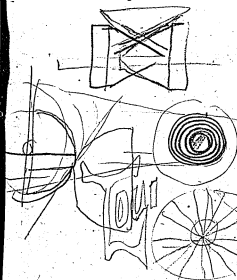
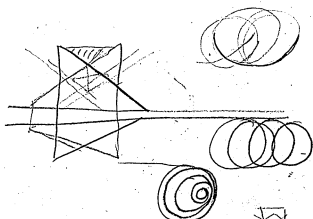
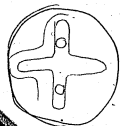
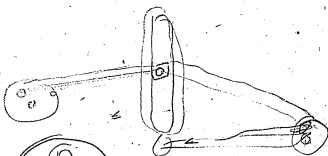
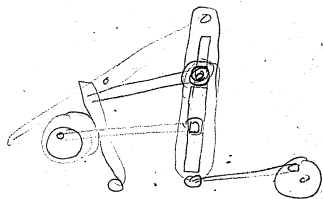
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 600 3.5
 3 1.75
 150 - .85

12 - 6 - 3 - 1.50

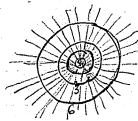
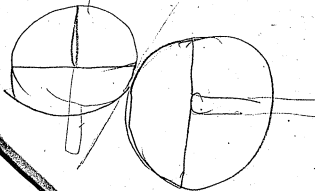
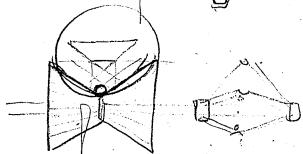
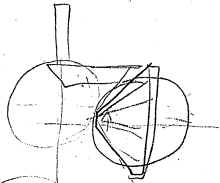
1200
 .30
 2 | 36000 | 3

12
 3000
 36000

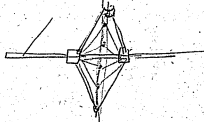
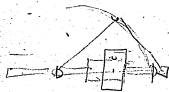
3.



Wachung



6.3.185



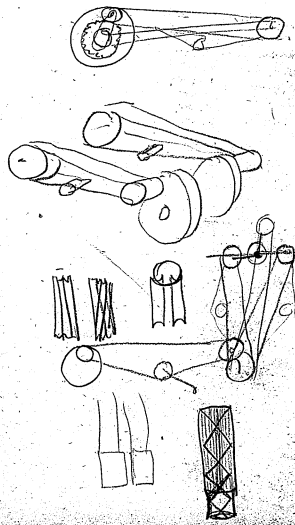
200 16
200

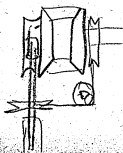
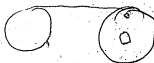
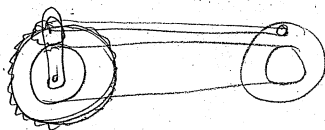
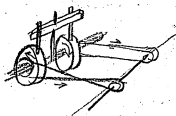
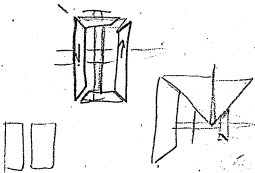
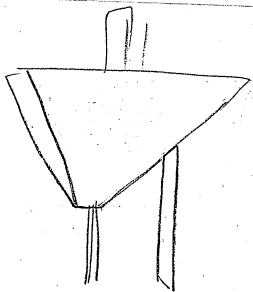
Kennelly = 15.00
 Boy - 3.50
 Office boy - 2.00
 Ebdell - 5.00
 2 watchmen \$ 1. 5.00
 Frady - 6.00
 Frady - 15.00
 John Oll - 12.00
 Fred Oll - 12.00
 Kayser - 11.00
 Goward - 12.00
 Wurt - 12.00
 Thomas - 12.00
 Walker - 11.00
 Engineer - 12.00
 Alwood - 8.50
 3 workmen - 13.50
 Chemist - 4.50
 Dixon - 15.00
 Shellyberg - 5.00
 Aylesworth - 9.00
 Kloor - 4.50

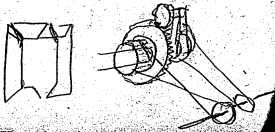
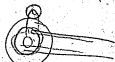
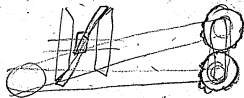
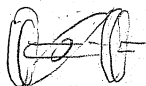
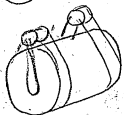
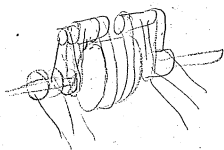
16 | 54
 71360 0 2
 4 5 0 0 0
 2 8 6 0 0
 3 3 0 0 0
 1 1 0 0 0
 3 1 0 0 0
 3 1 0 0 0
 1 0 0 0 0
 3 4 1 0 0

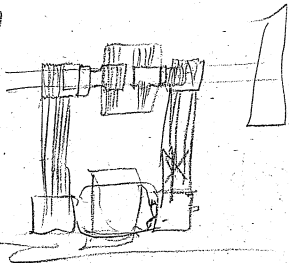
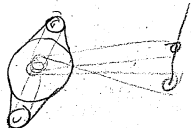
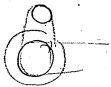
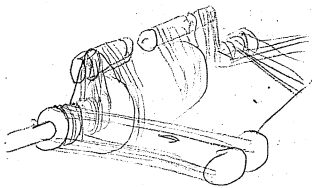
Cool -
 Tax -
 gas -
 Mfg. m. -
 Sup. pte. -
 Conf. -

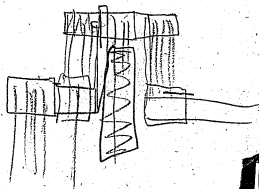
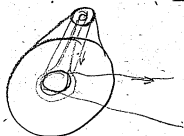
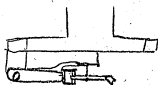
12 80
 968
 768

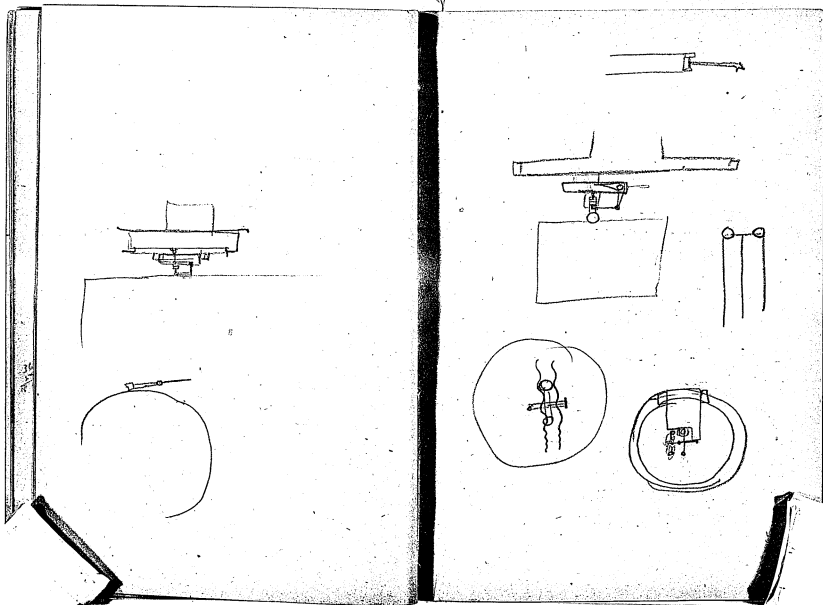


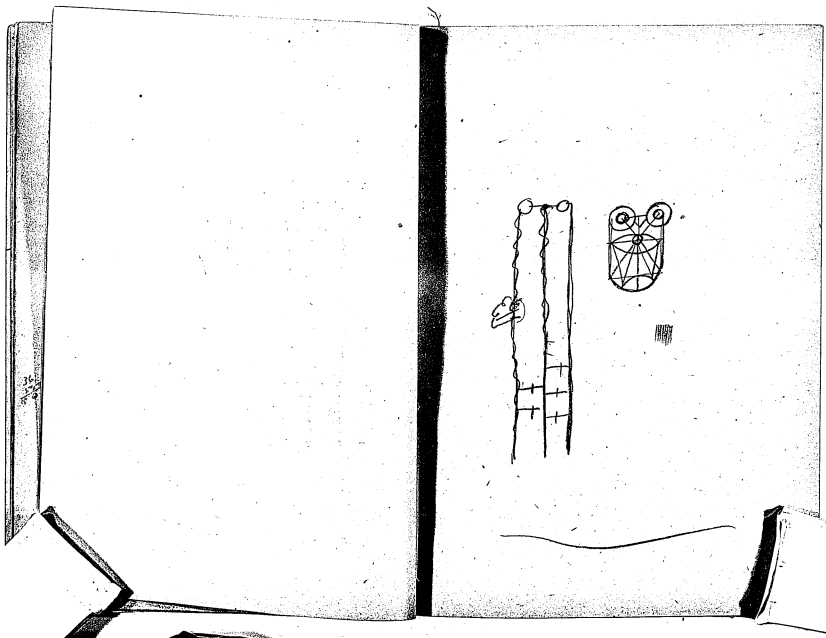




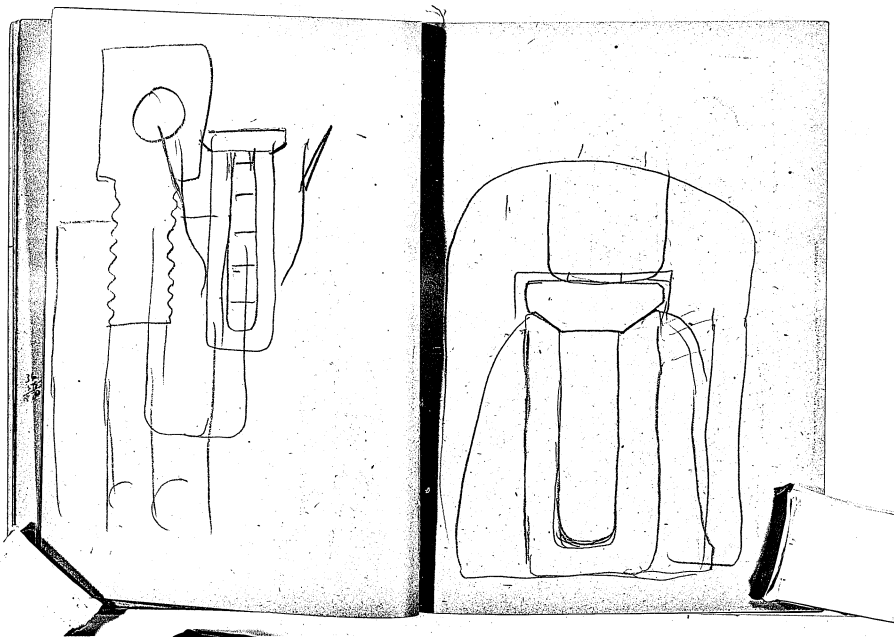








16
2
8





$$\frac{16}{\frac{20}{3.2}}$$

Construct Dynamos for Compulsion
purpose at $\frac{1}{2}$ the Cost of present
Machines -

Increase the present Lamps from

240 C per horse power to 320

without any more cost ^{per Lamp}

+ capable of used with the ~~present~~

~~of present~~ as the present 240 lamp 20 pct

less copper per lamp than ~~present~~

To Eradicate from the present

Lamp without diminishing its

life or economy the defect of ~~gross~~

drop loss of CP to an extent ~~is~~

greater than any known Lamp

and in the case of 20 per hp

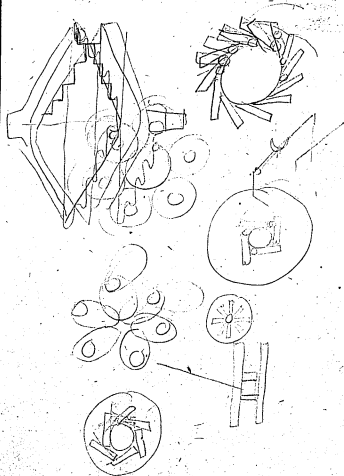
the average CP. for 1000 hours

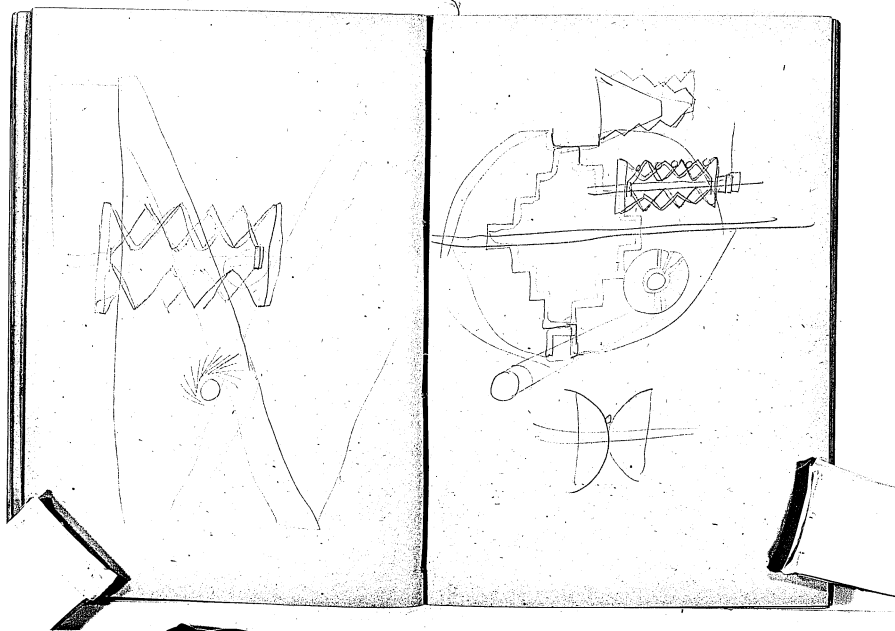
not to be exceeded by any
known Lamp - higher than
~~is~~ higher than 14 per hp -

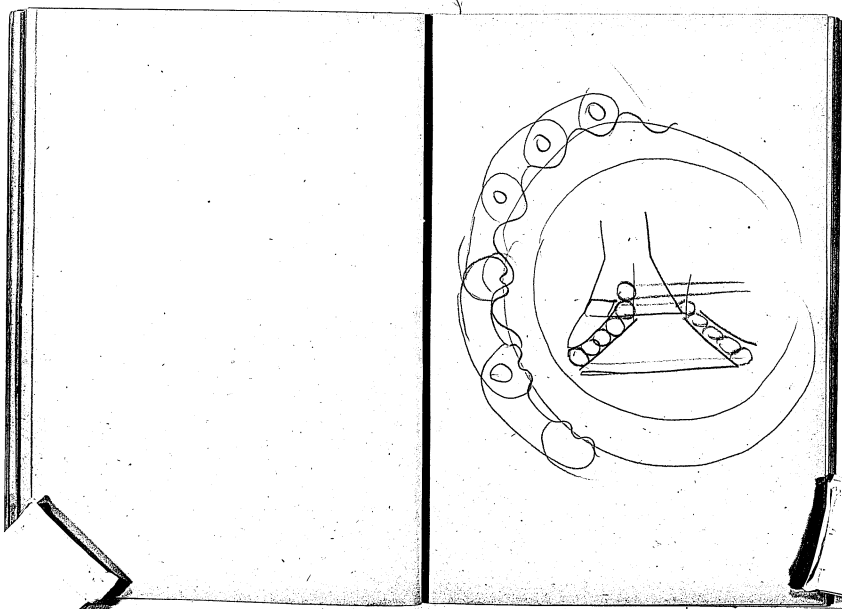
To diminish the Cost
of present Lamp. to
16 Cents ~~per~~

to make a ~~200~~ ²⁰⁰ ~~alt~~
20 CP Lamp ^{equivalent}
respect to economy
CP & life to any 16 CP
Lamp known

To draw a complete system
of patterns







$$500 \overline{) 1400} \quad (2.8$$

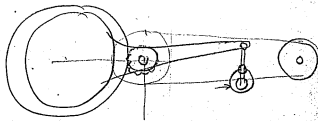
$$\begin{array}{r} 1000 \\ 4000 \end{array}$$

$$250 \overline{) 700} \quad (2.8$$

$$\begin{array}{r} 500 \\ 2000 \end{array}$$

500-

14'

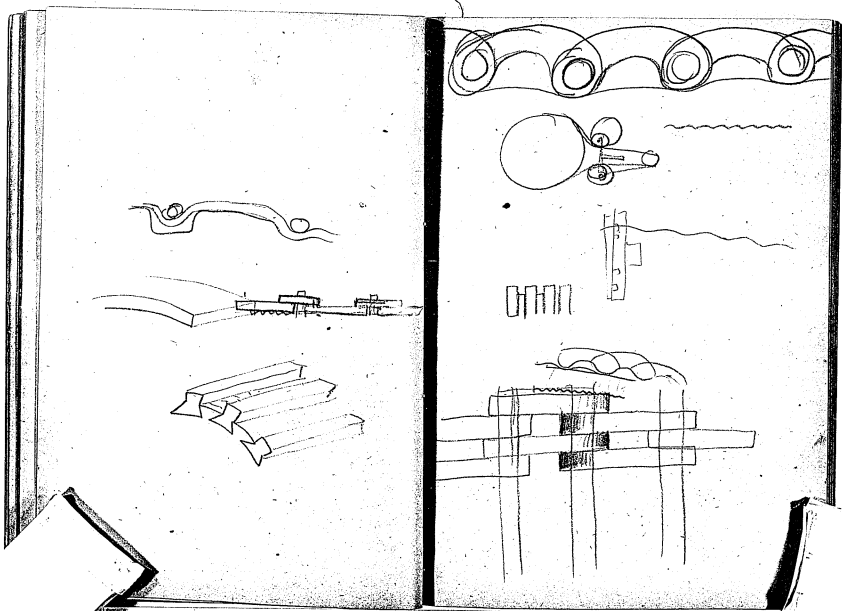


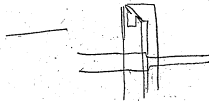
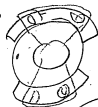
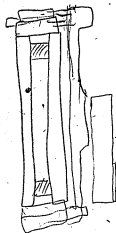
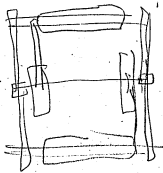
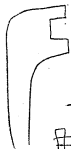
27.

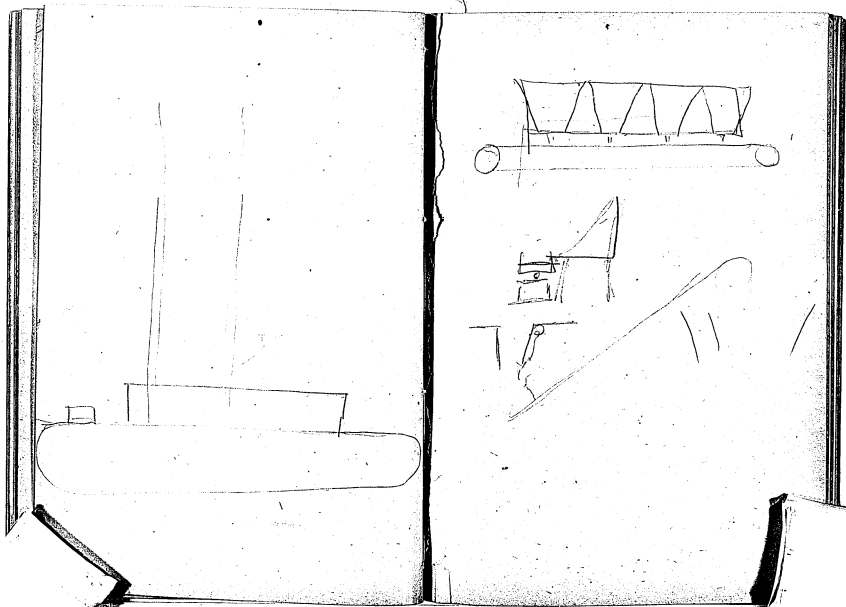
$$\begin{array}{r} 15 \\ 7 \\ \hline 22 \\ 50 \\ \hline 7 \end{array}$$

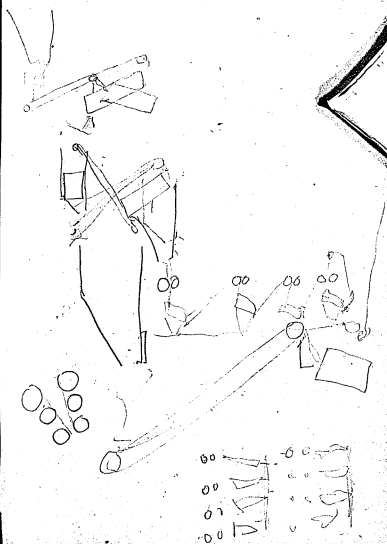
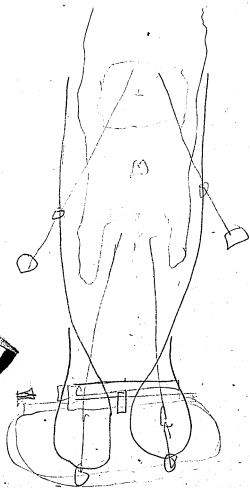
36.

$$\begin{array}{r} 80 \\ \hline 123 \end{array}$$









[THIS BOOK WAS USED IN BOTH DIRECTIONS.
THE FOLLOWING PAGES WERE FILMED FROM
THE BACK END FORWARD.]

$$\begin{array}{r} 120 \\ 12 \\ \hline 60 \\ 120 \\ \hline 1800 \end{array}$$

184-

$$\begin{array}{r} 522500 \\ 271400 \\ \hline 2571100 \end{array}$$

$$\begin{array}{r} 52 \\ 260000 \\ 10400 \\ \hline 10800 \end{array}$$

271522

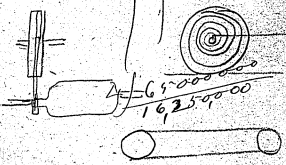
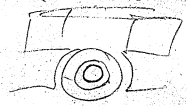
$$\begin{array}{r} 522500 \\ 338000 \\ \hline 784500 \end{array}$$

$$\begin{array}{r} 2700 \\ 2431 \end{array}$$

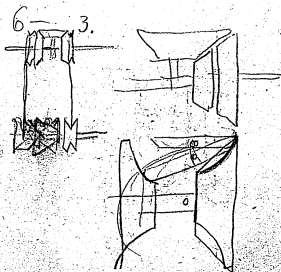
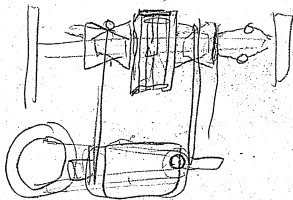
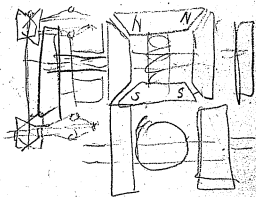
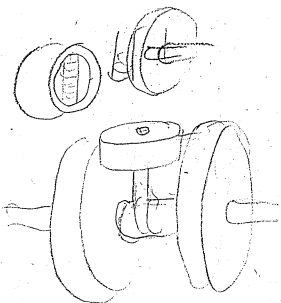
$$\begin{array}{r} 2600 \\ 400 \end{array}$$

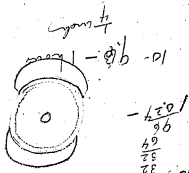
$$\begin{array}{r} 65 \\ 3250000 \\ 13 \\ \hline 338000 \end{array} \quad 251000$$

361000

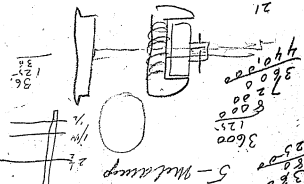
$$\begin{array}{r} 28 \\ 900 \\ \hline 25200 \end{array}$$


650000
1625000

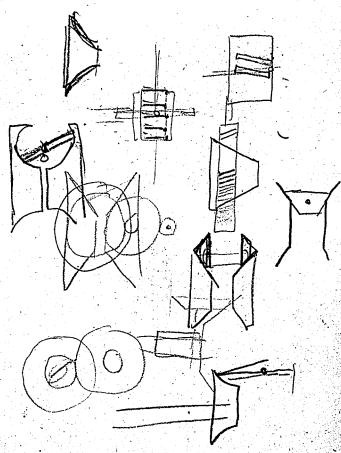
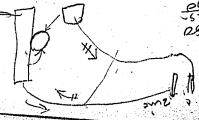




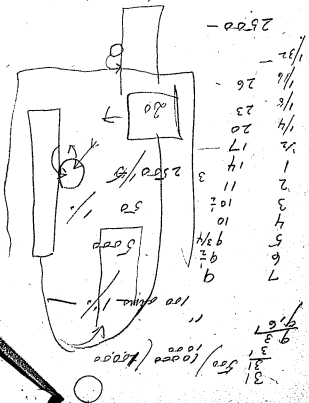
1. width - 32
 64
 32
 96
 1024 -



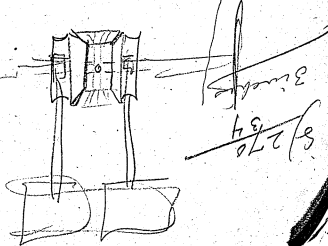
5-Mud pump



1000 - 10000
 $\frac{10000}{10000}$ of camp
 00095
 373.



1000 of camp
 83 mchms
 00025 camp
 $\frac{10000}{10000}$
 932
 4/373
 8388



$$\begin{array}{r} 120 \\ 12 \\ 40 \\ \hline 12 \\ 20 \\ \hline 134 \\ \hline 165500 \end{array}$$

$$\begin{array}{r} 365 \\ 200000 \\ \hline 73000000 \end{array}$$

$$\begin{array}{r} 69 \\ 4 \\ 260000 \\ 455000 \\ 1602650 \\ \hline 3071000 \\ 48300 \\ \hline 65 \end{array}$$

$$\begin{array}{r} 35 \\ 472500 \\ 307000 \\ \hline 1655000 \\ 315000 \end{array}$$

$$\begin{array}{r} 472500 \\ 315000 \\ \hline 1575000 \\ 1575000 \\ \hline 4725000 \end{array}$$

$$\begin{array}{r} 12 \\ 14 \\ 120 \\ 28 \\ 14 \\ \hline 168 \end{array}$$

$$\begin{array}{r} 312 \\ 2 \\ \hline 624 \end{array}$$

$$\begin{array}{r} 365 \\ 39000 \\ 522 \\ 307 \\ 215 \\ \hline 1095 \end{array}$$

$$\begin{array}{r} 31 \\ 522 \\ \hline 184 \end{array}$$

$$\begin{array}{r} 365 \\ 200000 \\ \hline 73000000 \\ 63000000 \\ \hline 136400000 \end{array}$$

maximum load =
100,000 - lights averaging
3 hours each steady =

300,000 lamp hours -
15 per hp basis
100,000 light station, costing
Day 1,200,000. -

100,000 lights actually burning
at one time in Dec at 530 pph -
average during year, 2 hours,
365 days 2 hours daily -
or 730,000 lamp hours which
on gas basis is 365,000 M³
if sold at 1.50 per M³ the
average price gas - 70000 receipts
\$4,425,000 - If we assume
that Operating Expenses are
65 per cent then the saving would be
2880000 leaving 1,545,000 profit.
which is say 13 1/2 per cent
1.5% c div

$$\begin{array}{r} 7500 \\ 15 \\ \hline 37500 \\ 7100 \\ \hline 11250 \end{array}$$

$$\begin{array}{r} 472500 \\ 245000 \\ \hline 227500 \end{array}$$

$$\begin{array}{r} 7300 \\ 15 \\ \hline 36500 \\ 7300 \\ \hline 109500 \\ 19 \\ \hline 7000 \\ 105000 \end{array}$$

$$\begin{array}{r} 25 \\ 22 \\ \hline 47 \end{array}$$

$$\begin{array}{r} 52 \\ 20 \\ 36 \\ \hline 244 \end{array}$$

$$\begin{array}{r} 520 \\ 7 \\ \hline 3640 \end{array}$$

20 per hp basis -

Cost station for machinery
 $\frac{1}{3}$ less for building ~~and~~
 ground $\frac{1}{4}$ less for installing
 $\frac{1}{4}$ less Underground $\frac{1}{2}$ less
 or about 300,000 less

Making investment \$900,000.

The general expenses will be
 about the same. The actual
 expenses for coal oil water
~~insurance~~ will be $\frac{1}{4}$ less
 or say the ^{operating} expenses will be

7.52 per cent of receipts,
 as receipts will be the
 same as in the other case

to wit: \$522,500 then deduct
 271,000 less 257,000 profit
 or ~~257,000~~ ^{257,000} profit
~~15%~~ with 15% per hp -

$$\begin{array}{r} 120 \\ 1 \\ \hline 120 \\ 1200 \\ \hline 1320 \end{array}$$

16.

$$\begin{array}{r} 489 \\ 35 \\ \hline 136 \end{array}$$

$$\begin{array}{r} 459 \\ 46 \\ \hline 537 \\ 337 \\ \hline 144 \end{array}$$

$$\begin{array}{r} 60. \\ 489 \\ 573 \\ \hline 352 \\ \hline 209 \end{array}$$

338
 $\frac{15}{353}$ Expenses recpt. 489 000
 given: 136 000 profit or say
 11 pct at 125 recpts be 609 250
 profit 256 250, or say 21 percent
 or if sold at 110 give same div
 609 250
~~489 000~~

$$\begin{array}{r} 2445000 \\ 978000 \\ \hline 489000 \\ \hline 6092500 \end{array}$$

$$\begin{array}{r} 1200 \\ 2400 \\ \hline 2400 \\ \hline 6400 \end{array}$$

15

$$\begin{array}{r} 1200 \\ 2400 \\ \hline 1200 \\ \hline 25200 \end{array}$$

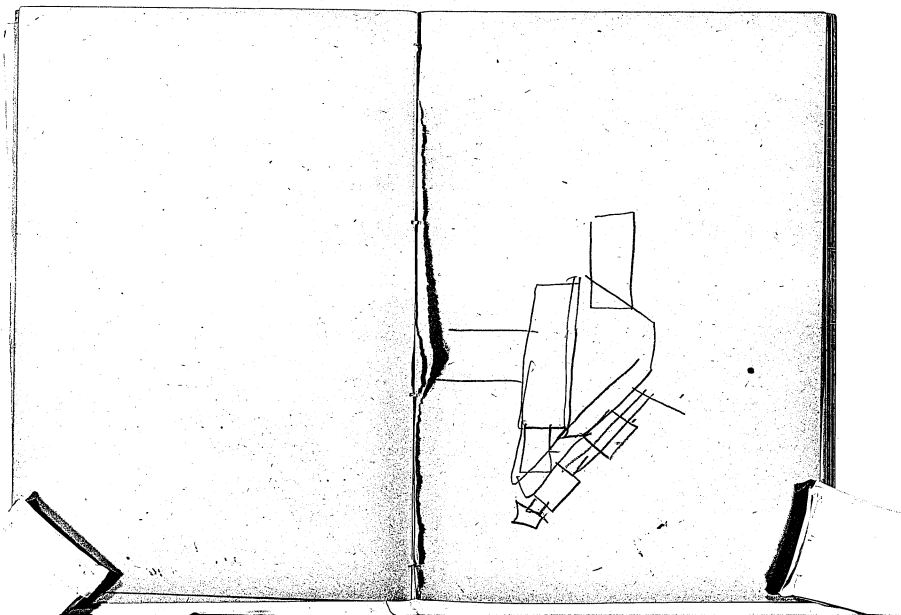
$$\begin{array}{r} 609250 \\ 353000 \\ \hline 256250 \end{array}$$

$$\begin{array}{r} 1200 \\ 6000 \\ \hline 12000 \\ \hline 18000 \end{array}$$

Co to pay for experiments in
the regular method, of charging
at laboratory, but the total
yearly

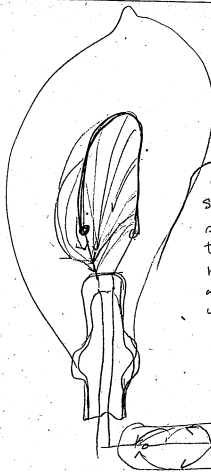
~~$\frac{1}{4}$ of the total, was to be~~

The whole of the expenses of the
laboratory w^o be paid by the
Co and $\frac{3}{4}$ of all the work
done at laboratory shall
be for the Co, ^{the other $\frac{1}{4}$ of the cost of the work} and ~~present~~
adapted by the Co. and made
use of Edison shall get
 $\frac{1}{5}$ of the benefits accruing to
the Co - to be mutually
agreed on or submitted to
arbitration, and a new delimitation
of benefits may take place at
the Eschylora of either party.



Boil SO₄ in 10 bulbs then
 wash in distilled -
 pass current through fil in
 strong SO₄ - with platinum
 weight on fil - Watson amp
 meter - do it in pump
 room if ok make 10
 lamps with highest amp
 move fil around while
 doing it -

Make lamp on spoon phos
 Run up to get red blue in
 globe & then to get the
 black spot near blue clamp
 & then notice its gradual
 spread over whole of
 carbon



Try 5 Lamps
 spoon phos -
 heating bulbs
 before why fil -
 heating in amp meter

2nd 5 Lamps
 Spoon phos - get
 red blue bring
 to visible red
 run $\frac{1}{2}$ minute
 at this. then work
 up quick -

Try lamp that has run on a
 curve but not blocked with
 HCl + rec of scale comes off -

Try 5 Lamp Exhaust spoon
phos. then bk Pac ~~heat~~
lake (amp off + heat bulb
very hot just as if you were
to burn off scale of block
& Re-exhaust)

Have Dishler pick out
20 lamps. Each having
same life - say 100 200
3 5 7 10 15 20
30 + 4000 hours not any
selected -

Then I will study
each set for peculiarities

find drop of Cp on M3 -
Selected specials -
-

Try dip fil in ~~20~~ 17 then
bring up with 40 Lamp
bank - dets dipped sugar
do Carb in chloro pipid -

Hang Freddy preliminary
50 ffls up to 65° Fahr
in charcoal & nickel box
8 hours - Take out
dip 17 at 100 - then
send to Mat - at same
time Run the 64 etc we last
squeezed through in charcoal,
some without charcoal tube
out

Load Nickel box up as follows: box with 10 + 00s -
lost squashed fibres 1/2 of each
in charcoal 1/2 without,
also box with 50 fibres Reg.
Gambos another box or boxes
50 in charcoal - another
box 50 tied together,
Run to 650 in 8 hours
take out all ~~boxes~~ +
Examine if squashed ok put
back - if others ok -
take ^{10 of the 50} ~~50~~ with charcoal
+ put them back -
Take 15 of the 50 without
charcoal dip in 17 at 100 deg
dry and put in charcoal.
Take the bunched ones
dip dry ~~and~~ to Mart also the
with 2 35° -

26 Castles oil acts on phos
it was in the new spoon
device -

$$15 \overline{) \begin{array}{r} 300 \\ 500 \\ 10000 \end{array}} \quad \begin{array}{r} 15 \\ 45 \\ 500 \end{array} \begin{array}{r} 10000 \\ 10000 \\ 10000 \end{array}$$

1000.

$$\begin{array}{r} 10000 \\ 10000 \\ \hline 109000000 \end{array}$$

$$\begin{array}{r} 20000 \\ 10000 \\ \hline 1000000 \end{array}$$

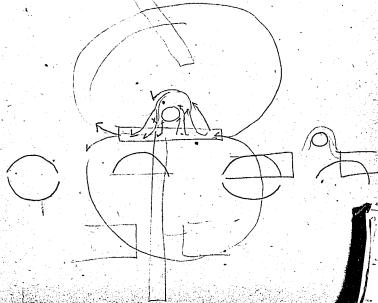
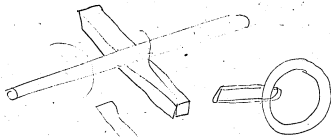
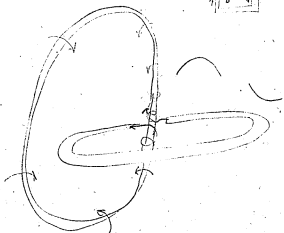
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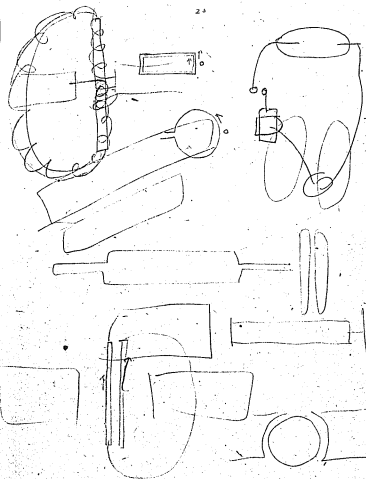
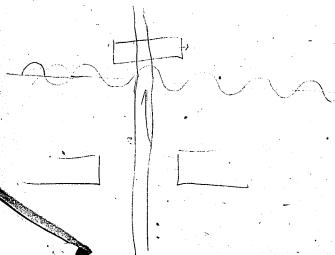
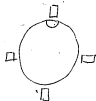
$$1000 \overline{) \begin{array}{r} 365 \\ 200000 \\ 7300000 \\ 36500000 \\ 36500000 \\ 36500000 \\ 36500000 \\ 36500000 \\ 36500000 \end{array}} \quad \begin{array}{r} 365000 \\ 365000 \\ 365000 \\ 365000 \\ 365000 \\ 365000 \\ 365000 \\ 365000 \end{array} \quad \begin{array}{r} 365000 \\ 365000 \\ 365000 \\ 365000 \\ 365000 \\ 365000 \\ 365000 \\ 365000 \end{array}$$

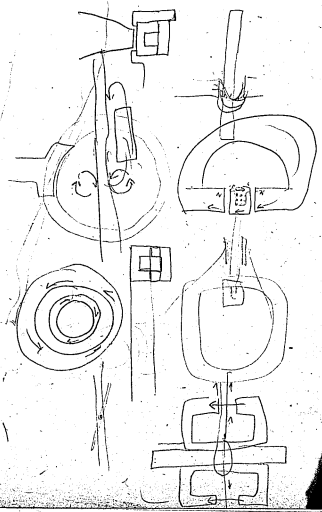
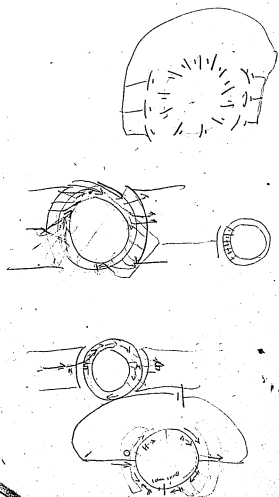
10



20





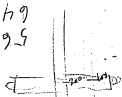


$$\begin{array}{r} 47 \\ 36 \\ \hline 113 \\ - 95 \\ \hline 18 \\ - 15 \\ \hline 3 \end{array}$$

1.25
 100
 1.25



-049
 095



$$\begin{array}{r} 1.5 \\ 2.5 \\ \hline 4.0 \\ - 2.8 \\ \hline 1.2 \end{array}$$



Doubling the number of lines of force cut by a wire whereby the amperes are doubled in a wire - doubles the number of whirl line ~~or coils in the same thing~~ and ~~the other offers which are~~ as ~~second~~ to ~~reverse~~ the lines without increasing their velocity or your kind of ballen it must double their amplitude ~~in~~ ~~increasing~~ their number in ~~both~~ ~~are~~ ~~probably~~ ~~two~~ ~~lines~~ ~~but~~ ~~the~~ ~~wire~~ ~~double~~ ~~in~~ ~~amplitude~~

If one turn of wire encircle a bar of iron, this will pass into it & be combined to certain number of whirl lines from the Copper ~~bar~~ if two turns this will be twice the Condensation with any turn, the number of Condensed whirls will be in proportion to the iron enclosed by the wire - two amperes turns, double the Emf of the coil line ~~two~~ ~~turns~~ ~~on~~ ~~a~~ ~~Dynamometer~~ ~~double~~ ~~the~~ ~~Wells~~

In an iron wire carrying a current
a part of the whole is contained in the
wire itself and thus retards signal -
this is not so with Copper -

In pure copper the molecules go with the
whirl nearly, but not quite, hence heat.
If the orbital path of the Copper molecule
is disturbed by foreign molecules,
vibrating in different times collisions
occur a more heat is given out and there
is a fall of pressure due to the work
absorbed. If the Copper molecule
vibrates or moves with the local
velocity of a wave in the Ether there
would be no collisions & no work
no heat & hence no resistance -

Resistance of Electric Conduction
more or less is a function of the
orbital velocity or vibrating period
or perhaps elasticity of the rigidity
of the aggregated molecules of the
conductors. It varies a metal approx
to elasticity of the Ether the better it conducts

All Combustion between Molecules take place with the same velocity as light c is a function of the speed of stresses through the Ether.

For battery where the Combustion gives 1 volt, then where the Combustion gives 2 volts twice as many individual molecular Combustions take place in the same time - or if this is not the case the amplitudes of the whirls are doubled, owing to the double amount of energy set free at each given molecular Combustion.

As iron is a conductor like Copper but its molecules ~~being~~ bonds being held together with an elasticity less than the Ether & less than Copper the latter is the better conductor, therefore magnetic Condensation of the whirls of

Copper must be not a molecular
ferri-ism of the iron but a Crystallographic
or rather larger & more sluggish
posical nature, while Magnetic lines
may have the same velocity as
light when propagated through the
Ether. It is not so with iron

~~2nd theory~~

~~All matter condenses ether on its
Crystallographic surfaces -~~

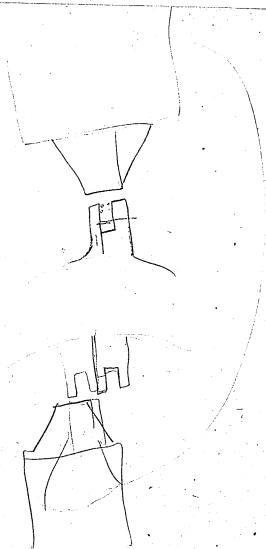
The iron molecules are capable
of motion in two directions
on following the whirl when
acting as a conductor & are
at right angles to this when the
whirl passes into iron

Reversal causes the energy absorbed in starting the molecule to be given out as heat hence hysteresis. Heat saturation is produced when the speed of rotation equal the speed of the ether wave. no saturation is rotation not up to speed or long -

permanent magnet is the molecule up to a given speed or amplitude + prevented from going to zero by the intervention of matter which prevents the system coming together in collisions. Carbon for instance. (10) Combined. These lines of molecules once set vibrating make a closed circuit probably Williams is the more closed the chain (11) the lower the resistance the

longer they will continue to rotate
 + eventually mutually react a
 rotation comes in steel the rapid
 system produced by. Carbon keeps
 these molecular lines of circuit
 separate hence they keep on without
 loss or diminution of amplitude of
 the wave that is obtained.

It may be the reason that
 so many stress lines can be
 packed in iron as against copper
 is that ~~iron~~ copper forms but one
 chain of molecular circuit
 while iron has thousands of these
 whirlwinds. Around the circuits
 in other words the magnetic
 metals have groupings in straight
 lines each molecular connecting
 with the other while non magnetic
 substance have no such property
 perhaps the very opposite =



Some small action as this is implied by the lengthening of a magnet without alteration in its bulk - that there is air space when magnetic is shown by the sharp sound on breaking.

Certain alloys of iron, manganese + iron with 70% of iron are non magnetic hence this must prevent the formation of these molecular currents of Cartegian iron particles & have no threading. It also proves that individual iron molecules are not of themselves magnetic but the phenomenon is due to mutual action so that chains of molecules giving closed currents can be had or else surround them

$$\begin{array}{r} 12000 \\ 3000 \\ 2000 \\ 1000 \\ 500 \\ 200 \\ 100 \\ 50 \\ 20 \\ 10 \\ 5 \\ 2 \\ 1 \end{array}$$

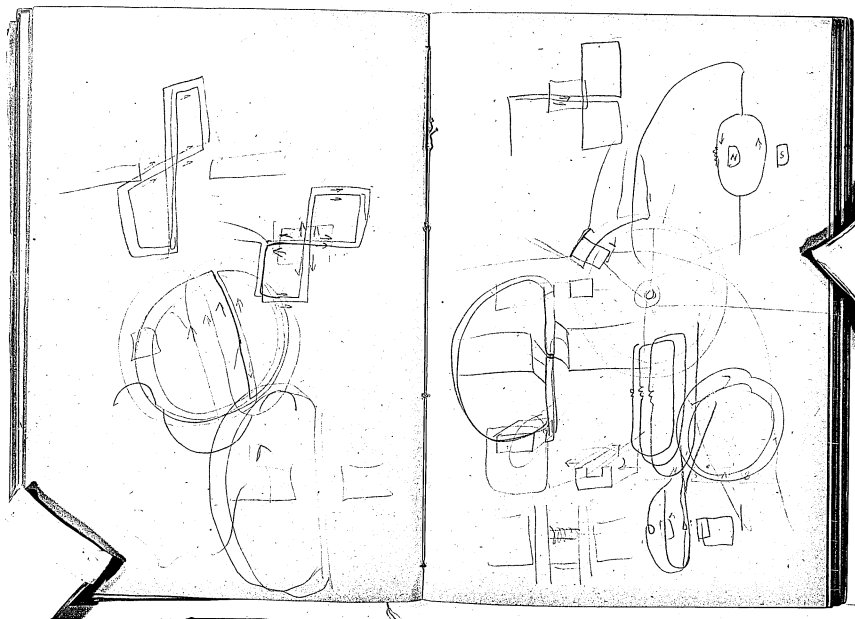
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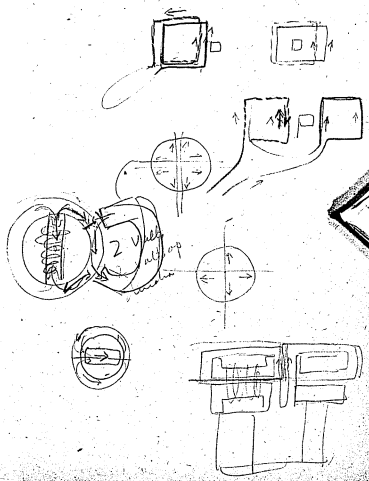
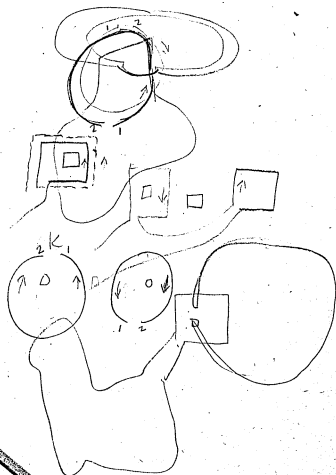
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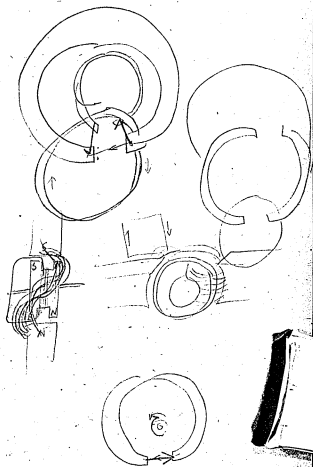
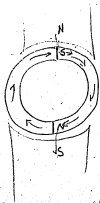
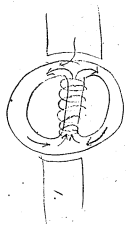
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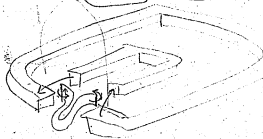
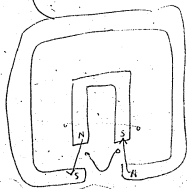
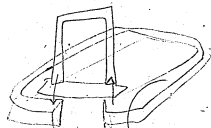
- 5

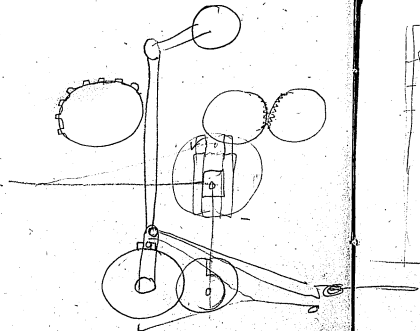
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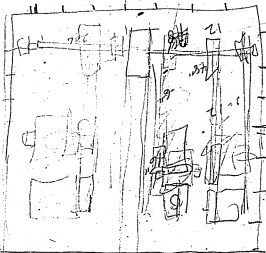




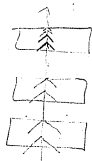




making eye size



$$\begin{array}{r} 15 \\ + \\ 17.5 \\ \hline 2.25 \end{array}$$
$$\begin{array}{r} 15 \\ - \\ 3 \\ \hline 12 \end{array}$$



Notebook, N-90-01-04.4

This notebook covers the period January 1890-March 1891. All entries are by Edison. The book contains notes relating to numerous mines and methods of ore separation for iron, copper, gold, and nickel. Also included are drawings regarding phonographs, generators, and electrical distribution systems. Near the end of the book are cost calculations for the North American Phonograph Co. The front cover is labeled "#582." The inside front cover is inscribed "#582 T A Edison Jan 4. 90." The pages are unnumbered. Approximately 110 pages have been used.

#588

N-90-01-04.4

Edison
Jan. 2. 90

Sudbury - Ironton - Cu Fe Ni.
ore plant -

Notes -

Jan'y 4 1890

Also Gold + Ni Cu

Cu Au ores -

Marhès reports convert Fe Cu Matte to get rid
Fe Min Res US 1883 p 540 - same page
Garnier, Bessemerizing Nickel Matte

Eugene Hermite Rouen France.

puts Ni Cu oxide from Matte in closed chamber
rotatable, puts ammonia in under pressure
60 lbs egg inch, amount proportion to ore
rotates 30 @ 60 minutes - after settling,
decants - reammoniates if necessary
dignand subjected to electrolysis in closed
vessels containing electrodes. Carbon &
Cast iron metals in solution collects on iron
Cathode. Na added to solution lower resistance
Ammoniation of Cu & Ni formed. Containing
Yessell may be of Cast iron as ammoniation
of Ni Cu don't attract, further details see
London Mining Jnl & also p 542 US Geology
1883 -

Heat roasts powdered matter to form sulphate
or then dissolves & oxidizes Fe by blowing in air or
use bleaching powder to oxidize iron & form a
insoluble Ferric Sulphate

Rivot & Phillips -

Heat roasts while white hot
put in bars of iron to reduce O of CuO
American Encycloped 693 article Copper.

Edison -

Heat roasts reduce Cu to metal state
then in lead to form Cu-Pb alloy the Fe or
Ni not alloying at the low temperature
also by autolysis -

See what Mankie's process is -
also Doetscher Rio Tinto process
purify Fe.

In 1882 US Geology says there is a
nickeliferous limonite ore in Michigan which
has produced wonderfully good iron.
p 416 — also states iron ore at Antwerp MI
contains Ni.

M D Cheney & Roberts - Mass Inst Techn
method estimating Ni in pyrrhotite or in the
phos N. Considerably soluble while phos Fe
almost insoluble in acetic acid in
presence of excess phosphatic solids.
US Geol 1882 p 417 —

See Ni Process Eister & Howe of
Boston - Trans Amer Inst Min
Mining Engrs 1882

1882 US Geol - says immense deposit
pyrites good for SO₄ + Clay + Coosa Co
Ala some containing 10 pct Cu not
very accessible

1883 USGCo says there are within few
miles of Jilly Foster Mine immense
deposits now not yet developed.

Blue Saphires found near Spartan,
Franklin + Vermont = Sussex Co. N.H.

Magnetic Iron Sand West shore of
Lake Champlain.

An Aluminous Magnetic iron ore
occurs near Peekskill ny self flowing
not worked immense deposit.

Warty Ancient Iron Age S 1858 +
Sullivan Co. N.Y. 1859 Repton N Carolina, Hadon
+ Lincoln Co. N.C. ore esp. abundant
on Western slope of Cross Mountain -
covered with gossan

In Phillips Work Called Ore deposit
says p 596 -

On north side Island of Michipicoten
Canada cliff 300 ft high is a soft
amygdaloidal bed containing Native
Copper. Tracible some Miles - in this bed
an attempt was made number yrs ago to work
a remarkable bed of deposit of Native Cop
with grains native Silver disseminate through
quartz composed of Hydrous Silicates of
Nickel - ore was stamped & Ni ore the
value of which not then known washed
away leaving residue Malathic Ag & Cu
Shaft was sunk to depth 12 fathoms but
after considerable outlay was abandoned

Get Windt on the pyritic
deposits of the Alleghenians

Process by Sudbury ores -

Powder to 120 mesh separate
pyrite, then want to make
Calcsparite magnetic.

then put Ni + Cu Sulphide
into brix separately. using

Say 20 30 50 + 75 pct
of fine 100 mesh Connellville
Coal - See if Sulphur all

grows off as Bi Sul Carbon +
there is Gricking - also

bring up high to make
Sponge flat & under hammer
also fuse - Σ

Have Dixon try following

Roast a Calcopirite until
all magnetic - then dead roast
until all nonmagnetic
then with fine coal or
reducing flame again
roast to make it magnetic

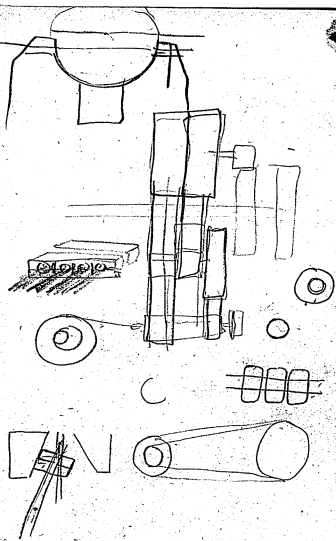
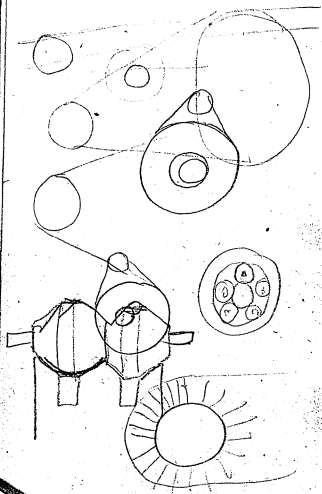
If this will, can get a
double twist on some
ores - also roast Calcopirite
& Iron pyrites under same conditions
see which becomes magnetic first
or if Coles can be easily roasted
to magnetic without effluvia
pyrite

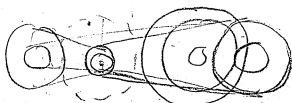
Nickel Mine Warren Co N Jersey
at Brass Castle mine named Witty Mine
operator. C F Staates - produced in 1880
nothing -

Nickel Mine operated 1880 by
Draent Nickel Mining Co - local
Draent Middlesex Co Massachusetts
product in 1880 - 13400 lbs -

Mica Mine - Broadway
Warren Co N.J. named Broadway
Mica Mine - Operator J Merrill
no product in 1880 -

~~South~~ Virginia Cop belt. Ore Knox
Vinegarque principally Magnete pyrite &
Sphalerite bearing Calcoppyrite
1 bet Nickel in Ore -





$2\frac{1}{2}$ to 1

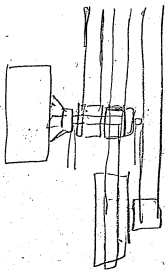
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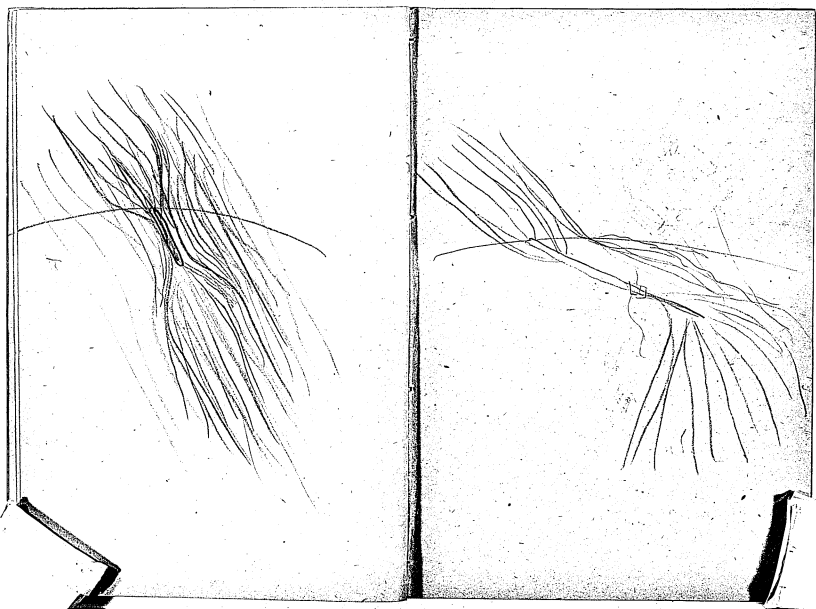
33.

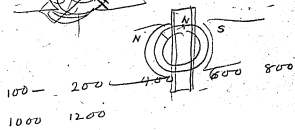
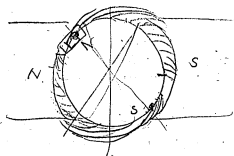
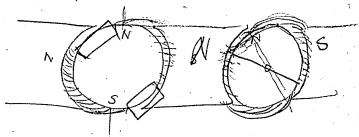
$$33 / \begin{array}{r} 300 \\ 297 \\ \hline \end{array} \cdot 9$$

$$L. 25 / \begin{array}{r} 30 \\ 25 \\ \hline 50 \end{array} \cdot 1.2$$

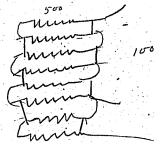
60/520 (4.







~~~~~  
v



4000  $\frac{1}{2}$  -  
3000 300  
2000 600  
1000 600  
500 - 1200

$$\frac{16\frac{3}{4}}{\frac{2\frac{1}{2}}{4}} = \frac{16\frac{3}{4}}{\frac{1}{2}}$$

$$\frac{29}{8\frac{1}{4}} = 2$$

$$\frac{29}{8\frac{1}{4}} (7\frac{1}{4}) = 5000$$

$$3 \frac{25}{12} (7\frac{1}{4}) = 5000$$

$$\frac{29}{24} \frac{1200}{25} = 3000 \text{ feet}$$

$$29 - \frac{1430}{100} = 6.4$$

$$225 - \frac{1430}{100} = 4\frac{1}{2}$$

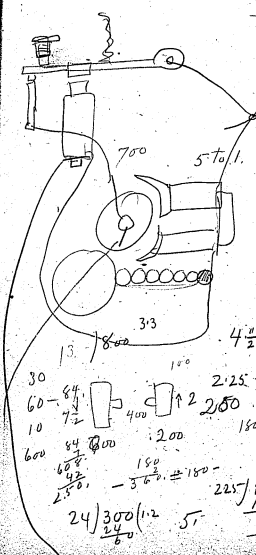
$$4\frac{1}{2} \cdot 2 = 9$$

$$480 - \frac{2400}{4} = 47$$

$$250 \frac{800}{750} \frac{1000}{100} = 82\frac{1}{2}$$

$$4\frac{1}{2} \cdot 16 = 72$$

$$3\frac{2}{3} \frac{2\frac{1}{2}}{5} = \frac{14\frac{1}{2}}{10} = 1.45$$



$$225 \frac{800}{675} = 3.65$$

$$\frac{1230}{1230} = 1$$

$$\frac{290}{1} = 290$$

$$\frac{1450}{1} = 1450$$

$$3 \frac{335}{118} = 2.837$$

$$2.37$$

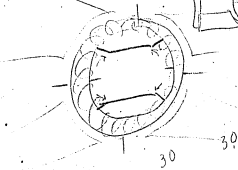
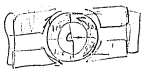
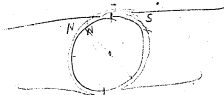
$$14\frac{1}{2}$$

$$2:25$$

$$225 \frac{1450}{1350} \frac{6.44}{1.65} = 7$$

$$24 \frac{300}{240} \frac{12}{60} = 5$$

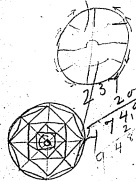
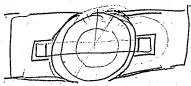
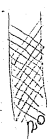
$$\frac{1900}{1900} \frac{540}{4} = 1.35$$



60  
14200



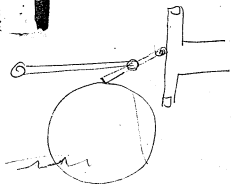
*Proctor*



237  
200  
7400  
942 - 2

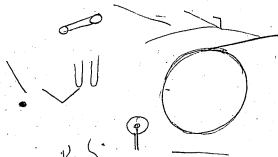
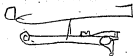
237  
300  
11  
1422 - 300

237  
230  
1850  
474  
59250  
1784



$$\begin{array}{r} 70 \\ 240 \\ \hline 310 \end{array}$$

310  $\frac{800}{100}$



17.15

8.7

4.3

11



July Aug 88

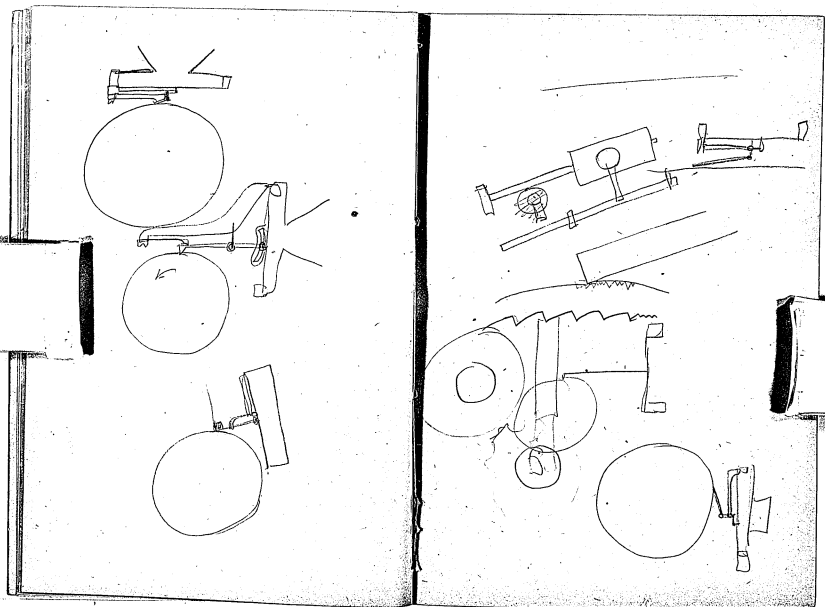
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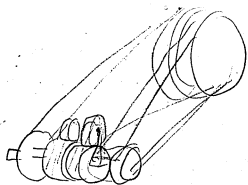
1

3

4

41 months





60  
120  
250

500  
250  
125

A.O

2  
16  
18,

Next unit is Lig

69.

300

600

48hp



2 mi

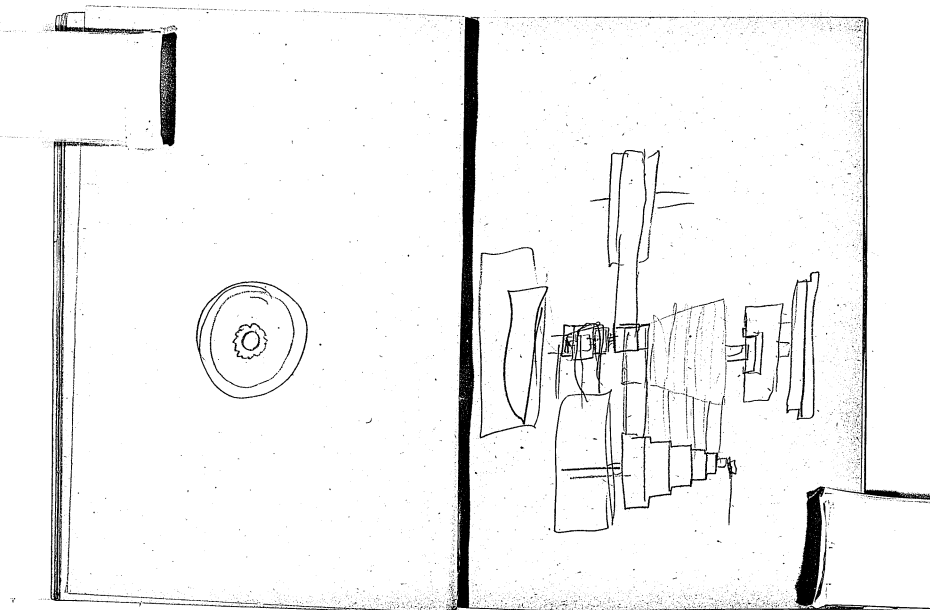
65.

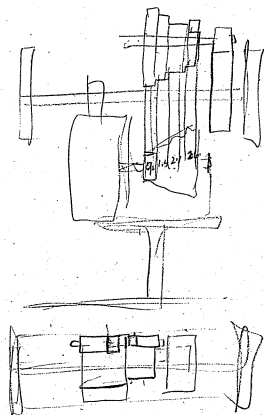
16

81 - Mod cost.

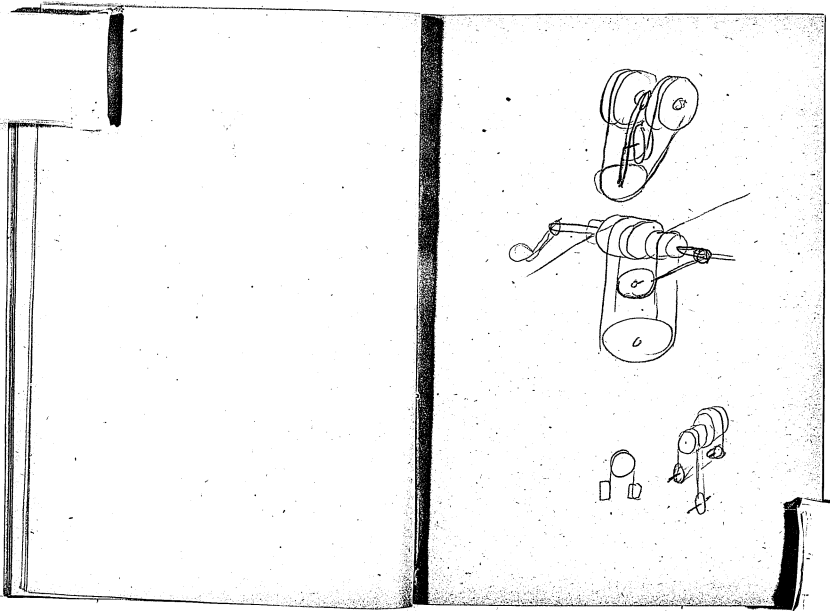
15  
12  
30  
18  
18

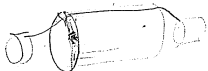
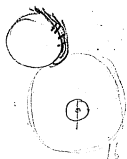








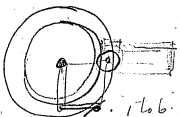




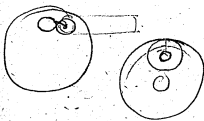
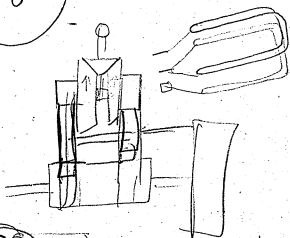
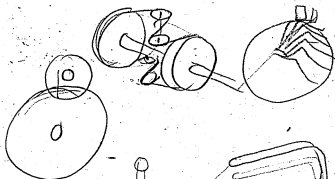
12

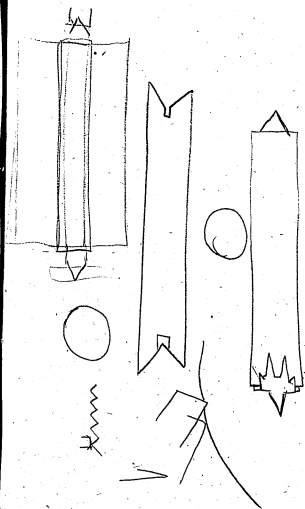
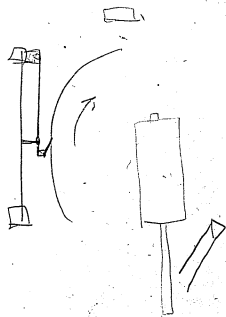
611. 600.

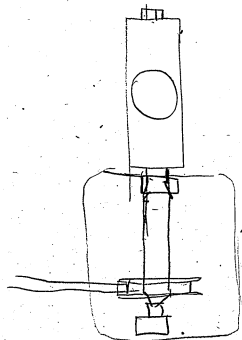
24  
3  
72.

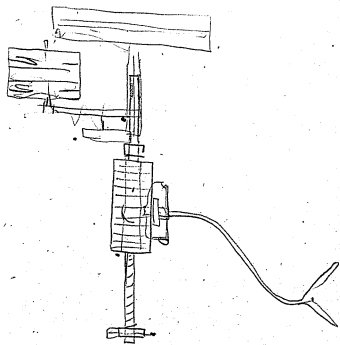
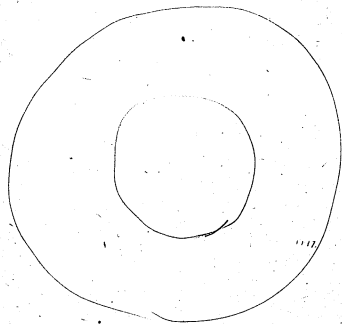


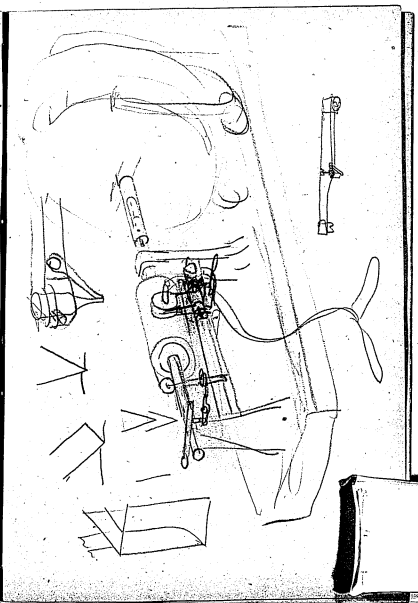
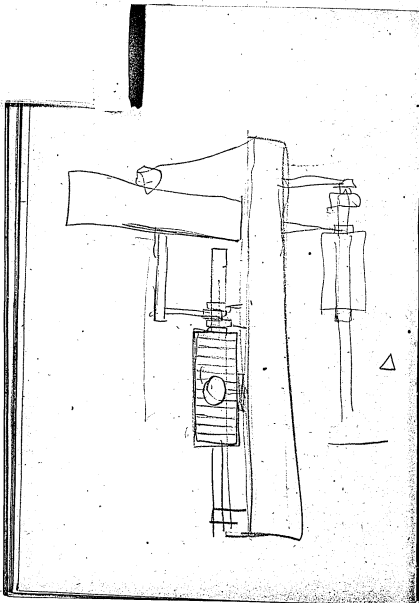
1 to 6.

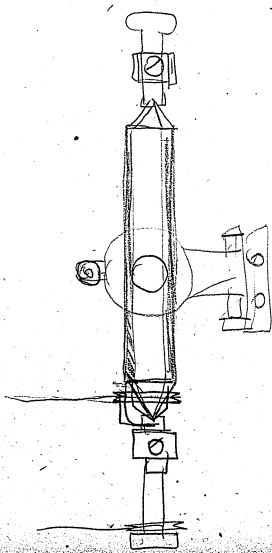








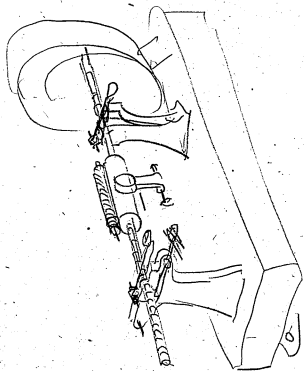
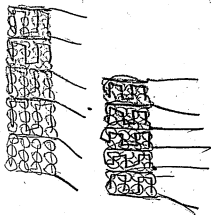
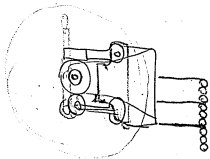




321 / 1,000,000 (512 10-  
 96,000  
 417  
 3260  
 88

10  
 3/2  
 2/0  
 3/0

Mo. 3m  
 1





March 15-1911

Non Expt 1

Chloride Ammonia

Sulphate "

Carbonate "

Borate "

Sodium Amalgam -

Anhydrous Caustic Soda

Carbonate Soda

lining Ladle with Marble  
make  $\text{CO}_2$  which pushing  
thru iron reduced  $\text{CO}$  -

$$\begin{array}{r}
 33^{00} \\
 35^{00} \\
 \hline
 33 \\
 412 \\
 64 \\
 67 \\
 68 \\
 69 \\
 70 \\
 71 \\
 \hline
 3582 \\
 1792 \\
 \hline
 7164 \\
 1152 \\
 \hline
 3582
 \end{array}$$
  

$$\begin{array}{r}
 125 \\
 \hline
 \end{array}$$

32-

Impurities

March 15 '91

Mixture 1 part Buchron Pal 2 Chlamme  
 heating gives Nitrogen -

Sodic Nitrate -

Iron pyrites -

Sulphate Potash -

Cyanide Potassium

Sulphate Barium

Magnesian Sulphate look  
 out for water -

Sulphate Potash no water  
 hot water H<sub>2</sub>O -

~~Bisulphate~~

Strontic Sulphate no Water  
Phosphide of Iron -

Uranium

Wolframum - Sulphide  
also of Lead -

Arsenic

Triglypton

Chromium

Cerium

Manganese

Black oxide Manganese  
in part also Carbide

Dihydrate Sulphate of  
Iron or bradley

Oxide of Lead - say Letharge  
in large dose -

Baryta is sulphate Ba

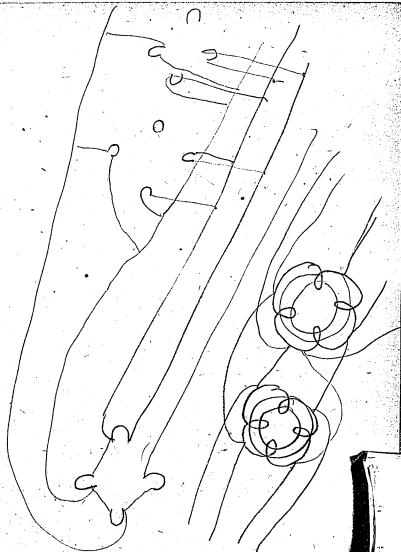
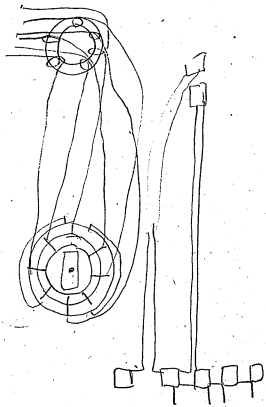
Basic Carbonate high heat to  
Decomp -

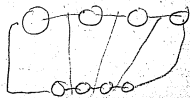
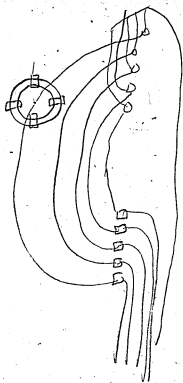
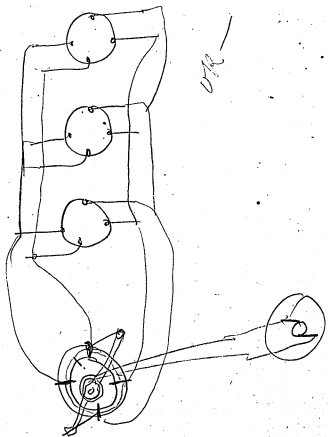
Zinc Sulphide -

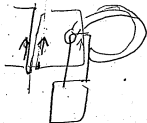
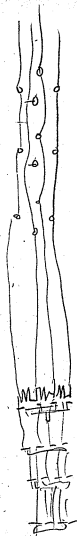
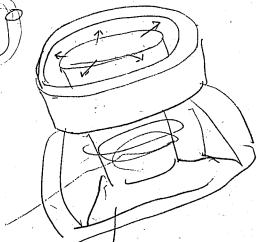
Oxide Zinc -

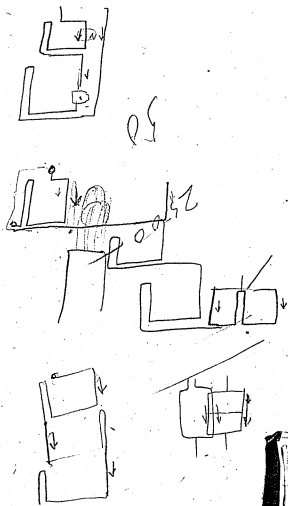
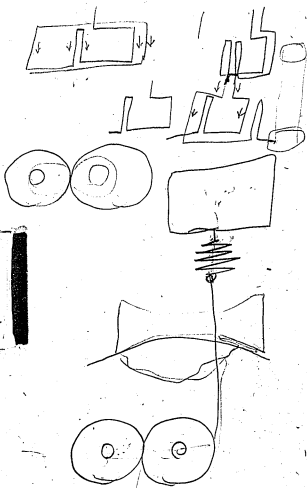
Nickel -

Uranium yellow a  
pigment (Coral)

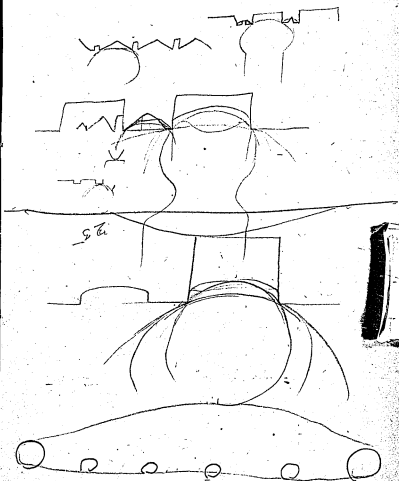
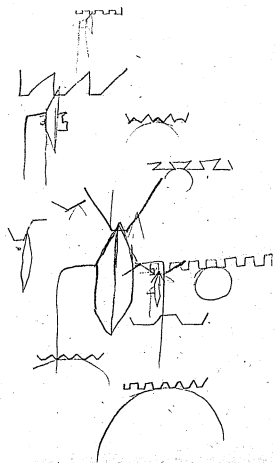


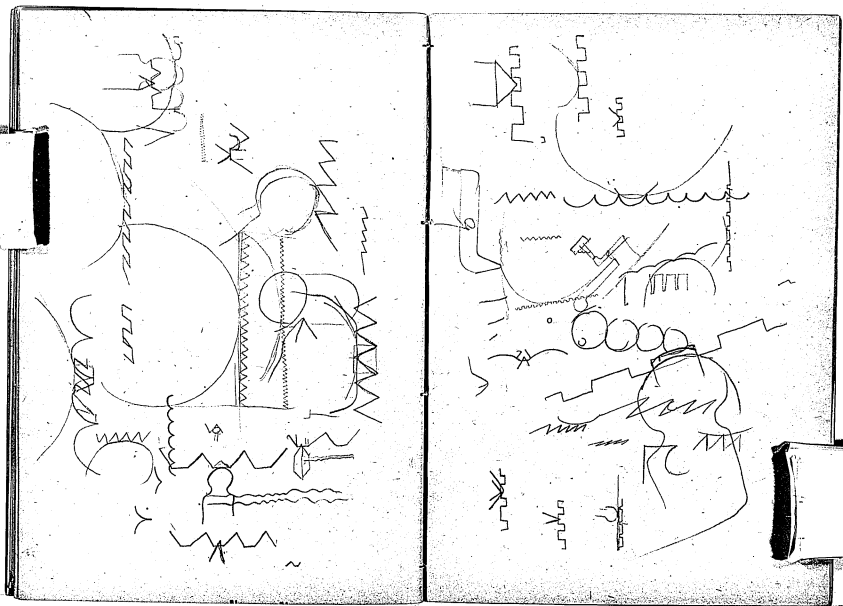












$\frac{2.500}{2.510}$   
2.500  
2.510  
4437.500

$\frac{8437.500}{2812.500}$   
8437.500  
2812.500

$\frac{1250.00}{1562.5}$   
1250.00  
1562.5

$\frac{1825}{125}$   
1825  
125

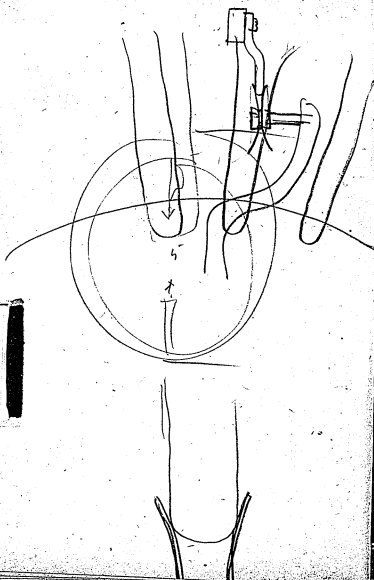
$\frac{2625}{125}$   
2625  
125

27.925

$\frac{300}{2700000}$   
300  
2700000  
2779.900

$\frac{48}{1350000}$   
48  
1350000

$\frac{125}{125000}$   
125  
125000

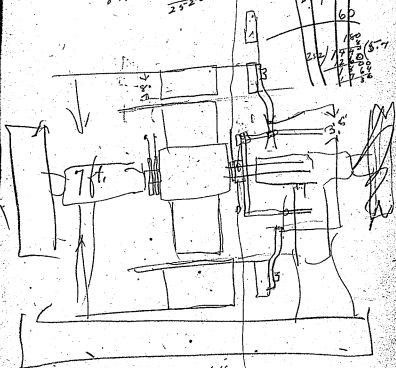


$$\frac{21}{4} = \frac{5.25}{}$$

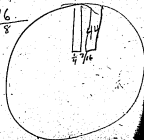
$$\frac{21}{42} = \frac{0.5}{}$$

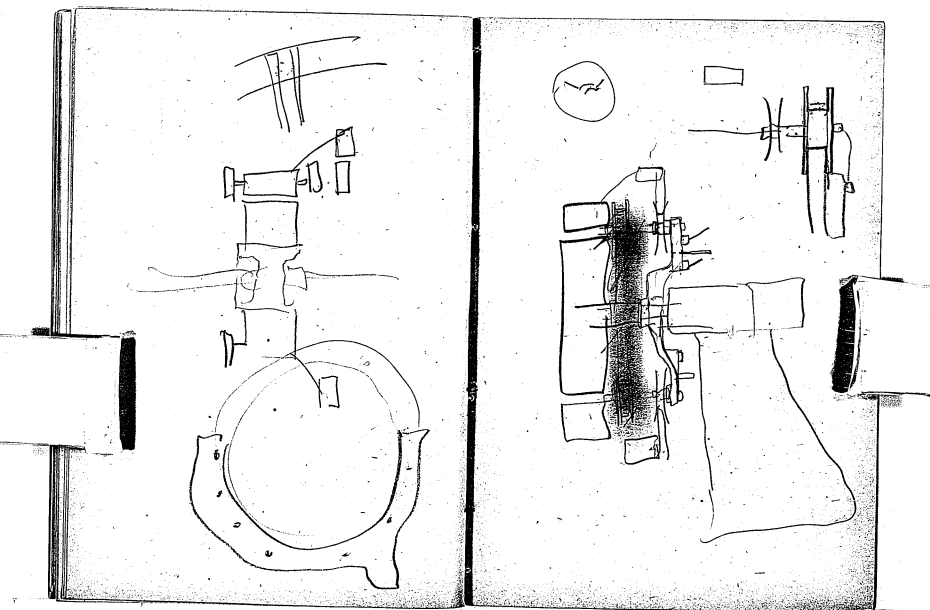
252.

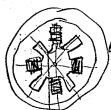
|   |     |      |     |      |      |
|---|-----|------|-----|------|------|
| 2 | 5/8 | 1/4  | 1/8 | 1/16 | 1/32 |
|   |     | 60   |     |      |      |
|   |     | 100  |     |      |      |
|   |     | 150  |     |      |      |
|   |     | 200  |     |      |      |
|   |     | 250  |     |      |      |
|   |     | 300  |     |      |      |
|   |     | 350  |     |      |      |
|   |     | 400  |     |      |      |
|   |     | 450  |     |      |      |
|   |     | 500  |     |      |      |
|   |     | 550  |     |      |      |
|   |     | 600  |     |      |      |
|   |     | 650  |     |      |      |
|   |     | 700  |     |      |      |
|   |     | 750  |     |      |      |
|   |     | 800  |     |      |      |
|   |     | 850  |     |      |      |
|   |     | 900  |     |      |      |
|   |     | 950  |     |      |      |
|   |     | 1000 |     |      |      |



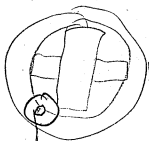
$$\frac{84}{16} = \frac{5.25}{}$$



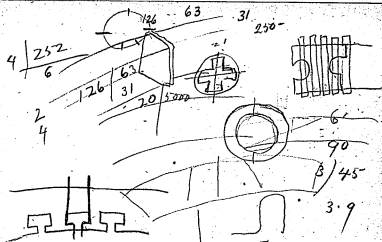
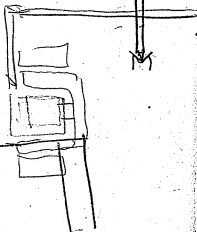




off pole

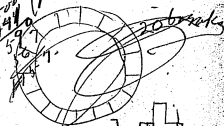


5



$$\begin{array}{r} 21 \\ 4 \\ 4 \\ 16 \\ \hline 16 \end{array} \begin{array}{r} 15 \\ 2 \\ 90 \end{array}$$

$$\begin{array}{r} 63 \\ 5000 \\ 4900 \\ \hline 100 \end{array}$$

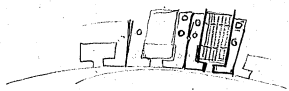


150-

150-

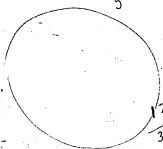
150-

150-

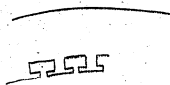
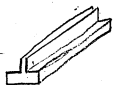
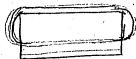
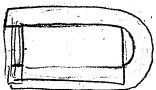




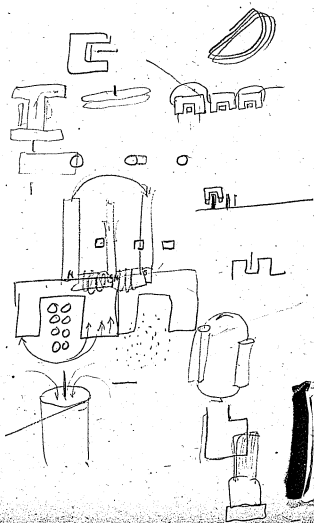
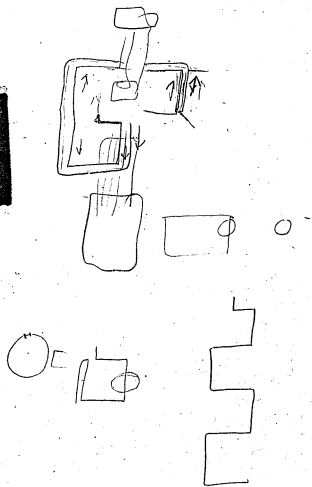




12  
300  
360



6  
19 X 200

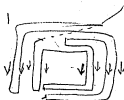


$$\begin{array}{r} 522 \\ 252 \\ \hline 16 \\ 16 \\ \hline 32 \\ 32 \\ \hline 64 \end{array} / 21$$

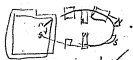


115

107



206  
20  
20

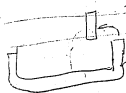


$$\begin{array}{r} 222 \\ 222 \\ \hline 444 \\ 444 \\ \hline 888 \\ 888 \\ \hline 1776 \end{array} / 16$$

$$8) \begin{array}{r} 2206 \\ 2205 \\ \hline 1 \end{array} / 09$$

9

067



$$\begin{array}{r} 121 \\ 059 \\ \hline 180 \end{array}$$



$$\begin{array}{r} 913 \\ 672 \\ \hline 2455 \end{array} (38)$$



33 - 38



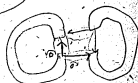
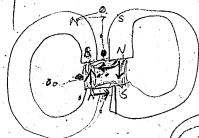
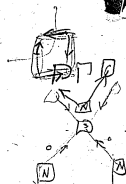
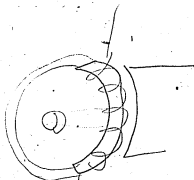
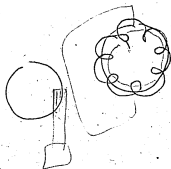
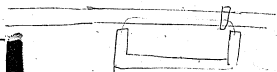
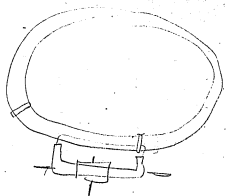
$$\begin{array}{r} 100 \\ 200 \\ \hline 300 \\ 300 \\ \hline 600 \end{array} / 33$$

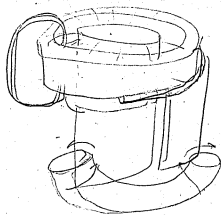
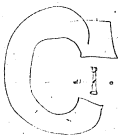
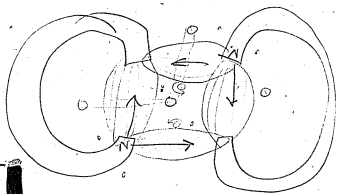
$$\begin{array}{r} 027 \\ 271 \\ \hline 298 \end{array}$$

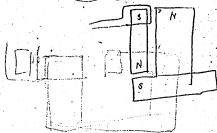
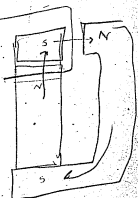
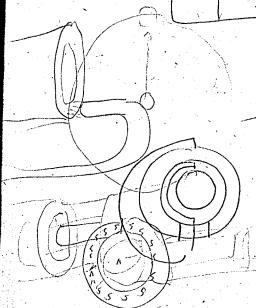
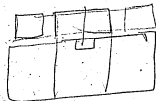
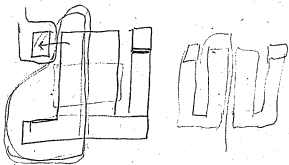
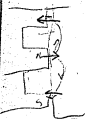
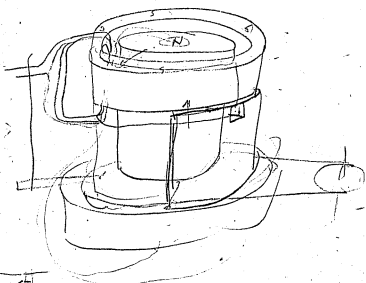
$$(0.2) \begin{array}{r} 367 \\ 22225 \\ \hline 22582 \end{array} (33)$$

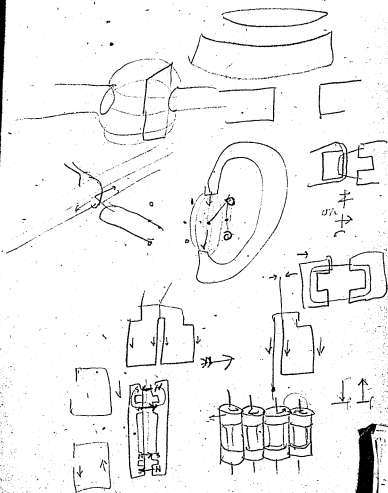
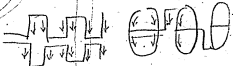
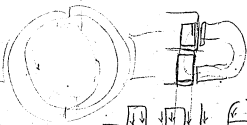
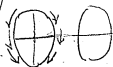
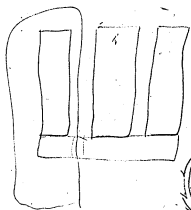
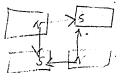


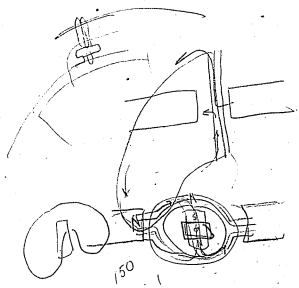
$$\begin{array}{r} 25 \\ 25 \\ \hline 50 \end{array} - 38$$





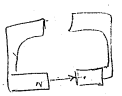




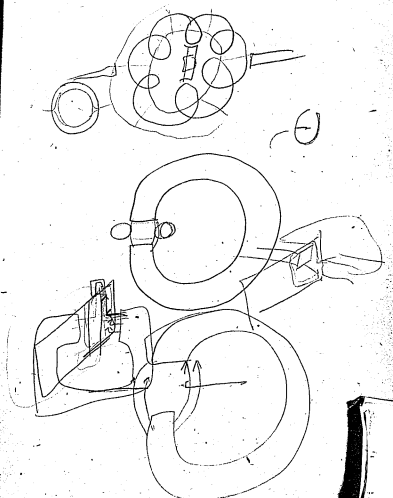


16

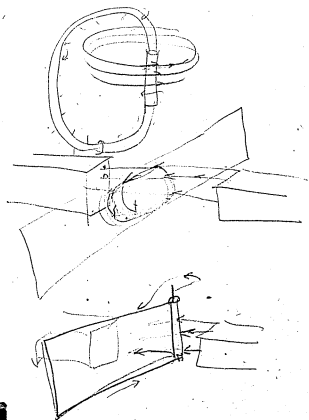
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 $\frac{198}{12} = 16.5$

198  
 $\frac{198}{12} = 16.5$

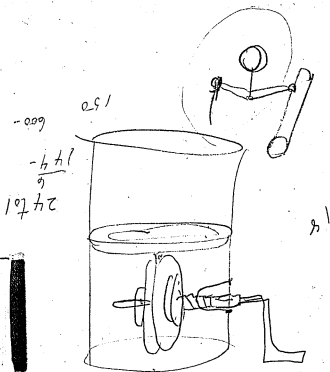
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$\frac{1406}{12} = 117.1666...$   
 $\frac{1406}{12} = 117.1666...$

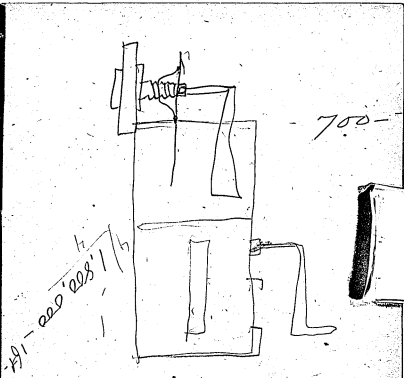
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 $\frac{12}{12} = 1$

$\frac{12}{12} = 1$   
 $\frac{12}{12} = 1$

$\frac{12}{12} = 1$   
 $\frac{12}{12} = 1$



$\frac{25.261}{0.31} = 81.34$   
 $\frac{20.522}{2.1} = 9.77$   
 $\frac{20.061}{2.2} = 9.12$   
 15- 750-  
 9.75-  
 75-



20 material 150  
 15 labor - 60  
 10- 20. } 25  
 15- } 15  
 25 } 40  
 25 } 44  
 53-

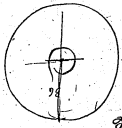
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 $\frac{204}{12} = 17$   
 $\frac{210}{15} = 14$   
 $\frac{216}{18} = 12$   
 $\frac{222}{11} = 20.18$   
 $\frac{228}{12} = 19$   
 $\frac{234}{14} = 16.71$   
 $\frac{240}{16} = 15$   
 $\frac{246}{18} = 13.66$   
 $\frac{252}{21} = 12$   
 $\frac{258}{22} = 11.72$   
 $\frac{264}{24} = 11$   
 $\frac{270}{27} = 10$   
 $\frac{276}{30} = 9.2$   
 $\frac{282}{33} = 8.54$   
 $\frac{288}{36} = 8$   
 $\frac{294}{42} = 7$   
 $\frac{300}{50} = 6$

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$\frac{198}{18} = 11$   
 $\frac{204}{12} = 17$   
 $\frac{210}{15} = 14$   
 $\frac{216}{18} = 12$   
 $\frac{222}{11} = 20.18$   
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 $\frac{234}{14} = 16.71$   
 $\frac{240}{16} = 15$   
 $\frac{246}{18} = 13.66$   
 $\frac{252}{21} = 12$   
 $\frac{258}{22} = 11.72$   
 $\frac{264}{24} = 11$   
 $\frac{270}{27} = 10$   
 $\frac{276}{30} = 9.2$   
 $\frac{282}{33} = 8.54$   
 $\frac{288}{36} = 8$   
 $\frac{294}{42} = 7$   
 $\frac{300}{50} = 6$

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$$\begin{array}{r} 36666 \\ 6733 \\ \hline 65-29933 \\ 3 \overline{)55000} \\ \underline{18333} \end{array}$$

(9.10) 5-33666  
 6733

18333  
 18000  
 3330

29933  
 25000  
 4933

19330

50. Cost sold 150-  
 5 freight - possible  
 profit \$95- to agent

who personally sells it  
 \$35. from  $\frac{95}{65}$  - left.

~~Attention~~  
 Basis of sales 2000. 1st  
 year. profit 120 000 -  
 general expense if NA ~~of~~  
 \$50 000 - leaves 55 000  
 profit. - of this 33 1/3 to

Local Co - 18333 -  
 leaves 36,666 to  
 NA - 20% Com from the  
 6733 leaves 29933  
 net to NAMEN -

this on 2000 machines  
 is \$9.10 per machine to  
 Local Co & 14.92 to NAMEN -

in case trouble with graphs  
 any royalty we may be  
 compelled p. ay be drawn  
 from these same in proportion  
~~Supplies~~ Rentals  
 to be abandoned as soon  
 as possible better  
 Rental Contracts are ended  
 but in numerous.



NAmen is to attend to  
the same where purchasable  
& after deducting Costs  
of maintenance so to  
divide the profits  
from rentals 33% to Local  
66 to NA -

Nickel in Slot, same  
way -

Supplies -

Local agents to get  
20% profit on Supplies  
for attending to business  
of the profit left  
40% is to be deducted

as general expense for  
conducting NA Co + ~~20%~~  
~~Consumption of~~  
~~bal. due bal + bal~~  
Net profit divided  
33 to Local 66 to NA

for instance, Cost of a  
Cylinder 10 cents.  
freight 1c Total 11c  
Sold at 25 Cents to  
Consumer 20% of che to  
Local agent is ~~20~~ leaving  
~~4~~ 20c deducting 11  
9 left  
9 profit deducting 20% gen  
exp for NAmen or 1.8c leaves  
~~8.2~~ 8.2 divided 33% to Local 66%

or 2.4 to local 4.8 to N.Amer  
other supplies figured same  
way

Thus the proportions are  
33% to local 66 to NA  
but from the 66% 20%  
can be paid or net 53% to  
N.Amer -

$\frac{66}{100}$   
 $\frac{13}{100}$   
 $\frac{53}{100}$

All musical Records

Except to be sold at  
uniform price East of  
Denver to the public

~~4.00 - 4.00~~  
~~from net~~  
for 50 cents. ~~plus freight~~

~~Cost~~ Cost now 33 but can  
be got to 30 c net freight

~~leaves 17 cents~~

~~leaves~~ Leaves 20%

cents profit. this to 62  
divided just a profit  
on blanks are divided

$$\begin{array}{r} 3 \overline{) 72} \\ 21 \\ \hline 24 \end{array}$$

$$\begin{array}{r} 4 \overline{) 90} \\ 18 \\ \hline 72 \end{array}$$

$$\begin{array}{r} 90 \\ 36 \\ \hline 54 \end{array}$$

$$\begin{array}{r} 3 \overline{) 54} \\ 18 \\ \hline 36 \end{array}$$

$$\begin{array}{r} 160 \\ 8 \\ \hline 20 \end{array}$$

Some area with mountain or desert  
 Alden M...  
 near house Mr. Little (Barton) Express

near house Mr. Little (Barton) Express  
 do near Wadsworth rd -  
 Alder mine - (Barton) Express  
 Little mine -  
 mountain mine 125 ft. E of Alder mine  
 Mary Moran opening further north  
 Fred mine  
 both holes opening  
 Do mine in the canyon  
 Alder mine  
 Fred mine  
 Alder gold mine  
 Alder gold mine

Very common for old site 1858  
Dumoulin's 1857 map -  
Dumoulin's 1857 map -  
M. R. on map by Dumoulin  
or west - Slope Green Mountain  
Crest with grass  
Nicket Mine at Barrow Castle  
Women's map named with  
mine open for 50 years,  
Nicket Mine, Barrow Mountain  
opened - 1858 probably 1800s  
Kearney Grant with address  
N. W. on  
Dr. K. B. K. mine 1 1/2 mi. on  
gravelly property - 1/2 mi. S. of  
Amherst - good field -  
geo. of Mt. P. 272 - Barrow  
by Com. Ross & by town of Barrow  
ETS by gravestone on  
Kendall's 1860s map  
of Barrow & Barrow

This is the idea of town  
of Kuller's Mines, Kymon, Barrow  
part of Atwood's 1860-66 & dist.  
Chap. 1000 - 2 mi. S. of  
Kymon - Barrow  
Advised good mine 1 mi. S. of Kymon  
Village -  
Amherst - 1858  
The day before of Village  
1500 ft high - 25 ft per  
Dodge Hill height 1100 - 25 ft per  
all suggests to say mine  
Barrow Mountain - 1/2 mi. S. of  
generally down in that  
Green Village in mine  
at Barrow - in shifts  
opened in 2 places -  
in same field - Mt. good mine - also in  
New England - in  
New England - in

phyllites deposits Lewis Co Va. Vol 12 mining logs

p. 529 - Map -

outlet 500-600 ft. - Va. - un-ly. - Md. - c. 1800  
from West. - from West. - from West.

W. Va. - Blue Hill Mt. - Va. - un-ly. - Md. - c. 1800

W. Va. - Blue Hill Mt. - Va. - un-ly. - Md. - c. 1800

W. Va. - Blue Hill Mt. - Va. - un-ly. - Md. - c. 1800

W. Va. - Blue Hill Mt. - Va. - un-ly. - Md. - c. 1800

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W. Va. - Blue Hill Mt. - Va. - un-ly. - Md. - c. 1800

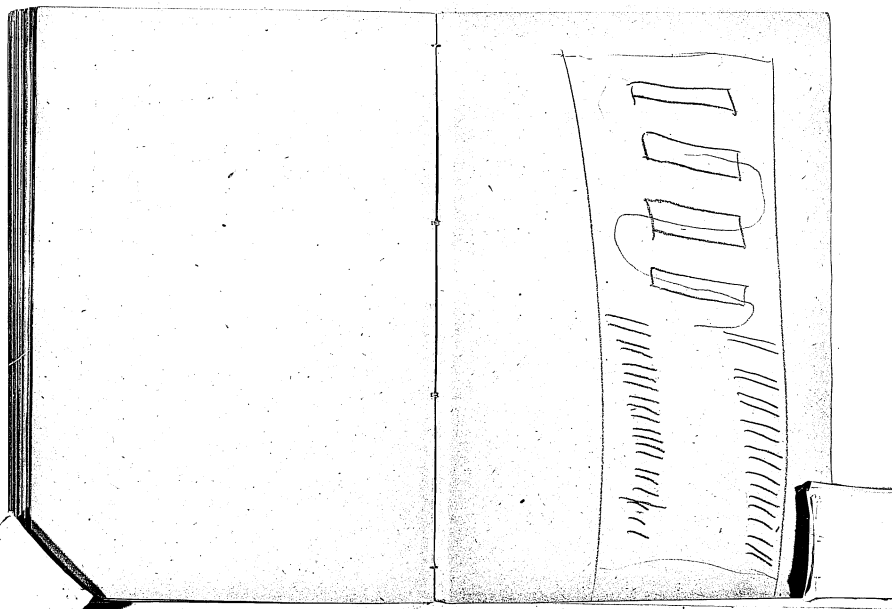
W. Va. - Blue Hill Mt. - Va. - un-ly. - Md. - c. 1800

W. Va. - Blue Hill Mt. - Va. - un-ly. - Md. - c. 1800

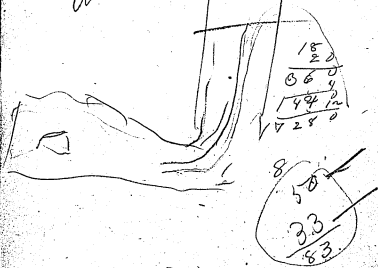
W. Va. - Blue Hill Mt. - Va. - un-ly. - Md. - c. 1800

W. Va. - Blue Hill Mt. - Va. - un-ly. - Md. - c. 1800

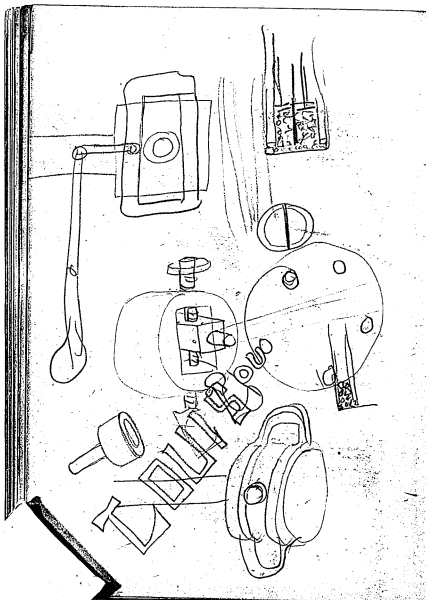
On kind mine No. 1  
Jallopore in mine ga.  
St. John's College  
Charleston W. Va. - Thomas Co. denton  
mine. also -  
Murchison  
Liberty mine Md. - Cotten from state  
table of sections of ga. mby.  
Clay + Co. Va. - US Geo Survey. 1852  
mining grant private claim 10% Co.  
not using records -  
from one of Adams map of Adams Mt.



Revision every year in  
 this proposition -  
 (books open to accredited  
 affirms each Co -



We do all the buying  
 We sell phones 150  
 plus ft - we pay  
 Comm 30. to  
 sell + 20% on  
 supplies we pay  
 20% Com to qual  
 tpt on net profit  
 to N.A. man, bal profit  
~~made~~ goes 50/83 to us  
 33/83 to Local -  
 Cost phone 50 - Cyls 10  
 mine 30 -



$10 -$   
 $2 /$   
 $2 /$

$X \frac{1}{\text{Divide}} =$   
 multiply - Subtotal  $\frac{1}{2} +$   
 all

$150$   
 $150$   
 $10$

Hand-drawn sketches of mechanical components on the right page of a notebook. The sketches include a cylindrical component with a handle, a large cylindrical component with a handle, and a complex assembly of parts including a gear-like structure and a cylindrical component with a handle.

*Blank*



**Notebook, N-90-01-04.5**

This notebook covers the period January 1890-April 1893. All entries are by Edison. The book contains notes, drawings, and calculations relating to ore milling, including what appears to be a list for a patent caveat. There are also electric railway drawings and notes for a series of chemical experiments. The front cover is labeled "#583"; the number has been crossed out. The inside front cover is inscribed "#583 T A Edison Jan 4. 90." The book contains 182 numbered pages and has been used in both directions. Two pages were removed before the book was paginated.

Filming order: 1-147, 182-152.

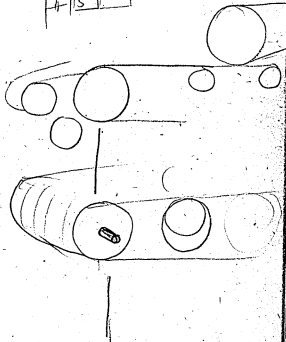
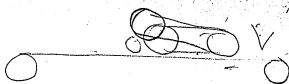
Blank pages not filmed: 2-3, 48-139, 148-151.

#583

N-90-01-04.5

J. E. Edison  
Jan 4. 90

XE-5286



|                                                                                                                                                     | sp grav    | Hardness | Streak                        |
|-----------------------------------------------------------------------------------------------------------------------------------------------------|------------|----------|-------------------------------|
| Pyrite -<br>FeS <sub>2</sub>                                                                                                                        | 4.83 @ 5.2 | 6 @ 6.5  | greenish or<br>Brownish black |
| 53.3 Fe 46.7<br>no Co<br>Cu Thallium sometimes replace<br>a little of iron -<br>Closed like S sublimation Resolves Magnetics<br>In HCl Bec Nitric - |            |          |                               |
| Chalcocite<br>53.9 Cu 20 Fe 3%                                                                                                                      | Sq. 4.02   | H 4      | Dark Reddish<br>Bronze black. |
| Closed like sublimation.                                                                                                                            |            |          |                               |
| Chalcopyrite<br>37.5. 34 Cu 30 Fe                                                                                                                   | 4 @ 4.3    | 3.5 @ 4  | greenish black<br>Sulfur      |
| Thallium + Selenium<br>Closed like description - Sub of S<br>dis in Nitric except green solution<br>Not in excess change to deep blue               |            |          |                               |

| Fracture                          | Phenomena                                                        | Color             |
|-----------------------------------|------------------------------------------------------------------|-------------------|
| Conchoidal<br>uneven<br>Brittle - | Streaks with steel                                               | Pale brown yellow |
| Conchoidal<br>uneven              | Subject to tarnish<br>after treatment<br>Can cut<br>with a knife | Brown yellow      |

Stannite

H. 4 - G 4.3 @ 4.5 Luster metallic  
streak blackish. Color steel gray to  
blackish - the former when pure, sometimes  
bluish - after effluvia from Calcopryte.  
Opaque - Fracture uneven brittle,  
S 30 Sn 26 Cu 30 Fe 12 @ G.

Zn 7 @ 8. Closed tube decryptite.  
fracture bluish. open sub of Sn oxide,  
Dec Nitric blue Sol separates 5+ oxide Tin

Stannite -

H 5.5 G 4.8 @ 5 Luster metallic

1600 mesh



20 tons

50 tons hour

1000 lbs -  
200%

10+

27  
4

100.

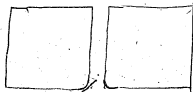
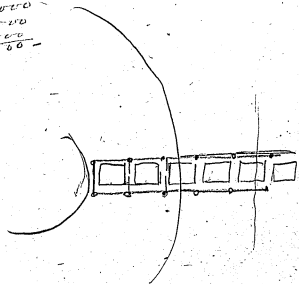
10800 -

1 pair rolls 50 hrs - hour  
50 screens - } net 10 tons / 100 m  
in 20 hours

10 pair rolls - 500 screens -

25 000 -  
10 000 -  
5 000 -  
12 000 -  
30 000 -  
10 000 -  
65 000 -

15 000 -





10 m. l.

78.3

Mining 25-  
 Crushing To 40 m.l. - 25-  
 Concentrate  $\frac{10}{60}$ -

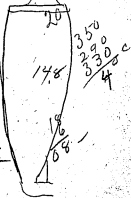
Must have \$/ to pay  
 concentrate - 220 lbs pyrite

or

#6 per ton Conc

Conc-

Drifing 3  
 Semerosty 10  
 Magnets - 10  
 Dead Roasty - 75-  
 Re Concly 9 - 50  
 148.



Bridging Iron 35-  
 Making Copper 125-  
 148  
 308

Total 908.-

get 75% of ore - at 1/3.  $\frac{908}{2.25}$   
 Total 683  
 1000  
 Total 783

908.  
 $\frac{600}{3.08}$

60000

18000.  
 18000  
 $\frac{36,000}{2}$

2-

5000-

$\frac{20000}{12000}$

$\frac{62}{8000}$

250

$\frac{25}{225}$

$\frac{33}{22}$   
 $\frac{66}{66}$

2240 275000 (123)  
 $\frac{2240}{3140}$   
 $\frac{4480}{6200}$

$\frac{9}{172}$   
 $\frac{185}{185}$

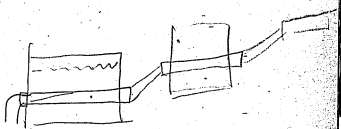
$\frac{3}{225}$   
 $\frac{75}{75}$   
 $\frac{3}{375}$

375- Cn  
 225- Fe  
 $\frac{600}{2}$

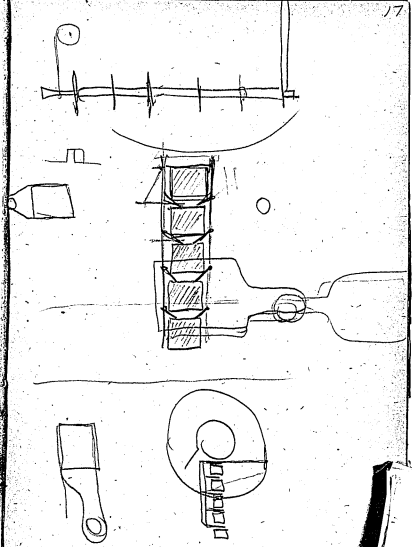
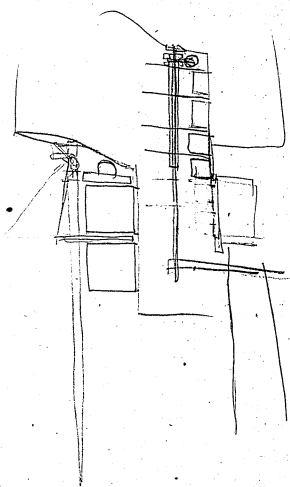
$\frac{60}{8000}$

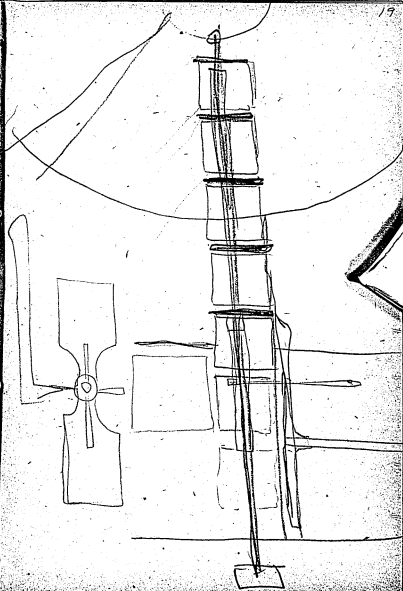
10. 30

$\frac{175}{5250}$

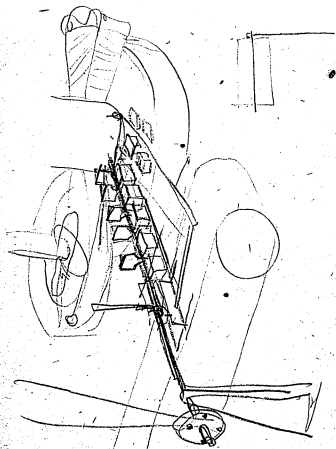


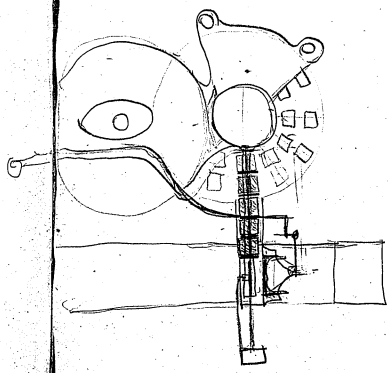






36-

$$\begin{array}{r} 49 \\ 78 \\ \hline 392 \\ 343 \\ \hline 3822 \end{array}$$
$$\begin{array}{r} 36 \\ 78 \\ \hline 288 \\ 252 \\ \hline 2808 \end{array}$$




Note - Vol. 4 to V.  
on pgs 25-47 refers to  
Watts, Dictionary of Chemistry  
1983 Edition.

N.R.S.  
10/6/83

Miscellaneous -

By action of ammonium sulphur on acetone a number of products are obtained of indefinite character -

Acetone Aconitic acid - aconitate Calcium  
gelatinous when nearly dry than Resin. -

Albumen Combines with Lime, product dried  
after while hard as stone also Combines  
with Na K Baryt. etc.

Chloride Cal combines with Alcohol to white  
amorph mass see Vol 1 - 80 -

Sulphide of Alloy form with aqua regia of  
gold beautiful yellow pp - calcite Resinous  
mass as becomes covered with films gold.  
Vol 1 - 144 -

Sulphide's Platinum - pp - when dried at 100 C  
takes fire & glows until reduced to Pt -  
Lump -

~~Antimonides~~ of Methyl -  
Hydrate = Slovego bat. Vol 1 392

Arsenic alloys w/ Cu white Malleable  
Pbly no focaults,

Arsenite of Sb, transparent fused  
Vitrim mass

As-Arsenate Aluminum  
Stannous Arsenate, gelatinous

Sulpharsenite Tin - gummy  
Sulpharsenate Silver grey ductile cake  
Inhibiting metallic dusts,

Cyanide of Cacodyl - for War -  
Bromide gold reds so Pt. dec by slight  
heat = (Heat photology) -

Chrysammate KO Crystals  
peculiar effect of light,

Cinchonidine electric by  
friction.

Sulphate Cinchonine becomes  
phosphorescent at 100 C.

Sulphate of Todeo cinchonine  
Crystals act like tourmaline to  $\frac{1}{2}$   $\frac{1}{4}$

~~Cocoa~~ Cocoa wax strongly  
Electrified by friction

End Vol 1

31

Cobalt cyanide, w/ Hydrogen  
heated to 250C leaves blue  
powder & higher temp Carbide  
of Cobalt, black is it a Cinder

---

Carbide gold 214 2nd -  
" 21m - 222.

Platinocyanide strontium - white  
but so sensitive to monochrome that  
breath turns crystals blue -

---

Sulphurate Calcium takes hair off  
Formate Hg decomposes in dark  
also by pressure on crystals or 100C heat



33  
Formate Silver eventually  
black in the dark see Kennedy

Fulminate Silver  
gradually blackens when  
exposed to white or blue light.

Chem compounds of gelatin  
& phos Calcium is obtainable

glucinum don't combine  
with sulphur at any  
temp - forms alloy  
with Silicon -

glycerides in Chloroform  
& alcohol -

36  
Guaiacum Violet rays  $\text{FeCl}_3$   
turn green - Red ray

restores the yellow color

End 2nd Vol

Protoclauride of Iodine - Chlorine  
removes an H & Iodine goes in -  
to Iodine places -

Barium Salt of Iodine acid  
swells up over 100 times its  
bulk -  $\text{K}_2\text{O}$  salt melts  
350 & solidifies to a material  
like porcelain - without  
loss weight

Copper amalgam expands on setting  
it highly plastic like clay

Iron amalgam - magnet (10) Hg  
globules magnetic - Ampermeter -

Mercuric oxide the weakest of all  
basic oxides being only salt  
pp from its salts by water alone

Boric Methide, takes fire  
spart in air bright green flame  
if assm through tube  
in air amount of smoke  
disproportionally great. 2 or 3 Cubic  
inches of gas burnt this way  
fills large room full Carbon  
flocks. flame vis & hard  
can be held in it Vol 3 986

Methylamine is the most soluble  
in water of all known gases

Molybdenum harder than  
Topaz —

Phosphide of Molybdenum  
Conducts Electricity diffuses

Volatile Oils when cold dissolve  
Oxide Copper & deposit the O<sub>2</sub>  
on heating = O<sub>2</sub>

Olivine - Reson electric

Oxalate Calcium dry  
highly electric

41  
Palassio-Manganic Oxalate  
purpled by <sup>very early</sup>  
decomp Light & heat

do Ammonio-mercuric Oxalate  
do Palassio-mercuric "

See - 432 - IV Reservoir the 2<sup>nd</sup>

Hydrochlorate of Chloraniline  
the Chloroplaticate of a-chloraniline  
gives violet film on ft. to light.

Ethylaniline turns brown  
on exp to light or air -  
amylaniline odor of  
Roses

See - ~~phosphite~~ Bisphosphite  
Ethyloxy - 479 - IV  
+ acetate " "

Phosphite Chromium dry salt  
on heating gives off pure

Hydrogen

Thioisulphide of phosphorus in oil,  
shines in dark - 599 - IV

Pyperine - See 659 IV slide

Tripotassamide - for red + sandy  
else for white delting in air  
in vac by cracks - Vol IV 696

Pat + Phos under Naptha  
must only boiling Naptha

See B-8510  
Watts Dictionary of  
Chemistry

45

Tartaric or Citric acid ignited  
~~in a dish~~ make a  
pyrophoric stuff - might seal  
in vac when slowly permit

O in =

Sulphides KO Na Lithium  
~~bluish~~ obtained by igniting

Corresponding sulphates with  
Lamp black -

Alum or Hamburgs pyrospoon  
by heating alum with  
Lamp black, starch, or flour

Ichthyol & potassium inflam

IV 765-6 -

Hydroferrocyanite Quinje  
Resembles Mosaic gold

20. Val V

Zodoginini - effect of light fuming  
Val V 25

with oxide lead  
 Pyrotartaricamide, dissolves paper  
 + Can't be filtered

gun cotton when ignited in small  
 quantities in rarified atmos.  
 Exhibits 3 diff luminous phenomena  
 in high vac beautiful green glow  
 as vac gets lower, yellow etc.  
good = see pyroxylin IV 780

3rd Sup pt 2nd obsvalhite  
 See oxidation p 1430 —



April 10/13

- 1 The use of an odd number of rolls -

Applying pressure to the Conchings

- 2 Rolls by a multiplying rope with 2 or more ends getting sub as described -

- 3rd arranging one roll above the other

- 4th first passing the one through the other then the other.

- 5 Important in fact of Conching  
Conching in producing great pressure between the two or more Conching rolls by the multiplying action of moving ropes or equivalent device arranged sub as described so that such pressure without force transmitting such pressure through fixed bearing surfaces -

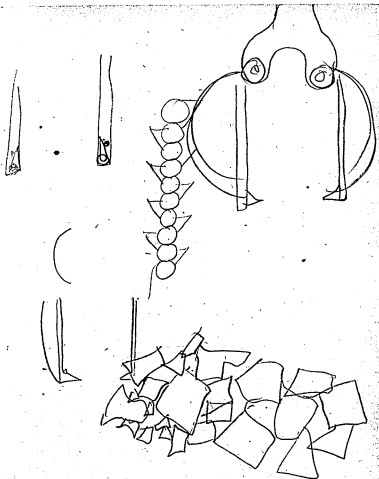
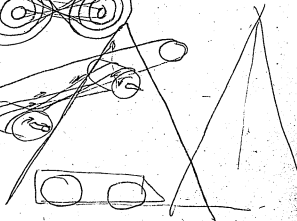
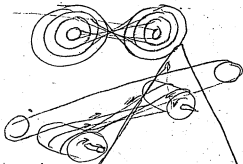
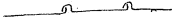
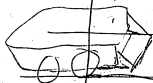
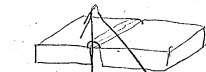
The use of a spring or equivalent weight producing a constant but relatively small tension on a traveling rope or multiplying such tension many times to produce great pressure between rotating Conching rolls sub as set forth,

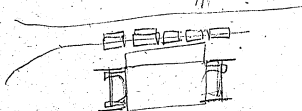
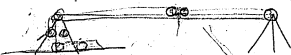
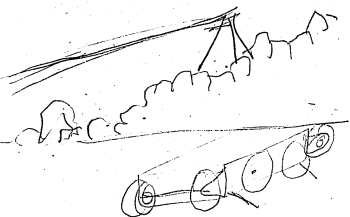
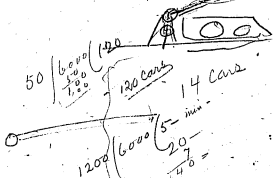
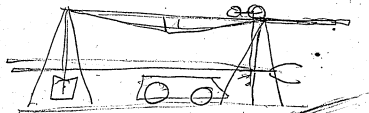
The fixed sheave ~~is~~ removable sheave takes up.

Two rolls above could be used the ropes on the sheave crossing at each turn, but the method of application is not so effective as with the 3 rolls -

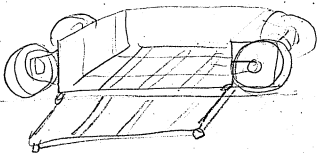
Sign

142





146



[THIS BOOK WAS USED IN BOTH DIRECTIONS.  
THE FOLLOWING PAGES WERE FILMED FROM  
THE BACK END FORWARD.]

300

14820  
9490  
5330  
1,597,009490)2250(

280  
350  
1350  
84  
850

290  
350  
1450  
870  
8150

240  
350  
640  
98000

280  
350  
40  
84  
98000

2280  
650  
1140  
1360  
1482000

285 - 650  
10

8, mills.  
opening min.  
PR - etc.  
800 000  
80 000  
200 000  
1,080 000

8, Mills.  
8000 tons daily  
000 3200  
1200  
570  
570  
250  
570

285  
800 tons daily  
4 | 2280  
370  
freight.  
18,000, 6360

2280.  
12  
2280 Crude etc.  
570  
250  
3130  
40  
800000  
10000  
120000  
6360  
3130  
9490

1600  
333 / 10900 (333)

285  
333  
531

285  
44  
143  
143  
285

27  
160  
216  
216  
48

285  
333  
2664

333  
333  
666

265  
333  
435

265  
10900  
10900  
21800  
21800  
35500

2664  
300  
799200

2.400.000 / 66.666.666 (27)  
48000000  
18666000  
16800000

269.

2664  
269

23976  
14884  
3328  
715616  
3500  
146800

6666  
801 / 66660000

80900  
2700000

27  
12 feet

400  
200  
12 / 80000 (666)  
720  
72  
78  
72  
78

40

$\frac{350}{2800}$

60

$\frac{75}{300}$   
 $\frac{23,500}{300}$

500 | 1090 - |

450

400

5

$\frac{2,60}{2,25}$   
 $\frac{4,25}{4,25}$

333 | 1090 - | 3.

390

400 | 1090 |  $\frac{7,50}{900}$   
 $\frac{400}{900}$

450.

2496  
6800  
1890

436  
6800  
1890  
2490

23 -  
521  
1400  
1700

32  
574  
1810  
1270  
640  
1500

72  
720  
2230  
660  
1570

30  
125  
625  
935

250  
312  
1500  
6800  
651  
400

72  
720  
2230  
660  
1570

1000 60c

3c per 100 lbs - Mining & Moly -

15 pc ore 12 net. 220

Expenses \$600.

|            |          |        |       |      |              |
|------------|----------|--------|-------|------|--------------|
| 10 pct ore | 125 tons | at 500 | - 500 | 6000 | 180 -        |
| 15 "       | 150 "    | "      | 500   | 525  | profit 225 X |
| 20 "       | 200 "    | "      | 500   | 1000 | " 400 X      |
| 25 "       | 250 "    | "      |       |      |              |

|            |     |       |     |     |             |
|------------|-----|-------|-----|-----|-------------|
| 10 pct ore | 125 | - 500 | 20c | 625 | - profit 25 |
| 15 "       | 187 | -     |     |     | 335         |
| 20 "       | 250 | -     |     |     | 650         |
| 25 "       | 312 | -     |     |     | 960         |
| 30 "       | 374 | -     |     |     | 1270        |
| 35 "       | 436 | -     |     |     | 1580        |
| 40 "       | 498 | -     |     |     | 1890        |

125 / 5500 / 2 unit on every 1000  
 125 / 1000 (8.  
 250 / 2500 / 2.5 3 / 1000



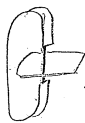
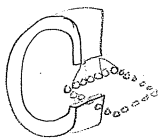
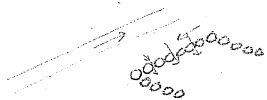
10 picture

2.5-C Raw Ton

9600

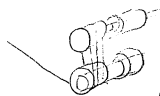
127 1 ) 6500 6

1000 ) 3350 ( 3.35  
3000  
3500  
3000  
5000

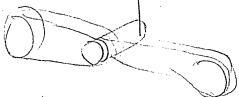


1-1/2" x 1" with  
2-1/2" x 1" with cut.

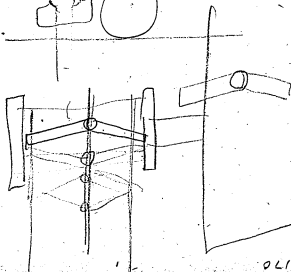
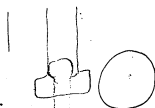
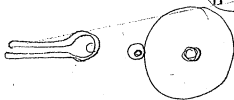
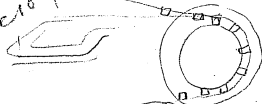




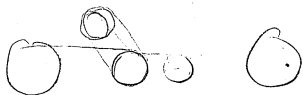
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Dec 10, 90



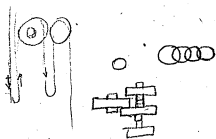
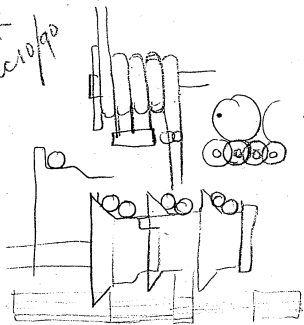
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Dec 10, 90



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Dec 10/90

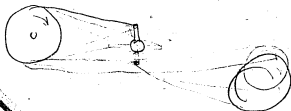


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Dec 10/90

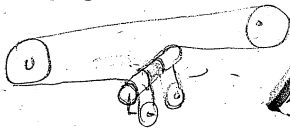
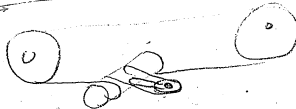
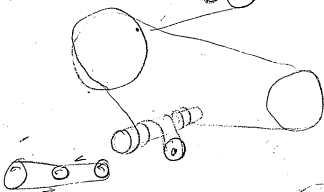


11/2

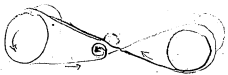
ERK Dec 10/90



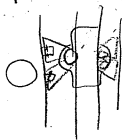
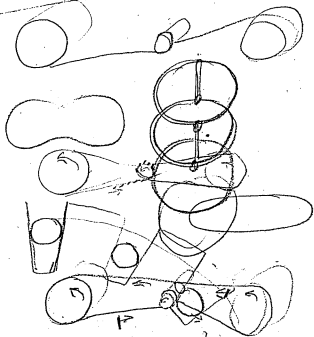
ERK Dec 10 90



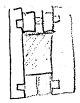
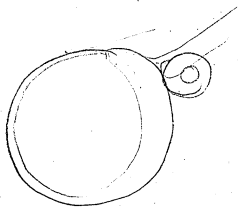
SKR Design



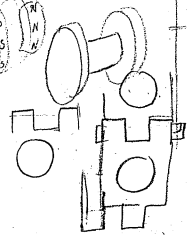
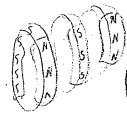
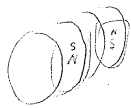
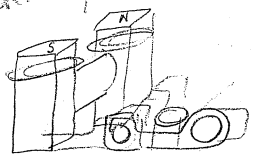
SKR Design



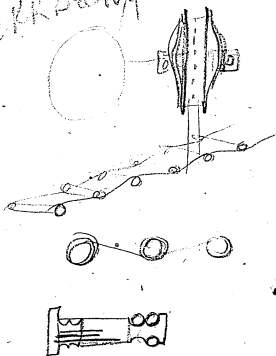
ERR Dec 10/90



ERR Dec 10/90



SKK 250-1040



$$\begin{array}{r} 60 \overline{) 1696} \\ \underline{120} \phantom{0} \\ 496 \\ \underline{480} \\ 16 \end{array} \quad \left| \begin{array}{l} 28 \\ 28 \end{array} \right.$$

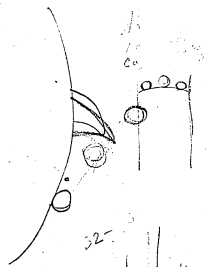
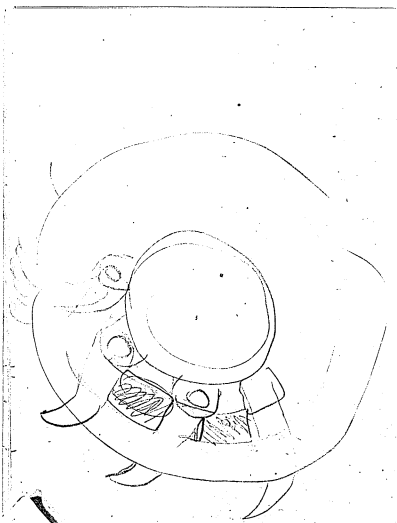
28000 ft in hour

12-

$$\begin{array}{r} 28000 \\ 33000 \overline{) 56000000} \\ \underline{330000} \phantom{00} \\ 2300000 \\ \underline{240000} \phantom{0} \\ 200000 \\ \underline{200000} \\ 00000 \\ \underline{297000} \\ 230000 \end{array} \quad \left| \begin{array}{l} 2050 \\ 1696 \end{array} \right.$$

0917

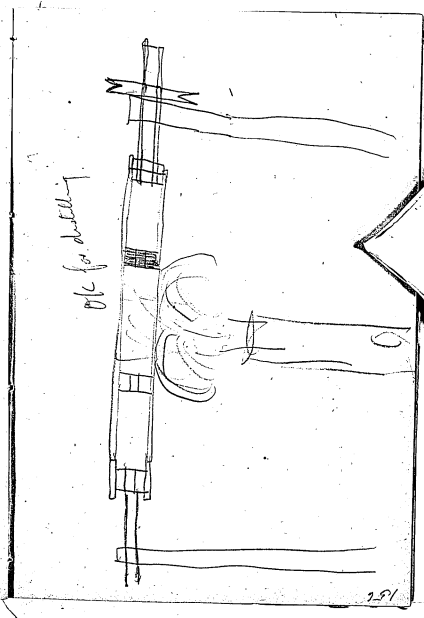
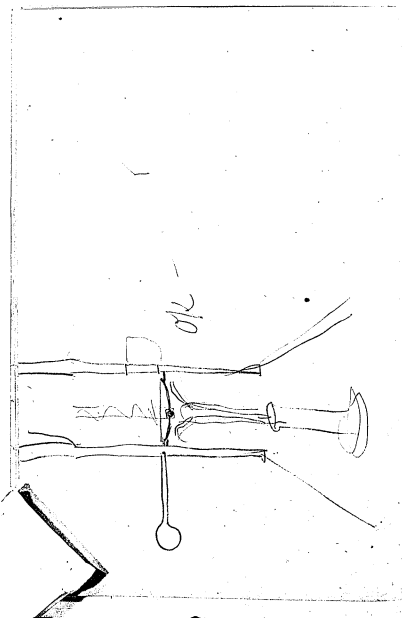


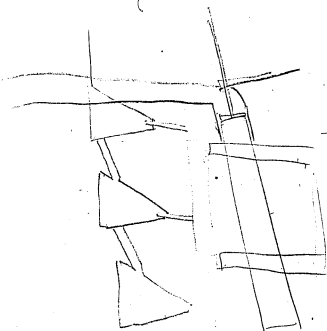


800-  
650-

$$\begin{array}{r}
 15 \overline{) 493} \quad 329 \\
 \underline{45} \phantom{0} \\
 43 \phantom{0} \\
 \underline{30} \phantom{0} \\
 130
 \end{array}$$

$$\begin{array}{r}
 25000 \\
 25000 \\
 15000 \\
 16200 \\
 18200 \\
 30500 \\
 29700 \\
 108000
 \end{array}$$





$1000 \frac{3}{4}$   
 $\frac{20}{2000}$

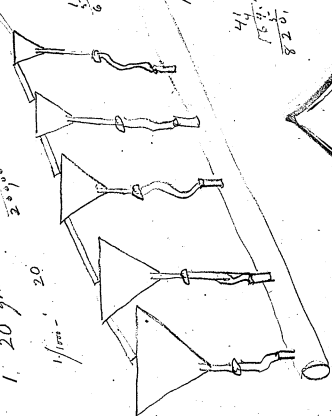
$1. 20 \text{ gram}$   
 $1/1000 - 20$

$\frac{1}{4} X \frac{1}{4} 2 \text{ gram} = 5 \text{ gram (rad)}$

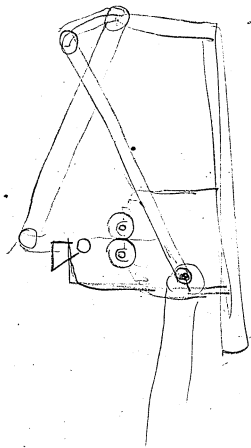
$2.30$   
 $2.30$   
 $1.00$   
 $6.31 \quad 4/1000 - 5000$

$120 \quad 5000 \quad 41.$   
 $\frac{4800}{2}$

$41$   
 $\frac{1}{2} \quad 75c$   
 $15 \quad 20.$   
 $820.$



451



**Notebook, N-90-01-04.6**

This notebook was begun in January 1890. All entries are by Edison. The book contains notes, drawings, and calculations relating to ore milling, including cost estimates based upon labor and energy requirements for separating iron, copper, and nickel ores from the Sudbury mining region. There are also notes regarding the testing of numerous chemical compounds for potential use in incandescent lamp filaments or phonograph cylinders. Edison's notations indicate that he assigned many of these chemical tests to an experimenter named "Joe," possibly John Joseph Force. The front cover is labeled "584." The inside front cover is inscribed "584 T. A. Edison Jan 4. 90." The pages are unnumbered, and at least one page has been removed from the back of the book. Approximately 130 pages have been used.

90-01-07.6

# 582

J.A. Edison

la 4.90

64 53  
10- 40  
5-vakt.  $\frac{44}{176}$

32  $\frac{44}{22000}$

32 amp -  
44

$\frac{1}{400}$  of ohm

40)  $\frac{1296}{1200}$   
 $\frac{96}{96}$

400 -  $\frac{32}{128}$   
 $\frac{128}{408}$

11 of 16.  
 $\frac{44}{4400}$   
1 1/2 hp  
32

1. 44  
2. 176  
4. 704  
8. 2816  
16. 11264  
32. 45056

8 ohm - 3 sqmches -

3 ohms per sq inch = .55 vakt.

$\frac{144}{296}$  -  
1 3  
2 1.500  
4 .750  
8 .375  
16 .188  
32 .094  
64 .047  
128 .0235  
256 .01175  
512 .00562  
1024 .00281

15  
 $\frac{1}{2}$  sq -  $\frac{1}{2}$  hp  
pukku  
1 lb per  $\frac{1}{2}$  hp

32 amp - 39  
15 ohm  
11 ft lbs  
44  
176  
704  
2816  
11264  
45056  
1  
2  
4  
8  
16

300  
2 1/2 1/2 1/2 1/2  
240 400 amp 1 amp  
12 lbs

1 volt 1 ohm 44  
24 lbs per day per  
100 hp per ton.  
1200 hp  
 $\frac{400}{17600}$

25  
 $\frac{6000}{150000}$

2

4400)  $\frac{10000}{2000}$  - 23 -

25 - 6  
 12 - 3  
 9

75000  
 27000  
 10000  
 8000  
 $\frac{120000}{1}$  72

30.  $\frac{130 \text{ hrs}}{32}$

400. amp  
 800 L.H.

64  $\frac{1200}{600}$  14  
 $\frac{560}{312}$

7  
 $\frac{1000}{35}$   
 12

2000 hp Trip Expn -

|                   |               |
|-------------------|---------------|
| 1/25 gun hp -     | 75000         |
| 1/9 gun hp boiler | 27000         |
| Bldg -            | 15000         |
| Pump, piping etc  | 10000 -       |
| Drawings -        | 54000         |
| Beltshopfy .      | 10000         |
| Wiring -          | 12000         |
| Units -           | 25000         |
| Crane &           | 5000          |
| Cost Mater -      | 2000          |
| Restor & Sln -    | 15000         |
| Water ptg etc -   | 5000          |
|                   | <u>255000</u> |
| Vat Home -        | 20000         |
|                   | <u>275000</u> |
|                   | <u>10000</u>  |
|                   | 450000        |



1 million grams

24  
 $\frac{24}{4800}$

16  
 $\frac{16}{768}$

100.

320  
 $\frac{320}{6400}$

2000  
 $\frac{2000}{4800}$

24  
 $\frac{24}{1308}$

2000  
 $\frac{2000}{200}$

96  
 $\frac{96}{200}$

2000 per year

320  
 $\frac{320}{200}$

amp  
 57160

30 ton load

11 pector

24  
 $\frac{24}{144}$

4  
 $\frac{4}{216}$

21,000

2.

54000

6  
 $\frac{6}{48000}$

20 pcp per yr -

117

30

75000  
 27000  
 15000  
 18000  
 5000  
 132000

Coal - 3 lbs stock, 50¢ 4 tons hour -

#2 per hour \* 48. 300 16000-  
 15000 7500

12 fire 5000  
 4 engine 2400  
 4 oiler 2000  
 oil 7000  
 Repair 1000  
 grate bars 1500  
 Insurance 2000  
 Taxes 4000  
 Chief Engr 5000

3,53400  
 16000  
 21000  
 9000

3000  
 53400 17  
 3000  
 25400  
 21000  
 24001

$$\begin{array}{r} 40 \overline{) 650000} \text{ amp} \\ \underline{40} \phantom{000} \\ 250 \phantom{00} \\ \underline{240} \phantom{0} \\ 100 \phantom{0} \\ \underline{80} \phantom{0} \\ 200 \\ \underline{200} \\ 0 \end{array}$$

16250 - 25 vatts.

3- 5  
 15 30.  
 40-  $\frac{24}{4.8\%}$   
 10- 4 tons Muller -  
 100  $\frac{200}{200}$

Coal 32,000  
 Chief Engr - 3,000  
 2 Engr - 25,000  
 1 oilers - 15,000  
 oil 12,000  
 10 fremm 6,000

$$\begin{array}{r}
 3000 \\
 562.00 \\
 \hline
 3000 \\
 \hline
 262.00 \\
 240.00 \\
 \hline
 22.00
 \end{array}
 \left( 1875 \right)$$

~~60~~  $\frac{75}{15000}$

127,000  $\frac{25}{16000}$   
 225,799  $\frac{16000}{16000}$   
 26(

$$\begin{array}{r}
 16000 \\
 \hline
 64 \\
 3000 \left) 32000 \right/
 \end{array}$$

140 -  
35 per hour

190  
4

50

12

250  
2244  
8960

35  
1750

3500  
1050000

50000  
70000  
920000 -  
8650  
3344  
12400

280  
1400  
5600  
7000

70  
3750

7500  
375000

14000  
28000

4900  
3750  
8650

280  
1700  
14000  
19600  
280  
49000

280  
44000  
32000  
132000  
8470

4700  
19000

2249  $\frac{12000}{4000} (3.0)$

$\frac{10000}{125} = 80$

$\frac{12000}{12000} (1)$

$\frac{12}{24} = \frac{1}{2}$

32  $\frac{365}{730} = \frac{1}{2}$

$\frac{10000}{12000} (8)$

288  $\frac{365}{252} = \frac{1}{2}$

$\frac{2920}{730} = 4$

6000 hp - 24 hours 365 days -

|                     |                   |
|---------------------|-------------------|
| Coal 2 lbs at 2.25  | 105,000           |
| Chief Engr. -       | <del>4,000.</del> |
| 4 Engineers -       | 6,000.            |
| 2 oilers -          | 1,500.            |
| 2 Laborers -        | 1,200             |
| 16 firemen -        | 10,650            |
| 6 laborers -        | 3,500 -           |
| 2 firemen foremen - | 4,000             |
| Grate bars -        | 2,500             |
| Oil,                | 2,500             |
| Repairs -           | 5,000             |
| 24000 Interest 6pd. | 14,500            |
| Depreci -           | 9,500             |
| Insurance           | 1,800             |
| Taxes -             | 3,000             |
|                     | <hr/>             |
|                     | 174,650           |

365) 174000 (477  
 1465  
 2585  
 2450

50 tons

200-

150 hp for Co. Corp

150  
 60  
 2150

6000  
 6000

40-

365) 477 (119  
 427  
 50  
 370

365) 644  
 2380  
 2900  
 1800

174000 total cost 6000 hp -

6000) 174000 (29 pump  
 12000  
 54000  
 54000

Steam only

|                                     | Daily |    |
|-------------------------------------|-------|----|
| Helpers for 200 tons Matte 40 men - | 20    | 20 |
| Melters -                           | 10    | 10 |
| Copper strippers -                  | 5     | 10 |
| Roasters, _____                     | 5     | 10 |
| Strip makers _____                  | 2     | 4  |
| Solution mix _____                  | 2     | 4  |
| pumpers - _____                     | 2     | 4  |
| Dynamo men & rollers -              | 6     | 12 |
| <del>Other</del> Inspectors -       | 2     | 8  |
| Electrician _____                   | 2     | 10 |
| Chief Electrician _____             | 1     | 10 |
| " Chemist. _____                    | 1     | 20 |
| Chemist. <del>4</del> _____         | 4     | 10 |
| Daily                               | 142   |    |

|                        |               |            |
|------------------------|---------------|------------|
| Salaries Contin'd      | bratford      | 142        |
| Melters of pure Copper | 6             | 12         |
| Brickers of $M FeO^2$  | 6             | 12         |
| laborers leadpile      | <del>40</del> | 70         |
| Genl Managers          | 1             | 30         |
| Coal for roasting      |               | 75         |
| loading                |               |            |
|                        |               | <u>341</u> |

240000 - Interest 6

|  |              |
|--|--------------|
|  | 365 days     |
|  | <u>1705</u>  |
|  | 2046         |
|  | <u>1023</u>  |
|  | 124468       |
|  | 14500        |
|  | 2500         |
|  | 500          |
|  | 9600         |
|  | <u>10000</u> |

Taxes;  
 Ins -  
 Depen - 4 pct.  
 Repairs  
 Electrical  
 Eng

|  |               |
|--|---------------|
|  | 162563        |
|  | <u>174000</u> |
|  | 336565        |

$$\begin{array}{r} 934 \\ 3724 \end{array}$$

$$\begin{array}{r} 931 \\ 1862 \\ 2793 \end{array}$$

$$\begin{array}{r} 2) 931 \\ 465 \\ \hline 1396 \end{array}$$

$$6000) 365000 \left( 68 \frac{1}{2} \right) \begin{array}{r} 50000 \\ 7200 \\ 4000 \\ \hline 74000 \end{array} / 66$$

$$\begin{array}{r} 4) 2793 \\ 932 \\ \hline 698 \end{array}$$

$$365) \begin{array}{r} 68 \\ 3400 \\ 3285 \\ \hline 11150 \\ 550 \\ \hline 11396 \\ 349 \end{array} (9.31$$

$$\begin{array}{r} 4) 1862 \\ 465 \end{array}$$

$$\begin{array}{r} 4) 1396 \\ 349 \\ \hline 1/3 \end{array}$$

In round Noo

11000 per day -

~~Over the year~~

or 68. per hp per year 24 hours

If 1 ton Cu takes 50 hp <sup>per 24 hours</sup> then 1 year gives 365 tons per 50 hp - or \$9.31 per ton with the Fe in bricks or cans - Cu in ingots with interest and municipal paid -

|                        | Tim Cu       | Tim Milled |
|------------------------|--------------|------------|
| If 75 hp required then | Cost. 1396 - | 349        |
| If 100 hp " " "        | 1862         | 465        |
| If 150 " " "           | 2793         | 698        |
| If 200 " " "           | 3724 -       | 932        |

So



$$\begin{array}{r}
 75 \\
 \hline
 37,500 \\
 44 \\
 \hline
 86,500
 \end{array}$$

$$\begin{array}{r}
 280 \\
 \hline
 175 \\
 140 \\
 \hline
 190 \\
 28 \\
 \hline
 49,000
 \end{array}$$

1250

freight from Sudbury - to Orange say  
 1/5 penton matte - or 1/20 for the 25 pct  
 of Copper matte.

Matte Cost 8650.  
 + 500  
 9150.

Cost matte . 9150  
 Cost wh<sup>at rate</sup> 75 hp penton Cu - 349  
 9499.

at rate of 100 hp  $\frac{465}{9615}$

at 150 -  $\frac{698}{9848}$

at 200 -  $\frac{932}{10052}$

$$\begin{array}{r} 1680 \\ 260 \\ \hline 1084 \\ 400 \\ \hline 1504 \\ 131 \\ \hline 613 \end{array}$$

$$\begin{array}{r} 1300 \\ 6500 \\ \hline 6500 \end{array}$$

613

$$\begin{array}{r} 2500 \\ 300 \\ \hline 58400 \end{array}$$

$$\begin{array}{r} 2500 \\ 230 \\ \hline 5200 \\ 6440 \end{array}$$

$$\begin{array}{r} 250 \\ 200 \\ \hline 400 \\ 26000 \end{array}$$

$$\begin{array}{r} 120 \\ 140 \\ \hline 480 \\ 120 \\ \hline 600 \end{array}$$

at the worst // 100 per ton

$$\begin{array}{r} 500 \text{ lbs Cu at } 13 \\ 250 \text{ lbs Ni at } 30c \\ \hline 6500 \\ 8400 \\ \hline 14900 \end{array}$$

$$\begin{array}{r} \text{Ni at } 25c \\ \text{Cu } 6500 \\ 7000 \\ \hline 13500 \end{array}$$

$$\begin{array}{r} \text{Cop at } 10c \\ \text{Ni at } 23c \\ \hline 5000 \\ 6440 \\ \hline 11440 \end{array}$$

Lowest profit 14 per ton

Surplus profits above interest  
at full Capacity or highest  
est 200 hp per ton Cu  
gross 30 tons daily or  
120 tons month at 14 prof 1650  
per day or 613 000 per year  
or 150 per day -

$$\begin{array}{r} 435000 \\ 105000 \\ \hline 171000 \\ 175200 \\ \hline 135200 \end{array}$$

$$\begin{array}{r} 11380 \\ 9040 \\ \hline 2440 \end{array}$$

$$\begin{array}{r} 8650 \\ 8200 \\ \hline 450000 \end{array}$$

692000

$$\begin{array}{r} 120 \\ 365 \\ \hline 7600 \\ 26500 \\ \hline 34100 \end{array}$$

$$\begin{array}{r} 365 \\ 36000 \\ 41200 \\ \hline 36500 \\ 43700 \\ \hline 12000 \\ 12000 \\ \hline 113 \end{array}$$

8000 tons covered in vatts at 8650  
 \$ 692000 - used at rate of 120  
 tons daily (200hp per ton) lasts 66 days  
 say 80 days @ 113 per day 6 pct  
 9040 for 80 days or 110 per ton in vatts  
 or 440 per ton Copper -

Coal bill 105000 -  
 2 lbs per hour per hp at 23¢  
 If moved where Coal was  
 112½ saving to 52500 yearly  
 Cost per ton in vatts 105000 rate for  
 Coal (200hp per ton C) 240¢ for Coal  
 112½ Coal 120  
 56 C Coal 60

## Copper Elect -

Try Alumina Hydrate  
to form aluminate of  
Cop or Ni. to separate  
from Fe -

Try Rosin which is  
cheap & powerful acid  
to sep Cop or Ni from Fe

Acetal. liquid -  
glacial acetic acid. liquid

acetamide -  $C_2H_5NO$ . solid  
melts. 78 boils 221. C.

Acetic Ether

Acetate of allyl. liquid.

" " Benzyl-

" " Ethyl.

" " methyl.

*any lac is an alk.*

Acetonitrile -  $C_2H_3N$  (eq of methyl)

Hydrate of phenyl- or phenylalcohol

Cetyl alcohol

Acetic Aldehyde - or Hydrate  
of acetyl - liquid -

act on oil amber  
with Nitric acid  
do on amber by  
Nitric.

acetic Aldehyde -

Propionic

Butyric - "

Butyric acid

Formic acid

Benzoin aldehyde - OK

Allyl alcohol

Oil Garlic - or Sulphide of  
Allyl -  $C_6H_{10}S$ .

Allyl Mercaptan.  $C_3H_6S$  OK

Allylamine.  $C_3H_7N$ . OK

Oil Almonds OK

Amyl or Fusil oil  $C_5-H_{11}$

Hydrate of Amyl- $C_5-H_{12}O$ , or Cornel Fusil

Hydride of Amyl- $C_5-H_{11}H$ .

Amylic Ether.  $C_5-H_{11}O_2$

Hydrate of Amylene. or Amylene Glycol  
 $C_5-H_{12}O_2$

Oil anise oil

---

Amaric - When HCl is poured on Amaric  
Oil formed, which gradually solidifies on drying  
& may be drawn into threads on heating. It  
distills without decay, as an oil solidifying  
to brownish mass - Pharaoh,  
Watts Vol 1 162 -

---

In distillation of Amber, there is a white  
Substance produced as well as a  
Bitumen - White Sul has 95.6 per Carbon  
Cal H - ~~it~~ it melts Nitric Residue, &  
Watts Vol 1 162 - Fils

Anchoic Acid (stry) -

Anchoate Ammonium dis water soly non-  
Electric. Amorphous - Barium Salt.  
Transparent films - dried over  $H_2SO_4$  form  
white opaque mass like Porcelain -  
Watts 1 290 Phenol.

---

Anchusin - Coloring matter of alkanet  
root, (*Anchusa Tinctoria*). Amorphous  
does not fuse softens gives off iodine like  
Vapors at 60 C. - & carbonyl at higher temp  
dis ether - Turpentine & fixed oil.  
Fils Watts 290 - Val 1

---

Angelica Wax - phenol Watts 1 292

---

Nitranic acid. ~~completely heated~~  
it melts at 120 C. heated further  
sublimes - partly blackens & decomposes  
Fils - Val 1 302



Anisic alcohol. Converted by  
Salicyl into Resin: hard transparent  
Phono - Val 1 303 p. 64 N. 9 -

Anisal -  $C_7H_8O$  - 10% dig.

Anisulmin - Brown product from  
extracting Anise seed after previous  
treatment with Alcohol, Water, + HCl.  
with aqueous solution of precip. alkali  
by acetic acid - Fils Val 1 306

Hydrate of Anisyl pentachloro  
alcohol. thickens it + finally gives  
black pitchy mass. Watts Val 1 307  
Fils

Annetto = Nitric acid. pitches it - files  
Watts Vol 1 308

Creosote, 308 <sup>OK</sup>

Asphalt Oil 308

Formate of Methyl 308

Ethylamine.  $C_2H_7N$  - 308

Tetramine.  $C_4H_{11}N$  "

Amylamine -  $C_5H_{13}N$  "

Octylamine -

Phenylamine -  $C_6H_7N$ . want the disac

Umic substance its aniline oil I think.

Cyanide Ethyl  
" methyl.

Nitrite amyl-

308

Caminal ~~ok~~ Joz

Tetryl.

Wood Spirit, ~~ok~~

Phenyl alcohol.

get Sulphof/Bismuth  
in form of that don't  
melt + find solvent

Carbonyl sea + bean material  
see if pub much. + can be squinted

Di Nitro benzene - Joz

Benzil - "

Benzoin oil - "

Squint Mushrooms.

Deppels oil - Joz

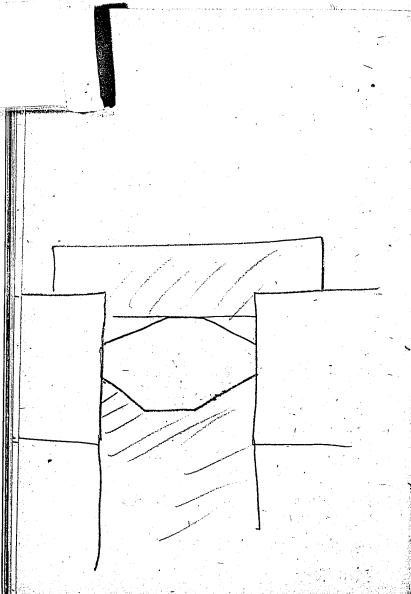
pyridine. get

Nicotine ✓

Lutidine ✓

Collidine ✓

Liq. in Deppel oil Joz



Black crystalline Compound left in  
retort Hexaglyceric Bromhydri.  
Fils - Watts 1 668,

Bromform - have it pass through  
tube containing amalgamated copper  
to stop-effect on Hg of pump's  
500

Butyric acid 700K E

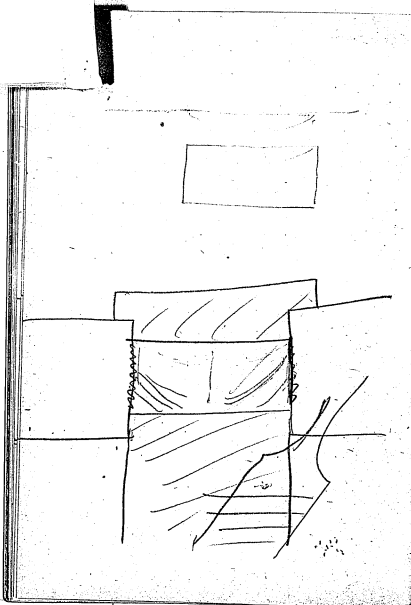
oil Cajuput. OK 1000

oil Juniper OK 1000

oil Thyme. OK 1000

Camphal - 1000

Tetrachloro-camphor resembles  
white wax - photo -  
Watts Vol 1 728



~~Watts~~  
 Camphoric acid -  
 melts to transparent mass  
Phane Watts Val 1 729

Camphor Resin - a non volatile  
 Resin due to action of alcohol Camphor  
Phane file Watts 724

Camphorone - - 90%  
~~Watts~~ - liquid -

oil Caoulchoue - 90%

Caproic acid - 90% - 90%

Capronitride - 90% - 90%

Tetrachloride Carb. in Lig -  
 and Cap in Tube - Watts

Bisulphide Carbon - Joe  
lookout for discolored Rubber:

Cardamon seed oil Joe

Oil Caraway - Joe of

Ceropic acid - waxy mass.  
phone - Watts 1 836

Cerosin - wax of sugar cane  
very hard. brittle MP 82 Cent.  
Watts 1 836 Phone

Insulation - Chlorinated Carotene  
for Chinese wax Watts 1 836

Chlorocerotic acid - Chlorinating  
cerotic acid for Beeswax. transparent  
gum - phone - Watts 1 837

Cerylic alcohol. gor

Cetraric acid - formula of Umic a  
by  $H_2SO_4$  - watts 1 839 fil

Cetyl sulphate. of potassium  
not fusible - fully mg fil -  
watts 1 842 -

Trichloroacetic acid gor  
+ for dissolv Carbmit.

Chloral - <sup>thin</sup> oil - gor  
+ for dissolv Carbmit -

Chloralide - dis Carbmit. solid.

Mix Syran with Iodiform will  
a fil heated melt. If Iodine  
Dissolved in Chloroform to Syran or in  
Turpentine or Bisul Carbon or Syran Evap  
will fil melt, or powdered Iodine or  
Syran just mixed with Chloro or  
Chlty - melt. 5

Cinnamene - Valili oil of Liquid  
storax = J52

Balsam Peru can be saponified  
phoro - try all balsams -

Chlorine passed ~~into~~ in diffused  
daylight over dry Cinnamic  
acid produces tough greasy  
substance, phoro - Watts 1, 984.

Oil Cassia - J52

Citrate of Magnesium gummy  
amorphous mass - phoro  
Watts 1, 998.



Oil Citron  
" Citronella Joz  
Oil Cloves

Petrolene phlyd use Carbonite.

Comenic acid or Meconic acid  
& Comenati - good base  
for fcls - see Watts 1, 1104

Oil Rosemary - Joz

oil Coriander seed Joz

oil Cress -

Cumal Joz or cuminal

Cyanide of ammonia -  
deposit Algalenic acid Joz

Cyanide of Ethyl + propionitrile J<sub>2</sub>  
with hydrocyanic ether,  
very poisonous

Cyclamin - may be good for fish  
Waltz Vol 2 274.

Furfural - J<sub>2</sub> Lij

Oil Wintergreen J<sub>2</sub>

Gentianic acid Carb.  
without much decamp W 2 830

Glycocine - can be carb'd before  
much decamp - W 2 902

Haematoxylin - from logwood - we have  
some Conc Extract sold - Sol alcohol -  
good for J<sub>2</sub> & also possibly extract  
in an anaphro stuff against fish -  
try Benzene to dissolve out everything most  
anaphro - Waltz 3 p 4

Haemaloylin has many reactions &  
is good base for soft files - see water

Benzene - Naptha, Regalene, 908 - Emmons

Heat Ammonia Diphenylamine in  
side tube - Emmons

Hydroquinone - Watt III, 213 fil  
do 214,

Imasatin - watts III 245 - fil

Indifuscin - don't fuse - Watt III 249

Common indigo - fil

Sulphendigatic acid - fil  
Watts 259 - Vol 3

Indichumin - Watts 3 264 fil

Japonic acid  $C_{12}H_{10}O_5$  - the  
black stuff which produced by  
exposing Catechin in caustic pot. HCl  
throws it down to sol water in sal  
alcohol with 3 442 pts.

---

Legumin - leave shiny  
Carbon - give dip - Val III 569

---

Lignin - base for soft file Azobenzene

---

Luteolin Residue file soft Val III 736

---

Malate of lead phos Val III  
p 796. 707 amorp - basic salt  
with Crystalline

---

Melanic acid resembles  
damp black very sol alcohol  
Ether & alkaline lqds - perhaps good  
for soft file - Val III 867

Melassic acid - black - action of  
Heat on Caustic alkali & glucose -  
Soluble in alcohol -  
Spt files with 3 868

Melenic - ~~not~~ spt files III. 868

Methilmic acid - base for  
Spt files Vol III 983-84

Moric acid dec without meltg  
Vol III 1048 salal

Murexan dont fuse  
Spt files Sal alkali III 1063

Oleamide - phno  
Vol 4 191

Pectic Acid sgt file Vol 4 367

Hx melophosphate of Lead  
a Resin - phos - 4 566

Use phos anhydride for dryer  
& put dry Cy of K in side bulb -  
sheet when ready flesh - Emmons

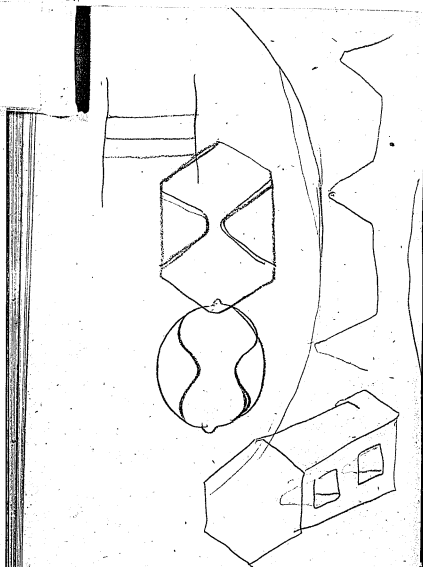
Dry Stearate Aniline phos

Oxide paraffine oil with hot nitric  
for a film specimen -

Uranine - 70% a fel

Fluorescens, " "

Thallene " "



Physalin - Vol 4-634 - 715

Picolini - gas - flesh - all  
the C.H.N. depends to be used

Stearic acid pitch - black stuff  
boil down to a very thick  
asphalt. then dissolve +  
used for dip file - all C.H.N.s  
poly

Try this stuff with Na  
but to make a solid for  
phone

Picolinic chloride - Vol 4 638  
Phono  $\text{H}_2\text{O}$  -

All the solid Medicinal  
Extracts - see MCKTR Catalog  
most are amorphous by  
Carbon direct  $\text{CO}_2$  &  
dip of  $\text{CO}_2$  film, from  
stock bottles  
If they such as show semi  
fused shiny  $\text{CO}_2$  - by  
& separate by solvents the top  
easily fusible material -  
starchy matter -



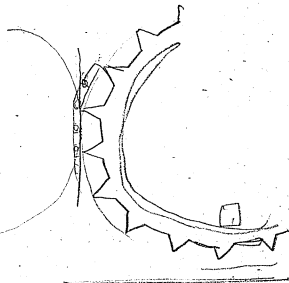
Picrotoxin Dream without  
from file 643 val 4 -  
Cocculus Indicus

Picroyl - file 644 val 4

Nelromarin or Azomarin acid  
carb. without. Meltg.  
addn Nit a on Pimaric acid  
from Gallipet.

Piperic acid file - 654 - val 4

Piperidine a C.H.N. dig -  
watts 4 655 - first -



Propionitrile - flash - Val 4 736,  
CHN or cyanide ethyl -

Pseudoveratrine - dip - Val 4 743

Urates of the CHNs - say  
Urafiandini - for dip -

Alloxan - dip -  
Murexide - good dip

Alizarin

Purpurin. Watts 4750

Purp. a Camel Color - a CHN,  
Watts 757 - 4 Val

Pyridine - CHN - Val 4 754  
flash -

Chlorine Brown & Iodine  
act on the C<sub>2</sub>H<sub>4</sub>N<sub>2</sub> liquids to  
give solids & perhaps may  
will be salinable & good for  
dip or sqpt files,

Pyrolantramide combined  
with Oxide lead - get larger  
paper - C<sub>2</sub>H<sub>4</sub>N<sub>2</sub> = thread  
files & other uses Insulation -  
Watts 775 - Val 4

Pyrolantramic acid files,  
Val 4 775 -

Pyrral C<sub>2</sub>H<sub>4</sub>N<sub>2</sub> - flash -  
Pyrral Red - Val 4 784 6L  
dip sqpt -

Carbohydrates treated  
with ammonia take up N  
give gelatinous deposits.  
Disint by heating glucose,  
Milk sugar with aqueous  
ammonia. obtained nitrogenous  
products which precip by  
alcohol in Tennessee thread  
fils: Val 5-467

only Copying marks  
in 4-th Val & chance  
on - without recant

Verantien fils Val 5-995  
do Halogen Compd. & dilute

Viscin - arabic residue -  
good base = fils - Val 5-1003

acculmic acid Val 6 29 -

<sup>Kennell, Tall</sup>  
Teraphthalic acid not fusible

Hydroxyteraphthalic acid  
not fusible -

Umbelliferone not fusible,  
 $C_9H_6O_3$  —

Cholecamphoric acid  
not fusible -

Naphthalene-dicarboxylic  
acid - not fusible.  
 $C_{12}H_8O_4$ .

Hexahydroxydiphenylene  
ketone - decamp without melt  
Anthraflavic acid  
non fusible

Tetrahydroxynaphthoquinone, non fusible

Anthrachryson " "

Anthral, " " "

Diphenic acid (iso) - one of them  
infusible -

Tetrahydroellagic acid non fusible

Benzoylterephthalic acid " "

Aurin - " "

Phloroglucalphelein " "

Anamintin - " "

Resorcinophalein - " "

(Fluoresorcin) " "

Melaphloretin " "

Calcein-Tetrahydrate " "

" Dianhydride " "

$CH_5N^+$  -

Formoguanamine - nonfus.

Anhydroformaldehydamine "

Dianthramine "

Diphenanthrazolide - "

Perchlorobiphenylamine - "  
 $C_6Cl_5-N$  -

~~Comp.~~ Acolyetine  
for fets - from Herbarium  
Extract. Vol. 6. 5-4

Acrolein - combines with  
aniline from amorp base  
Vol. 6 56 -

Flaming: -

Acetonitril, (Methylcyano)  
 $\text{CH}_3 \text{CN}$ . good - Liq -

Propionitril, Ethylic Cyano.

Isopropylamine -

Propylamine.

Ethylic Cyano.

Diethylamine -

Butylamine.

Isobutylamine.

Pyridine.

Pyrrrol

Piperidine

Amylamine.

Picoline.

Benzonitril.

Parvoline for Dippels oil

Naphthalamine. a Salt



get Schuchert's Fremmsdorf +  
marks Cal also gives + get all  
the  $C_4H_6$  -

Have atwood try  
Naphthalene as Lubricant

Naphthalene dissolves  
Sulphide of antimony  $As_2S_3$   
Iodide of Oxalic Succinic  
+ Benzene acids - try  
these for thorough experiment

Hot Phenol is great solvent  
for organic insoluble substances

Quinone  $C_6H_4O_2$  this stuff decomposes  
brown and exposed air especially if  
alkalies something in this group for  
files - ~~Atwood~~

Wüllers Chem =

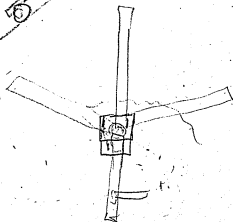
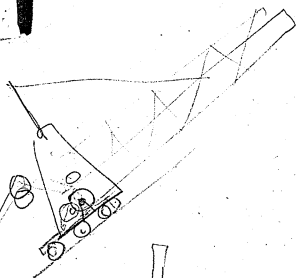
p 712. (1617) Ethylenic Sulphide

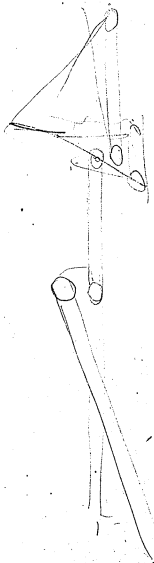
$C_2H_4S$ . 50 p 713 all  
amorphous. action of

$\alpha$ -dibromethane or Ethyl bromide on  
Alcalalis Salubris Palassie Sulphid  
white amorphous substance

Methyl iodide & propyl bromide  
treated with alcalalis Pot  
sulphide - gives amorphous substance  
insoluble in all neutral Solvents  
fils & phons

Dinitroanthraquinone - or Fritzsche's  
Reagents - Combines with all  
Known solid Hydrobns in Coal  
Tar = action of Dilute nitric  
acid on Anthracene. for phons  
to get rid of Crystals  
Try Dipul Oil Solut Carbides





after acting in Carburettor with  
Methyl alcohol - try other solvs, but  
Solvents, to get out all puffy stuff.

Metacinnamene polymerized  
by heating in sealed tubes to  
392.7 Amorphous Calorless  
Solid high refracting power  
disoluble smaller taste -  
quite transparent softens on  
applied heat. Drawn out in  
threads - Make some for  
pheno - Insulation - +  
physical experiments.

B-Dinitro Anthraquinone - film  
with 1st vol 3rd sup 99-

Action of Nitric Acid on a  
phenol - a Resin pheno  
for mixing -

Alcoholic solutions of Coumarin  
Soda or Potash produce kindly  
action on Dinitrobenzene + black  
amorphous substance formed  
which are partially soluble  
in Alkalies - Wirths 1st & 3rd  
Exp 175 -

Action of Arsenic Trichloride  
on Azobenzene is converted  
into black amorphous substance  
forming conchoidal fracture  
Metallic lustre Sol in Water  
more easily in Alcohol.  
Wirths, 1st & 3rd Exp 212

Melamidobenzene sulphonic acid  
melts without ~~Discoloring~~

ant this Sulphanilic acid  
Sulfurety hütswal. 15<sup>56</sup> 27<sup>7</sup> sup  
p 227

See 228 Diamidobenzene sulphonic acid

Dibromomelamidobenzene sulphonic acid  
Dont fuse, 14<sup>14</sup> vol 5<sup>5</sup> sup 232  
also 233 and 234

1st Vol 3rd Sup  
Tribromomelamidobenzene sulphonic acid  
p 234, Dont fuse

Tetabromomelamidobenzene sulphonic acid  
234 Dont fuse

→ Tribromomelamidobenzene sulphonic acid  
234 Dont fuse

Methylammonesulphate 235 - 1<sup>st</sup> 3<sup>rd</sup>  
Don't fuse

Amide of Tribromobenzene sulphonic acid  
241 - 1<sup>st</sup> 3<sup>rd</sup> - Don't fuse

Amide of Tetra bromobenzene sulphonic acid  
1<sup>st</sup> 3<sup>rd</sup> 243 244 Don't fuse  
Sol alcohol easily

~~Amide~~ Pentabromobenzene sulphonic acid  
Don't fuse  
1<sup>st</sup> 3<sup>rd</sup> 244 Sol alcohol  
245 - Nitrode brom ditto

There are a lot of these  
sulphonic acid benzene  
that don't melt / dissolve  
in 1<sup>st</sup> 3<sup>rd</sup>

thiocarbamido benzoic acid  
fils & phos - 30 lines down  
another Compound -  
1st 3rd 269

Azory benzoic acid  
Not fuse

1st 3rd 277

Diazoy benzoic acid good -  
p 277, 278, 1st 3rd  
just like Melissic or Azuline  
acid salt in ammonia  
ammonium salt dries to a  
Varnish brown & transparent  
see Barium salt most soluble Lab known  
gumps in dish for hours  
also below - Monobromo diazoy benzoic

See Carbohydrate from Coat w/iron  
achin Nitric acid causes solubility  
in it = fil, 1st 3rd 405

Chryseal 1st 3rd 463

Benzoyl-methoxym-carboxyl.  
• Part fuse

Carboxy-benzoyl-azoxim, propionyl.  
Carboxylic acid - Part fuse -

Walt New Dec Val 1 424

Don't forget to dip fil in a Carbon  
Compound of lime or Magnesium to  
produce a semi- or non-conductive surface  
in fil to prevent electrical action,



Chlorine passed in diffused daylight  
over dry Cinnamic acid forms a  
tough-greasy substance Walts  
1st ~~at~~ 954 -

Citrate of Lithium phos }  
Magnesium phos } 2  
Nickel phos } #

Di Potassic Citrate acid - prevent  
Electrification phos -  
See Val 1. 999 - Walts file Inster Phos  
1001 - 1002 1004 -

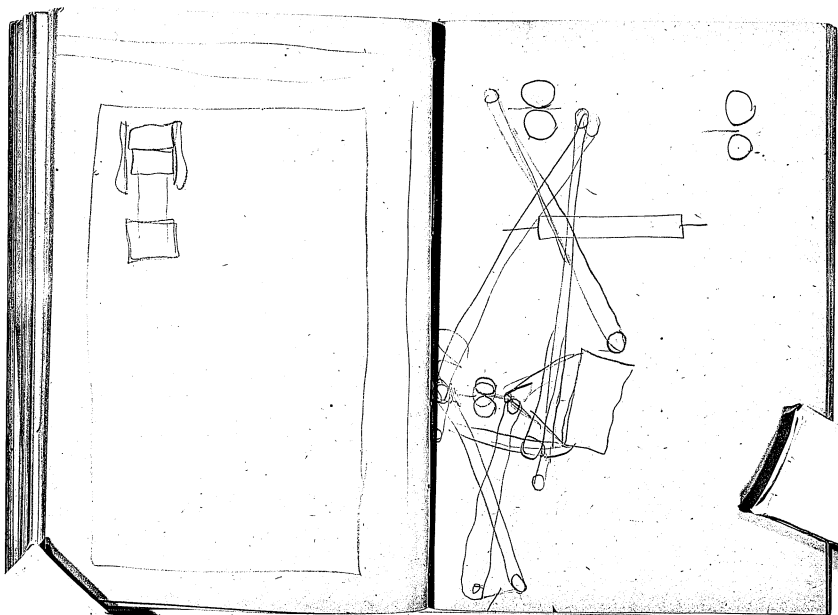
Acrylic aldehyde forms a  
white Amorphous Compound with  
"Ammonia Val. 1 112

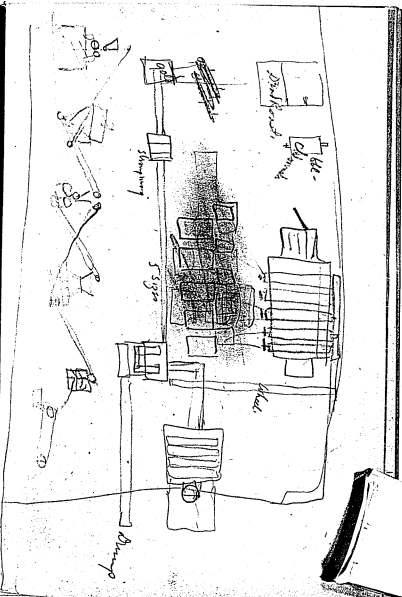
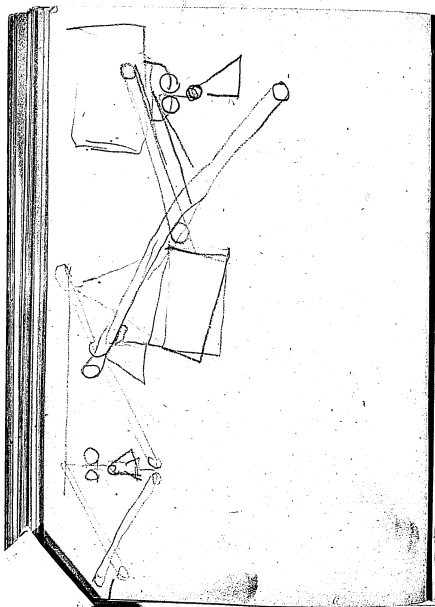
Allophanate Glycerol - phos  
Watts 1 p 134

Invis alcohol actin of Sal acid  
phno - Walth Val 1 303 + chzn

Anisulman filo Val 1 306 -

Arachidic acid - like proclan  
phno - Val 1 353 see 354  
" susp made in France"





$$\begin{array}{r} 45 \\ 31 \\ 45 \\ \hline 135 \\ 1395 \end{array}$$

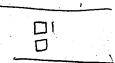
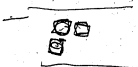
$$\begin{array}{r} 1395 \\ 2240 \\ \hline 845 \end{array}$$

800 lbs - ✓

380 lbs from org

$$\begin{array}{r} 31 \\ 20 \\ \hline 62 \end{array}$$

$$\begin{array}{r} 2240 \\ 620 \\ \hline 980 \\ 122 \end{array}$$



$$\begin{array}{r} 17 \\ 33 \\ \hline 50 \\ 118 \end{array}$$

$$\begin{array}{r} 434 \\ 124 \\ 140 \\ \hline 170 \end{array}$$



$$\begin{array}{r} 16 \overline{) 380} \\ 320 \\ \hline 60 \end{array}$$

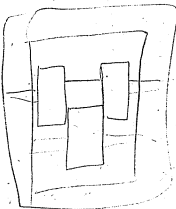
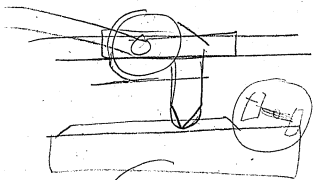
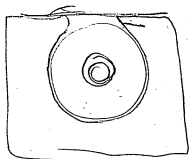
$$\begin{array}{r} 75 \overline{) 2240} \\ 1500 \\ \hline 740 \\ 650 \end{array}$$

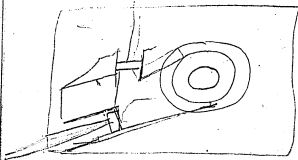
140 lbs per ton. - 23 lbs -  
75" operation -  $\frac{41}{64}$ . 2 units



$$\begin{array}{r} 140 \overline{) 2240} \\ 1400 \\ \hline 840 \\ 840 \end{array}$$

$$\begin{array}{r} 30 \overline{) 1240} \\ 1200 \\ \hline 40 \\ 20 \end{array}$$





Pol Na

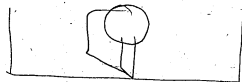
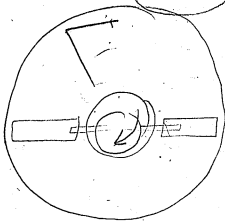
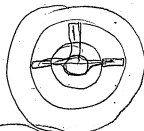
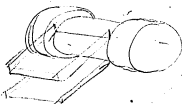
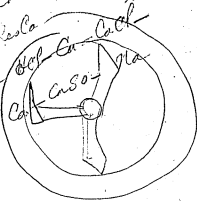
CaCl<sub>2</sub>

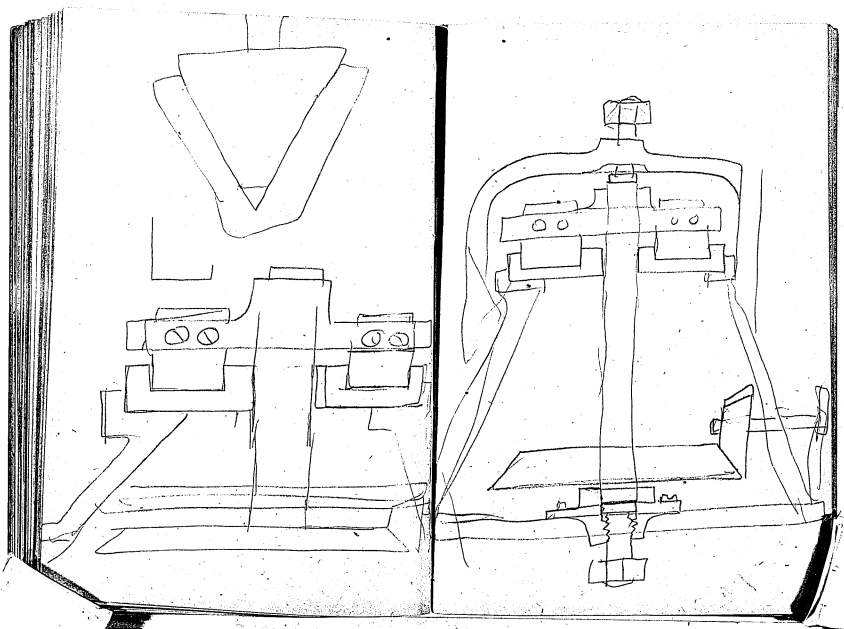
Pol Na - CaCl<sub>2</sub>

NaCl - K<sub>2</sub>CO<sub>3</sub>

804-Na

504-Na









135 tons over  
33 1/2 days

1382 40  
55-2 81  
1382 2646  
500  
88.3

60 tons  
44 tons  
55-33  
20 | 2646 (132)  
60  
46

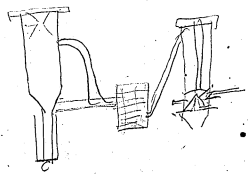
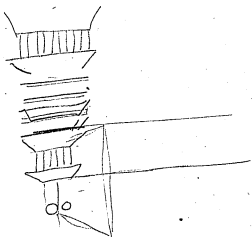
1382.

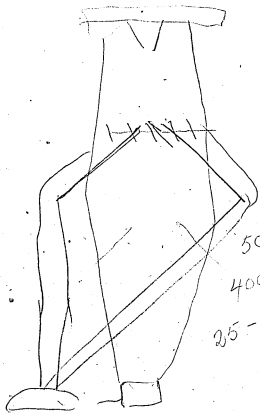
4000  
20  
55 | 80000  
550  
250  
22  
300  
2750  
250  
220

1457  
72  
1382

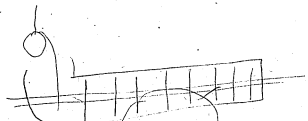
55 | 4000 (72)  
385  
150  
190

1454





50-  
400-  
25-



86

10<sup>02</sup>

16 \$5<sup>00</sup>

50-

7-

100<sup>03</sup>-

30-

12-

50%

30-

75-

145

50

195

100

295

125<sup>000</sup>

20- 15<sup>00</sup>

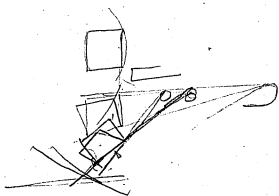
17 \$750

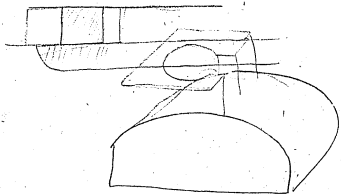
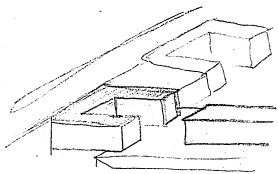
7500 175<sup>000</sup>  
7500

500 750

17560  
" 2

strings  
4 trains





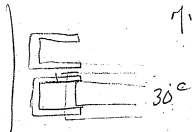
2.

40°  
12°

360°

45°

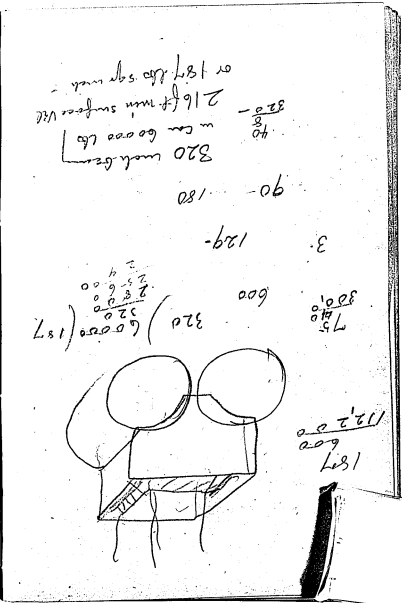
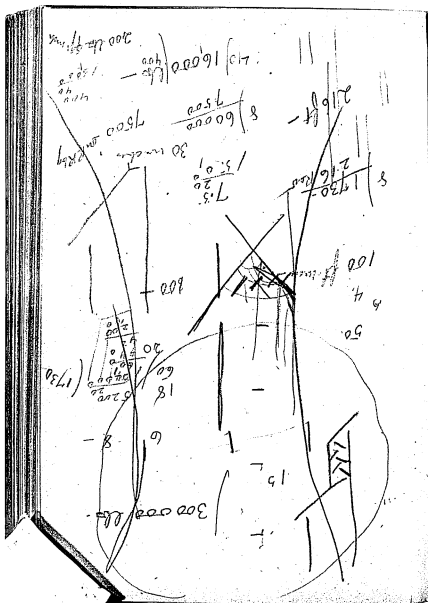
7.

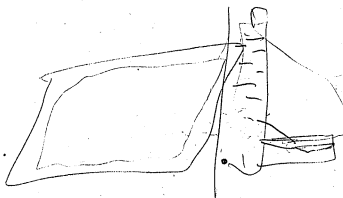


36°

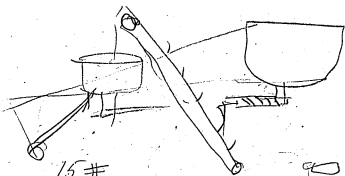
275°



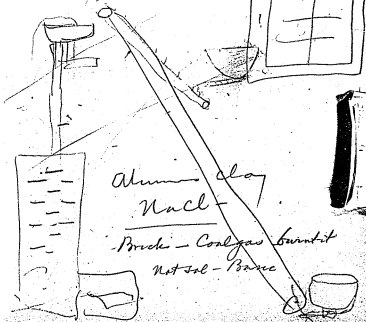




100  
120—  
10—  
C. 5—  
137.5 120—



15#  
137

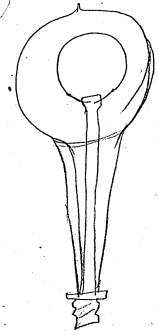


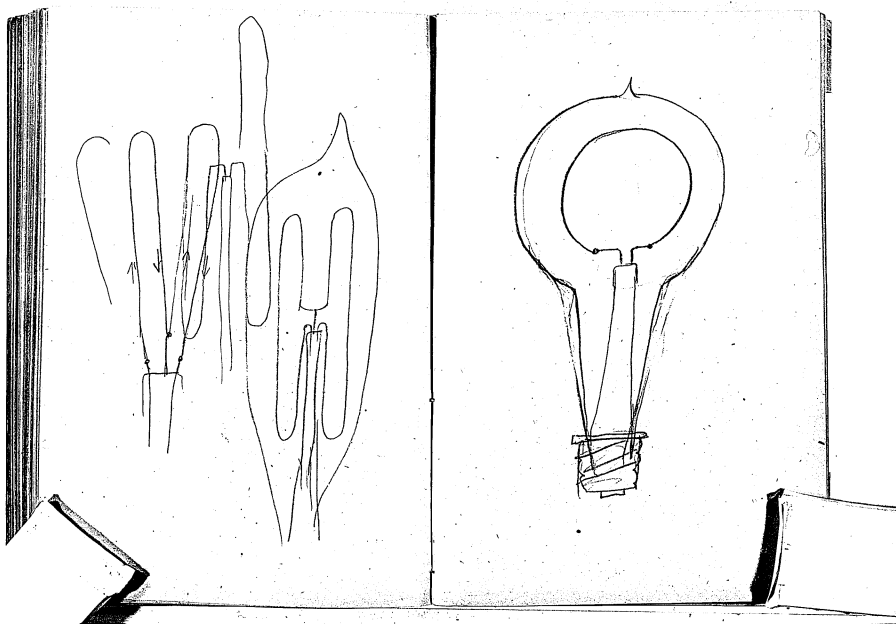
Alumina clay  
NaCl

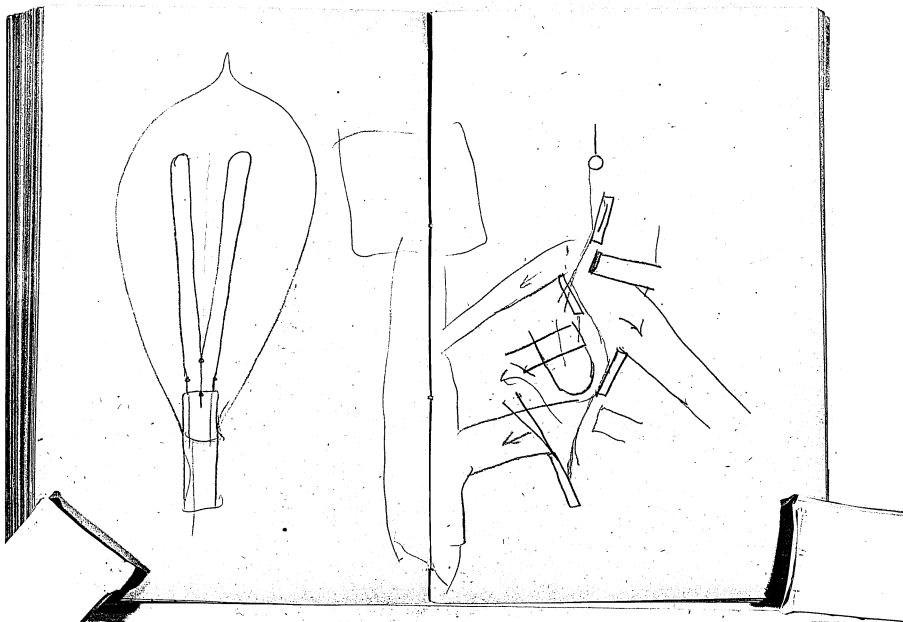
Procki - Coal gas burnt  
Not sol - Base

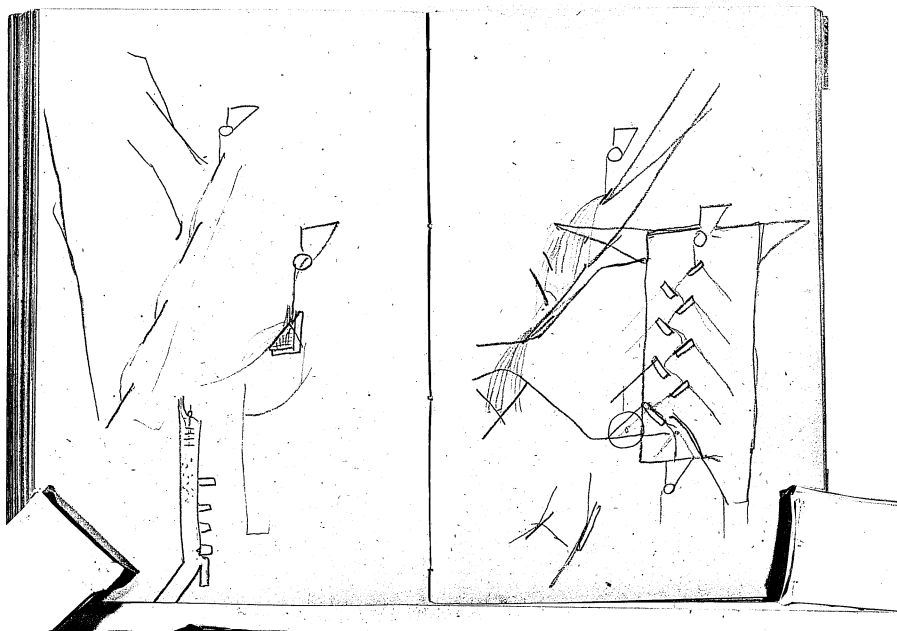


3 lbs  
 700 - 10  
 5 10 - 50 - 31  
 10 19  
 1/2 - 3 1/4 Cents  
 176 / 3522  
 58  
 36  
 34  
 1748  
 209  
 60.  
 120  
 67  
 2.47  
 2  
 80  
 100 Cents  
 12  
 352  
 4200  
 352  
 600  
 392  
 2200  
 3168  
 247 / 352  
 210  
 378  
 318  
 200  
 Potchefstroom  
 187





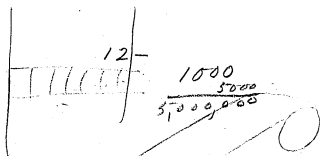




If a procedure that  
 also get permanent  
 surface velocity - and  
 a feel to gas 1 inch diameter  
 or 65 lbs for 10" + the mass  
 of 20 inches or 130 lbs of  
 the use of 4 times the pressure  
 will produce the best results  
 -  
 Assume 1 inch pressure as  
 the best thickness -  
 then the best results will be  
 obtained by allowing the  
 rolls to 300' + increasing the  
 pressure until the product  
 grows 45% the usual gas  
 500' 90 lbs -  
 If a procedure that  
 also get permanent  
 surface velocity - and  
 a feel to gas 1 inch diameter  
 or 65 lbs for 10" + the mass  
 of 20 inches or 130 lbs of  
 the use of 4 times the pressure  
 will produce the best results

you had to open you can  
 denser the feed denser the  
 pressure and the temperature  
 will be just the same -  
 get higher now have turned the  
 pressure on each particle -  
 a greater grinding effect  
 in addition -  
 If of 40 lb hp for 120' in 23%  
 than it will take 80 hp to denser  
 the pressure to get 45 -  
 the rate however according to  
 gas but 100 lbs -  
 you increase to 20 each  
 you get 800 lbs but you  
 must increase the pressure  
 in proportion hence 800 lbs  
 160 hp 45%



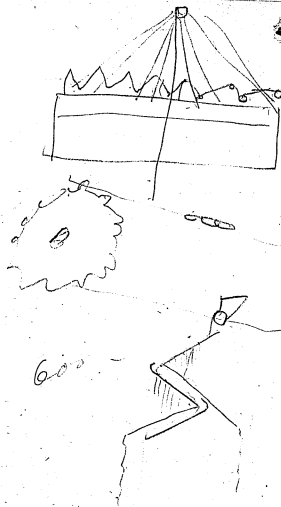


14000

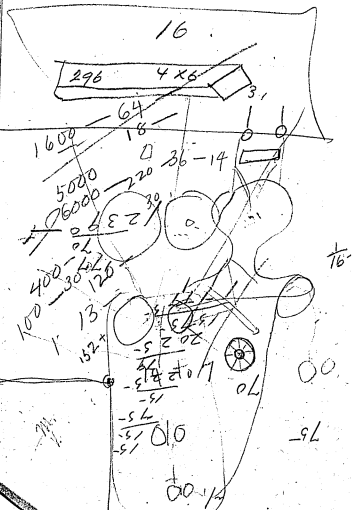
$$\begin{array}{r}
 36 \\
 210 \overline{) 60} \\
 \hline
 420 \\
 \hline
 180 \\
 \hline
 180 \\
 \hline
 0
 \end{array}$$

$$\begin{array}{r}
 7500 \\
 210000 \overline{) 1575000} \\
 \hline
 420000 \\
 \hline
 1155000 \\
 \hline
 1050000 \\
 \hline
 105000 \\
 \hline
 0
 \end{array}$$

$$\begin{array}{r}
 60 \\
 300 \overline{) 18000} \\
 \hline
 6000 \\
 \hline
 6000 \\
 \hline
 6000 \\
 \hline
 6000 \\
 \hline
 0
 \end{array}$$



Gas



Madaganne, 150 000  
 Humboldt - 150 000  
 Cliffs -  
 Split Rock - 200 000  
 Br. chalcidella 500 000  
 Croton 350 000  
 Pterostelea 100 000  
 Pachygnathus 10 000  
 St. Henry 175 000  
 400 Kienm 300 000  
 Mosiac 100 000  
 Pt Orange 25 000  
 Wilder 25 000  
 1643,000

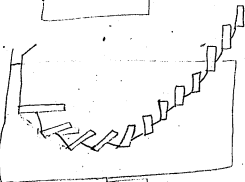
31  
 52



1 / 15000.  
100- 15- 3/4-6 1/4-



15



2.

10  
70  
80  
28  
30  
300  
800

2000000

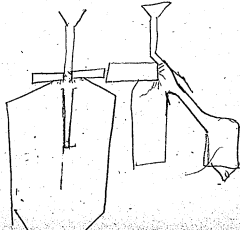
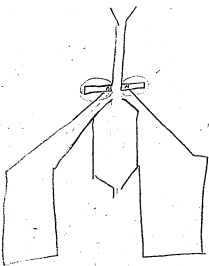
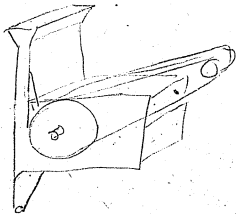
50°  
5 6- 8° 120 1/4

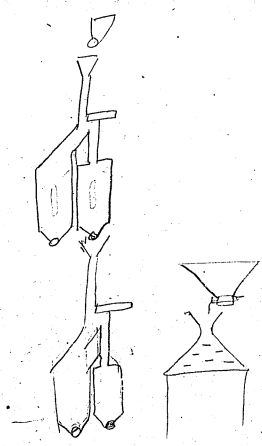
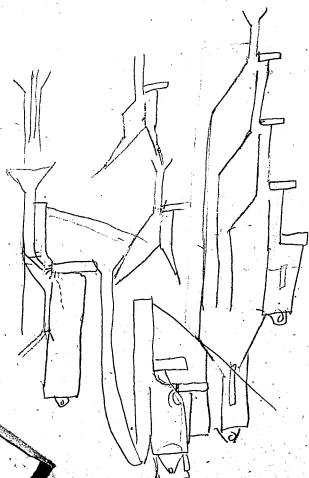
2

5

3000

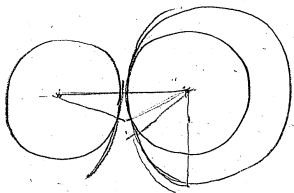
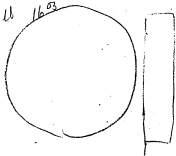
240







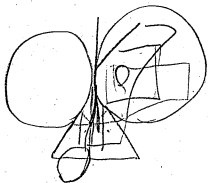
20 lbs  $\frac{1}{2}$  cup of water  
- 1 lb 16oz



120

110

220-





15 000 - Eng  
 22 000 boiler  
 3 000 pump  
 3 000 condenser  
 10 000 - Light Dynam  
 6 000 piping  
 10 000 Dynam  
 7 000 Motors  
 20 000 Rollo -  
 25 000 belt screw  
 12 000 - Separators  
 60 000 Truss  
 20 000 furnace water gler  
 5 000 Dryer  
 25 000 Rollers  
 18 000 bundles  
 50 000 Bldg -  
 15 000 Lumber  
 8 000 RR  
 10 000 Shops  
 35 000 Belt Conveyer  
 30 000 Extn

200  
 $\frac{200}{500}$   
 $\frac{200}{500}$

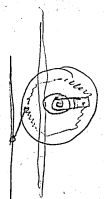
Buddhi

25

25  
 $\frac{500}{12500}$

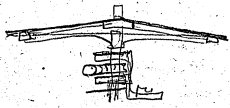
119 650

500 000



240 / 10 000 / 4

2



Mining 235  
 40 mile ml - 25,  
 Concrete Trac. 025  
 Drilling - 02  
 Rebar 10  
 Manholes 1  
 Sewer 3  
 Roadway 5  
 Slopers 2000  
 2350 (2000)

75 c  
 100 all cost

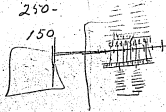
2370

- 200 -

13

250 -

175  
 13  
 525  
 - 175  
 2273 -

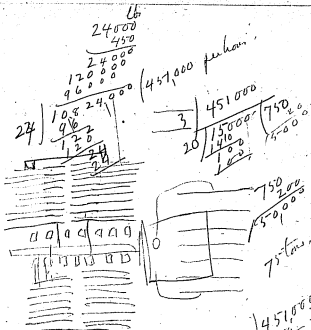


500 - 2000

24

2,500,000  
 09  
 04

175  
 24  
 700  
 350  
 4200



6 - 451,000

40  
 24  
 120 long  
 150 across



450 451,000 (1000)

451

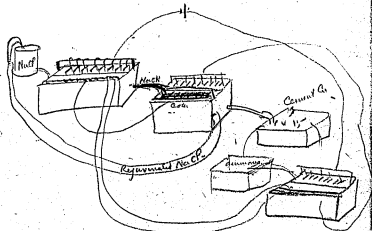
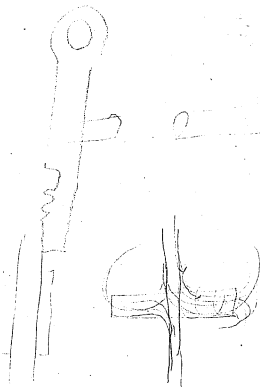
**Notebook, N-90-01-04.7**

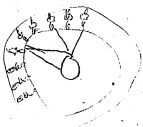
This notebook covers the period January-May 1890. All entries are by Edison. The book contains extended theoretical notes about electricity and electromagnetism, including an entry in which Edison compares his own theories with those of James Clerk Maxwell. There are also notes and drawings relating to unidentified electrical equipment, ore milling and bricking machinery, electric traction, and electric lighting experiments for both incandescent and arc lamps. In addition, there is a list of chemical compounds to be tested. The front cover is labeled "#580." The inside front cover is inscribed "#580 T A Edison Jan 4. 90." The pages are unnumbered, and the book has been used in both directions. Approximately 160 pages have been used.



#580

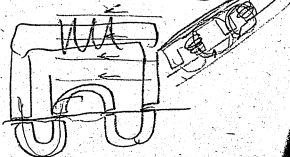
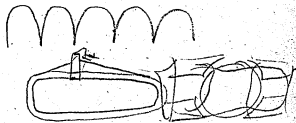
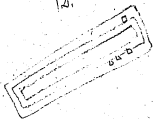
Edison  
Jan 4..90





8

12.



[THIS BOOK WAS USED IN BOTH DIRECTIONS.  
THE FOLLOWING PAGES WERE FILMED FROM  
THE BACK END FORWARD.]

N-90-01-04

12 - 55 c  
 $\frac{64}{114}$   
 160. 500/320p (64)  
 $\frac{320}{2000}$

1300  
 $\frac{8}{1040}$  (40)  
 $\frac{350}{350}$   
 $\frac{500}{2000}$   
 $\frac{35}{1400}$  tons  
 $\frac{1400}{400}$   
 $\frac{10}{25}$   
 $\frac{15}{13}$   
 65 15

147  
 $\frac{147}{180}$   
 $\frac{20}{180}$   
 180  
 180  
 180

Side tube in Lamp - + Carb  
 Compound

Albumen about 1/20 before goes  
 to Lamp

alcohol (absolute) - dissolved fused  
 Chl Calcium sulfate, then  
 Chill out. ~~use~~ a Calcium  
 alcoholate, in Crystals which  
 on heating give off only  
 pure Carburetted Hydrogen  
 Watts Vol 1 p 80 alcohol must have  
 no water

ammonia aldehyde - Benz

Analine oil

Cumidine  
 Cyanide

Naphtylamine

phenylamine

Triphenylamine

Picoline

Piperidine

Pyridine

Amarine

amarone

Amylete Potassium

do Sodium

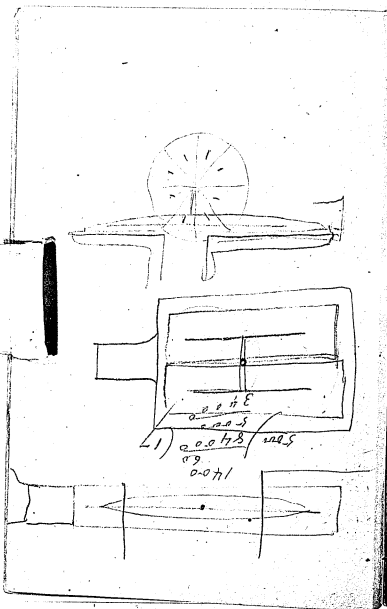
anise Seed

Benzoyl acid heated with Caustic lime  
fibre of Benzol

Phthalic acid heated Caustic lime

Dry Distillation Quinic acid

Benzoyl formic acid



Camphor -  
 Caoutchouc, dry  
 gullo percha dry

Mellone

Idioform

Chinoline

Rosin -

Amylate Soda

Coinine · C.H.N.

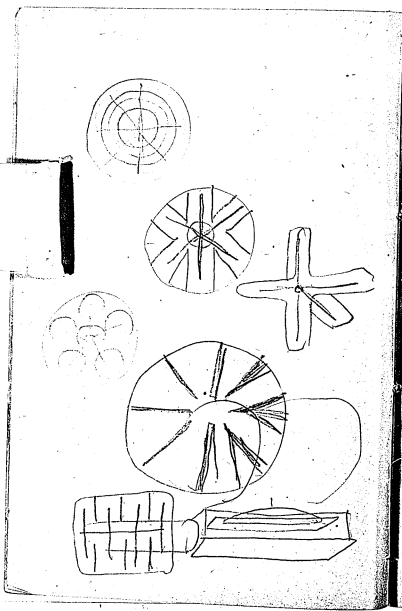
Copaiba -

Julu - Balsam

Peru

Copal -

ambur che



Cyanamide. C.N.H.

Cyanide of metal.

Cyanide Amyl-Capronitrite  
C.H.N.

Solid Chloride Cyanogen  
" Bromide "

Dammar Resin Goudie gum,  
gum off Dammaral C.H.

Benzonitrile

Elemi Resin

Formate of lead <sup>Barium</sup> Zinc Polish

Buytrale

gum Euphorbium

Cholesterin

& Ethal -

gum galbanum

Hydrobenzamide C<sub>11</sub>H<sub>9</sub>N

Hydrocinnamide "

~~Hydrobenzamide C<sub>11</sub>H<sub>9</sub>N.~~

Pyrene,

Iodoform - Cyanide of Silver

gum 'Borohade' Cyanide of <sup>Hg</sup> Silver

Napthalamine Alpha Beta

Linseed oil -

Lophene - C<sub>11</sub>H<sub>9</sub>N pyridine-like

Lupulin -

Mastic





Gum Myrrh -  
Carnauba Wax

Glycerate Soda  
" Alumina  
" Lead -

Phenyl Carbamic acid,  
ques. of aniline C.H.N.  
do Salicylamide,  
" azoxybenzene } distilling  
Indigo -

aniline Salivable. in Colophony  
hence makes a salad of Rosin + Amber

Diazobenzene C.H.N. or azoxybenzene

Aniline Blue - COTN  
" Purple -  
" Red  
all the Aniline Colors but black

Cyanide phosphorus

Venice Turpentine

Burgundy pitch

Coal Tar

Wood Tar

asphalt pitch

Syrian asphalt

Piperidine - COTN

$$\begin{array}{r}
 5400 \\
 7500 \\
 12950 \\
 \hline
 1500 \overline{) 25400} \quad (17.2) \\
 \underline{15000} \\
 10800 \\
 \underline{10800} \\
 0
 \end{array}$$

$$\begin{array}{r}
 300 - 18 \\
 500 - 15 \\
 350 - 20 \\
 350 - 17 \\
 \hline
 1500 \quad 2/37 \\
 \underline{1500} \\
 0
 \end{array}$$

$$\begin{array}{r}
 185 \\
 100 \\
 \hline
 129500
 \end{array}$$

129500

618 tons,

$$\begin{array}{r}
 2400 \\
 16800 \\
 \hline
 120 \overline{) 24000} \\
 \underline{24000} \\
 0
 \end{array}$$

$$\begin{array}{r}
 40800 \\
 39600 \\
 \hline
 66 \overline{) 40800} \quad (628) \\
 \underline{39600} \\
 1200 \\
 \underline{1200} \\
 0
 \end{array}$$

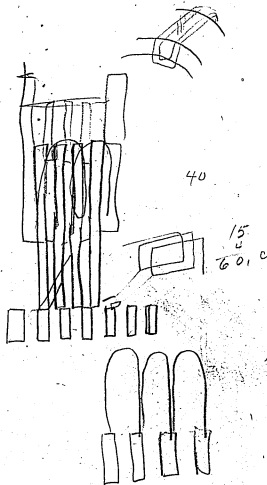
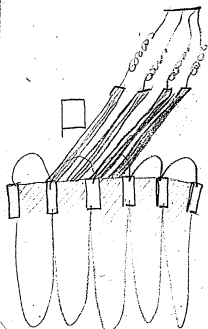
$$\begin{array}{r}
 450 \\
 400 \\
 350 \\
 \hline
 1500 \overline{) 26500} \quad (17) \\
 \underline{15000} \\
 11500 \\
 \underline{10800} \\
 700 \\
 \underline{700} \\
 0
 \end{array}$$

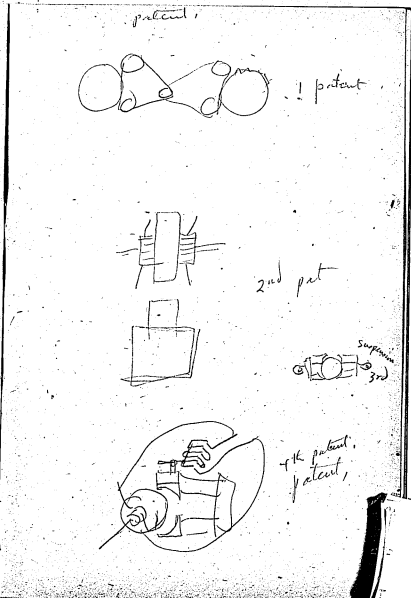
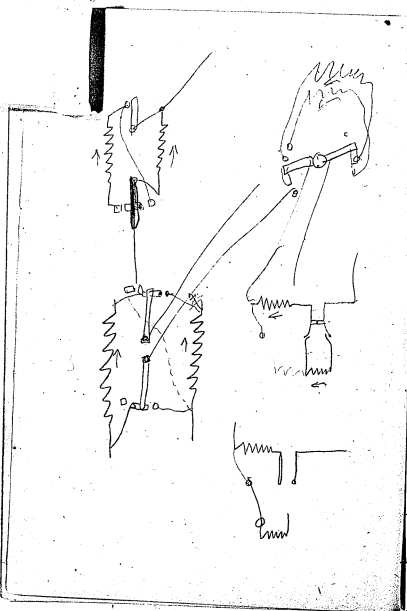
$$\begin{array}{r}
 45000 \\
 16000 \\
 \hline
 27000 \\
 45000 \\
 \hline
 72000
 \end{array}$$

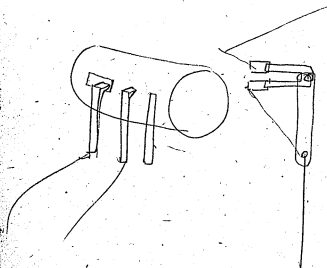
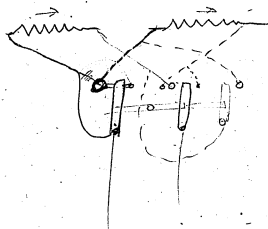
$$\begin{array}{r}
 350 \\
 17 \\
 2400 \\
 \hline
 24500 \\
 3500 \\
 \hline
 59500
 \end{array}$$

19%

$$\begin{array}{r}
 9200 \\
 8000 \\
 5950 \\
 5400 \\
 \hline
 1500 \overline{) 28550} \quad (19) \\
 \underline{15000} \\
 13550 \\
 \underline{12800} \\
 750
 \end{array}$$





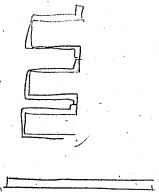


Quad  

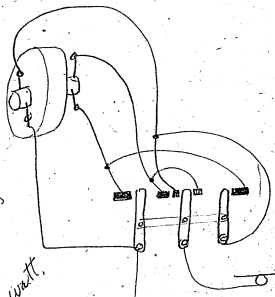
$$\begin{array}{r} 2 \overline{) 22.3} \\ 4 \phantom{0} \\ \hline 2 \overline{) 11.15} \\ 4 \phantom{0} \\ \hline 2 \overline{) 5.57} \\ 4 \phantom{0} \\ \hline 2 \overline{) 1.8} \end{array}$$

10 deg - 1 watt  
 20 "  
 40 "  
 80 "  
 2" per 100 deg 1 watt

1 watt  
 223"  
 each deg John



180-  
 100 of fact  
 10 watt  
 480  
 10 deg John 1 watt  
 223-  
 100 of fact  
 10 watt  
 100 of fact  
 10 watt



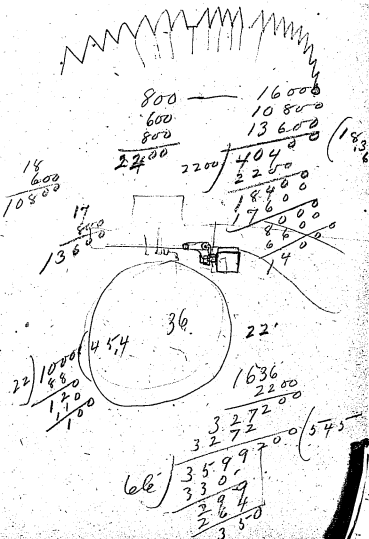
1 watt  
 10  
 .25-4 Volt

$$\begin{array}{r}
 16 \\
 4000 \\
 66 \overline{) 64000} \text{ (970)} \\
 \underline{5740} \\
 462
 \end{array}$$

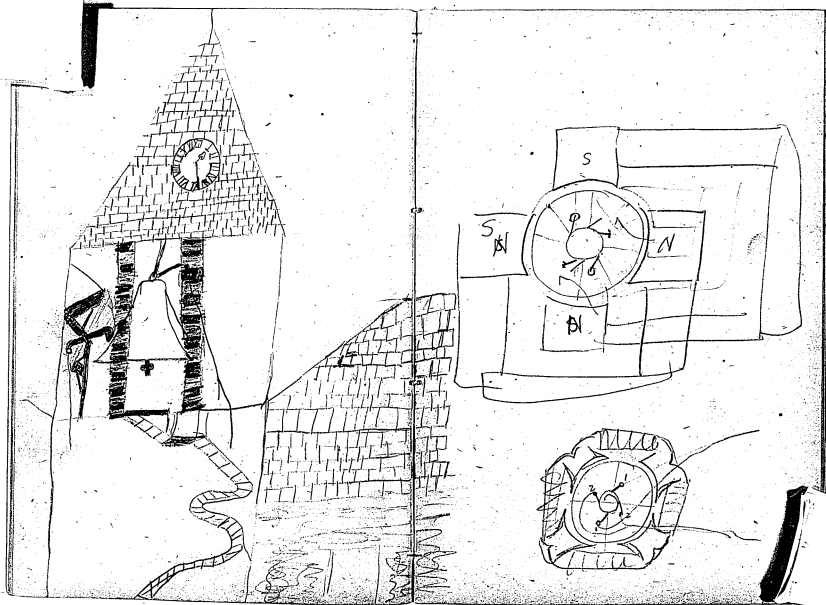
$$\begin{array}{r}
 4000 \\
 17 \\
 28000 \\
 40000 \\
 66 \overline{) 68000} \text{ (10.30)} \\
 \underline{6600} \\
 2980
 \end{array}$$

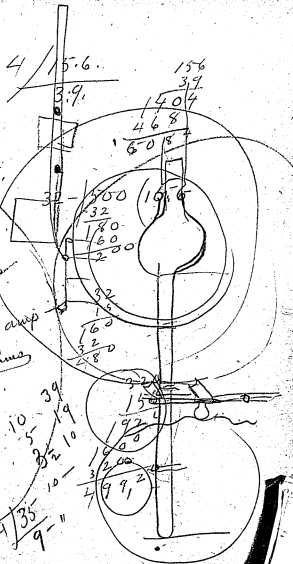
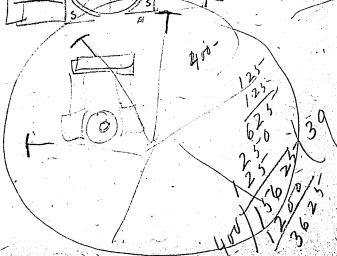
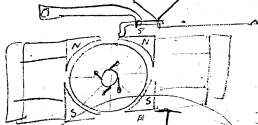
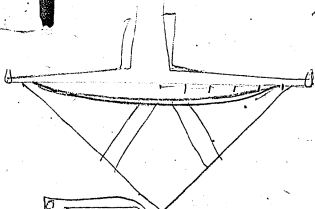
$$\begin{array}{r}
 2200 \\
 19 \\
 17600 \\
 22000 \\
 66 \overline{) 39600} \text{ (600)} \\
 \underline{396}
 \end{array}$$

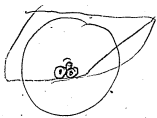
$$\begin{array}{r}
 400 \text{ } 205 \\
 4500 \\
 27000 \\
 45000 \text{ (1090)} \\
 66 \overline{) 72600} \\
 \underline{6600} \\
 594
 \end{array}$$

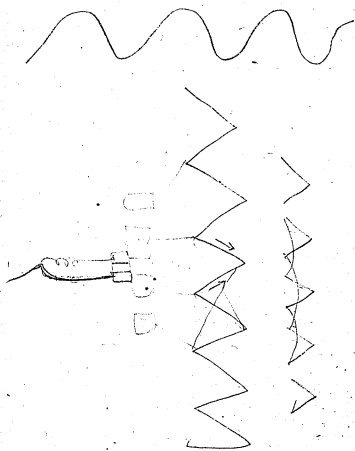




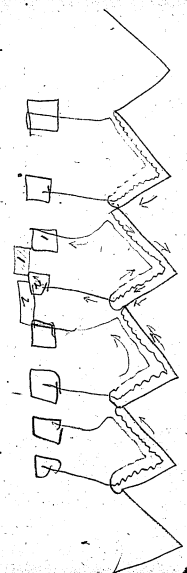


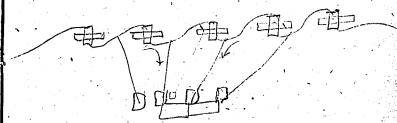
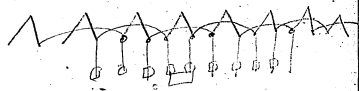
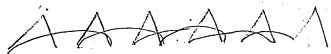


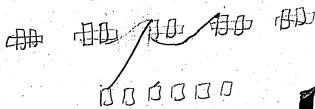
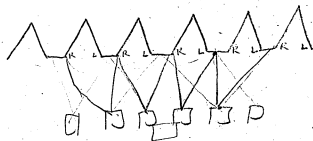




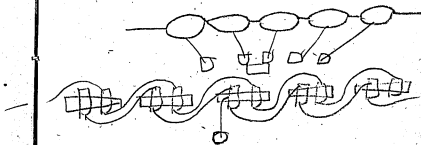
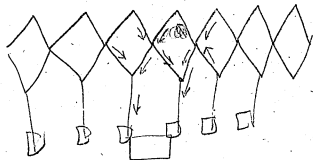
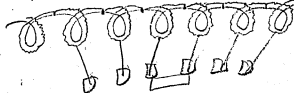
1 of 1 second revision

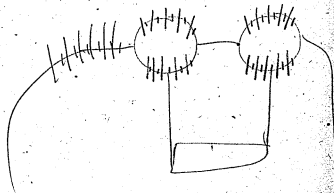
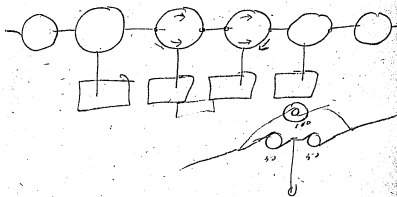






5.7 16010 Coko-







125  
 175  
 100  
 50

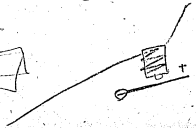
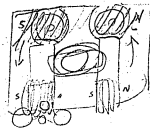
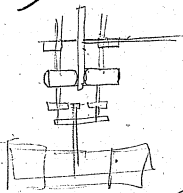
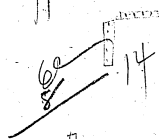
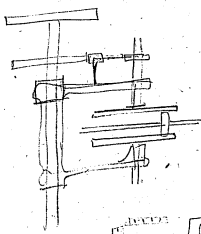
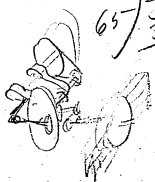
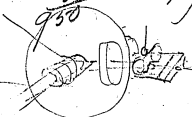
250  
 125  
 25  
 50  
 50

175  
 125  
 30.0

100 33  
 30

70/950/13c  
 70  
 250

13  
 20  
 26  
 130  
 390  
 32

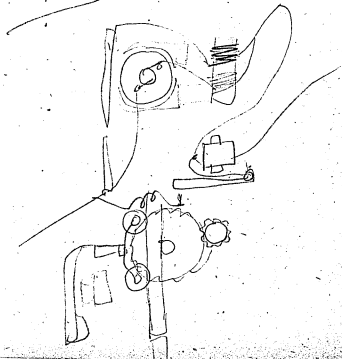


$$\begin{array}{r}
 49 \\
 20 \\
 15 \overline{) 980} \quad (65 \\
 \underline{45} \\
 530 \\
 \underline{45} \\
 85 \\
 \underline{85} \\
 0
 \end{array}$$

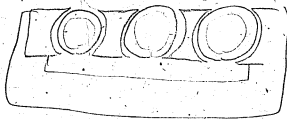
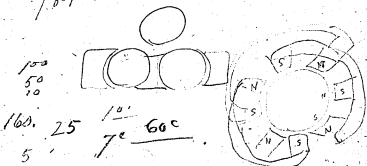
4 4. 23 2

$$\begin{array}{r}
 13 \\
 20 \\
 25 \overline{) 130} \\
 \underline{50} \\
 80 \\
 \underline{80} \\
 0
 \end{array}$$

10 62



$$\begin{array}{r}
 1803 \\
 88 \\
 \hline
 1891
 \end{array}$$





### Proven

you are to keep a book in which you enter  
each day the following -

Total No of Each Volt of lamps of a days  
photonium run - This will show the average  
Volts of the lamps;

2 The Volts used in exhausting lamps,  
The first peaking in should be 110 Volts  
and finishing 160 or 175 volts.

3<sup>d</sup> Report if any lamps go to dull red  
on 1st peaking in - if so what proportion  
Note the Resistance box that does this  
its number on the scale -

Note how many peags they go up on  
average before filament becomes red

Note general Condition of boxes if they  
are regular & do not have places where  
there is too great a jump of candle  
power =

4<sup>th</sup> Note average time before getting  
Vacuum before peaking - also if they

(2)

peg - before tube is solid - If they seal off when before tube is solid - ~~then~~ and what percentage roughly they do this = also what they do when pumps lag =

5th = State of Mercury - of swabs for cleaning tubes - if oily = test Mercury by running in smooth writing paper I will show you about this -

6th = Note everything done in pump room - from the 1st day you take hold, and then every change thereafter that takes place -

7th Describe in detail how they work the blur off etc -

8th Inspect the phosphorus Cup note average of those Cups which are gummed over on top of phos - of different amount etc -

9th = State of phosphorus in the  
mechanical dryer - also if fine flakes  
plus are liable to be carried up with blots -

10th Note if they try to run a pump  
too long -

11th Note if the Valts used in  
pump room vary much -

12th Note in room where Carbons  
are unspoked or put on wires how  
they handle them if no injury is  
done.

13th = Note if the temperature of  
Baking oven when lamps are  
heated before putting on pumps  
of temperature is greater than  
necessary to slightly brown  
paper. -

|       |     |       |
|-------|-----|-------|
| 100   |     | 40.   |
| 60    |     | 24.80 |
| 35.20 | 600 | 14.76 |
| 20.44 | 369 | 8.67  |
| 11.77 | 217 | 5.07  |
| 6.70  | 126 | 2.88  |
| 3.82  | 130 | 1.06  |
| 2.70  |     | 76    |
| 2.00  |     | 30    |
| 1.60  |     | 15    |
| 1.00  |     |       |
| 5.0   |     |       |
| 5.0   |     |       |
| 3.0   |     |       |

146.89  
25.1

9845  
40  
40  
2

146-

60 lbs-

50  
40

14-

Take 20 lamps each day just as the Come bring them to dull red in dark room and record @ pots - devise a system of your own to give the degree of @ patting =

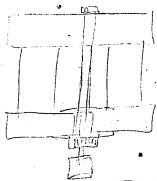
15th - See that in Sealing in machine flame does not strike filament & also if any injury from handling -

16th see if ~~the~~ bulbs are generally clean before sealing in -

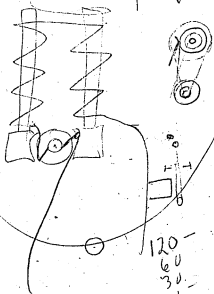
17. Report defects in clamping material & all other defects of whole output. The factory





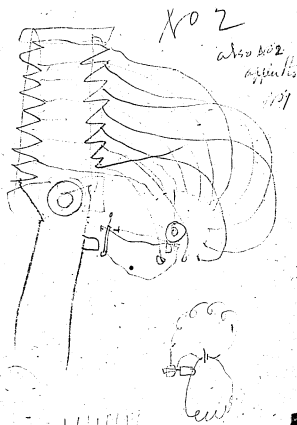


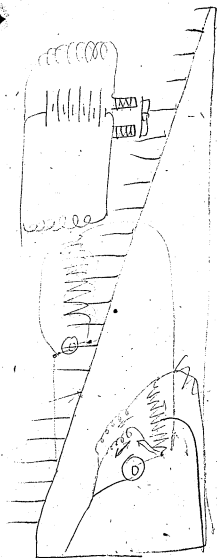
NOT



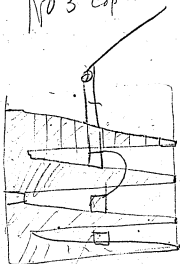
50  
100  
200  
400  
800  
1600

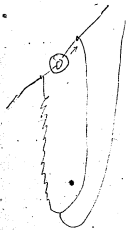
120 -  
60 -  
30 -  
15 -  
7.5 -  
3





short slit suture wound  
Copper sheath ~  
also there is Res  
No 3 Cop sheath suture wound





N04



iron  
and shaft

N05



increase in torque  
moves knee to  
more bristles on  
Res for feed-

$$\begin{array}{r}
 382 \\
 35 \\
 \hline
 1915 \\
 1146 \\
 \hline
 13370 \\
 670 \\
 \hline
 288 \\
 382 \\
 \hline
 382
 \end{array}$$

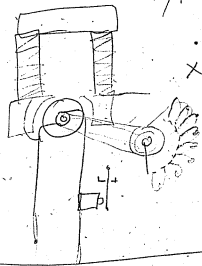
$$\begin{array}{r}
 1177 \\
 507 \\
 \hline
 670
 \end{array}$$

$$\begin{array}{r}
 670 \\
 35 \\
 \hline
 3350 \\
 2010 \\
 \hline
 23450 \\
 126 \\
 \hline
 380 \\
 380 \\
 \hline
 28800
 \end{array}$$

$$\begin{array}{r}
 1177 \\
 35 \\
 \hline
 5885 \\
 3531 \\
 \hline
 41195 \\
 217 \\
 \hline
 628 \\
 80 \\
 \hline
 50240
 \end{array}$$

$$\begin{array}{r}
 382 \\
 106 \\
 \hline
 276
 \end{array}$$

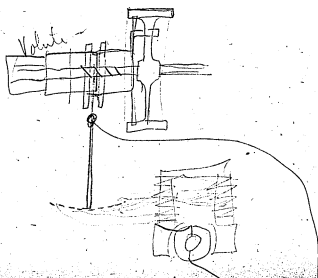
$$\begin{array}{r}
 133 \\
 80 \\
 \hline
 10640
 \end{array}$$

$$\begin{array}{r}
 276 \\
 35 \\
 \hline
 380 \\
 828 \\
 \hline
 9660 \\
 628 \\
 502 \\
 \hline
 128 \\
 .96 \\
 \hline
 17680
 \end{array}$$


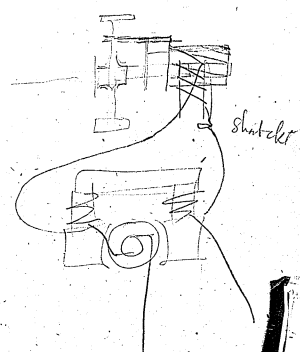
No. 6

X feet.

No 7



No 8



$$\begin{array}{r} 2044 \\ 867 \\ \hline 1177 \end{array}$$

$$\begin{array}{r} 3520 \\ 30 \\ \hline \end{array}$$

$$\begin{array}{r} 6000 \\ 2480 \\ \hline 3520 \end{array}$$

10-

$$\begin{array}{r} 50 \\ 80 \\ \hline 130 \end{array}$$

$$\begin{array}{r} 2044 \\ 35 \\ \hline \end{array}$$

$$\begin{array}{r} 3520 \\ 1476 \\ \hline 2044 \end{array}$$

50

$$\begin{array}{r} 60 \\ 35 \\ \hline 95 \end{array}$$

$$\begin{array}{r} 35-10220 \\ 35-6132 \\ \hline 7154 \\ 369 \\ \hline 7523 \\ 1909 \\ \hline 9432 \end{array}$$

$$\begin{array}{r} 21 \\ 10 \\ \hline 31 \\ 248 \\ \hline \end{array}$$

$$\begin{array}{r} 180 \\ 210 \\ \hline \end{array}$$

$$\begin{array}{r} 175 \\ 125 \\ \hline 300 \end{array}$$

$$\begin{array}{r} 715 \\ 369 \\ \hline 1084 \\ 80 \\ \hline 1164 \end{array}$$

$$\begin{array}{r} 3100 \\ 2480 \\ \hline 620 \end{array}$$

$$\begin{array}{r} 1225 \\ 620 \\ \hline 1845 \\ 80 \\ \hline 1925 \end{array}$$

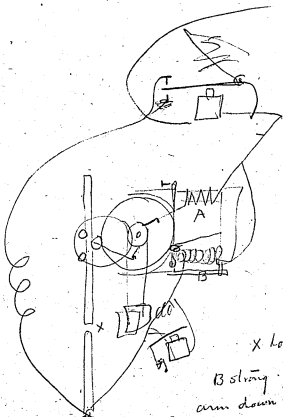
$$\begin{array}{r} 1084 \\ 80 \\ \hline 1164 \end{array}$$

$$\begin{array}{r} 1845 \\ 1476 \\ \hline 369 \end{array}$$

$$\begin{array}{r} 1084 \\ 867 \\ \hline 217 \end{array}$$



9 inches  
7 inches



X long

B string  
arm down

X short  
A string arm up

Current off:

34.7

34.7

$$\begin{array}{r} 255 \\ 8 \\ \hline 1338 \end{array}$$

$$\begin{array}{r} 138.8 \\ 153.0 \\ \hline 291.8 \end{array}$$

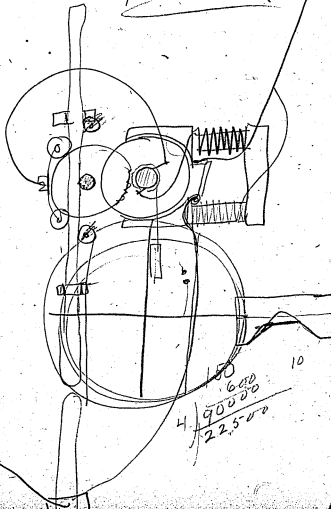
$$\begin{array}{r} 2000 \\ 347 \\ \hline 2000 \\ 147.00 \\ 140.00 \\ \hline 7.0000 \end{array}$$

$$\begin{array}{r} 160 \\ 153 \\ \hline 313 \end{array}$$

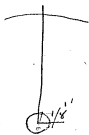
$$\begin{array}{r} 2000 \\ 347 \\ \hline 1735 \\ 2650 \\ \hline 429 \end{array}$$

$$\begin{array}{r} 57 \\ 2240 \\ \hline 228 \end{array} (4-$$

101 arc lamp



$$\begin{array}{r} 150 \\ 600 \\ 90000 \\ \hline 4 \\ 225000 \end{array}$$
 10



$1/32$

$1/8$

$13/16$

16

15

$1/4$

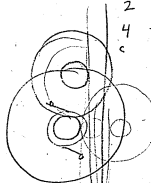
$1/4$

2

4

5

$24/32$

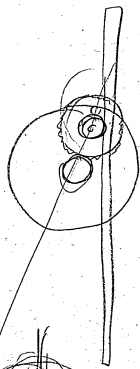


$12/16$

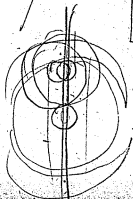
3

7

8

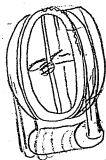
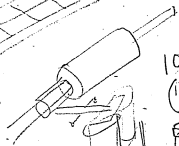
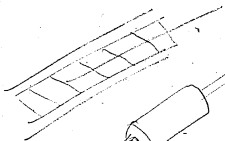


arc no 1



or this

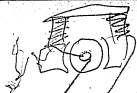




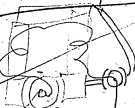
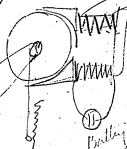
100 Vals CO-  
130 Vals CO<sub>2</sub>  
70 - CO-

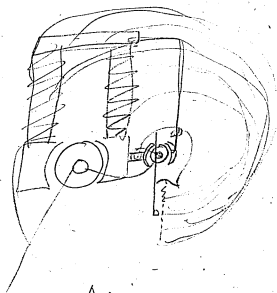
70  
60  
13  
30 CO<sub>2</sub> 100 CO  
60 CO 100

160-

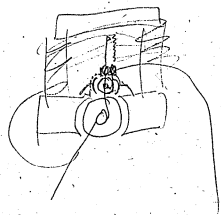


109





No 10 -



No 11

1st. The Ether is what we call the quality in  
Condensation or a stress in Ether manifest  
by increased magnetism -

Electricity is produced in any  
arrangement which can  
conduct and return an impulse to  
its place of origin by a condensation  
or diminution of the orbit of the molecule  
or displacement from one position to  
another position of the Molecules  
where any condensation can  
take place =

in a thin wire heat travelling  
Expands the distance of the molecules  
when they reach the metal wire there  
is a change in temperature by a  
poor conductor for heat. the  
distance between the molecules  
will suddenly diminish & this  
will produce a molecular wave  
throughout the circuit & produce  
what we call Electricity as with  
sound waves in a wire

with a battery, the effect is to  
produce a wave decompose water  
a wave is produced by rarefaction of  
O from H. & a condensation by combination  
of O with Zn & H with O of  
say Copper oxide

In passing a wire through a  
magnet, there is a rarefaction of  
of the molecules by the stress acting  
on the molecules. These stress lines  
of which there are billions in a Condenser  
exist between the molecules in a  
portion where there are also billions  
producing Condensations at one  
point & rarefaction at other,

Magnetic lines & Ether pervade  
all space but around the Earth there  
are Condensations & thus we have  
the Magnetic lines of the Earth.



$$\begin{array}{r}
 11) 140 \text{ (12.72)} \\
 \underline{110} \\
 30 \\
 \underline{22} \\
 80 \\
 \underline{77} \\
 3
 \end{array}$$

$$\begin{array}{r}
 8.75 \\
 \underline{16} \\
 5250 \\
 \underline{875} \\
 14000
 \end{array}$$

$$\begin{array}{r}
 140 \\
 1272 \\
 1272 \\
 165-47 \text{ (10.8)} \\
 \underline{16} \\
 34 \\
 484
 \end{array}$$

14

$$\begin{array}{r}
 16) \frac{1272}{25-44} \text{ (159)} \\
 \underline{164} \\
 90 \\
 \underline{88} \\
 144 \\
 \underline{144} \\
 0
 \end{array}$$

12.72 oz per Rev

140 oz 8:75: Lbs started

13 Rev would be 165.4 oz or

10.35 Lbs increasing temp

of hot blast stove should

tend to raise the pressure

~~of hot blast stove~~ The Briquette

just supplied the opening power

of the Coke & limestone taken

off at <sup>extra</sup> slag formed - the

raising of blast from 1000

to 1100 1200 faler -

Each Rev of mg made 25 tons

extra, 2 Rev 50 - 15 Rev

should make 200 tons

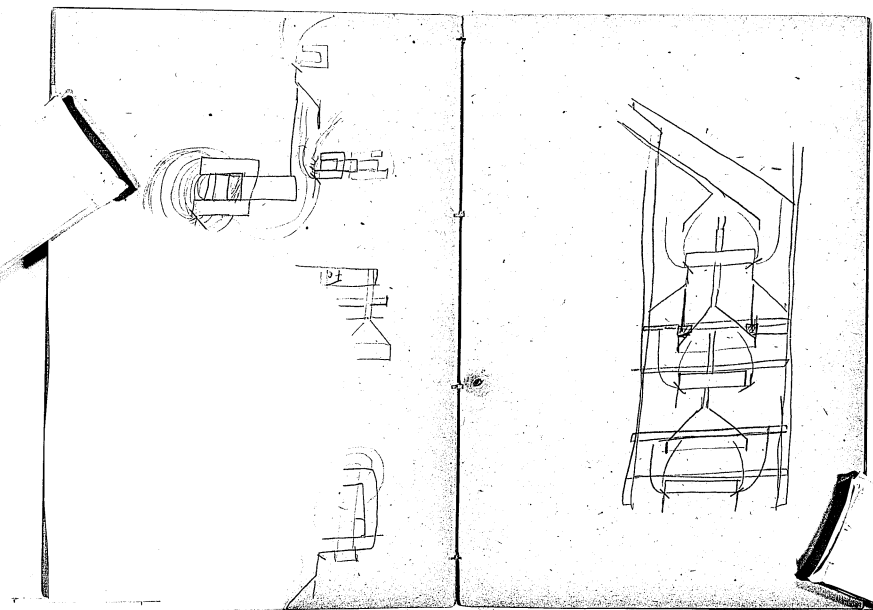
+ increase pressure 1.59 lbs

(10) from 1035 to 1194 say  
12 lbs - with gas to 1300



1035  
1.59  
1194





300 —————  
 225 —————  
 157 — 38 m —————  
 61. — 13 m —————  
 26 — 7 m —————  
 10 — 4 —————  
 4 —————  
 2 —————  
 6 —————  
 480  
 2

480 ←  $\frac{120}{480}$

$\frac{130}{90}$  — 70 hp  
 $\frac{75}{50}$   $\frac{35}{25}$   
 $\frac{70}{50}$   $\frac{28}{25}$

$\frac{130}{35}$   
 $\frac{650}{390}$   
 $\frac{4550}{4550}$

$\frac{245}{170}$   
 $\frac{415}{415}$

$\frac{26}{16}$   
 $\frac{10}{10}$   
 $\frac{26}{1300}$

$\frac{7}{20}$   
 $\frac{20}{1600}$

$\frac{225}{50}$   
 $\frac{11250}{11250}$

$\frac{112}{75}$   
 $\frac{187}{80}$   
 $\frac{14960}{14960}$

$\frac{157}{75.5}$   
 $\frac{3800}{11350}$   
 $\frac{80}{9040}$

$\frac{300}{149}$   
 $\frac{151}{151}$

$\frac{187}{149}$   $\frac{151}{90}$   
 $\frac{38}{61}$

$\frac{3050}{1350}$   
 $\frac{4350}{80}$   
 $\frac{34800}{34800}$

$\frac{6100}{348}$   
 $\frac{262}{262}$



3.4

2660

$$\begin{array}{r} 2261 \\ 566 \\ \hline 1995 \\ 665 \\ \hline \end{array}$$

$$\begin{array}{r} 108 \\ 20 \\ \hline 128 \\ 1995 \\ \hline 2223 \\ 665 \\ \hline 2888 \end{array} \quad (3.4)$$

$$\begin{array}{r} 66.5 \\ 50.0 \\ \hline 16.5 \\ 1.9 \\ \hline \end{array}$$

190-

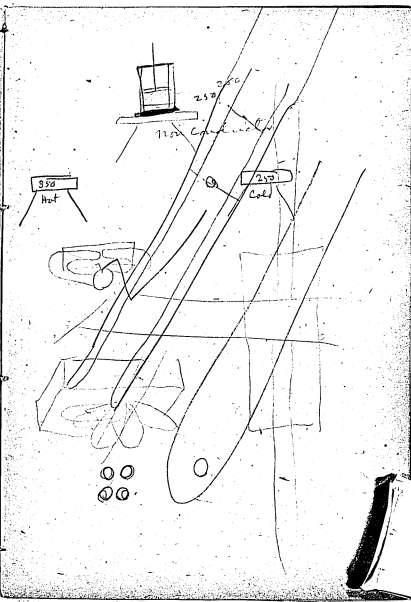
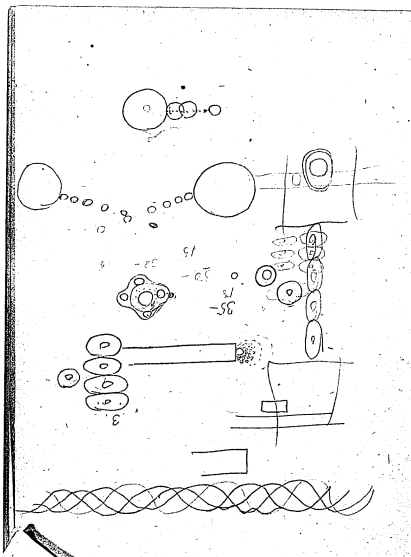
$$\begin{array}{r} 220 \\ 238 \\ \hline 260 \\ 34 \\ \hline 60 \end{array} \quad (1.86)$$

$$\begin{array}{r} 34 \\ 70 \\ \hline 2382 \end{array}$$

$$\begin{array}{r} 10 \\ 170 \\ \hline 271 \end{array}$$

$$\begin{array}{r} 3700 \\ 250 \\ 600 \\ 600 \\ 1000 \\ 300 \\ 350 \\ \hline 300 \end{array}$$

174



Iron & Carbon 100.0 ght 56.  
Equivalent ~~28~~ Fe = 28 Ferrum -  
Fe =  $18\frac{2}{3}$  - Ferrum  
All iron contains Carbon.

Pure iron - extremely tenacious  
softer than ordinary malleable  
iron - presenting a scaly  
conchoidal, or sometimes crystalline  
fracture. Specific gravity after  
melting  $\rightarrow 7.8439$  - in  
sheet or wire  $\rightarrow 7.75$  to  $7.60$  -

Iron by electrolysis spec  
gravity  $8.1393$   
slowly acted on by acids at  
ordinary temperature  
but by heat evolves H without  
the peculiar smell.

Crystallized from either the  
Cube or Octohedron or some other  
form belonging to the Regular  
system -

Specific Heat of a dry  
malleable iron  $0.113793$   
somewhat higher when an oxid carbon  
is mixed - (Regault)

Conducting power for heat  
gold being say 1000 is 374.3  
Property

Linear & Cubical Expansion is less  
than most metals -  
Increase  $\text{ft}^3$  for each deg between  
100 & 300 C. =  $\frac{1}{68100}$  (D & Platt)

An iron bar expands  $\frac{3}{560}$  when raised  
from a red heat to white heat  
 $+$   $\frac{7}{560}$  when heated from 20 C  
to whiteness. (Rumford)

Melting pt. that certainly determined  
 1550 C by Ponslet, 2000 C by  
 Scheerer.

Electric Conductivity -  
 Cu 100 then Fe 20 (Harris)  
 15.8 (Boggs) → 17.74 Leng -

~~Plate was at 0 C 91.8  
 72.8 } Matheson  
 84.6~~

~~Electrolytic  
 at 0 C = 100.~~

|           | Conductivity at 0 C | % of decrease in conductivity at 100 C | Sp. Heat at 0 C |
|-----------|---------------------|----------------------------------------|-----------------|
| Steady Fe | 100                 | 38.3                                   |                 |
| Plate Fe  | 91.8                | 36.0                                   | 102.7           |
|           | 84.6                | 34.7                                   | 99.2            |

Matheson

up to temp of 146 C Boiler plate  
 normally not gone p.c. (C. dimension)  
 at red heat reduced  $\frac{1}{4}$ th  
 tendency of good rivet now at  
 140 C w  $\frac{1}{3}$  quarter dim at  
 ordinary temp but at red heat  
 reduced nearly one half.

Malleability increases with  
 temperature. While heat is  
 pasty, its plastic + kneadable  
 far below its melting point.

Combines with O when heated  
 to 230 C in air

Dissolves  $H_2O$  at 360 C

Burns at white heat.

Sweeded sh iron

|         | No 1   | 2      | 3      |
|---------|--------|--------|--------|
| Iron    | 99.803 | 99.220 | 99.544 |
| Carbon  | 0.034  | 0.087  | 0.087  |
| Silicon | 0.025  | 0.056  | 0.115  |
| Sulphur | 0.055  | 0.632  | 0.220  |
| Phos    | Trace  | 0.005  | 0.034  |

Only iron free from C is called  
brant iron - this can't be  
welded, but some say this is  
because of oxide getting in

Sulphur makes Fe Red short  
Phos " " Cold short  
Si " " Rotten

Cu " " Red short  
reduces its capability by Welded

Manganese " " Wounds but not steel

Linear Exp of Cast iron

$\frac{7}{560}$  but 20 at 560C (fresh cast) -  
+  $\frac{12}{560}$  but 20 at white heat (Pinner)

Cast Fe mass - bulk when  
cooled from liquid

pieces of Fe thrown into fluid Cast iron  
sink but when near MP float

The greatest amount of expansion  
or contraction takes place just  
below the melting point in Cast iron  
on account of a contraction near  
MP its necessary work would  
for castings proportional to area  
then castings are required to be

Centradian grey C Fe 1 pct.  
white iron 2 to 2.5% of linear  
dimension

Cast Fe undergo permanent increase  
bulk under long continued heat

at slow and temp <sup>gray CFe</sup> oxidizes sooner  
than white Fe.

White CFe dissolved <sup>Cave</sup> in HCl hot  
entirely dissolved - gray CFe  
leaves residue of <sup>gray CFe</sup>  
- both cases containing C enters into  
Cm<sub>2</sub> with the nascent hydrogen  
forming Valerian Hydrogen.

Actin. 7. 1. 1. BCP-

White CFe - bushy, dark brown residue  
sat in KO, readily combustible, leaves  
black residue containing silica.

Gray CFe - Residue part graphite.  
partly of Carbonaceous sub as in White CFe  
+ some black Carbonaceous stuff which  
is magnetic takes fire in air & leaves  
Ferrous oxide.

Gray Fe 2' @ 415°C  
White 3' @ 517°C

Magnetic pyrites.

Slightly magnetic - sp gravity 4.15 @ 417  
ocean crystalline compound of pyrites  
60.149 - gram - 39.57 Sulphur

Ammoniac pyrites - Calcopryditis  
other than the Co pyrites of pyrites.

Alloys

3 pts Bismuth 1 Fe steel magnetic  
lead - 96.7 Fe 3.24 Pb - magnetic

Manganese 22% of it renders Fe  
non-magnetic

Mo. & W. in equal parts; magnetic  
Blue gray brittle, fine grained.

FeBr<sub>2</sub> Fe<sub>2</sub>B<sub>6</sub>-

FeCl<sub>2</sub> Fe<sub>2</sub>Cl<sub>6</sub>  
FeI<sub>2</sub> Fe<sub>2</sub>I<sub>6</sub>

Ferrous alkalis precip from salts  
as White Hydrate.

Fe forms 2 oxides

Primal FeO + Sanguiferous Fe<sub>2</sub>O<sub>3</sub>

Various colors like Cuprous which  
might represent an intermediate

The intermediate Oxid. products  
(Sanguiferous) Fe<sub>2</sub>O<sub>3</sub> is magnetic  
just as iron ore -

Oxoferrous Sulphide Fe & S magnetic

Hemiferrid. Fe 2S "

Proto, FeS non mag.

Sanguiferous Fe<sub>2</sub>O<sub>3</sub> non mag.

Ferrous-ferric Sulphide - magnetic pppts

Disulphide Fe FeS<sub>2</sub> non magnetic

action called iron  
Calden appeared Argel. line in fracture  
Wann. line " " fibrous  
all this has been denied - but  
proof given by RR records -

See Barnett Phil Mag (4) Xvii 51.  
also (4) XVI 478

Hankel Ann phy chem 201 283-

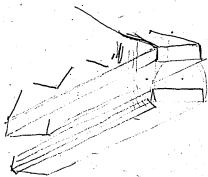
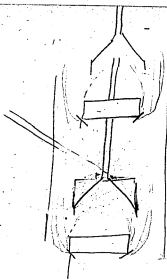
Rowland Phil Mag (4) Xvii 10  
Xlviii - 321.

A. H. Holz - Poggend - Cpi 69 -  
Cpi 67

Ann phy - Erganzbd - VIII 553.  
Jahresb f Chem 1847 171 -







Darwin T. Proc Roy Soc XXIII 445 473

G Weidemann paper

CXXVI 1 CXXV 177

of pr chan (2) IX 145 - Phil Mag

(4) XXX 366 XXXVII 314 -

(5) - IV 161 276 -

T. L. Phipson Bull Soc Chem (2) VII 322

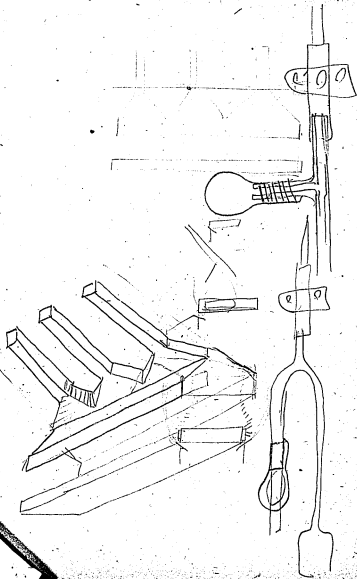
Chem News XXXII 162

### Maxwell's Theory

Magneto Electric phenomenon due to  
 substance matter under certain

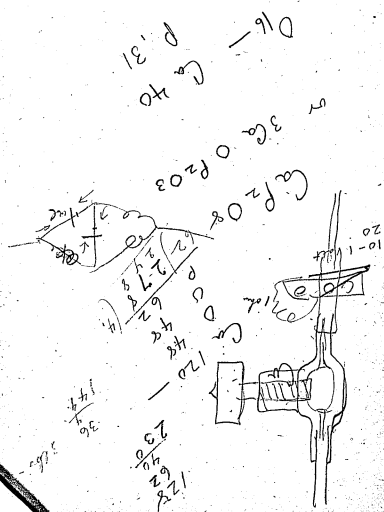
conditions of motion or of pressure in  
 every part of the magnetic field the substance  
 producing these effects may be a certain  
 part of ordinary matter or it may be  
 any other associated with matter

over



2 = Condition of any part of the field through which lines of magnetic force pass is one of unequal pressure in different directions, pressure being least along lines force so that this may be considered lines of tension -

3. This inequality of pressure is due to Vortexes Coaxial with the lines of force - The density of revolving matter is proportional to the Magnetic permeability of the medium. The density of rotation is related to the diameter of the line of force & the velocity of the circumference of the Vortex is proportional to the resultant magnetic force -



4th The Vortices are separated from each other by a single layer of round particles so that a system of cells is formed the particles being layers of these particles & the substance of each cell being capable of rotating as a vortex -

5th particles forming the layer are in rolling contact with both the vortices which they separate but do not rub against each other. They are perfectly free to roll between the vortices & so to change their place - provided they keep within one complete molecule of the substance but in passing from one molecule to the other they experience resistance & generate irregular molecules which constitute heat.

8000 - 1 deg C.

1 gm C heat 8000 gm wt H<sub>2</sub>O 10  
 1 gm C heat 48000 gm wt Iron or 1  
 1 gm C heat. 48-gm Iron or 1000. C.  
 $\frac{2}{6}$

$$\begin{array}{r} 15 \overline{) 1000} \\ \underline{90} \\ 1000 \\ \underline{1000} \\ 0 \end{array}$$

$$\begin{array}{r} 1000 \\ \underline{960} \\ 40 \\ \underline{40} \\ 0 \end{array}$$

$$\begin{array}{r} 200 \\ \underline{150} \\ 50 \\ \underline{50} \\ 0 \end{array}$$

$$\begin{array}{r} 1000 \\ \underline{900} \\ 100 \\ \underline{100} \\ 0 \end{array}$$

$$\begin{array}{r} 1000 \\ \underline{900} \\ 100 \\ \underline{100} \\ 0 \end{array}$$

These particles play the part of Electricity. Their motion of translation constitutes an electric current, their rotation serves to transmit the motion of the whole from one part of the fluid to the other. The tangential pressure thus set into play constitutes energy of the electric field of the circulating particles constitutes electric displacement.

$$\begin{array}{r} 1080 \\ \underline{1527} \\ 3654 \end{array}$$

$$\begin{array}{r} 5000 \\ \underline{4968} \\ 32 \\ \underline{32} \\ 0 \end{array}$$

$$\begin{array}{r} 11 \\ \underline{11} \\ 0 \end{array}$$

$$\begin{array}{r} 1000 \\ \underline{900} \\ 100 \\ \underline{100} \\ 0 \end{array}$$

$$\begin{array}{r} 212 \\ \underline{180} \\ 32 \\ \underline{32} \\ 0 \end{array}$$

19  
Discovery 2 - Edison

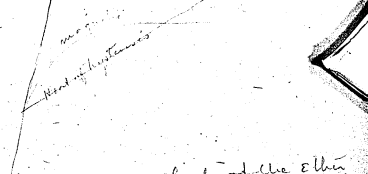
production of  $\mathcal{E}$  is a molecular disturbance - a forced disturbance of molecular architecture, it is never a continuous disturbance but momentarily all such disturbances produce waves in the ether of frequency comparable to light vibrations. Sound vibrations are immense, not molecular.

If molecular disturbances are made near a conductor, the only reason that the wave proceeds along a conductor is due to the molecular hysteresis of such conductor producing a lag between the ether in the smaller ether ether of the boundary.

Were there no lag, telegraphy would be impossible - this hysteresis produces heat - Each wave produces a movement of the molecules + a return ~~to~~ + releases freedom gives the heat.

The same holds good for iron  
The phenomenon of Magnetism is only

a question of degree, special amplification renders it striking but it is however only an accidental phenomenon hysteresis + heat due to reversal in iron will go one up to any degree being after the iron is saturated so called the heat of hysteresis will Calumie yield (the iron is red hot + above



There are no whorls of the ether around a Copper wire -  
The wave is like a wave put in the center of a phor funnel, it goes forward along the wire + not into space as with sound by reason of the lag in the ether in the wire

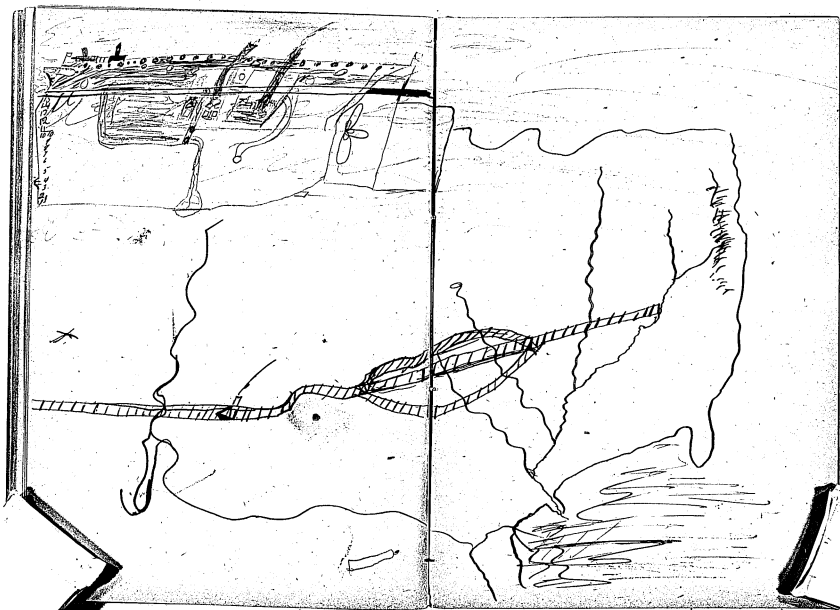
due to necessity of wave to  
do in disturbing the molecular  
structure.

Copper molecules return to zero  
at each wave while iron  
does owing to its molecular struc-  
ture.  $\therefore$  successive waves  
work the molecules under  
more or less strained conditions  
with a present state of strain  
depending on the wave when no  
further deformation of the structure  
takes place - when this condition  
is reached in hard steel, it  
is the permanent distortion of  
the normal condition of matter  
in connection with the ether.  
The 2 form a fixed elastic medium  
in which nothing moves -  
The passage of a wave through this  
fixed system in air simply serves  
to disturb the molecular condition  
of the wire & produce waves

which pass over the wire by the  
long due to hydraulic work in the wire.

Wire the internal molecular friction  
in copper as great as in iron. The  
shard have magnification

The resistance of Copper  
and Zinc is the same, but the  
work done in the Zinc by hydraulic  
is greater, there is a <sup>greater</sup> drop of <sup>energy</sup> if they  
have we say it has more resistance  
The difference in the friction is measured  
by keeping Zinc at 30 grains  
the <sup>weight</sup> of Cu until both have  
the same apparent <sup>weight</sup> then the  
heavier is to be represented the  
internal friction of the Zinc -



$$\begin{array}{r}
 373 \\
 \times 15 \\
 \hline
 1865 \\
 3730 \\
 \hline
 5595 \\
 \quad 24
 \end{array}$$

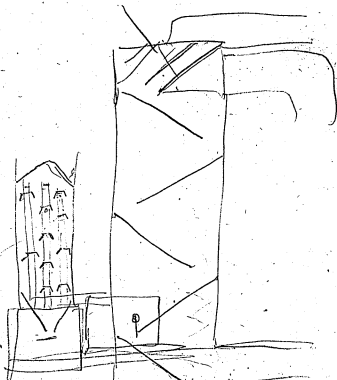
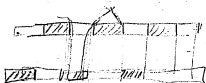
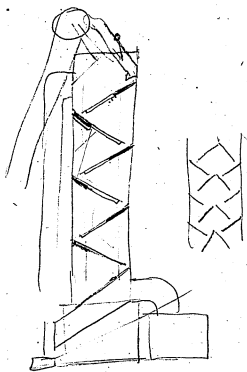


$$\begin{array}{r}
 .12 \\
 45 \\
 \hline
 60 \\
 480 \\
 \hline
 54
 \end{array}$$

100/5.4 amp

2/1000 ohms





60 / 300 / 5.

400) 3200 (

32) 400 (15  
320  
180

25

15 at 2500 -  
7.5 at 5000 -  
3.75 at 10000 -

4

13.5 at 5000  
6.75 at 10000

6

27  
3) 135  
90  
45

2240  
360) 1970 (5 1/2  
1800  
170  
1800  
170

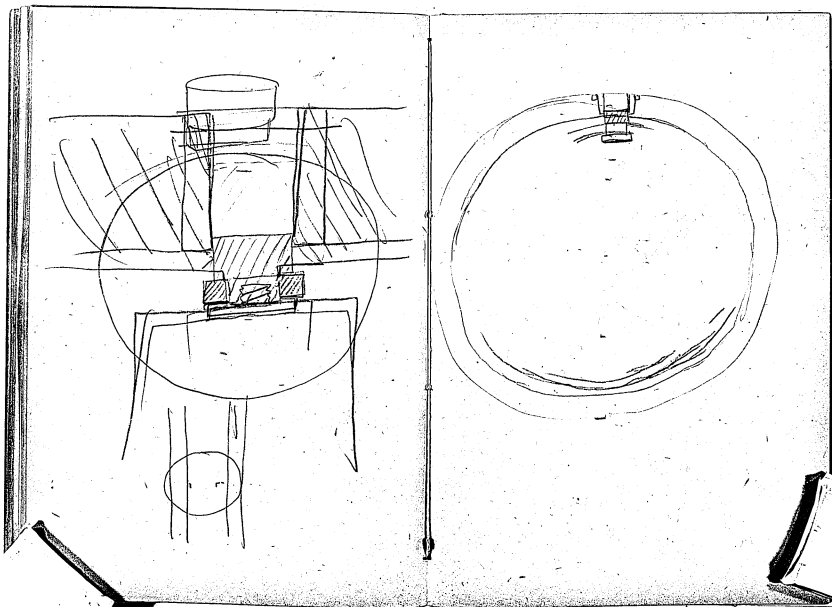
Theoretically 1 ton pure coal  
will raise 80 tons of iron  
ore to 1000 deg Fahr.

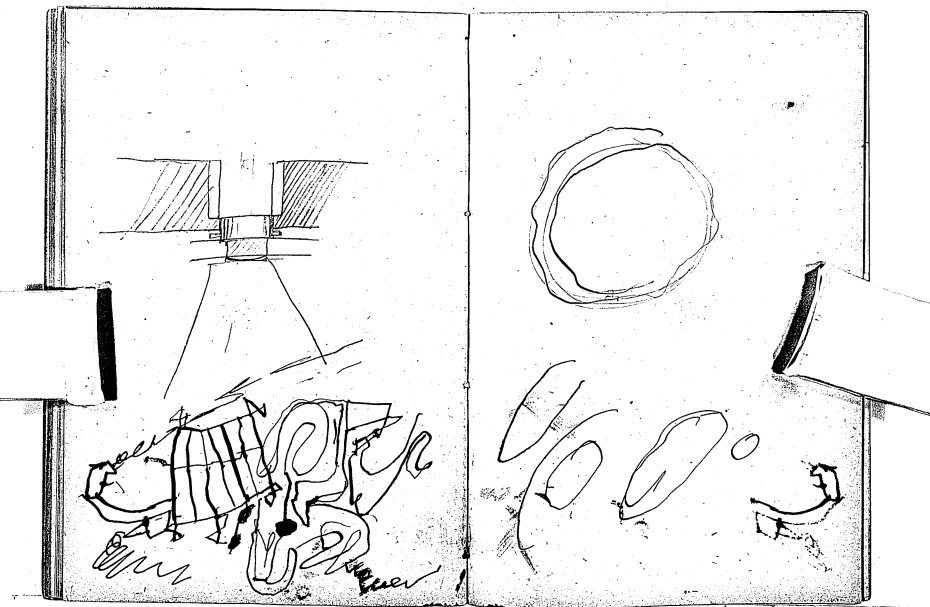
If all heat regenerated - & ejected  
at 200 Fahr finally assuming  
27 lbs air per ton coal burned  
& spec h of air .16 then

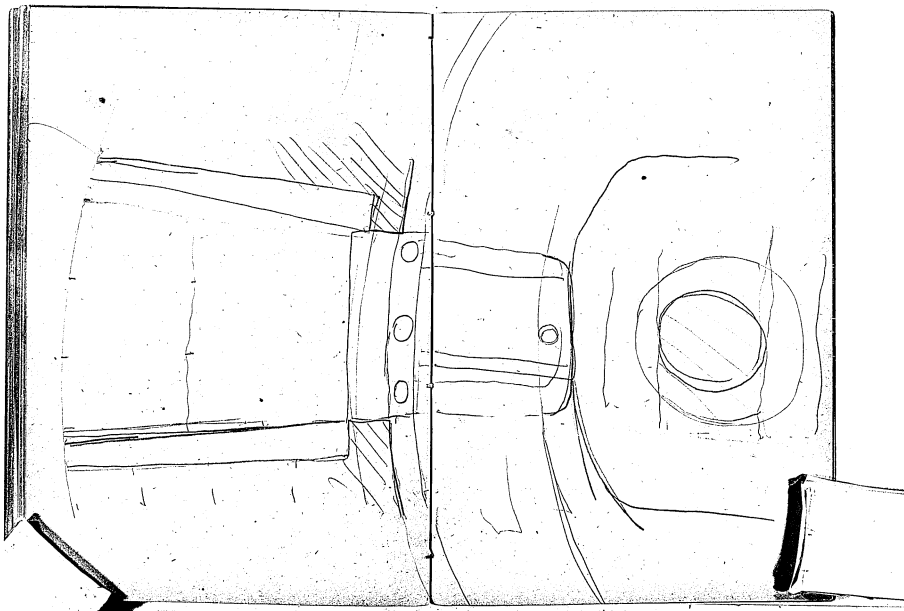
~~37 tons of ore~~ then the 27 ton  
air at 200 is equivalent to  
equivalent to  $6\frac{1}{2}$  tons ore -  
Reducing it from 80 tons to

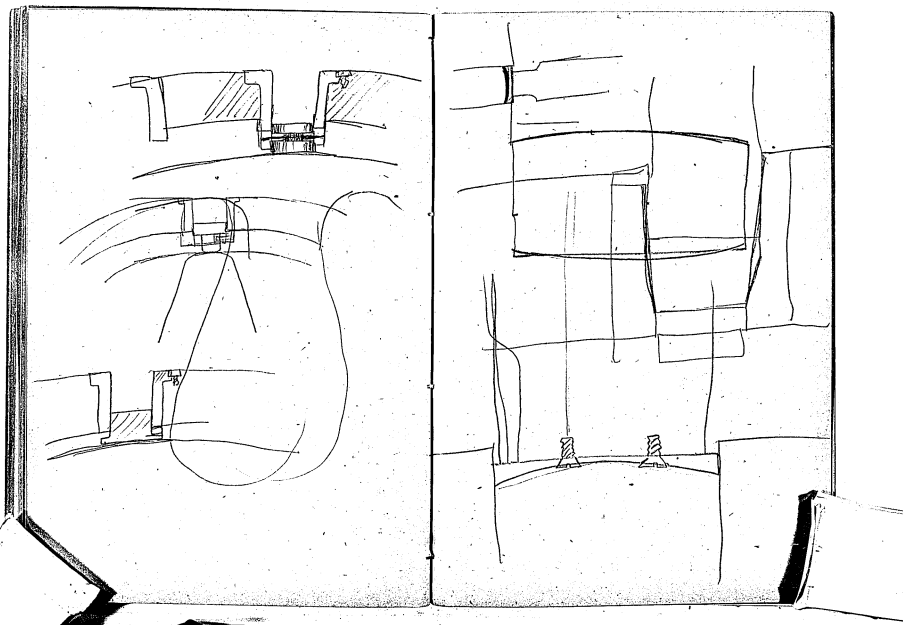
$73\frac{1}{2} =$

If 1 ton coal per hour is used  
then 27 tons of air ejected  
at 200 requires 10 lbs per ton  
or 270 lbs for the air -  
If we eject the ore at 300 - we can  
have 148 tons -

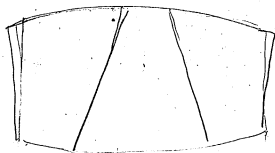




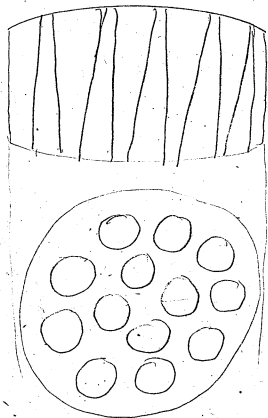
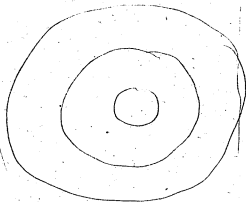


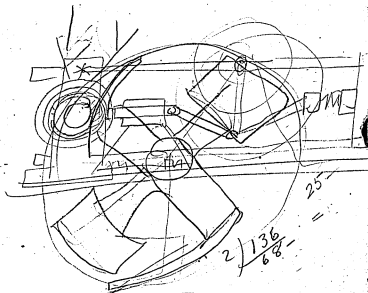


$$\begin{array}{r} 2 \overline{) 3.3} \\ \underline{1.6} \end{array}$$



$$\begin{array}{r} 65 \\ -103 \\ \hline 916 \\ \hline 94 \end{array}$$





$$2 \overline{) 368} \quad 198.$$

$$\underline{184}$$

$$4 \overline{) 49} \quad 4 \overline{) 76} \quad 4 \overline{) 61}$$

$$\underline{12} \quad \underline{19} \quad \underline{15}$$

$$37 \quad 57 \quad 46$$

$$\underline{80} \quad \underline{80} \quad \underline{80}$$

$$96 \quad 132 \quad 121$$

$$4 \overline{) 39}$$

$$\underline{12}$$

$$27$$

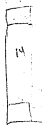
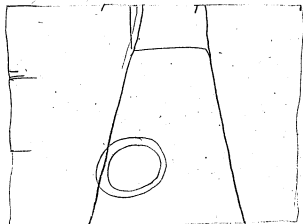
$$\underline{80}$$

$$80$$

|      |     |
|------|-----|
| 100  | 24  |
| 76   | 15  |
| 61   | 12  |
| 49   | 10  |
| 39   | 8   |
| 31   | 6.5 |
| 25   | 5   |
| 20   | 4   |
| 16   | 3.2 |
| 12.8 | 2.5 |
| 10   | 2   |
| 8    | 1.6 |
| 6.4  | 1   |
| 5    |     |
| 4    |     |
| 3    |     |
| 2    |     |
| 1    |     |

25  
0





48

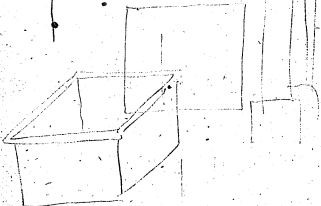
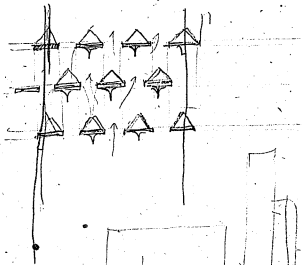
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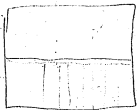
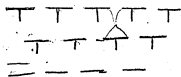
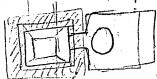
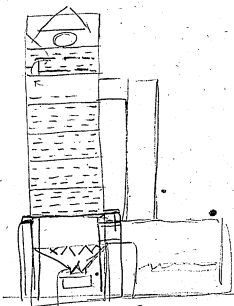
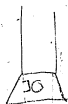
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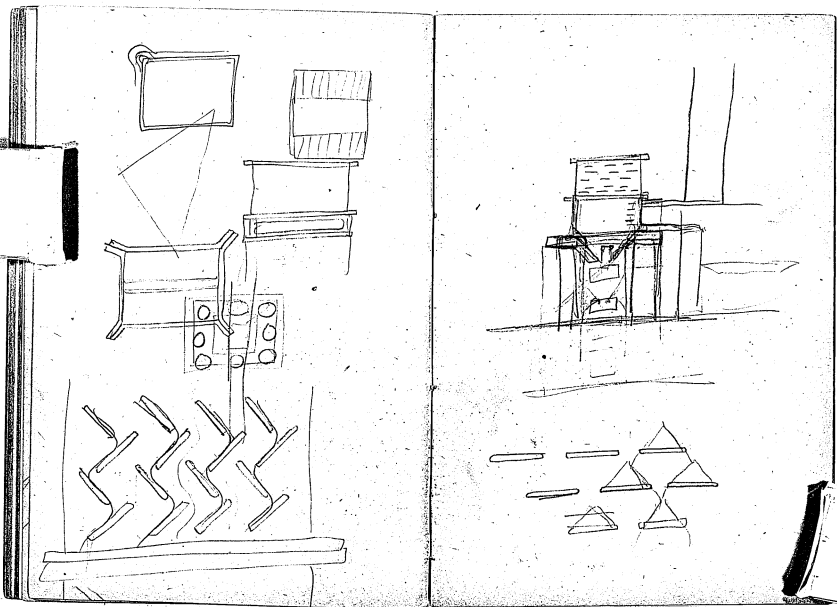


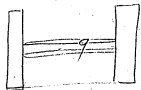
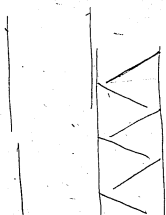
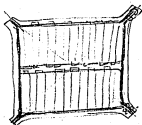
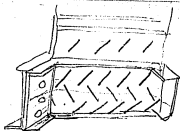
25

91  
25  
66

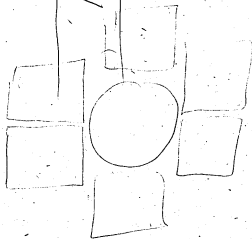




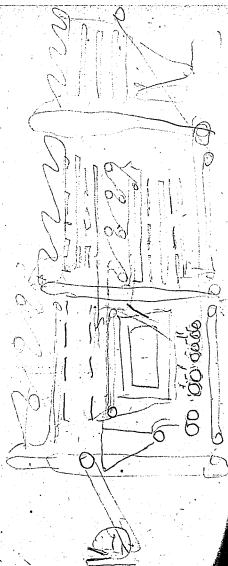
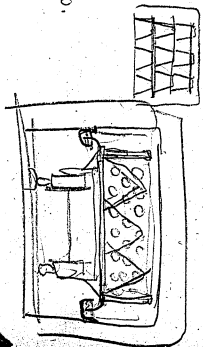


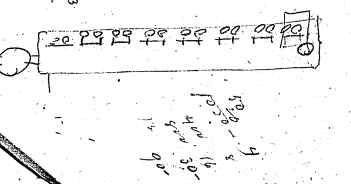
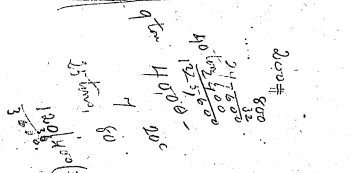
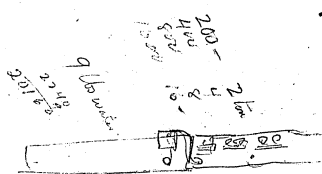


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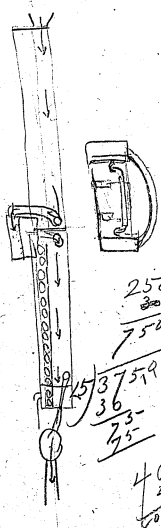


6.0 / 12.00 (20)





15 ft from top -  
 200 ft  
 75 ft from  
 1.6.12



2500  
 200000  
 750000  
 25000000  
 400000000  
 200000000  
 800000000



16  
4  
64

22000  
3000  
2000  
4000

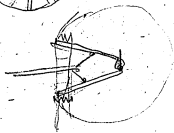
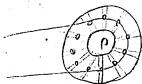
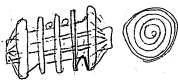
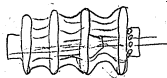
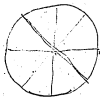
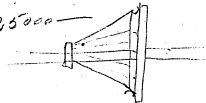
31000

|||||

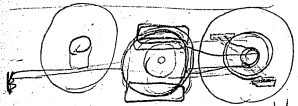
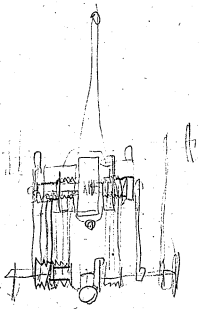
15<sup>th</sup>



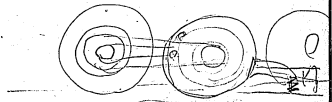
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3



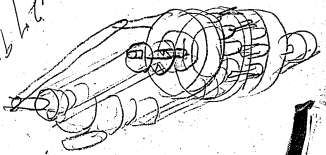


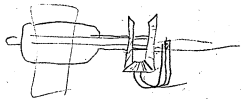
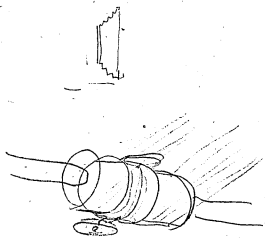


get a bit of light  
BRB

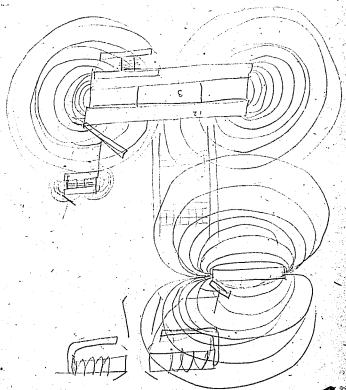


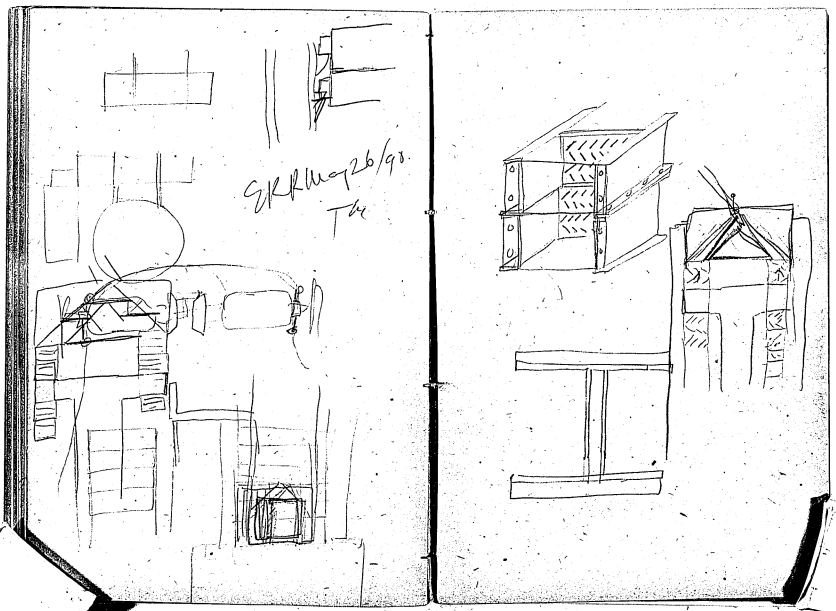
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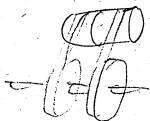
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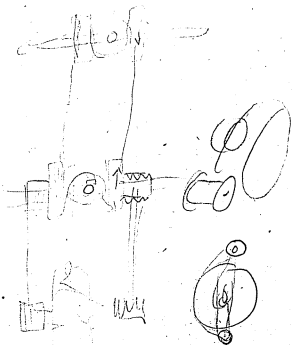




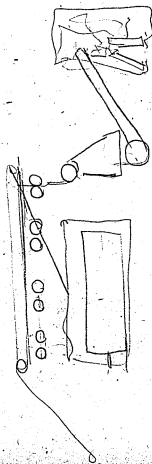
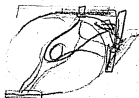


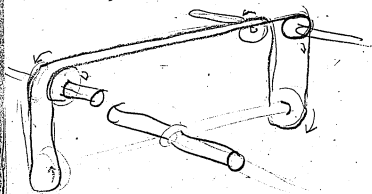
S.K.K. May 26 90



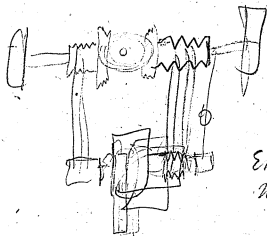


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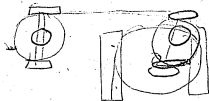


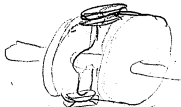
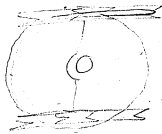


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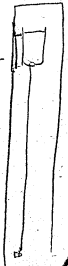
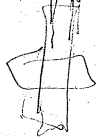
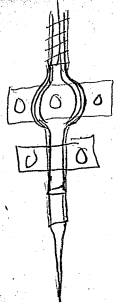
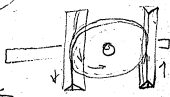
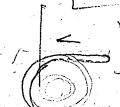


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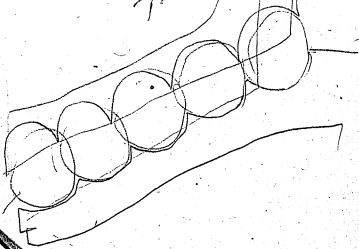


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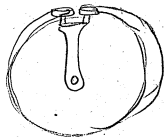
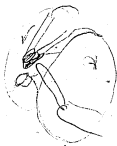
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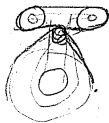


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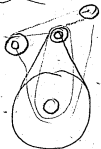




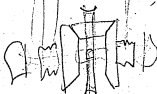
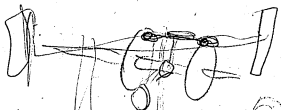


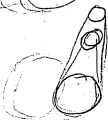
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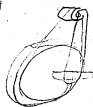
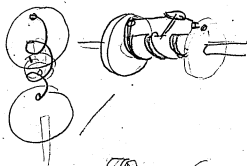
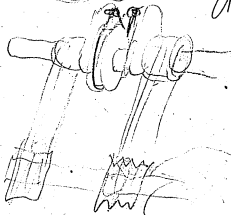


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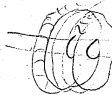




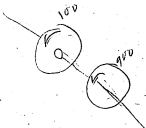
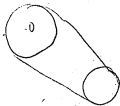
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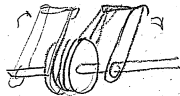
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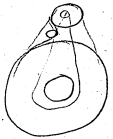
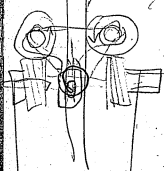
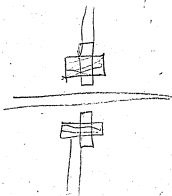
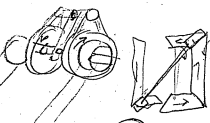
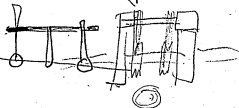
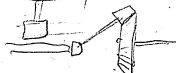


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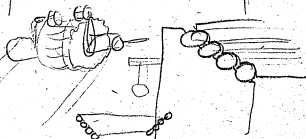
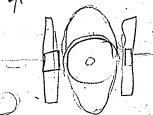
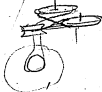


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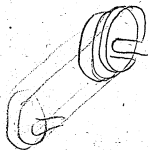
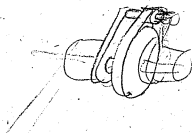
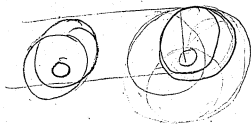
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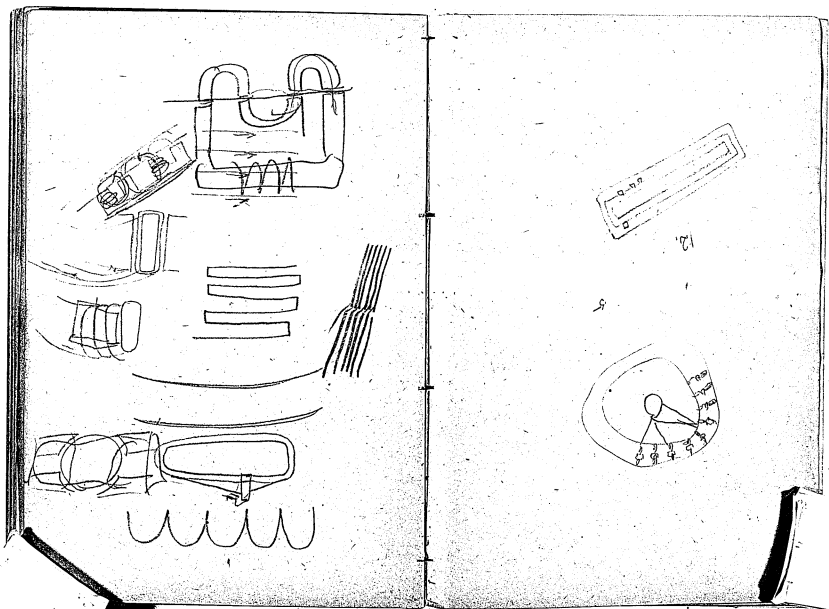


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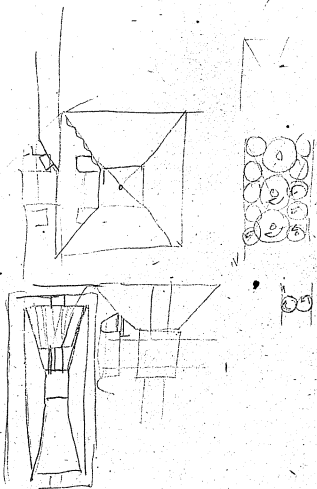


**Notebook, N-90-08-03**

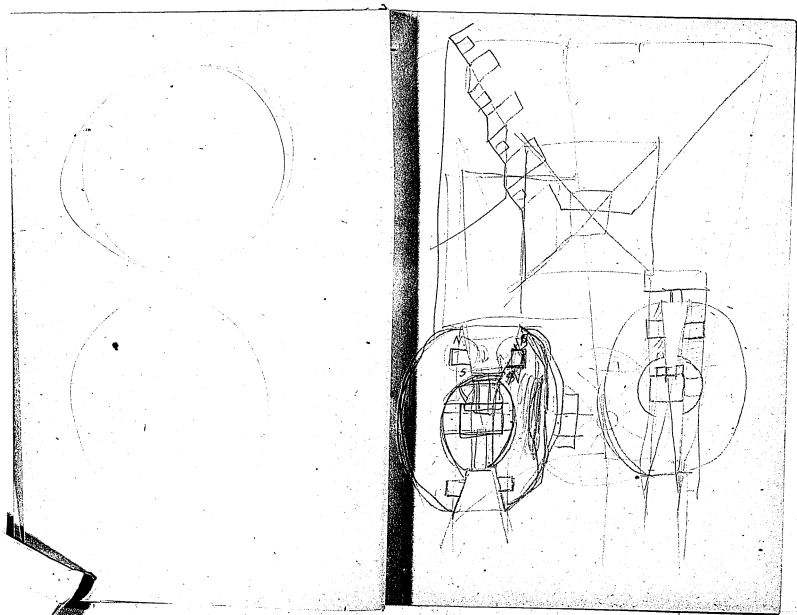
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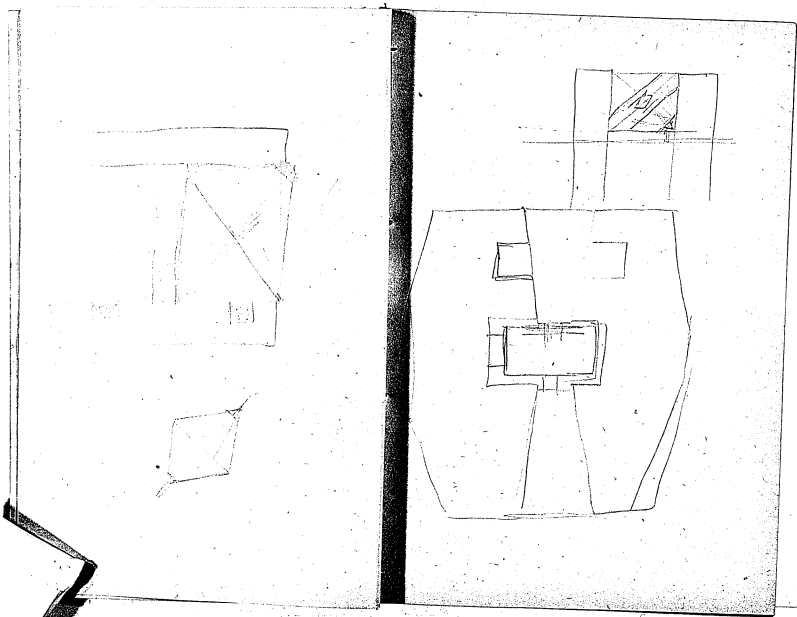
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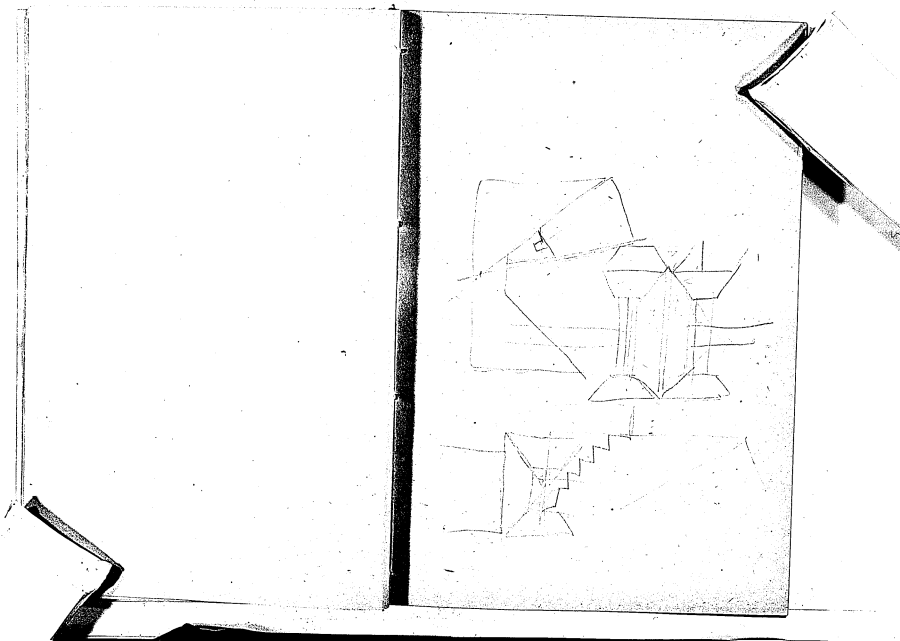
Electric RR July 23/90

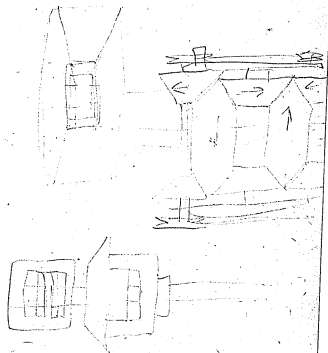


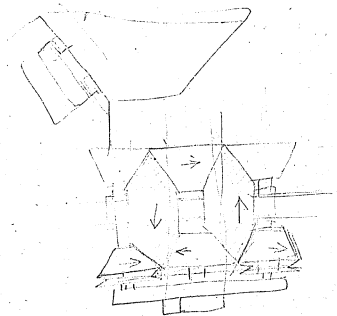
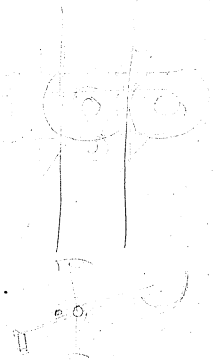


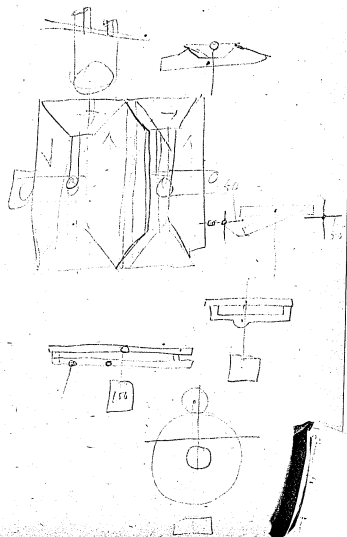


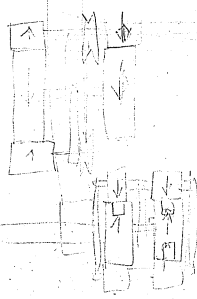
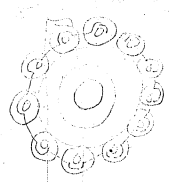
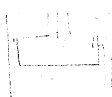
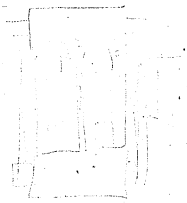


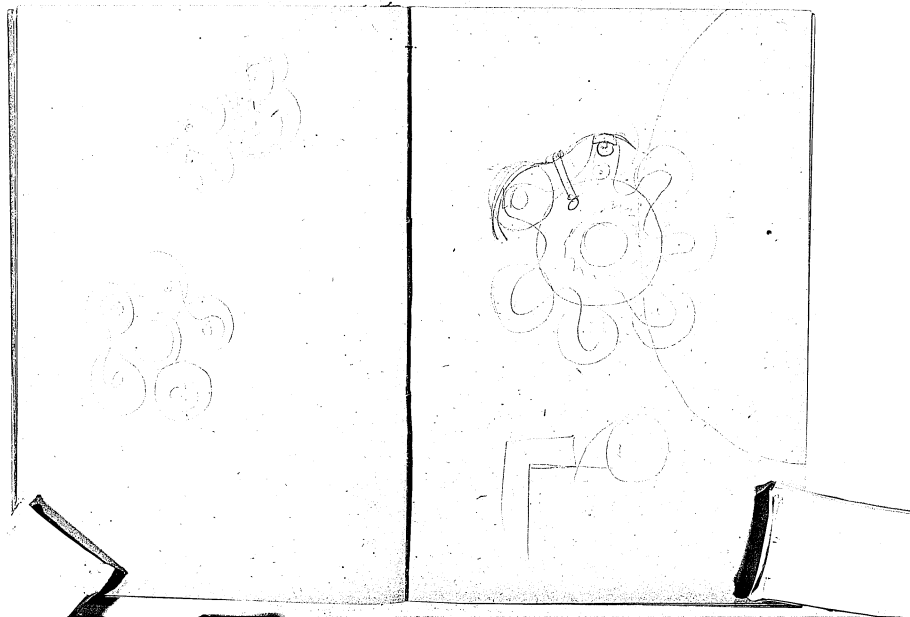




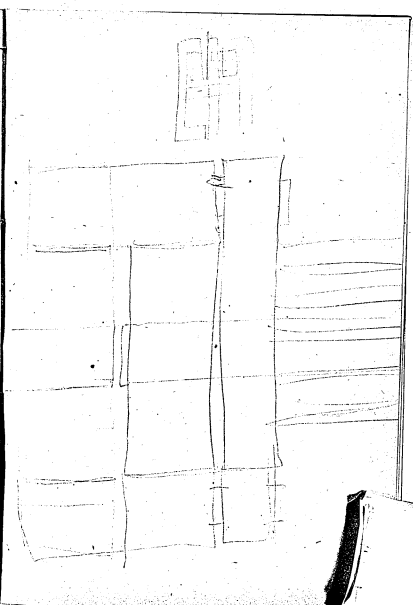
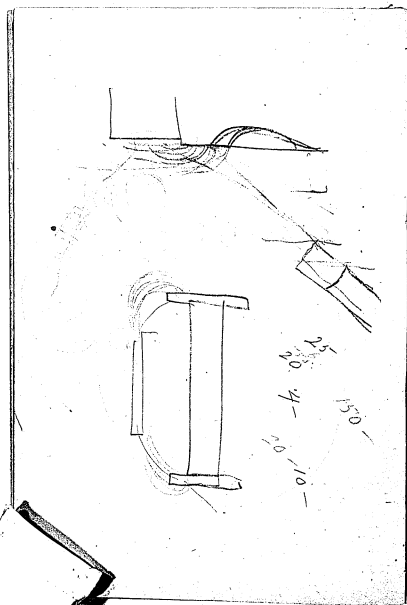


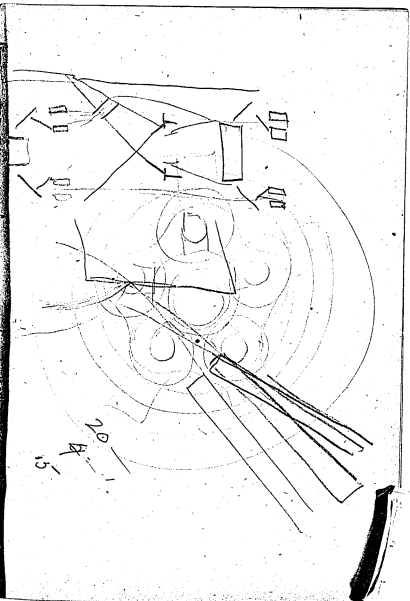
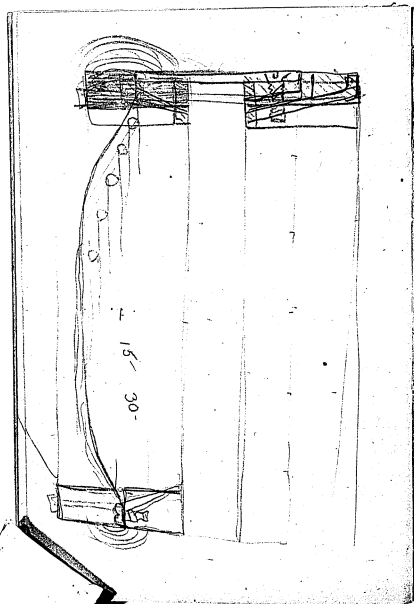


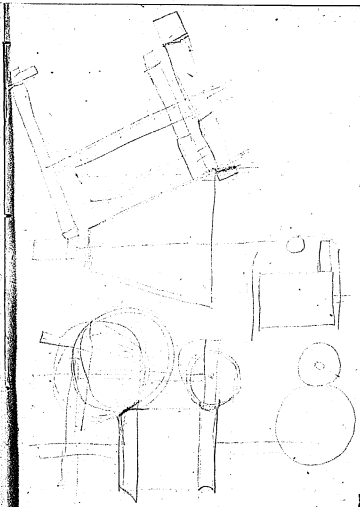


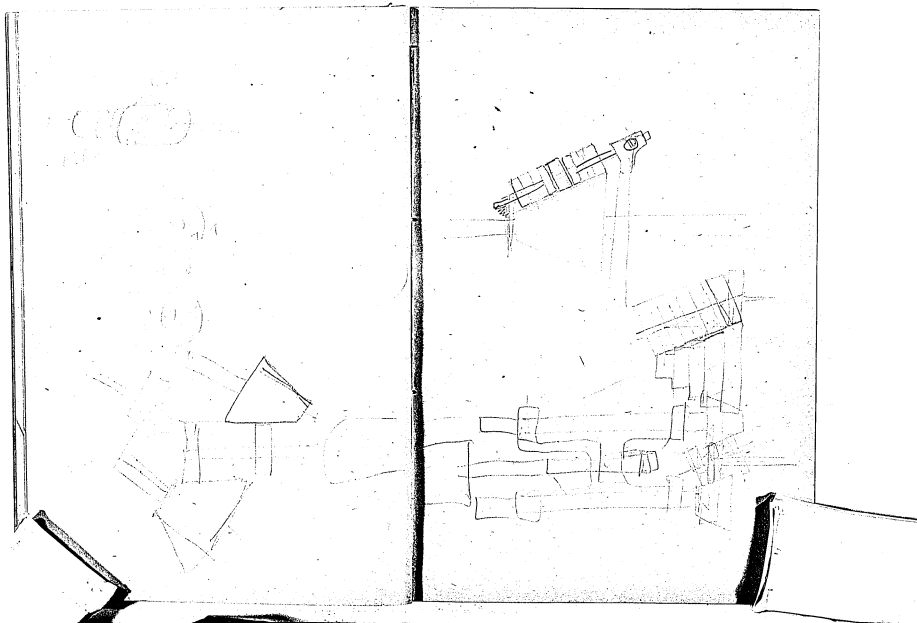


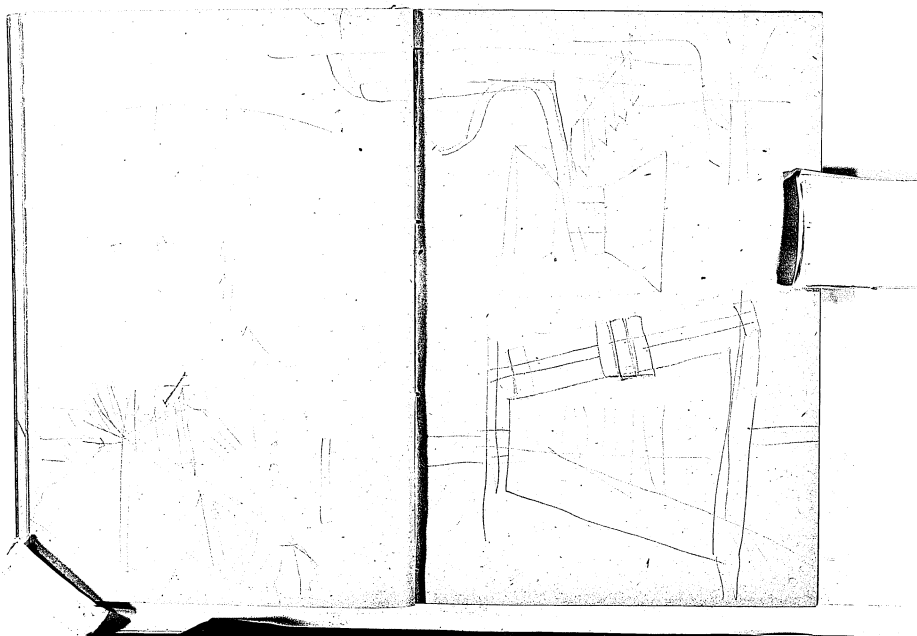


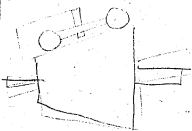
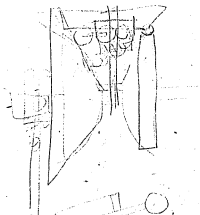
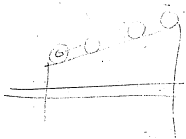


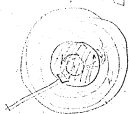
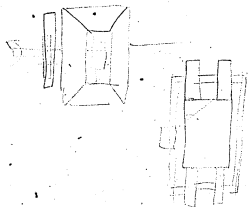


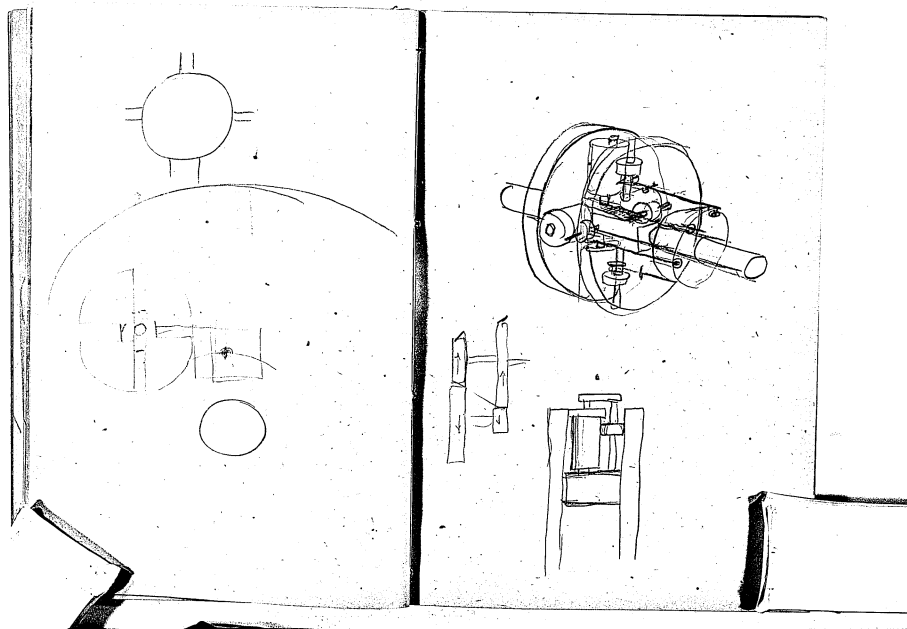




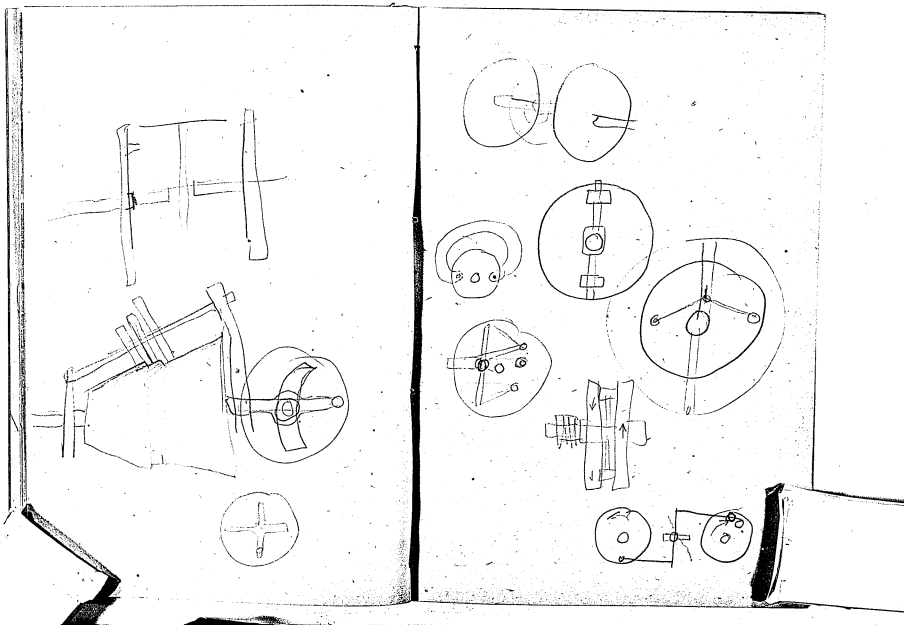


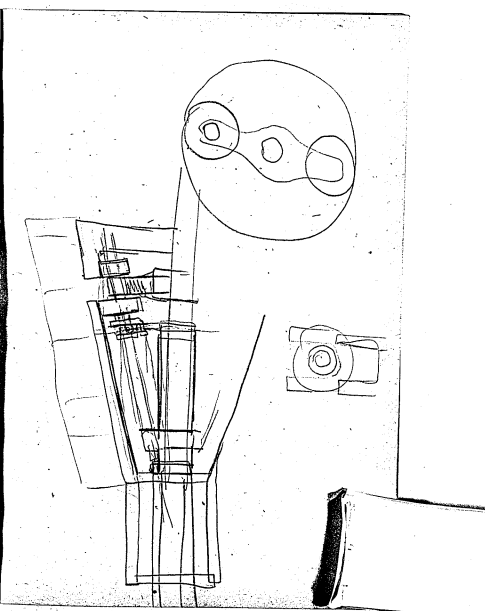
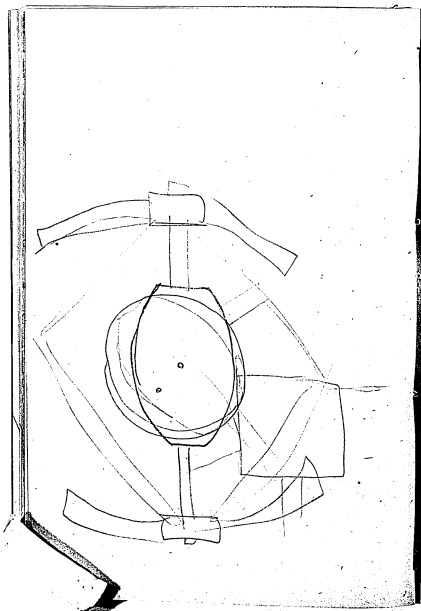


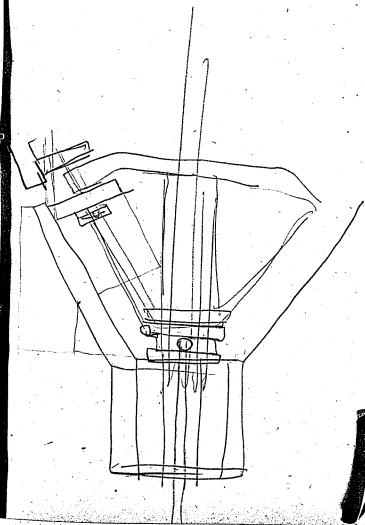
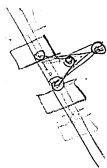


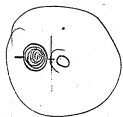
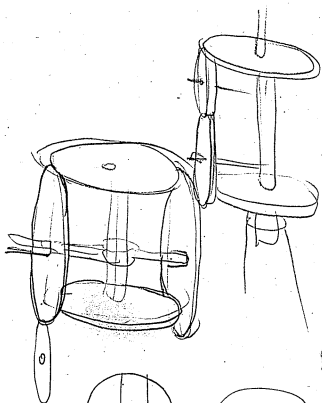


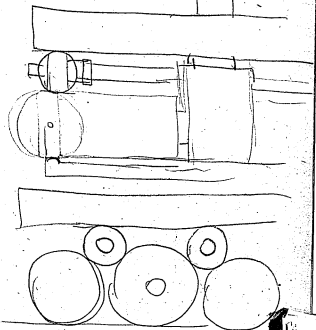
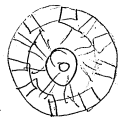
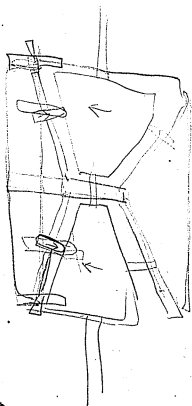


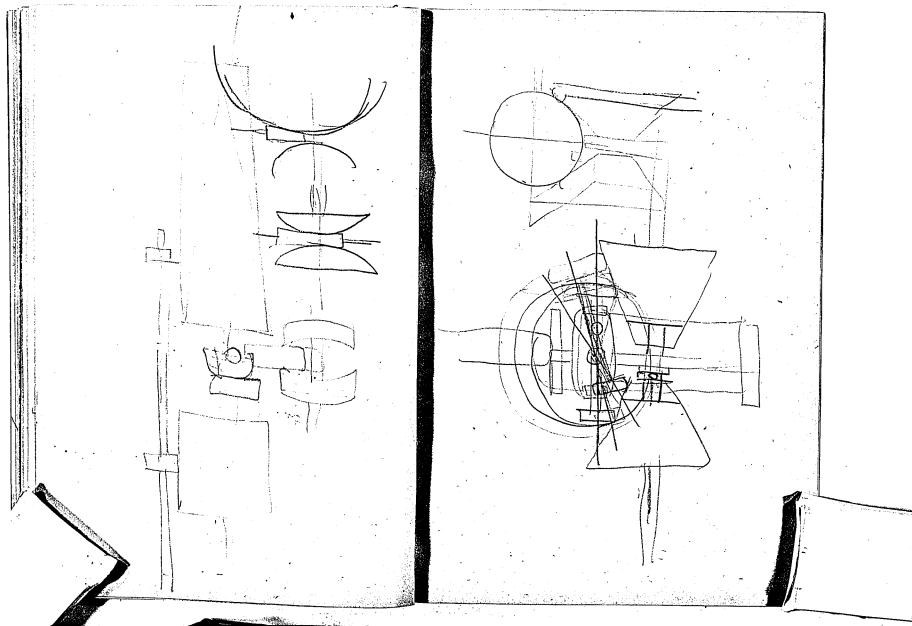


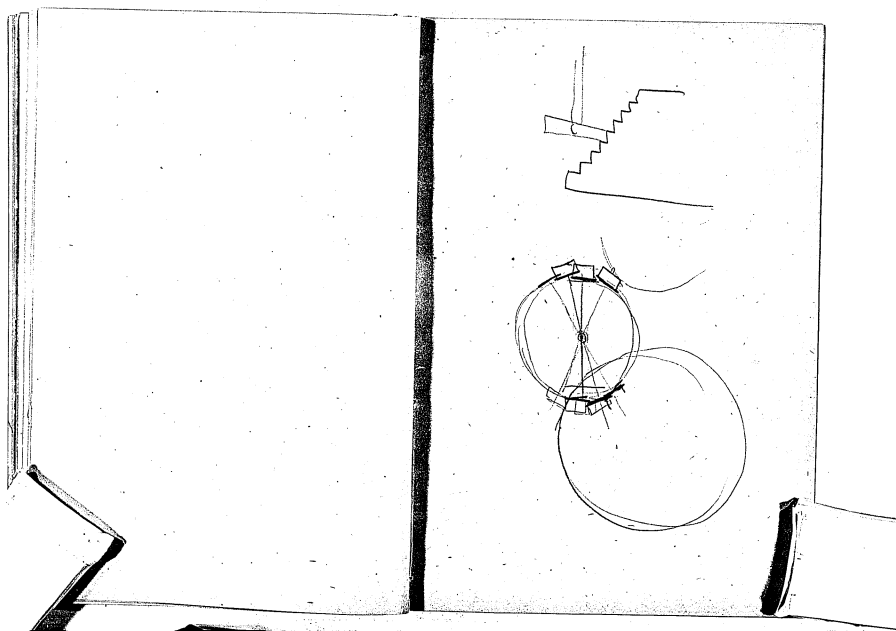


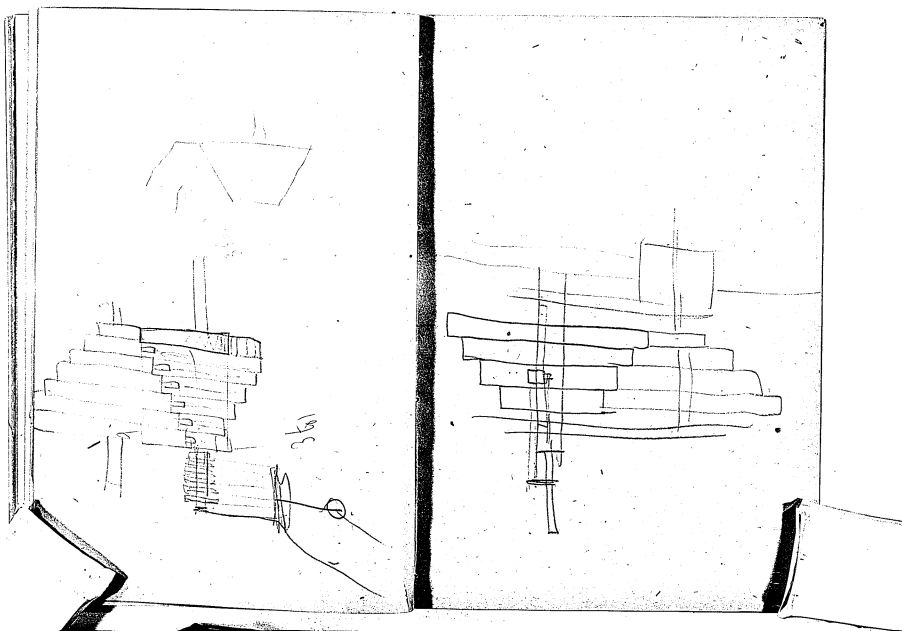




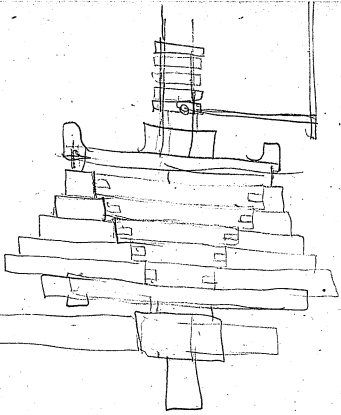
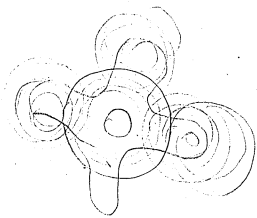


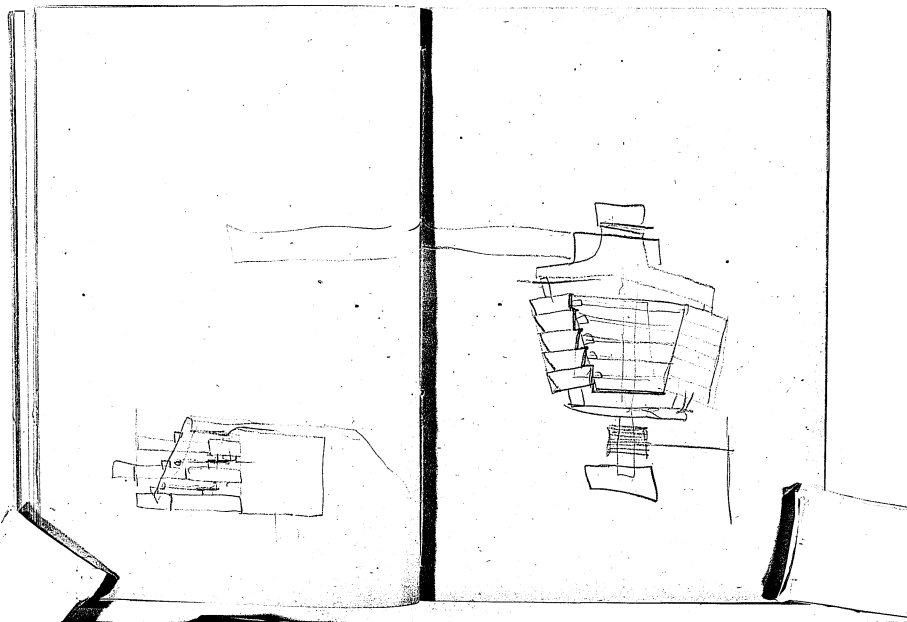


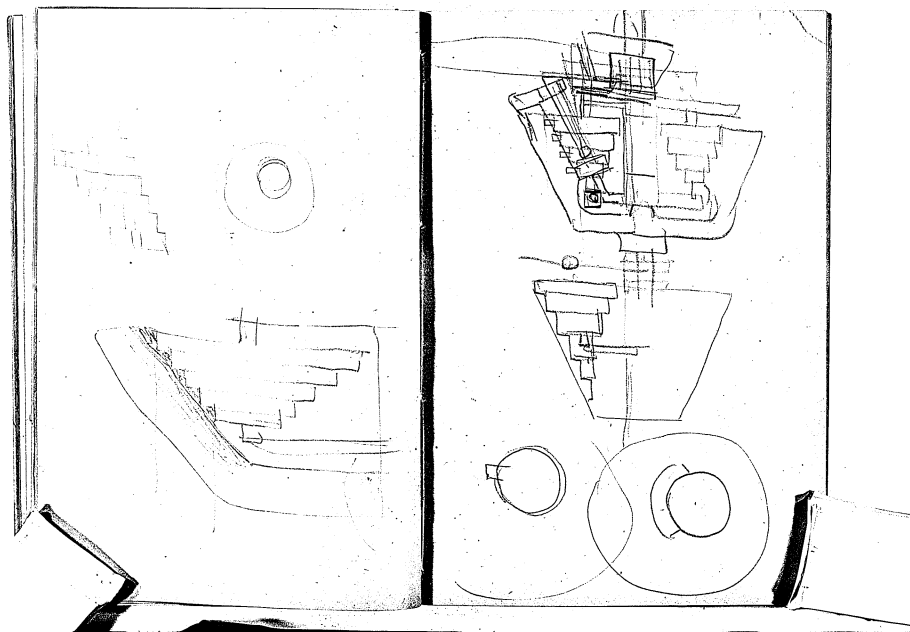


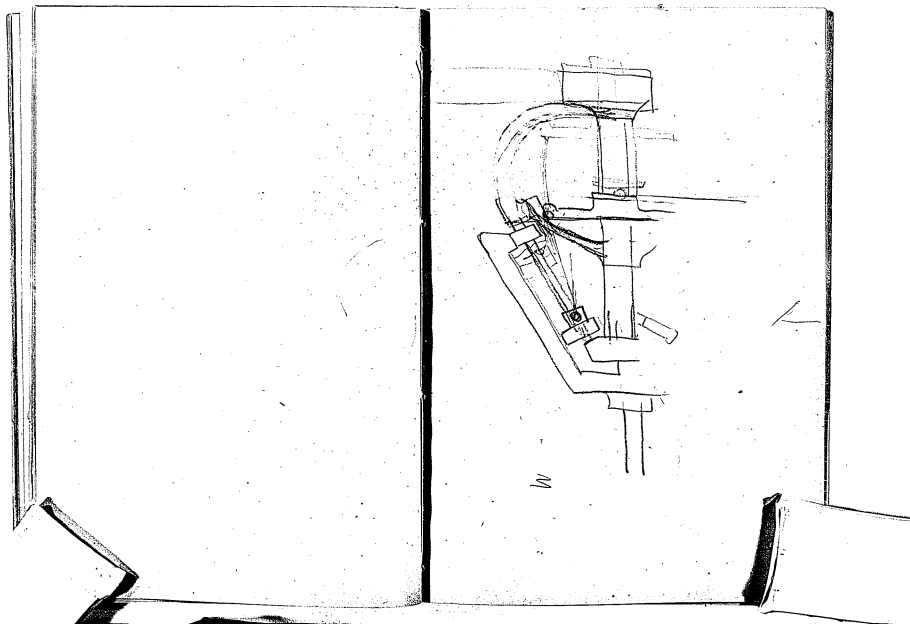


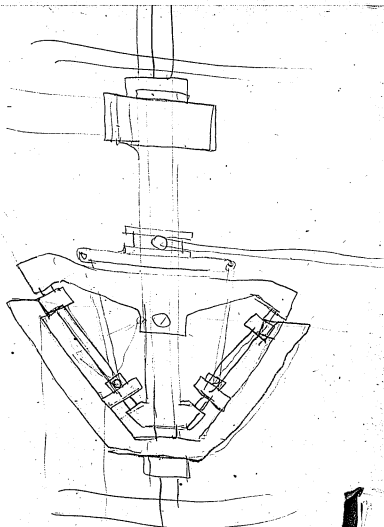
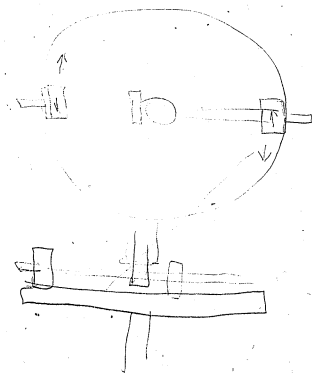


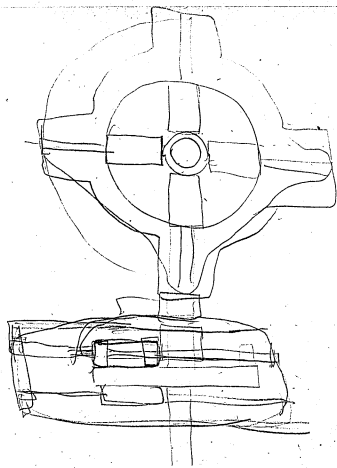
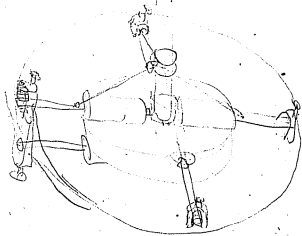
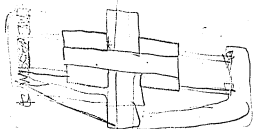


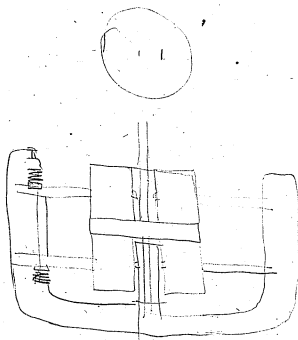
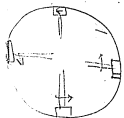


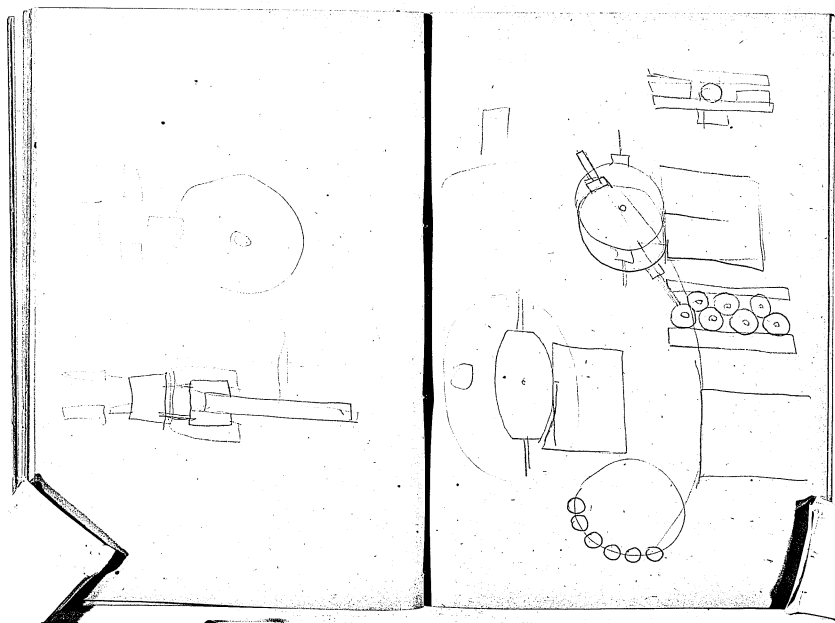




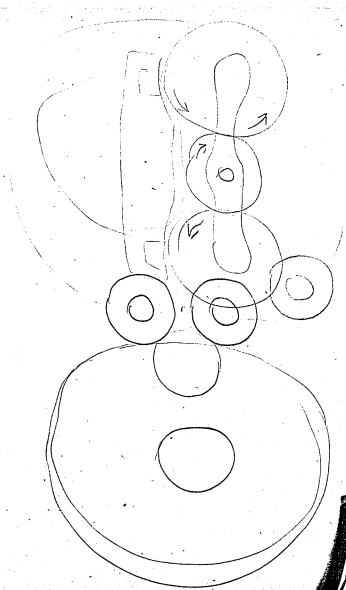
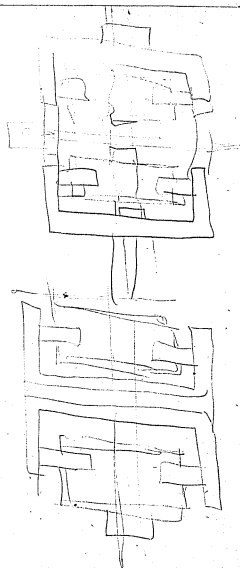


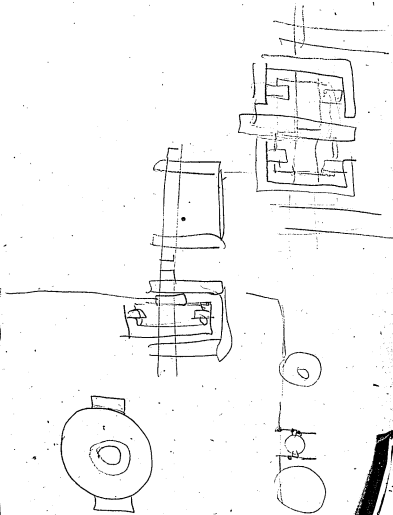
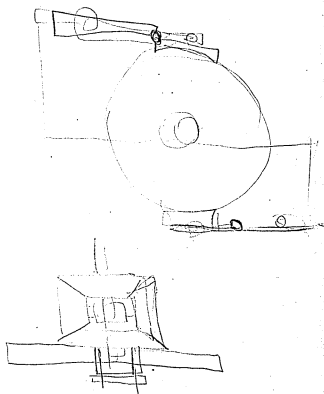




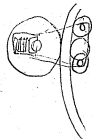
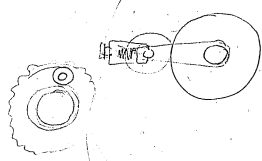
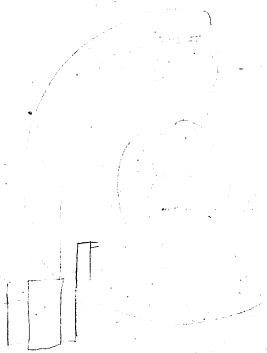




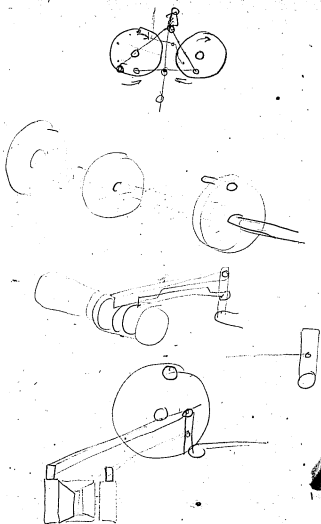


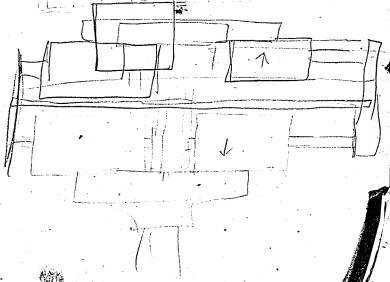
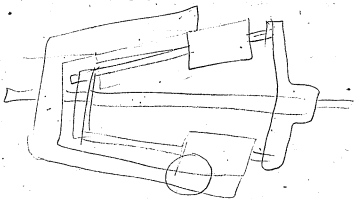
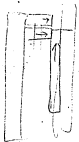
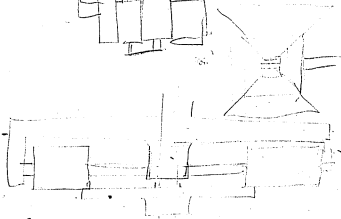
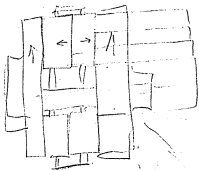


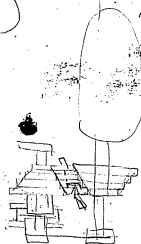
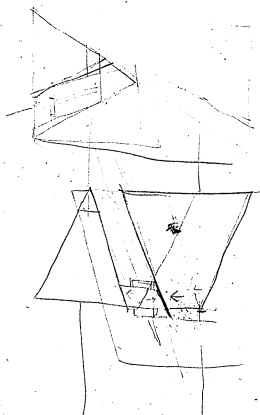


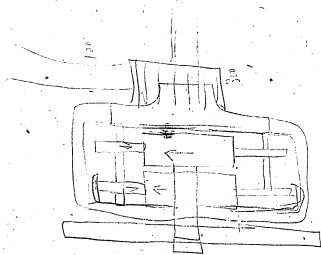
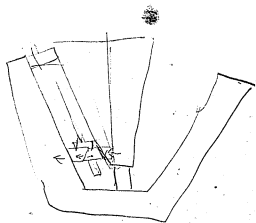


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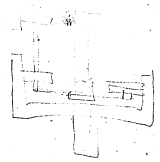
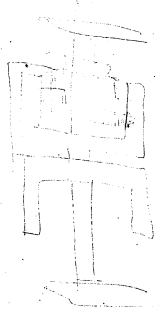


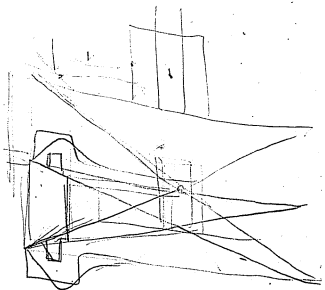




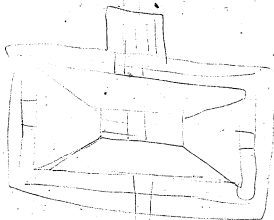


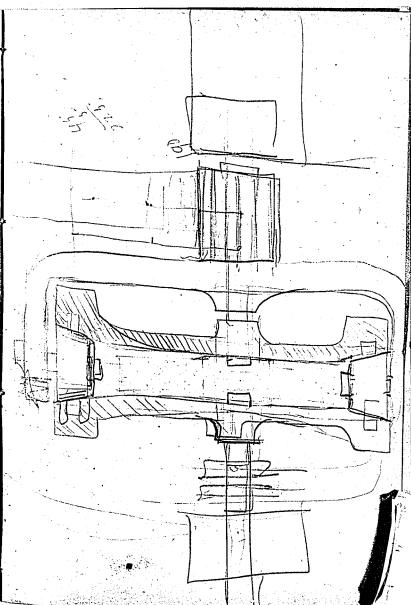
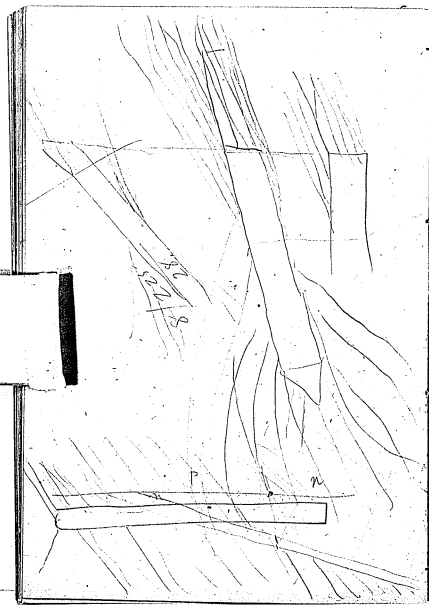


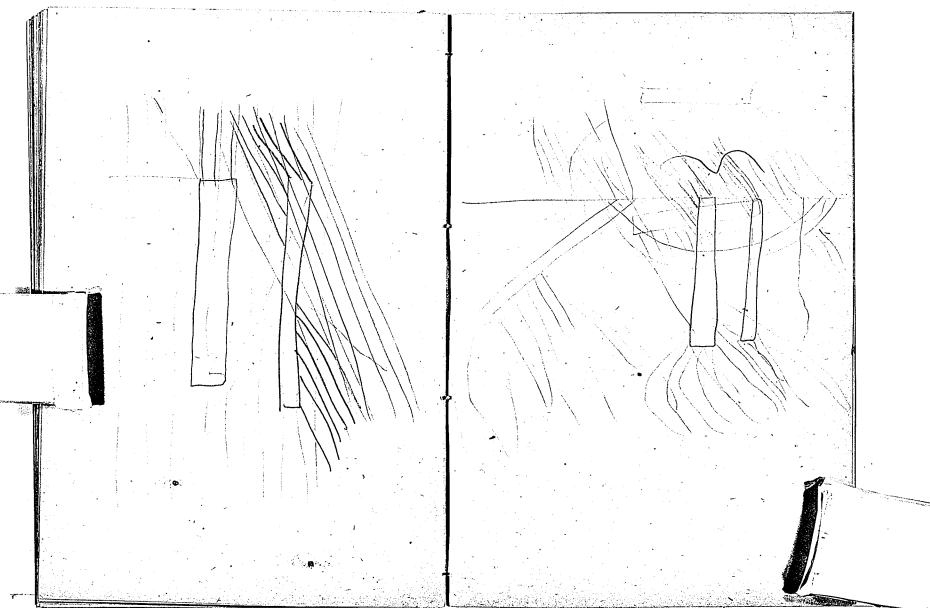


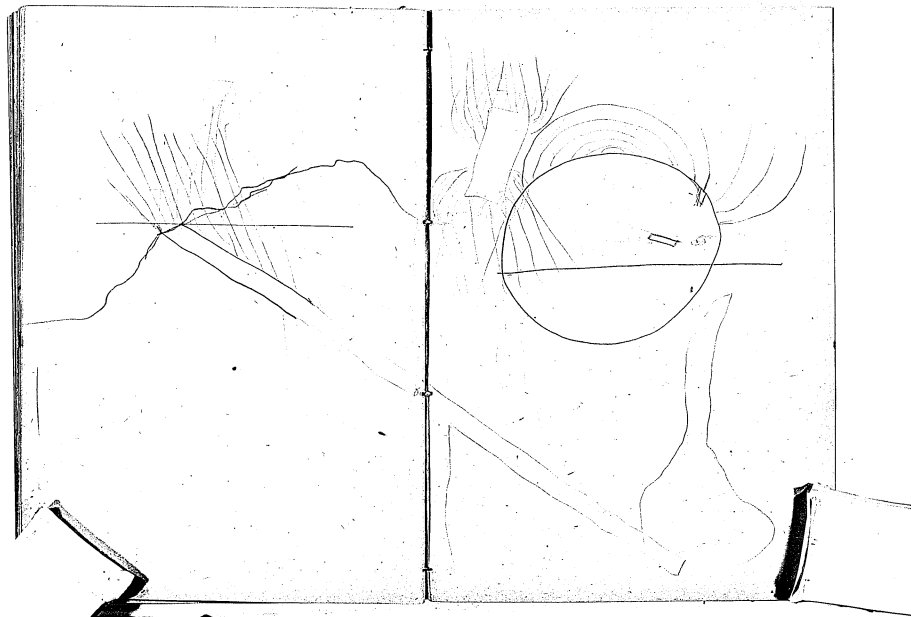


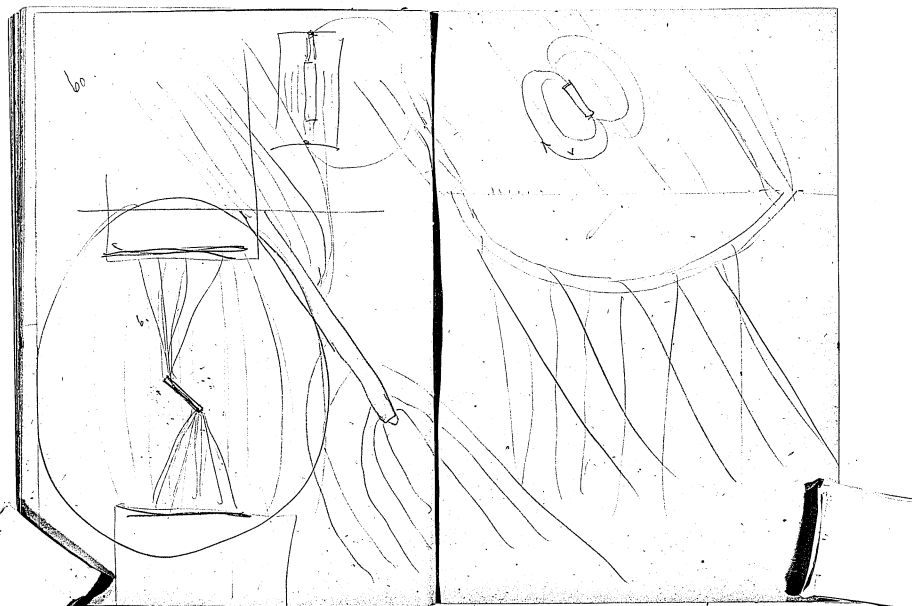
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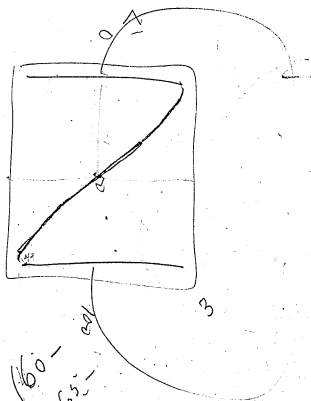




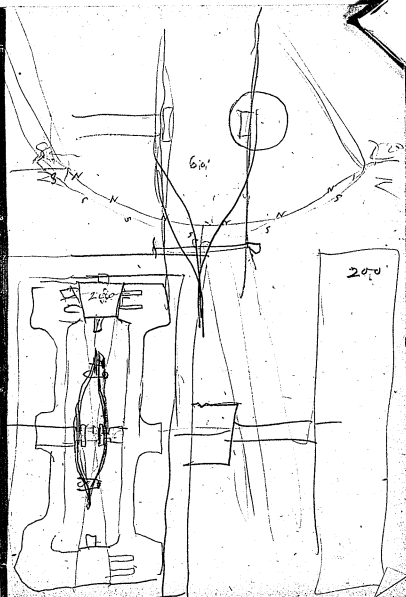


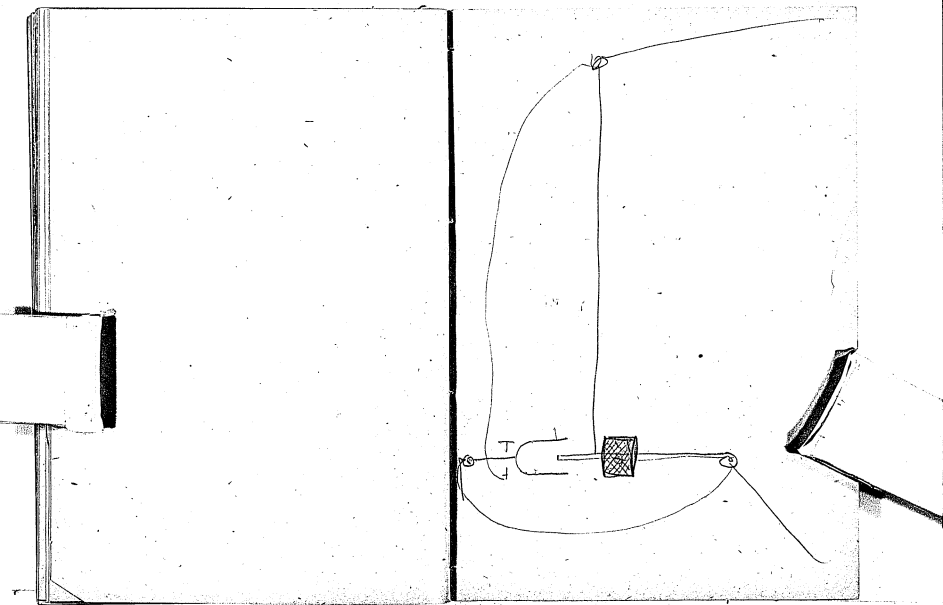




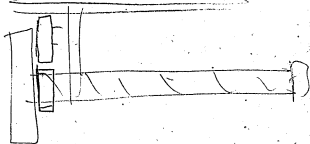
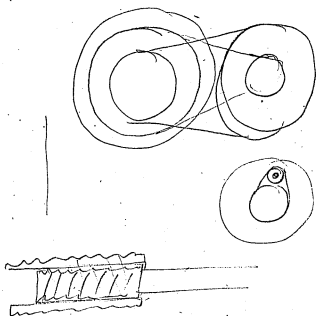
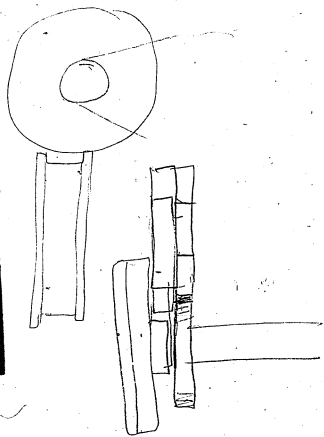


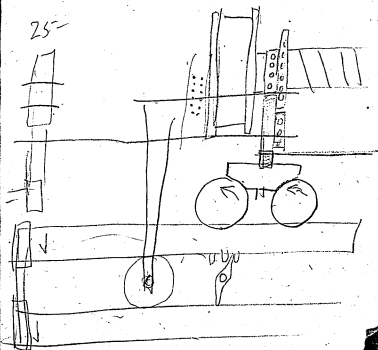
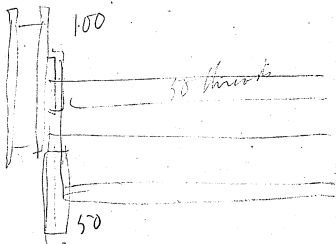
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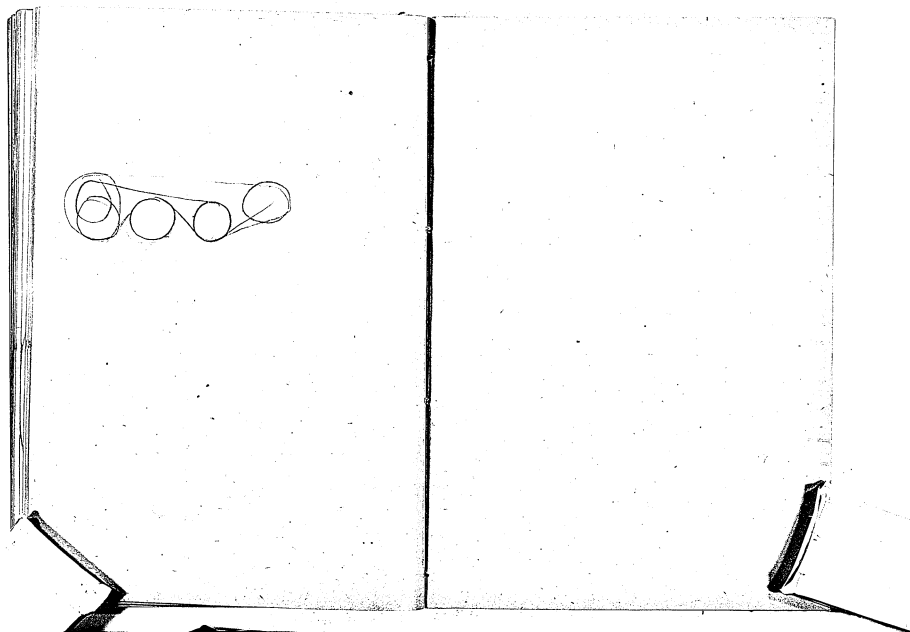


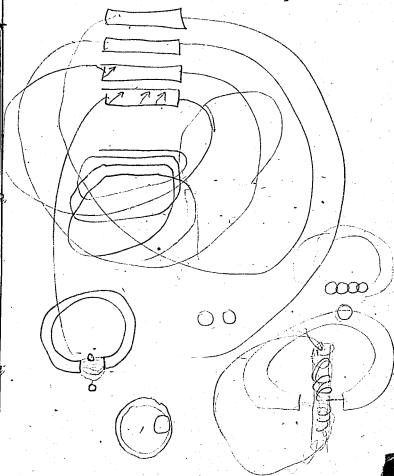
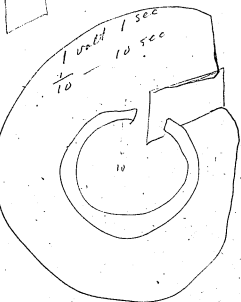
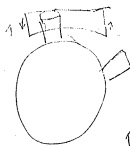




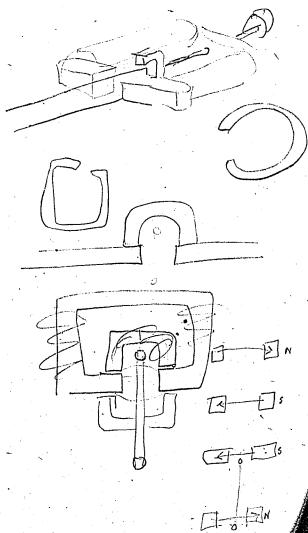
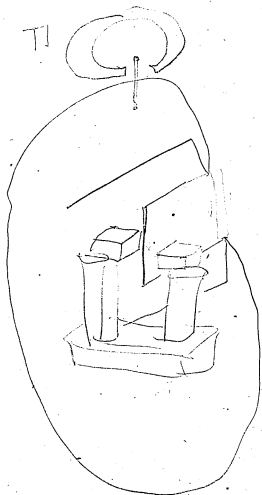


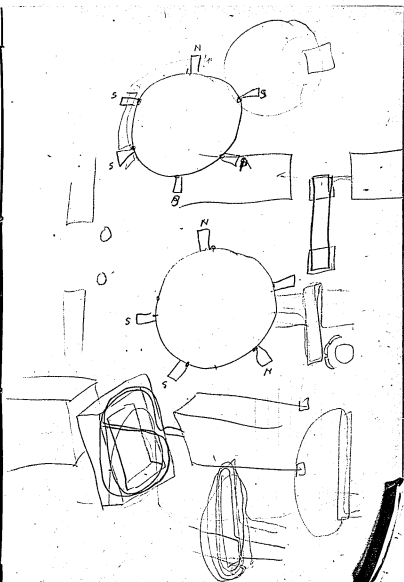
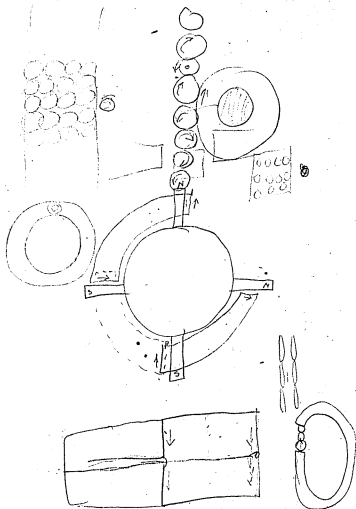


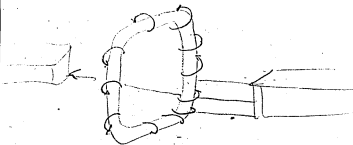
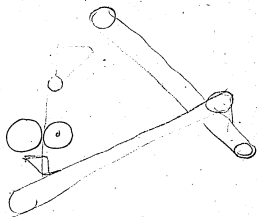
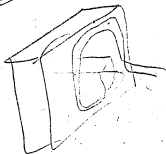
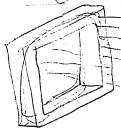
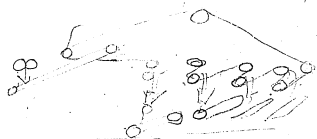




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Day

|                    |      |      |
|--------------------|------|------|
| 2 Messrs           | 3.00 | 3.00 |
| 2 Helpers          | 3.00 | 3.00 |
| 3 - Chemists       | 3.75 |      |
| 1 fassman          | 1.75 | 1.75 |
| 1 pig boy          | .80  | .80  |
| 1 brick boy        | .80  | .80  |
| 3. Loading boys    | 2.40 | 2.40 |
| 1 Furnace fire boy | .80  | .80  |
| 3 " men            | 4.00 | 4.00 |
| 1 Cast + man       | 2.50 | 2.50 |
| 2 Laborers         | 2.50 | 2.50 |
| 2 loaders          | 2.75 | 2.75 |

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|      |      |
|------|------|
| 3.00 | 3.00 |
| 2.50 | 175  |
| 175  | 175  |
| 50   | .50  |
| 250  |      |
| 400  | 400  |
| 375  | 3.75 |
| 250  | 2.50 |

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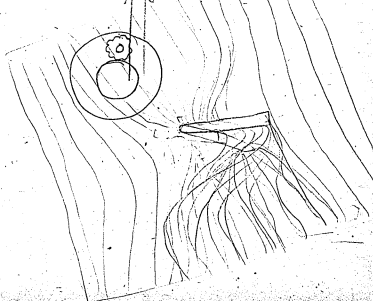
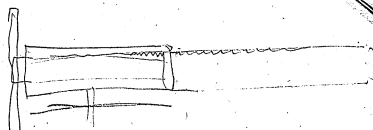
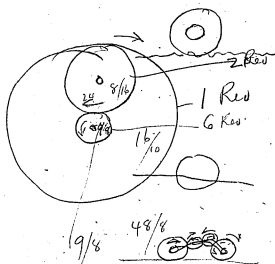
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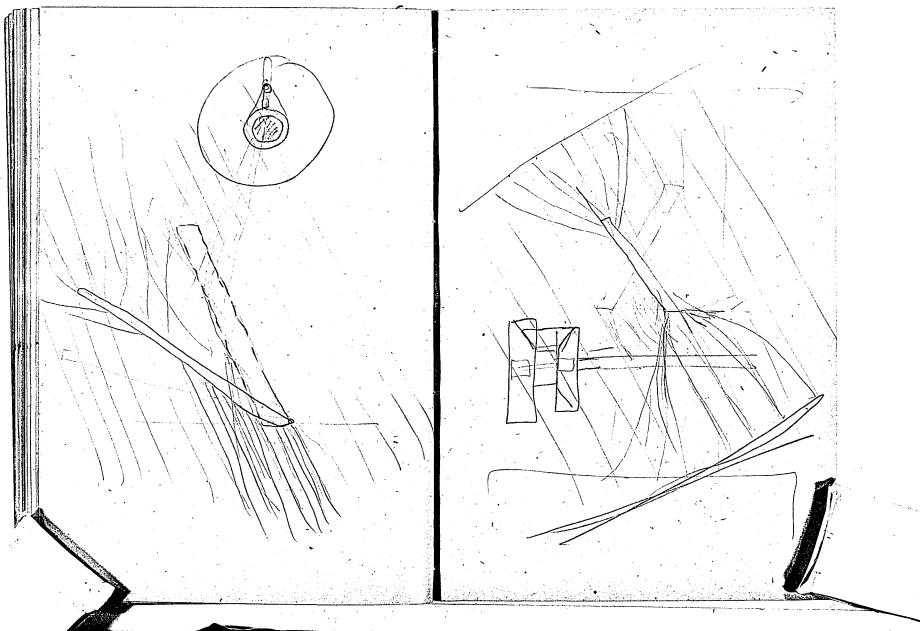
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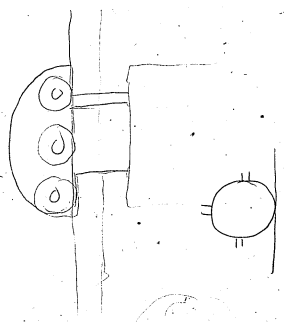
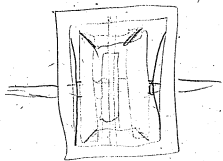
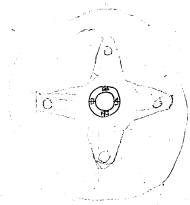
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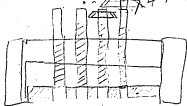
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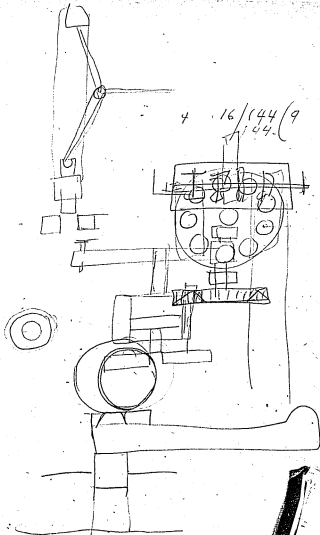
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$$\begin{array}{r}
 25 \\
 100 \\
 \hline
 60000 \\
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 \hline
 96000 \\
 345600 \\
 \hline
 441600 \\
 16 \text{ km}
 \end{array}$$


$$\begin{array}{r}
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 543600
 \end{array}$$

2000 9 200 15 33

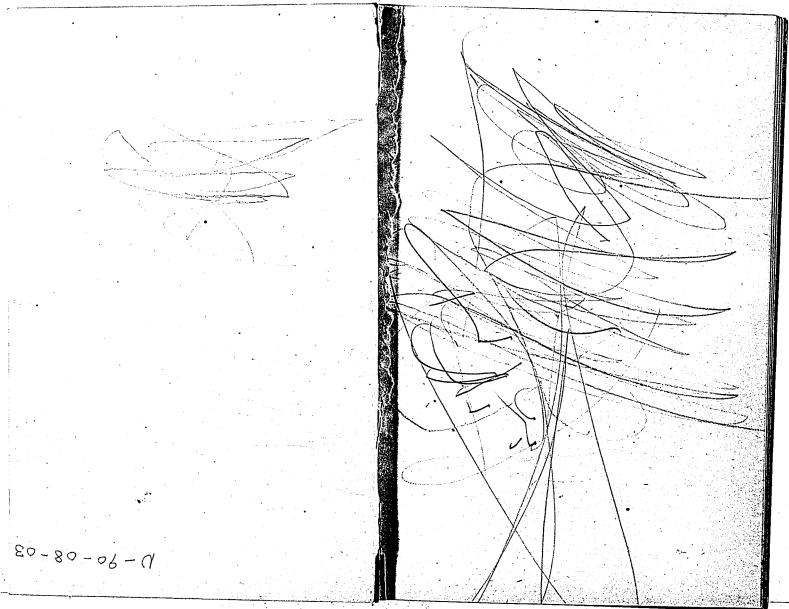


$$\begin{array}{r}
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 \hline
 1720 \ 28800 \\
 116000 \\
 \hline
 128000 \\
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 \end{array}$$


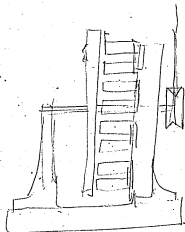
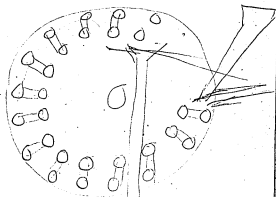
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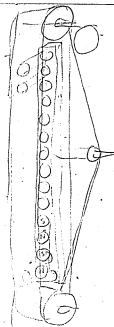
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THE FOLLOWING PAGES WERE FILMED FROM  
THE BACK END FORWARD.]

11-90-08-03



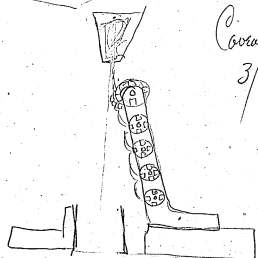
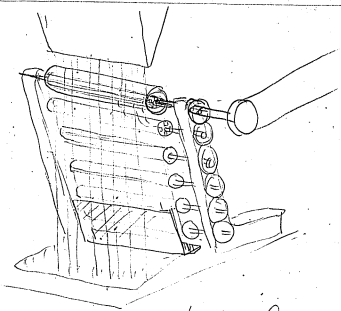
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Aug 8/90



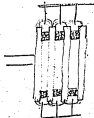
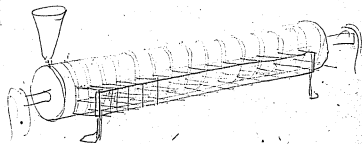
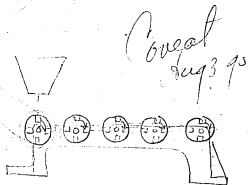


Coccat  
Aug 3/90

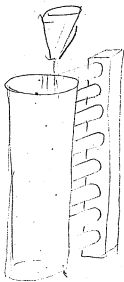




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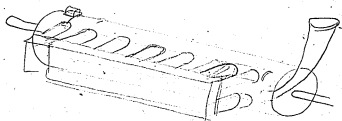


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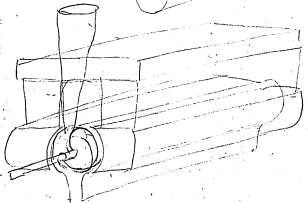
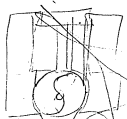


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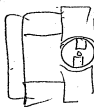
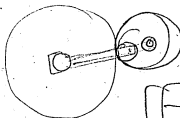
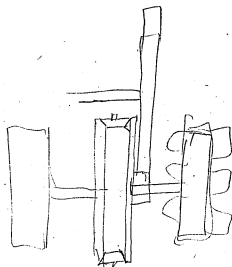


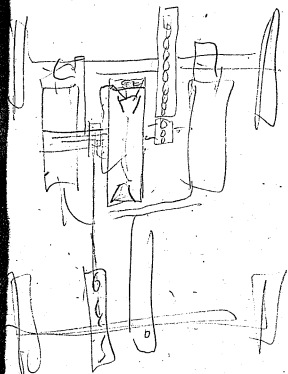
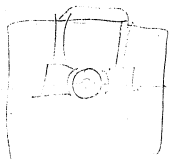
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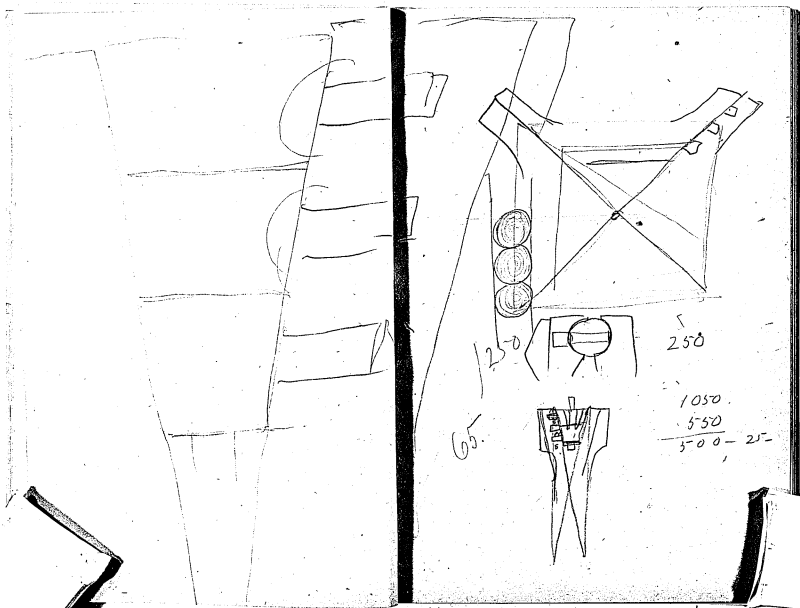


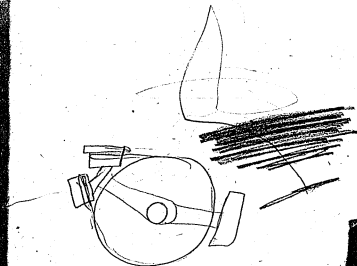
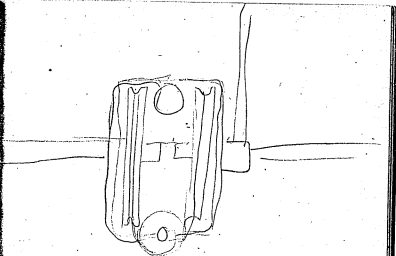
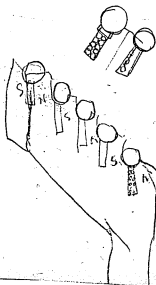
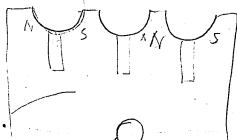
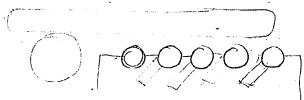
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C 21





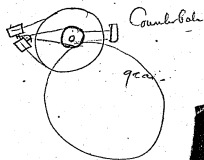
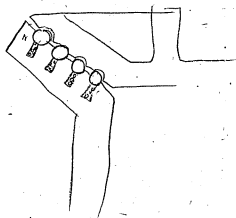




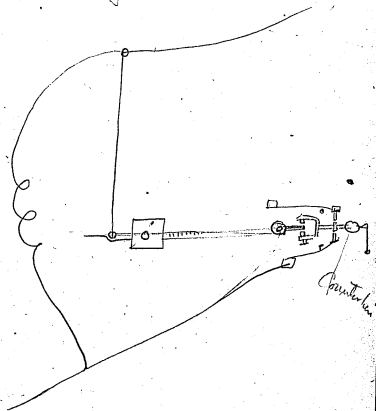


Conrad of Aug 5th.

Magnetized Belt

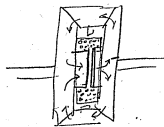
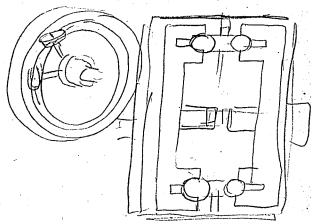


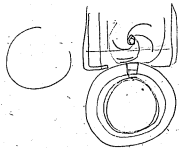
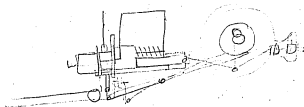
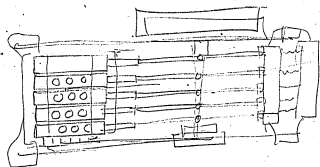
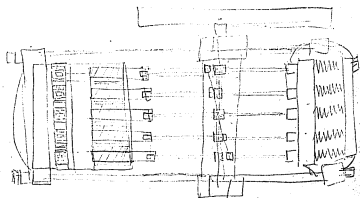
Alternating Motor, Revised Aug 5/90

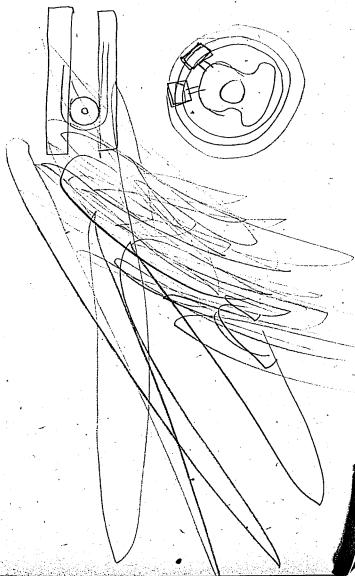


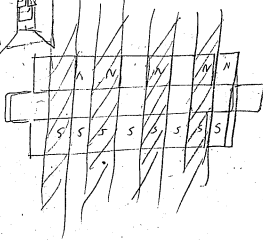
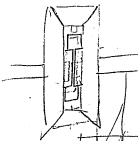
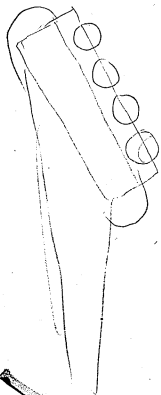
Case at Aug 5-90

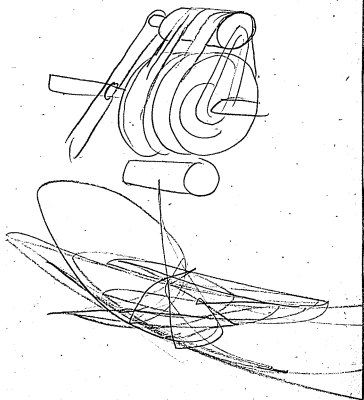
2 in di steel belt magnet

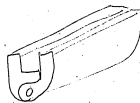
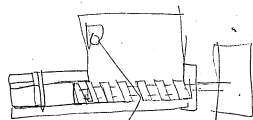




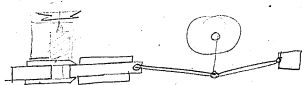
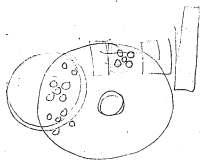
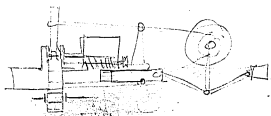




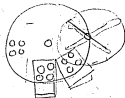
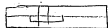


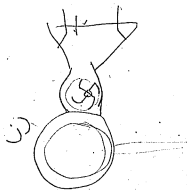
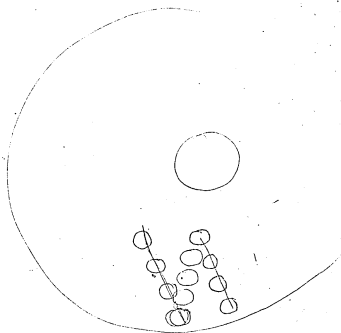






36-





**Notebook, N-90-11-07.1**

This notebook covers the period November 1890-January 1892. All entries are by Edison. The book contains a list of chemical compounds, nos. 1-28. There are also notes regarding carbonization tests for compounds; these entries are possibly related to Edison's filament experiments. The front cover is labeled "Book #6007 [607?] Nov 7/90." The spine is labeled "115." The book contains 71 numbered pages.

Blank pages not filmed: 36-43, 54-67, 70-71.

N-90-11-07

X E-172

|        |                       |
|--------|-----------------------|
| No 1 ✓ | Oil of Orange         |
| 2 ✓    | " " Spike             |
| 3 ✓    | " " Angelica          |
| 4 ✓    | " " Sandelwood        |
| 5 ✓    | " " Papaver           |
| 6 ✓    | " " Pennyroyal        |
| 7 ✓    | " " Laurel            |
| 8 ✓    | " " Pear              |
| 9 ✓    | " " Horsemint ✓       |
| 10 ✓   | " " Fennel Seed ✓     |
| 11 ✓   | " " Cloves ✓          |
| 12 ✓   | " " Juniper Berries ✓ |
| 13 ✓   | " " Riga Balsam ✓     |
| 14 ✓   | " " Cajapat ✓         |
| 15 ✓   | " " Carraway ✓        |
| 16 ✓   | " " Mustard ✓         |

No 17 ✓

18 ✓

19 ✓

20 ✓

21 ✓

22 ✓

23

24 ✓

25

26 ✓

27 ✓

28

29 ✓

Oil of Celery ✓

" *Patty Cloves* ✓ <sup>causa</sup>

" Myrbane ✓

" Janay ✓

" Wormwood ✓

" Cedrat. ✓

" Spruce ✓

" Peppermint ✓

" *Patty Jasmine* ✓

" Cedar Wood ✓ <sup>Feb</sup>

" Citronella ✓

" Balsam Fla. ✓ <sup>45</sup>

" Croton ✓

- 5
- |         |                      |
|---------|----------------------|
| No 30 ✓ | Oil of Birch ✓       |
| 31      | " Louaq root. ✓      |
| 32 ✓    | " Bitter Almonds ✓   |
| 33 ✓    | " Sweet Almonds ✓    |
| 34 ✓    | " Mace ✓             |
| 35 ✓    | " Pumpkin Seed ✓     |
| 36 ✓    | " Thyme Red-A. ✓     |
| 37 ✓    | " Star Anise ✓       |
| 38 ✓    | " Pinus Sylvestris ✓ |
| 39 ✓    | " Matico. ✓          |
| 40 ✓    | " Amber. ✓           |
| 41 ✓    | " Lemon ✓            |
| 42      | " Stoneblack ✓       |
| 43 ✓    | " Lavender Garden ✓  |

No 44

45

46 ✓

47 ✓

48 ✓

49 ✓

50 ✓

51 ✓

52 ✓

53 ✓

54 ✓

55 ✓

56

57

Oil of Fatty Violet <sup>7</sup>

" Hemlock

" Sassafras

" Savine

" Copaliba

" Bergamot

" Fireweed

" Pimento berries

" Male Fern

" Nutmeg

" Rosemary

" Wintergreen

" Bay

" Egeron

|         |                            |   |
|---------|----------------------------|---|
| No 58 ✓ | Oil of Olive ✓             | 9 |
| 59      | " Liquor ✓                 |   |
| 60 ✓    | " Origanum ✓               |   |
| 61 ✓    | " Valerian ✓               |   |
| 62      | " <u>Extract</u> Hemlock ✓ |   |
| 63      | " Apple ✓                  |   |
| 64 ✓    | " Ginger ✓                 |   |
| 65      | " Parsley ✓                |   |
| 66 ✓    | " Ergot ✓                  |   |
| 67      | " Stillingia ✓             |   |
| 68 ✓    | " Dill ✓                   |   |
| 69      | " Sage ✓                   |   |
| 70      | " Lobelia ✓                |   |
| 71      | " Canada ✓                 |   |
| 72      | " Comm. Seed ✓             |   |
| 73 ✓    | " Rhue ✓                   |   |
| 74 ✓    | " Coriander ✓              |   |



No 75

76

77

78

79

80

81

82

83

84

85

86

87

88

89

90

91

Oil of Horseradish

" Golden Rod

" Verbena

" Tinct. Ambrgruit

Aldehyde

Para Aldehyde

Terebene

Xylone

Oil of Linette

Acetone

Oil of Steubane

" Cinnamon

" Melissa

" Capsicum

" Chamamile

" Cardomon

" Beechwood

- |       |                                     |
|-------|-------------------------------------|
| 1092. | oil of Paraffin oil <sup>13</sup> ✓ |
| 93    | Creosote. ✓ X                       |
| 94#   | Sulpho-Vinsum acid                  |
| 95    | Butyric Ether                       |
| 96-   | Valerianic acid                     |
| 97    | Monochlorophenol ✓                  |
| 98    | Cresylic Acid ✓                     |
| 99    | Sulfo carbonyl amine Allyl.         |
| 100   | Anthrac Benzine                     |
| 101   | Amylum hydrate.                     |
| 102   | Eucalyptol                          |
| 103   | Styrone                             |
| 104   | Tyrolbauben Saure ✓                 |
| 105-  | Valeraldehyde ✓                     |
| 106.  | Methyl oxyd-acetic                  |
| 107   | Salicylox acid                      |

- 15
- No 108    Chemolin  
109        Sebacylic Ether  
110  
111  
112    ~~Naphthalene Alpha Mono bromat.~~  
113    ~~Naphthalene~~  
         ~~Naphthalene~~  
114    Dichloroacetic acid  
115    Undecylen Acid ✓  
116    Safrrol  
117    Amyl Iodide  
118    Propionic Acid  
119    Nitrosdimethylamin ✓  
120    Furfural  
121    ether Salicylic  
122    Carbohc Camphor  
123    Benzene Ether

- NO 124.      *amyl valerianate* ✓ - 17
- 125-      *Caprylic*      §
- 126      *Styrol*
- 127      *Carbon Tetrachloride*
- 128      *Valerianic Ether*
- 129      *Methyl Salicylate*
- 130      *Methoxy Benzoin*
- 131      *Toluene Ortho*
- 132      *Phenylhydrazin*
- 133-      *Oxalic Ether*
- 134      *Myrtol*
- 135-      *Monochloroethylen*
- 136      *Pyridin*
- 137      *Amyl Oxide Formic*
- 138      *Ethylamine*
- 139      *Amylum Chlorat*

- No 140 Amyl Nitrate
- 141 Hypan
- 142 Isopropyl Alcohol
- 143 Cymol
- 144 Cuminal
- 145 Abietic Pure
- 146 Butylsulfocyanat
- 147 Cumol
- 148
- 149 Aethylin Chlorohydrin<sup>et</sup>
- 150
- 151
- 152
- 153
- 154
- 155 Propyl Valerianic
- 156

No 157

- 158 Carbon Bisulphide  
159 Isobutylamine  
160 Pseudocumol  
161 Picolin  
162 Acetol  
163 Wood alcohol  
164 Allyl Sulfid  
165 Toluidin Meta  
166 Cinnidin-Isopropylamide Benzol  
167 Benzonitril  
168 Cinnamic acid Zimtsäure  
169 — Duplicated = (Isobutylamine)  
170 Butylchloral  
171 Isopropyl butyric  
172 Ethylpropion  
173 — Propyl formic  
176 — Nitrostyrol  
177 Isopropyl bromide  
173 Isobutylaldehyde  
174 Isobutyl bromide

## NO 178 ● Methylpropyl Ketone

- 179 Isobutylene Zinc  
 180 Amyl Propionic  
 181 Methylsulfonylacet  
 182 Isobutylisobutyrate  
 183 Isobutylacrylate  
 184 Amyl Ether  
 185 Butyl Nitrosium ✓  
 186 Propionitrile ✓  
 187 Isobutylacetate  
 188 Methylisobutyrate  
 189 Methyl Propionic  
 190 Di-methyl Acetic Carbonol  
 191 ✓ Isopropyl Alcohol *Diplo*  
 192 Isobutylbutyrate  
 193 Isobutylacetate *Diplo*

- No 194 Butyliso-formic 25  
195 Tetramethylammonium 10%  
196 Isobullersaine ?  
197 Di. Methyl Resorcin ✓  
198 Ethylmethylacetat ✓  
199 Amyl Rhodanat ✓  
200 Amyl Sulphat ✓  
201 Capronic Ether ✓  
202 Methyl Butyric ✓  
203 Pseudo Butyl. Alcohol ✓  
204 Diphenyl Methane ✓  
205 Iso Butyl Nitric ✓  
206 Diphenol ✓  
207 Amyl Mercaptan ✓  
208 Propionic Ether ✓  
209 Xyliden ✓  
210 Malonic Ether ✓



- NO 211 Trimethyl Carbonil ✓
- 212 Methylen Iodide ✓
- 213 Kresol-Meth ✓
- 214 Pyridin Chlorat, ✓
- 215 Isobutyl-Iodide ✓
- 216 Isobutylen-Bromide ✓
- 217 Benzyl Bromide ✓
- 218 Amyl Iodid - aus Mannit ✓
- 219 Isopropyl Iodide ✓
- 220 Ether Disulfuret ✓
- 221 Zimmtsaure-Alkohol.
- 222 Propylacetat, ✓
- 223 Propylpropionat ✓
- 224 Benzolmonochlorat ✓
- 225 Kresol-Ortho ✓
- 226 Ether Sulfuret,
- 227 Methylacet essig Ether

- 29
- No 228 Ethyl Trichloroacetat.  
229 Ethyl Dichloroacetate. ✓  
Methyl Jodat 2 no dat  
230 Methylcyanide  
231 Ether Pelargonic  
232 Amosaldohyde ✓  
233 Styron  
234 Propyl-Chlorat.  
235 Jod-Benzol  
236 Phthalylchlorid  
237 Salicylic Ether  
238 Methylamin  
239 Formamid  
240 Methyloxyd-Salicylic  
241 Dichlorohydrin  
242 Ether Rhodant Sulfocyanat.  
243 Methylamin 33%

- No 245 Xylol
- 246 Butyl Chlorat.
- 247 Carbamic Ether.
- 248 Nitromethan ✓
- 249 Butyron ✓
- 250 Nitropenthan ✓
- 251 Nitrosdiethylin ✓
- 252 Butyrim ✓
- 253
- 254 Aethylbenzol ✓
- 255 Propylaldehyd ✓
- 256 Benzophenon ✓
- 257 Aethylen Alcohol. ✓
- 258 Butyl Mercaptan
- 259
- 260 Propionitril ✓

- No 261. Toluolchlorat ✓
- 262 Octylic Alcohol ✓
- 263 Trimethylamin Hy - duplicate
- 264 Oenthal ✓
- 265 Diethylamin ✓
- 266
- 267 Nitro-oxylol Para ✓
- 268 Benzol ✓
- 269 Chloroform
- 270 glacial acetic acid
- 271 Benzene
- 272 • Naptha - also Tetraethylamin
- 273 Glycerin
- 274 Gasalene
- 275 Turpentine
- 276 Absolute Ethyl alcohol
- 277 Reg " "
- 278 Methyl Alcohol

W.P.X

H.P.S.

- 279 - Anthrac Benzene Apicol. 35  
280 Methyl Bichloride  
281 oil Cade-  
282 Nitrite Amyl.  
283 Benzoyl Chloride  
284 Xylof.  
285 Ammonia  
286 Kerosene  
287  
288 - Benzol Sulpho Chloride  
289 Pyro Galol Dimethyl Ether

Anilin Brun - Bismark -  
pharab considerable -  
little but more than Indulin  
dont melt like Indulin  
has considerable red ash  
& pieces quartz = def Comb  
Charcoal. —

Nitros - dimethyl anilin  
Hydrochlorat, to koke trace -  
boils & pharab some but  
Charcoal shiny -

Anilin Blue Cant. red (Kund)  
full arsenic - Dont melt

47

Tropaeolin - Red - can't read  
but - Don't phans scarcely  
any think aromatic in it.

---

Naphthalene alpha di  
Chlorate d - Melts easy  
may do as sublimat  
by heat.

---

Anthracolin - melts  
don't phans much C<sub>10</sub>s  
to very hard - may do as  
heat under great deal  
sublimat -

Hoffman - Viol. l. dont phase  
as much as Indulin -  
insects on 1st plant, by slow  
clyn small puff - as this  
is very soluble may be  
good -

Andin Red Congo - dont  
swell or stick together. Carbon  
dull - with best for ash -  
ash great, what is left soils

Andin Gelb - Marl -  
Explosive -

Andin Rott. fuchse. as some fire  
melts very well on crystals, by slow  
no fars - as but hot sublimate  
por - Carbon - ash -

49  
Negrosen Nat Water  
dont swell at all - ash

Andin Blue Methylen  
fars if quick, seems melt  
at bottom but slow clyn  
thinks it would be OK

Andin Gelb - Melts,  
fars porous, if done  
quicks -

Santonin Melts -  
no Carbon -



Amber Green Methyl  
for arsenic - melts fairly  
easy - practically no fash  
contracts to saled Carbon  
good binder of no ash  
or Dalm or gum -

Amber Yellow T.  
dust puff in slightest Carbon  
hard - pblly lot ash - full ash.

Amber Sulphate. Water  
goes off then Carbon  
Saled Carbon - not very strong  
pblly cause volting beats.

Amber - Rose Royal - 57  
dust melt, Violet sublimate -  
pblly lots ash - testing for it -  
no ash -

Amber Red Ruby Orange  
ng - fash horrible by ash.

Amber Blue - reddish -  
dust puff or melt, test  
for ash - full ash, horrible

Amber Green -  
Cant make out name  
mark + X - dust melt or puff -

fluorescein - melts, good binder  
for 20 -

---

Andrin Acetate semi melt  
good binder for No 20 -

---

Jan'y 24 1892.  
Expts on residues of files —

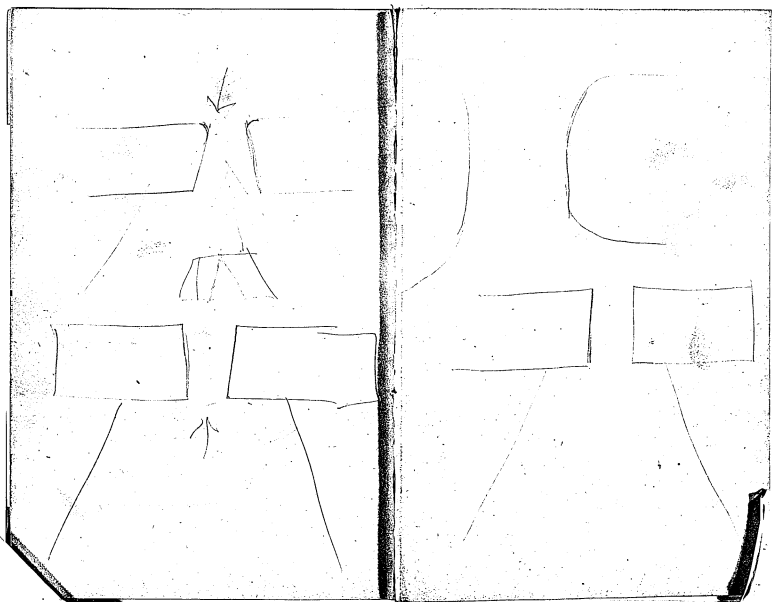
69

Citric acid in small Evap dish on sand bath  
scarcely any residue

Spalls and usually all Volatile  
residue black not Sal Chloro but  
Sal ~~Cl~~ dipi — about 2 pct Residue  
left —

Leaf shellac by residue —  
but appears insoluble — etc —

Bay Wax.



**Notebook, N-90-11-07.2**

This notebook, which covers the period November 1890-September 1892, is a continuation of N-90-11-07.1. All entries are by Edison. At the beginning of the book are notes regarding some of the 289 compounds from the preceding book, tested as solvents for syrian asphalt. Also included are notes relating to carbonization tests and to squirted filament experiments, as well as drawings and calculations, probably pertaining to ore milling. The front cover is labeled "Book #6008 [608?] Nov 7/90." The pages are unnumbered. Approximately 100 pages have been used.

12-90-11-07.2

$$\begin{array}{r} 225 \\ \hline 45 \end{array} \frac{200}{800}$$

$$\begin{array}{r} 200 \\ 475 \\ \hline 700 \\ 075 \end{array}$$

$$\begin{array}{r} 1075 \\ \hline 1300 \\ \hline 2600 \end{array}$$

$$\begin{array}{r} 450 \\ \hline 350 \end{array} \frac{300}{800}$$

33.

7-  
$$\begin{array}{|c|} \hline 4 \\ \hline 2 \\ \hline 1 \\ \hline \end{array}$$

$\frac{1}{2}$

$\frac{1}{4}$

$\frac{22}{13}$

34

360

13

14

14

280

8

275

400

490

101

102

103

104

105

106

107

108

109

110

111

112

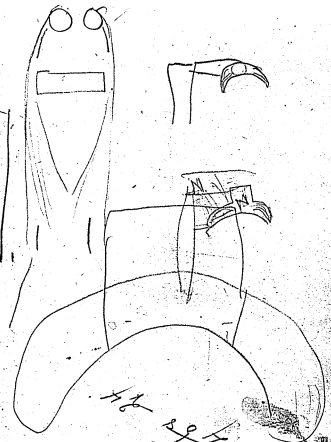
113

114

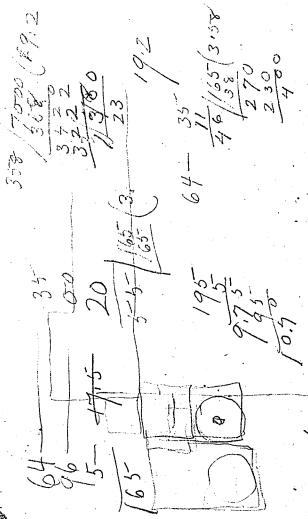
115

116

117



$$\begin{array}{r} 48 \\ \hline 28 \\ \hline 89 \\ \hline 17 \end{array}$$



Oxidized Linseed Oil. Solvents.

182. Isobutylacetat. Sem. solvent  
 disintegrates it but only 1/2 apparently  
 dissolved.

213. Kresol Meta\_ swells it up  
 very much -

221 Zimitsame - Alcohol, swells  
 it very much cant say it dissolves  
 as whole of liquid was absorbed  
 added more -

219 - Isopropyl Iodide - dissolves  
 wholly -

145 - Anilin swells & partially  
dissolves it -

135 = I think  $\frac{1}{3}$  to  $\frac{1}{2}$  dissolves

62. Extr. Hemlock don't dissolve  
turns it white.

132 Phenylhydrazin - perfect  
Solvent.

149 - Ethylchlorohydrin dissolves  
fair + disintegrates.

174 Isobutylbromide, perfect  
Solvent with very slight  
Sediment.



159 - Isobutylamine perfect  
solvent.

113. Neurin Lösung - good  
solvent.

244 - Methylamine 33% - dissolves  
but pblly combination = dirty -

245 - Xylenol - swells it greatly  
absorbed all solution cant say  
if any dissolved put more  
solution in -

---

246 - Butyl Chlorat. swelled +  
salubim all gone put more in

251 - Nitrodiethylben. swelled  
liquid gone - put more in

225 Kresol - Ortho - swelled  
liquid absorbed put some in

234 - Styrene - swelled  
dont think it has dissolved  
any -

233 Anisaldehyde - swelled  
liquid absorbed cant say it  
dissolved any - put more in -

237 - phly. (ch. m. k. swell)  
absorbed - didn't put any more  
w -

268. Benzal - good solvent  
for 80 pct of it which  
sediment appears to be  
unchanged luff + dirt.

272 Naptha - good solvent  
for 80 pct. - about same as  
Benzal -

160 - Pseudo animal - swelled  
& made light yellow dent  
think any dissolved -

270 glacial acetic acid  
 $\frac{1}{2}$  solvent - dissolving grades

276 - Absolute alcohol.  
nearly dissolves it - put fresh  
in + about  $\frac{1}{4}$  clear dissolve  
probably whole would  
dissolve with plenty liquid  
+ time. - This mixed with  
Naphtha or Benzol may do  
the biz =

166 - Cumidin - Isopropylamine Benzol  
swells - absorbed liquid  
filled it up again

fits -

Dec 16 -

Nitric 1.45 Baumé -

No 1            12 grams  
                  Sugar in each  
Water 300 cc - 100 acid

No 2 - 300 cc - 100 acid

No 3    300 cc. ~~240~~ acid

No 4    200 cc = 150 acid

No 5    200 cc 100 acid

No 6    150 cc - 100 acid

No 7    100 cc 100 acid

No 8    70 cc 100 acid

NO 9 - 50 cc 100 acid

NO 10 30 cc 100 acid

Started Soaking Syn  
at 1030 am Dec 16.

NO 10 softened and balled up  
Nitrous fumes comes off considerable  
quantity.

NO 9. Semi balled but floats  
over top whole dish - Red fumes

NO 7. 4, 8 dont. Balls some  
Red fumes - all other quiet.

No 11.

70 nitric  
70 water  
70 HCl.

fast work  
2 pm Dec 17.

12 grams Zn -

Nos 10, 9, 8, 7

Filtered at 3 pm Dec 17

6 5 4 3 2 1 - finished

at 10 pm 17<sup>th</sup> -

NO 10 Residue after Exhaustion  
with Wood Alcohol -  
6 grams. 7 Decagrams

---

NO 10 Before Exhaustion with  
alcohol puffed more than old  
unpurified.

---

purified NO 10 - decomposes all at  
once don't swell. large volume  
smoke comes off all in 2 seconds  
then ceases and don't show any  
Combustion when lowered in  
flame - don't puff at all -  
Smoke very inflammable,

---



No 9 Puff more than old regular  
when unpurified - when  
purified by alcohol weight  
of residue 7 grams & 8.5 cc  
purified by alcohol don't puff -  
decomposes like No 10 but  
takes 4 @ 6 seconds, no inflammable  
matter when lowered in flame.

---

108. Impurified swells  
about the same as ~~4.7.7~~  
4.7.7, but not so much  
combustible comes off by  
far as in 4.7.

Purified by alcohol dust  
puff in least. - Takes some  
time say 10 @ 15 seconds to  
decompose, then on lowering  
in flame no combustible,  
The particles scintillate showing  
no softening material at  
all in it to lock particles  
together. Weight 8.  
grammes

No 7. Unpurified puffs.  
about same as 4 -  
- purified. Weight of Residue  
7 grams of Diacrylamide

Purified. no puff whatever.  
takes long time to decompose.  
20 @ 30 seconds - afterwards  
lowering in flame no combustion  
scintillations, plentiful.

No. 6. Unpurified

Puffs more than old  
Regular -

Purified weight

8 grammes 4 ~~0~~ Decagram

Purified puffs slightly  
towards end when put into  
flame after I thought it  
was ok - until then it  
didn't puff - takes long  
time to decompose. Think  
from looks it has largest  
residue after decomp but  
I may be mistaken -  
don't scintillate. yet  
these don't appear much locking  
water as you can see parallel

No 4

Unpurified don't swell  
as much as old Reg

Weight of purified

8 grammes of Dica

purified decamp very  
slowly and don't puff  
except at end of heat

it swells up from underneath  
probably because I look too  
much Its swelled little

more than G. seems give great  
residue - scumlike little

on top =

105 Unpurified -  
Puffs greatly lost  
unchanged Syriam in it  
Weight after exhaustion  
alcohol 8.9 grams 3. Dec.  
Puffs some towards final heat  
at bottom. There is probably  
some unacted on Syriam which  
could be taken out with  
Benzol -

Carbonization in  
Evaporating dish to acetone  
Carbonaceous Residue -

No 4 -

1 gram purified by alcohol  
after ignition, residue weighed  
4.98 milligrammes - Carbon  
very slow

No 6

1 gm

Weight after Carbonize  
4.93 milligrammes  
Carbonize slow

No 8.

Residue 500 mgm  
decomposes quickly at first  
takes 1/2 time to finish as 4 & 6

No 7

Residue 497, milgms

Takes considerable time to  
decompose - requires high  
heat to finally decompose

No 9

Residue 485 -

Decomposes very rapidly  
at first = 9 water when  
weighing pyroperic -  
Charcoal fragments -



No 10 -

Residue 483

Decompose, all at once  
+ quickly finished -  
Charcoal for 80.

---

So far No 7 best

---

$$\begin{array}{r} 144 \\ 12 \\ 36 \\ 44 \\ \hline 192 \end{array}$$
  
 2240)  $\begin{array}{r} 3760 \\ 7520 \\ 6720 \\ 30 \\ 30 \\ 90 \\ 720 \\ 6300 \\ 7020 \end{array}$  (314)
   
 $\begin{array}{r} 1700 \\ 902 \\ 4212 \\ 3456 \\ 7560 \\ 6480 \end{array}$  (243)
   

$$\begin{array}{r} 243 \\ 23 \\ \hline 129 \\ 48 \\ \hline 5589 \end{array}$$
  

$$\begin{array}{r} 55 \\ 2400 \\ 2300 \\ 782900 \\ 23 \end{array}$$
  

$$\begin{array}{r} 1320 \\ 158 \end{array}$$
  

$$\begin{array}{r} 24 \\ 4 \\ 9 \end{array}$$
  

$$\begin{array}{r} 240 \\ 20 \end{array}$$

Action of solvents  
on purified by alcohol

No 10

Chloroform dissolves not scarcely anything  
 don't viscous - particles keep separate  
 Benzol scarcely anything -  
 Pyridin good but not perfect  
 I think =

No 9 -

Chloroform dissolves scarcely  
 anything, don't viscous - particles  
 separate -  
 Benzol nothing  
 Pyridin good -

No. 4

Chloroform sticks it - viscous -  
dissolves some to opaque yellow

Benzol colored same -

Waxes if ground = warm

waxed dissolves considerable

No. 6

Chloroform dissolves quite a lot  
more than 4 = Viscous (very) -

Benzol dissolves in one chain

4 if hot it would

dissolve very considerable

pyridine not perfect I think

No 7 -

Chloroform opaque yellow  
Viscous =

Benzol some - not much  
take considerable -  
Pyridin good -

No 4 -

Chloroform opaque yellow  
Viscous not extra so as

7 =  
Benzol dissolves good  
does look like unacted  
on Sy -  
pyridin good not perfect  
I think -

105-

Chloroform semi-viscous -  
dissolves to light yellow -  
Benzol dissolves  $\frac{1}{2}$  as  
much as Chloroform  
pp. in good but not  
perfect.

---

109 with Caustic Chloro

Viscous apparently OK

107 not quite so vis  
but good = Can evidently  
better for 9 than 7 but  
think 7 OK =

Syn - Chl waler

No 12

750 cc Chlome waler

by Aylsworth 10 grms

Syn - 3 pm Friday -

Raw Syrian Salvarsan

Naptha apparently dissolves  
it good though maybe  
fine specs

---

Ether takes out something  
not very much -

Chloroform 1st class

Benzol next,

problem same -

Naptha next,

Benzine - takes out something

more than Ether.

Methylic Alcohol may dissolve  
some transparent stuff -

Experiments oxyd<sup>n</sup> Sy<sup>n</sup> - <sup>11 am Saturday</sup>

7 A. 18 grammes Sy  
100 cc water 100 cc acid

7 B 24 grams Sy  
100 cc H<sub>2</sub>O 100 NO<sub>3</sub>

7 C 30 gms Sy  
100 cc H<sub>2</sub>O 100 NO<sub>3</sub>

7 D 36 gms Sy  
100 cc H<sub>2</sub>O 100 NO<sub>3</sub>

7 E 42 gms Sy  
100 cc H<sub>2</sub>O 100 NO<sub>3</sub>



24-

100 cc  
130 grams  
145  
166 150

Residue of 7. E. after Exhaustion  
alcohol, 25 grms 600 mlyms  
The residue dried all night on  
heater. The Exhaustion by Al  
not very good - will reexhaust  
in bottle over night.

after good purification. 23.500  
but some was lost say 24-

June 6 1891.

Squirted pellets from following

No 1. Trichlorophenol wal. 4 grms Carb. +  
about 2 grms trichlorophenol, worked in water  
till sticky, then 10 minutes kneading in finger  
squirts first class 008 die polished with  
diamond dust, smooth very shiny

Note - after 6 months pellets more than 1002  
when thrown on hot sand bath.

No 2. 4 grms Carb. + 1 Camp, Chlo -  
worked very dry - brittle and cracked in part.  
Crumbly when put in press - works fine in  
press - don't stick to die, file round and  
Even but surface not good polish. but good

Note 6 months after - pellets but little  
when thrown on hot sand bath, globular  
swelling only. Surface got polished  
but with old w carb. I think it will work  
ok - don't swell very greatly when put  
right in flame.

NO 3 Alcohol Camp's unpurified.  
amhu nitric - works fine in die.  
note after 6 months. ok but can see  
its rough & porous - very imperfect  
locking of particles & Cracky. Either must  
be used alloyed with Carb or  
dipped —

NO 4 - Purified by Alcohol amhu  
nitric dissolved by phenylammonia  
previously well powdered - works  
only fair in die.

Note 6 months after fearfully  
rough scarcely puffs at all  
must be alloyed dipped or to ng

NO 5 - Negroin with considerable  
Camp & methyl alcohol afterwards  
add some absolute ethyl al.  
went better. worked well in die

Note - after 6 months found  
fils all snapped in pieces  $\frac{1}{2}$   
inch long

N06. Same as N05 - with smaller  
Camphor & absolute Ethyl al.  
only - works bad in die, probably  
lacks Camp to Lubricate.  
Note 6 months after - all cracked  
as N05 -

N07 Fully native. by HCl slow +  
careful stemming every little while.  
very weak acid plenty H<sub>2</sub>O washed in  
H<sub>2</sub>O dried slow, powdered, then  
Ethyl al to which say 1/4 or 1/3  
Chlty added worked in mortar  
taken out kneaded - when held over  
flame the chunk softens & appears  
to dissolve better works well in  
die, soft, it comes out 008. very  
round but has a rough surface  
when may give it porosity -

Note 6 months after - round even but  
rough & particles not locked together  
most puff out considerably but not  
globular as N02 but more whole than  
increases - will have to have a better  
solvent, or used as alloy or dipped

No 8 Falcq Native, H Cl, Umi. same as  
7 but with. Isotulyamine above - Rough  
surface, dont squirt well, but could  
be made to if lumps got out. unph  
R 009 1/2 003 4-007. - dont pull hard

Note after 6 months. Rough  
fracture but even & round. need  
better solvent or locking material.  
alloy or dipping puffs more than  
2 =

June 9 1891 - new Expts  
No 10  
Old Carbut purified well by Methyl  
4 gms Carb 1 Cam - ~~plenty~~ plenty  
chloroform - heated Carbut slightly  
so I could powder it - Its a very  
hard thing to get lump out. in  
fact almost impossible - using  
10/100 die, plumb god inside & ingot  
as well - worked ingot to stiff  
dough - squirt well -

June 8-91.

No 11 -

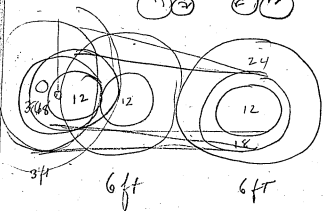
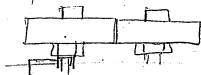
Same as 10 but a good deal of Common Lampblack worked in by fingers - seems to require an extraordinary amount of Chloroform and when you think it ok you find it in a second hard and brittle. While working in Lampblack must dip wgat in Chloro form time to time & start with very sticky wgat of Carbit as in No 10 proportion. It will probably be better to make the Carbit very thin and give it time as perfect solvent action take place to get rid of lumps due to difficulty of grinding fine and then knead it & work in Lampblack. Perhaps the Lampblack better be added to the very thin Salicin all at once & then crop part down & knead -

1012 Expurged Rog purged <sup>CaH<sub>2</sub></sup>  
but boxed in wood etc  
& just dried it then is easy to  
get a good pulverization in water  
the perfect drying with a  
little of the stuff left in which  
is taken out by methyl a  
sticks it is rather more  
difficult to pulverize  
but when the stuff is all  
or practically all taken out  
by methyl a & not dried  
hard the pulverization is  
fine & the stirring does it  
perfectly -- 1012 was not a  
weighed quantity I was  
trying pulverizing experiment  
& put considerably more  
than  $\frac{1}{4}$  Cam dry & tried to  
work it dry but had to add

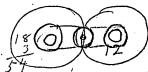
Chloro. I added a small quantity of fensol but got too stiff & I then added sufficient ground of pretty well & then worked to very stiff dough - plumbagoed & ground  $\frac{10}{100}$  die worked fine in die - fits round & polished -

A previous experiment with small quantity methyl added to Chloro. caused fets to work well but they were all rotten so lost all. -



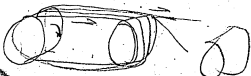


$$\frac{18}{3} = 5 \frac{3}{4}$$



36-

4:6



I have just completed Carboraz  
 2 moulds of squashed fibres  
 Freddy started prelin furnace  
 and ran from 75° fah to 638°  
 fah in 8 hours - I put them  
 in Muffle and was 7 hours  
 getting up to 638°, afterwards  
 took 6 hours to run to  
 Red hat door which I suppose  
 was fully yellow ~~with~~ with  
 perhaps tinge of white,  
 for the last hour full wind on  
 and as much gas as gave  
 slight flames 3 inch high  
 through hole in bottom door -  
 used Charcoal powder 1/2  
 inch over all -

No 1 was shiny a good  
 not puffed. Charcoal shiv  
 little. Brittleness little better

than Reg bamboo oxidized to  
point <sup>stage</sup> but could see there were  
holes due either to lumps or  
a fluffy part which oxidized  
quickly - The perfection of  
the work can easily as well as  
homogeneity be seen by the  
oxidation test,

Now in some ways rough  
not shiny ~~or~~ brittle, but not  
much more than Reg. bamboo  
oxidizes not to a point but  
to a framework with holes all  
through like a sponge -

In the 2nd round was  
No 2. This appeared to  
shrink much more than the  
others. I should say shrink  
in length  $\frac{1}{2}$  - They are  
shiny but would be  
brilliantly shining if the  
fine charcoal didn't stick.  
Perhaps the charcoal shined  
be 50 mesh not less than  
40 or more than 60 with no  
dust. - They are very tough  
bend almost to a knot &  
its difficult to get grip  
enough in fingers to break  
them by pulling - They  
oxidize to a minute point as  
fine as silk & show great  
homogeneity -

The original length was about  
 $3\frac{1}{2}$  & they shrank <sup>to about</sup>  $1\frac{3}{4}$  -  
The original die was I think  
from an 008. The Carbons are  
 $3\frac{3}{4}/1000$  -  
-

No 11. is dull and shows bad  
places not wells but sharp  
lumps sticking out as if  
lamp black was not well  
mixed - It oxidizes rather  
easy & not even although  
goes to a point but very  
bad point being full  
holes -

### On the pumps

NO 1 is exceedingly Even when lighted up - They are same length as regular Carbons not having contracted as much as NO 2 - in the dark and at the dullist red show no spots whatever - best I ever saw in this respect. They light up without peaking on 120 volts. but after working <sup>long</sup> don't seem to light up as bright as before when no peg in - apparently they go up in Resistance. NO 2 seems to act this way also. - 1 arc.

NO 7 - Is higher Res. than NO 1 -

NO 7 - 1 no spots NO 7 - 2 <sup>6d</sup> 5 spot  
1/2 inch inside from clamp -

NO 3 no spots NO 7 - 4 <sup>5</sup> do NO 7 no 4  
not spots -

NO2 beautifully Even at low  
red - 2 arc - 4 1 pump didn't  
work = we didn't dare to work  
too high for fear arcing -  
The vac was not lost & gas will  
by & save the 2 arc carbons  
as they are probably unimpaired

~~Condition~~ Adjust new file (June 10 1891)

No 13 4 Carb 2 trichlorophenol with Mal, ground down until Mal all gone, then there appears to be an oily substance which causes stuff not to adhere to ~~work~~ workbar. I notice that if solvent is put in raw Carb only ground a little, the lumps never come out no matter how much you grind where if you add Mal alone and grind until dead dry all the lumps go out then the agglutinating doses can be added it, (No 13) Knoeda will. The injot was not very stiff as it takes a long time to get it hard - 20/1000 die. The filament is not very shiny, only one row & then. Could see lumps of raw untouched by solvent, apparently - but was broken open by die. Exposing the red tinge of the original; trichlorophenol seems to be non-lubricating on the die & without something else mixed with it will be hard to use in smaller dies - There are many specs - when these specs strike side of die it spoils the fit for long distance by dragging -

No 14 -

June 10 '91

Req highly purified, was afterwards extra purified by hot Mal, then while still moist, ground dry in mortar. This makes it very fine and practically free of grit. No 14 is 2 gms Carb. 1 gm Trichlorophenol. with Mal, it was ground till Mal all gone & oil came to lubricate mortar. Picked it to the right consistency in mortar by pebble & rolled it up in fingers - didn't use plumbago Die 20/6000 - worked well but came out very rough. Trichlphenol is evidently not a lubricator, perhaps a higher percentage would make it shiny or just not so stiff. Present mixt might have worked OK if had not made it so stiff. Stiffness was not more & I think even less than Regular.



June 10 91

No 15

2 grms Carbonyl 1 gm Trichlorophenol Chloro. This works well (Carbonyl is xx purified groundwit.) very sticky in water but gradually comes to an ingat afterwards it must be kneaded @ 8 minutes in hand (or pug?) it doesnt dry fast on surface at all & gives plenty of time for kneading. Die 20/000 - its quite perfect, the surface is pretty shiny but sinuous waves  $\frac{1}{64}$  @  $\frac{1}{32}$  apart. all the fibres dia pply measured 10 @ 15 pct at high swell. The stick somewhat to glass rod - The ingat should have been stiffer or else wasted much

Trichlphenol - I find after pulling in box they stick together somewhat. took them out & then dried them separat. dont think I injured them very much = I find on dry  $\frac{1}{2}$  hour fil collapse pply to viscous not hard stuff



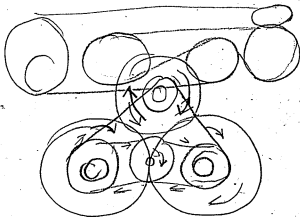
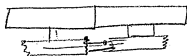
No 16 is like No 2 but 20/1000 die

June 10 91

2 gms Carb.  $\frac{1}{2}$  gm Cam-Chloro ingot  
 works well. (XX ground not used) best way after it  
 dry nuff plug it in mortar. Ingot used,  
 rather stiff not very viscous; requires perfect  
 (shiny) file broke of their own weight  
 so had to hold glass rod @ 8 inches  
 from die; not sticky - fits dead round  
 dont stick to glass rod

No 17, (XX pfl ground wcl)

20/1000 die - 2 gms Carb, 1 gram  
 Cam - Chloro - (plenty Chloro) works  
 beautifully in die Tougher than 16.  
 Shine is perfectly satisfactory.  
 ingot pugged in mortar; tend to  
 dry about like No 16 - fits dead  
 round. dont stick to glass rod



No 18.

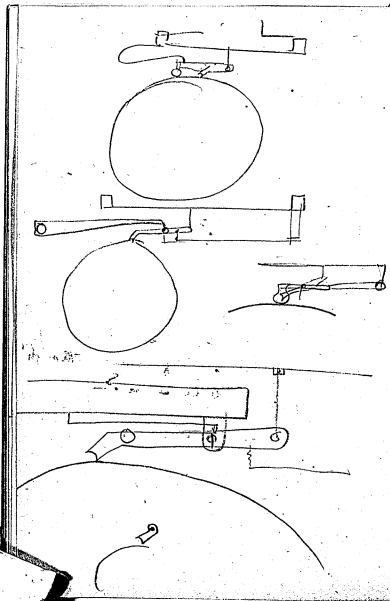
(XX pfd wet) 2 grms C 2 grm  
Cam - Chloro. Long time to pug  
film beautifully shiny but I think I  
did not pug it long enough, nearly  
all the film stuck to glass rod  
& together managed to get 3 or 4  
for test. Entirely equal quantities  
too much Cam, still extra pugging  
may do it, S -

No 19. (XX pfd wet mal)

2 grms C & Chloro only. Early pugged  
but must be got to die quick.  
must rather hard to surface  
somewhat dry before getting to  
mould - B. little film had to hold  
glass rod right up to die,  
shrink greatly after getting on  
glass - not least sticky shiny  
but not so shiny as 16 17 & 18. But nice  
Notes after few minutes film drawn in &  
all will pbbly go S and O

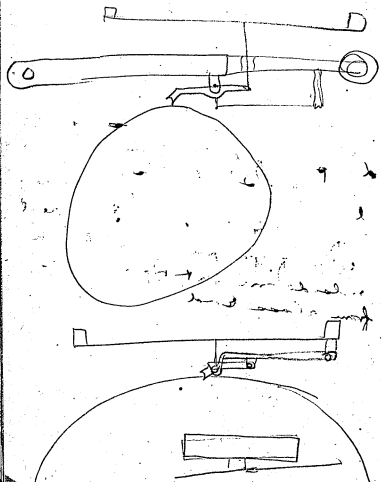
No 20 (xx ground wet)

2 gms C. 1 gm Thymol, - chloro-  
grants well takes long time to pry  
too unobjectionable & practical,  
stiff in get, yet could have been  
made very much stiffer & wk ok  
Die 20/1000 - works perfect in die  
tough - shears a rod slightly  
so had to grant slow  
yet all were saved by  
prying a little more would have  
been perfect. - It has the usual  
brilliant shine of all from  
12 to 19 - Exquisite  
Dart Contact like 19 - think there  
is some solvent action.

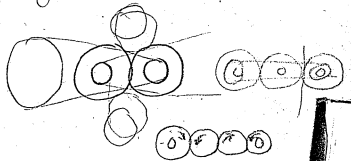


No 21. xx ground wet.

2 gms C - 1 gm menthol, chloro  
 sticks, takes long time to pug as it  
 sticks in spots to make when  
 pugging - even when quite hard  
 still so easy to get lost dry.  
 quick die 20/1000. die  
 squirt elegantly. tough -  
 not sticky at all in glass rod  
 dead round beautifully shiny  
 they seem to swell less than any  
 other when put in flame although  
 this may be a delusion? -



No 22 - Same as 21. Except .005  
 die, works fairly well although  
 die slipped up 1/2 doz times  
 from dried lumps in puffing perhaps?  
 puffing must be done with great  
 care so as to prevent some dry  
 lumps - fit very shiny -  
 round but involves some pl  
 diameters pblly due to hard &  
 soft wgt material? a great  
 deal of graphite catches on  
 outside might have come  
 from glass rod.



June 11, 91

We now commence carbonyzing

Mould 1

Bottom No 15 - ok


next 17 - stuck together somewhat  
and a little brittle.

No 18 3rd from bottom, very  
brittle and distorted, only got 3 or 4 in.

No 20 stuck together slightly,  
not troublesome though - not cracked but  
good ~~ext~~ perfect, - Top mould -

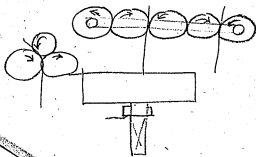
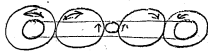
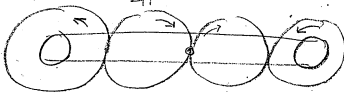
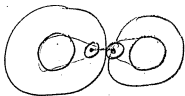
Mould No 2

No 21 - good - slight <sup>rust</sup> sick, bottom,

No 22 005 fibres - good some of  
them thus 

will run in prelim - 1 dig faber filament

No 19 + 16 all cracked - 15 all  
shrunk - 14. Rough shrunks 13 all  
shrunk



Result of Carbyn's

No 15 - shrank, irregular, warty - nq  
only 4 good - contraction not ~~was~~ so  
much as 17. - showed say they were nq

No 17 - only one broken, contracted  
the most by far of all these  
Carbyn - round good - diameter  
very much diminished compared  
with others.

No 18, shrank warty - nq -  
brittle.

No 20 - warty, only got 4  
- not shiny like 17. - larger  
diameters.

No 21 - all broken - most  
fibres good & round - chanc stuck  
found cracks or 2 ply the cause



of brillianess noticed in all

The oos - no 22 - only got 1/2 doz  
out of 20 or 30 very brittle not carboled  
very much in dia or length not  
near as much as 17-

Camphor - Melted - dissolves perceptibly  
Cbt - Very sublimable - nearly all  
sublimes - I don't crystallize - but gives  
glossy coat Cbt, where thin

Chl Hyd - Don't appear sublime  
dissolve perceptibly - on cooling  
don't give Cbt film - where thick  
all crystals but after a while  
when spread out thin - sticky  
viscous - good dissolve

Anthracene - seems to dissolve  
more sublimable than Camphor  
probably good - leaves sometimes  
very thin film Cbt

Chloralacapsol - Low mp  
don't sublime good solvent  
of Cbt - Crystallizes after cool  
crystals tinted with Cbt,  
no film - I doubt if this is good

Chloranil = High MP - exceedingly  
sublimable - scarcely able to melt  
it - it dissolves but perfectly  
to black shiny scale - hard - this  
would seem to be perfection.

---

Benzonilid - Melt mod Low,  
sparingly sublimable - dissolves  
little. Crystallizes strongly  
on cooling, no film - seems to  
run globularly.

---

Dibromoanthracene  
High MP - no signs dissolving  
slightly crystalline on cooling  
this might be good stuff to  
adulterate etc -

Nitrobenzaldehyde (meta)

Low MP - not sublimable - good solvent when very liquid + hot of Carb. often turning on gutter crish + while nearly cool to viscous but when rubbed with finger solidifies to non decomposable dull -

---

Dulpho-Carbamide -  
not sublimable - Mod high MP.  
good elegant solvent. dries to S - seems to be same chem action - dries to perfect shine  
Viscous phy OK if sublimable  
is unnecessary don't enjoy.

---

Anthracene sublimes very  
much before melting but finally  
melt. at very high temperature  
seems to have no solvent action  
on Cb. but perhaps throwing in  
Cb. chilled it. no film - this  
is good if more subliming is  
required.

Tarpen Hydrate  $\frac{1}{2}$  to  $\frac{1}{3}$  as  
sublimable as Camphor  
Rather high mp - perfect  
solvent gives beautiful shine  
but after drying crystallizes  
to dull color probably because  
of excess of Hydrate. This  
stuff may be as good a Camphor

Betal (Naphthalal)

Wetting paint moderate -  
disappears but perfectly  
don't appear. crystallize  
viscous after cooling.  
don't sublime would be good  
to dissolve at high temperature

---

Naphthalamin - M.P. low - very  
slightly & slowly sublimable.  
appears to dissolve but good  
after cooling solution then  
shiny acts curious but  
after long while of uses shines  
& actinometric crystals.

---

Di-Resorcin - Very high MP -

It changes from white powder to wet look  
stuff very little liquid comes out  
95% don't melt. sublimed slowly  
think it scarcely dissolves any  
Carbonate though it may - by expelling  
I clean Carbon pretty clean before  
Di-Resorcin melts. - this is abnormal  
suit for cap and on subliming

Obtain in g sticky pull out  
threads

Diphenylamine - form dimer  
on cooling actually crystal  
sublimed slightly -

Pyridine sulphate absolutely  
unsuitable - n<sub>D</sub>

Carbazol - M.P. high sublimes  
before melting, seems to dissolve  
but slightly - very sublimable.

Thymal M.P. low - very slight  
sublimable. good solvent, in fact  
extra good - thin after cooking.  
If Thymal used small quantity  
gives great waxes when you  
throw in. But spreads, & dissolves  
itself without stirring  
not viscous.

Ulenthal.. M.P. very low -  
not so good solvent as Thymal  
not viscous had to be stirred  
still its good solvent. not  
very sublimable



Hydroquinone - MP high -  
Seems melt as if water crystallization  
think it dissolves same about,  
sublimes quite rapidly.  
Chills instantly when poured.  
Surface not shiny last check to  
dish.

Ammonia Formate -  
absolutely insoluble

Ammonio Sulpho Cyanate  
Absolutely Insol

Acid Phthalic, sublimes  
readily - seems to dissolve  
fairly - mod high M.P.;  
Chills quick - dull -  
Crystallizes slightly.

Phenolphthalein -

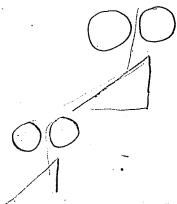
Exceedingly High M.P.  
don't appear to dissolve any - not  
sublimable.

Camphor Monosulfonate.

Low M.P. - pretty good  
dissolve. non - sublimable

Aniline Chloral. High M.P. -  
difficult get any liquid  
sublimable readily.

Quinidine - Very high M.P.  
Semi-dissolve <sup>very</sup> at high temp  
not sublimable



Methyloxide Oxalic

Mod MP Cbt acts greas.  
dont appear dissolves in liquid  
get it melt + globulate,  
thru in thread etc funny.

---

Turpentine oil Dihydrochlorid  
Low MP dont appear dissolv  
in tray just miff to color it

---

Monochlorhydrate same as  
above

Phenanthrene - Mod low mp  
dissolve Cbt somewhat  
Crystalline

Oxamide - Doesn't Melt, at all.  
Even on water slip - flame  
but silently goes away without  
smoke until all gone -

---

Trichlorophenol low mp  
of low density - Coal  
insoluble - hydrolyzes actually

---

Injots.

Regular dose chloro  
is 10 grains or  
2 on right side of head of the

23 - 2 qm C. 1 Cam. 2 drops  
pydn with reg dose Chloro

24 - 2 qms C. 1 Cam 5 drops  
pydn to reg dose Chloro

25 - 2 qms C. 1300  
250-mil gm Camp  
Reg Chloro ~~(10)~~

26 - 2 qms C 1 qm Cam - 10 drop Pydn  
Nada variposa Camp.  
pydn hard reg dose Chloro -

27. 2 qms C - 1 Cam 250 mil gm  
Sulphur Reg Chloro.  
note. (2 forget-cho Campher)

28. 2 gm C - 1 gm Cam, 2500 units  
Iodine - Req Chlora.

on pugging -

29. 2 gm C + ~~Cam~~ Req Chlora  
pugged through 1 gm anthracis  
+ 5 drops pyridin -

30. 2 gm C + 1 gm Chloranil - 5 drop  
pyridin - chlora  
delivered pugged as hand above

31. Methyl Cl with 20 drops  
pyridin in 2 gm C no chlora  
Camphor

300  
32 - 2 gm C. 1 Cam - ~~1000~~ mil gm  
photo jabo Reg Chlors, 5 drops pyridin

33. - 2 gm Carb + ~~1 gm Anthracene~~  
Reg dose Chlors in two Reg dose  
Chlors saturated with Anthracene  
10 drops pyridin

34 2 gm Chloranil Cbnt. 1 gm  
Chloranil - Reg Chlors

35 2 gm Cbnt. 1/2 gm  
Chloranil - Reg Chlors  
5 drops pyridin

36. 2 gms C - 250 mg  
Sulpho-Carbamid - Req Chlors

37. 2 gms C 500 mg  
Sulpho Carbamid - Req Chlors

38. 2 gms C 1 gm  
Sulpho Carbamid Req Chlors

39. 2 gms C - 1 gm Anthracin  
Req Chlors -

40 - 2 gms C Terpin hydrate  
250 mg - Terpin Hydrate.  
Req Chlors - natural in  
Chlors - must try other  
salts -



41- 2 gms C - 500 mg-Terpunhydral  
Req Chloro -

42- 2 gms C - 1 gm Terpunhydral

43 2 gms C 250 mg-Betal,  
Req Chloro

44 2 gms C 500 mg Betal  
Req Chloro

44- 2 gms C - 750 Betal.  
Req Chloro

45- 2 gms C 1 gm Betal  
Req Chloro

46 2 gm C 1 gm De Resorcin  
Req Chloro.

---

47. 2 gm C. 1 gm Carbazol.  
Req Chloro

48- 2 gm C. 1 gm Acid Phthalic  
Req Chloro

49- 2 gm C 1 gm Quinidine  
Req Chloro

50 2 gm C - 1 gm Oxamide  
Req Chloro -

51. 2 gm C 1 gm Camphor 1 gm Oxamide  
Req Chloro.

51  $\frac{1}{2}$  = 2 gm - Falcberg HCl - part oil methyl.  
1 gm Camphor; 5 drops pipridine - methyl  
alcohol solvent =

52 - 1  $\frac{1}{2}$  g. Cbit 1  $\frac{1}{2}$  gm residue  
after Exhaustion by methyl of falcberg  
Natives; chloro - + perhaps Mal -

53. ~~2 gm Falcberg oil + Mal.~~  
1 gm Camphor Chloro/Req -

54 = 2 gm Cbit. 1 gm. Meta-  
Chloraldehyde Req Chloro - how abt.  
Solvent?

55 - 2 gm Cbit.  $\frac{1}{2}$  gm Camphor  
 $\frac{1}{2}$  Methylal = Req Chloro

56 = 2 gr C. 1 gr Mmochlorohydrate  
Turpentine oil

on 11<sup>th</sup> June we put 75 grms sgm  
powdered 150 mesh in flat round  
glass dish - with 3 doses of  
200 cc each, of mixed acid + water  
Equal parts - This is at the  
rate of 24 grms sgm to 100 cc acid  
+ 100 cc water -

Each dish had same - very little  
puffing or nitrous fumes; reaction  
went on quietly - had to stir it  
well at first to get it wet;  
afterwards only stirred it 2 or  
3 times, on the 15<sup>th</sup> pm we  
poured the three dishes in 6 times  
their bulk of water from hydrant  
in glass jar - well allow it  
to soak for hour and then  
filter off - and ressoak in  
fresh water ~~the~~ ground -  
put in water. June 16 - 5 am after  
grinding - (note, I find particles have  
not stuck together but no stices  
about same size as original 150 mesh  
on.

on the 20<sup>th</sup> June final filtered the whole -  
dumped the whole 3 jars on the 15<sup>th</sup> into  
water, after soaking day filtered ground,  
it slightly, in water & put in fresh  
water, stayed 2 days, then filtered &  
put in fresh water 1 day then filtered.  
The jar water was 10 dia. 12 deep -  
used our hydrant water -

The resultant stuff ~~is~~ doesn't have the  
fine yellow of acid stuff but darker  
& greenish & dissolves better in Chloro  
- about as well in pipe -  
not very much come off with  
methyl - not  $\frac{1}{4}$  as much as with  
aldrey - Benzol gave practically  
nothing with aldrey while it  
gives very considerable with new  
stuff & the slubb dissolved out  
by Benzol doesn't pull in the least  
in fact the stuff before purification  
scarcely pulls any - There is  
very considerable ash in Benzol  
residue but perhaps not more  
than in rest. It must be  
purified by Absolute methyl =  
benzene =

No 23 = Think there was too much  
Methyl left in it from grinding - I also  
suspect that the wood alcohol  
got down stairs to night is common  
wood spirit full of pyrolicious acid  
& needs a little of it and finds that  
I need a little of it, exceedingly  
notwithstanding this, exceedingly  
fine Carb. It will not dry out on  
Evaporating dish constantly stirred  
for  $1\frac{1}{2}$  hours - There was strong  
alcohol (methyl) fumes come off at  
first but afterwards it gave  
continuously repulsive acid fume  
smelt in pyrolicious spirit.  
I concluded to wash it out  
by Ethyl alcohol which I  
am now doing - The filtrate  
(Ethyl) is just tinted yellow -  
I think I smell a Sulphurous  
acid as it smells & makes me  
cough as Sulphurous acid.  
No 23. with 10 grms Chloro is  
less much Chloro - 7 grms  
is plenty so I am going to  
change it from 10. to 7.

I found that when I had got an ingot that the repulsive smell of Methyl overcame the pyridin & Chloroform. It pugged beautifully in water & didn't stick once - I think I noticed some liquid squeezed out - however I am quite sure the Woodspirit prevented the sticking to water & hence sticking to material is a good sign - on squeezing the filament was wavy also the skin was broken at points & it was not good - the filaments all stuck together any way = 2  
am going to try 23 over again as soon as ethyl residue is ok. Dick is dissolving some Methyl & find that owing to the fineness of the material that the Ethyl takes a long time to fill - over 1 1/2 hours & then

I am Compelled to lay open  
the filter paper while very moist  
with Ethyl to dry it.  
It dries nice and yellow without  
lumps but it tastes still  
acid — I am afraid that  
while it appears dry there will  
be some water left in it  
as Ethyl is only 95% — however  
by time this too can be  
drained out = I find that  
Exhausting with Ethyl clears out  
the acid liquid hence it is best  
after working with Washyl, to  
finally Exhaust with Ethyl.  
The material is good & practically  
free from lumps =

Again I try 23' - 2 gm C 1  
from Cam 2 drops pyridine -  
1/4 on left Drognet glass chloro  
pugged hard until outside dries  
rather rapidly works beautifully

in die - filo - very shiny and round  
& even size. notice frequent lumps  
but not enough to lose more than 15%  
of filo. If you squirt too fast  
there is tendency stick to glass  
rod but by going slow they dont  
stick - hang up tube when full.  
to prevent them to dry - made  
3 tubes full = on taking off  
stick they dont stick - nearly every  
one saved - very little loss will  
occur with this formula =  
We put one tube full in Chzy  
box & save the other two for  
future Expt = 008 die -  
want in No 2 box side of box - Carbon  
are - 0045 - filo 007

No 24 = 2 gm C 1 Cam -

5 drops pyridine - ground well  
after ~~to~~ about 2 ~~hours~~ notices  
on Diquet glass chess -  
ground it to free it from  
lumps which was very sunny  
didnt use any dry sides on



Wester = on hand pugged little.  
afterward pugged very hard in water.  
It pugs well & easy. Squinted  
perfectly = scarcely any loss, very  
round and Even & very shiny  
little more than 23 - don't stick to  
glass rod as we squinted slowly.  
we put one tube full in box ~~24~~  
which is single box Use loops being  
placed at opposite ends ~~24~~  
No 24 is on No 4 side of a box -  
008 dia - Carbon 005 0075 - fl -  
original length 5 1/2 - Carbon 4 1/2 -

No 25 - 2 gm C. 1300 vulgum Cam  
Scent. 2 Dey Chlors' c doesn't dissolve  
well cut by any amount of working  
got the lumps out, worked long time  
afterward took ~~one~~ balls & got  
together rather wet & worked long  
time in fingers afterward  
pugged - stuck persistently to  
water but after while got it  
reasonably stiff = hard to  
roll the wet with pestle so

as to get it in the die - it cracked  
in pulling it in 1/2 broke off &  
was lost still we got a large  
quantity good filaments it works  
well get 80 pct. gblly but it doesn't  
work so nice as 23 or 24 -  
I think it sticks to glass but am  
not sure cos die - fits very  
round very few lumps -  
shiny but as I am now using  
electric instead daylight don't  
say if they are as shiny as 24  
but think not - but water soluble  
they are beautiful - 25 is on  
No 1 side of box - Carbons 004. file 0065

27 - ~~Sub~~ 2 gm C ~~250~~ 250  
Wolym Sulphur but only part the  
dissolved - I forgot to put in the  
Camphor = Dies worked fairly but  
stuck 2 or 3 times - file not shiny  
some collapsed inward - not a  
success I should have tried  
would have been better if  
Camphor was put in  
No 10 of box Carbon 005  
file 006

26 = 2 gm C 1 gm Camphor  
10 drops pyridin scant 2 drop on  
Drying glass Chloro =  
Takes long time finger pry - flex  
Took 15 minute to pry hard in  
Water - no fear drying - Can pry as  
hard as you please - This was  
pryged rather hard yet it was  
mobile - outside ingot when put in  
mould was not dried or cracked -  
The Combination admitt of perfect  
ingots being made & ply by  
machinery - It would be satisfactory  
if filaments work in Lamp  
The bulk of ingot is much smaller  
than when ~~no~~ pyridin used -  
have pply more Camphor in fil -  
50% die - It worked the finest  
in die of any we have ever tried  
not a mass or bad fil - didnt  
stick to glass - Dead round  
Extra shiny & perfect no  
Lumps or fractures notes  
occupies No 13 of Cox. Carby 005  
file 005

exp. 2. 2  
exp. 2. 2

32- 2 c 1 gm Can 5-drops  
pyridine - chloro - ~~at~~ 300 mgms  
photo plumbago, did not get all lumps  
out, needed about 10 drops pyridine  
although it is fairly good around the plumbago  
in with dry chloro then added  
salutarin - It pugged well in finger  
Cupule minute then pugged it say  
7 or 8 minute in mortar - work  
fine for pugging did not stick -  
008 die works well in die -  
don't stick to rod in slightest.  
Dead round; rather lumpy chloro  
not touched with salutarin +  
also effect of plumbago is to give  
a broken surface shine with  
pieces - ~~the~~ files are good but  
not perfect like Reg 2 c 1 Can  
10 drop pctn - at 5 in boys.  
Carbons 0045 files 0075

32  $\frac{1}{2}$  = 1 gram Carb.  $\frac{1}{2}$  gram Camphor  
1 gram photostyrene p690 10' drops  
pyridin - Grinds well-mixed p690  
with dry Carb = put plenty Chloro  
got clear discalcio = purged some  
time in fingers and lay time in  
water say 15 minute - got it moderately  
hard. always tendency to stick to  
water. Even at the end - think  
excess of pyridin in proportion to Carb  
made it so should have been 5' drop  
to the 1 gram Carb but works well  
in die - no stick on glass rod  
so turns difficult to hang them had  
to use an extra rod to guide  
them = didn't stick got all -  
no 8 die = Dead round Gut  
surface Rough. although the  
filament seems to be OK =  
put in No. 6. of box - Carbons 006  
file 0075

33- 2 gm C - 1/2 g Anthracene is  
not soluble enough to get 1 gm in  
the chloro - so look twice amount  
(2) 4 Dis in Dreyfus glass & saturated  
it with Anthracene, filtered, put  
10 drops pyridine in = Work OK  
pings well, 008, if it not  
round O oval rough on  
surface - looks as if Anthracene  
crystalized when pinging & thus  
always causes rough surface.  
in No 9. of Box. Carbon 004  
file 006

34 2 gm C - 4 div of Drug measure  
of saturated solution of Chloranil  
in Chloroform - couldn't get 1 gram in  
or 1/3rd that, used no pyridine, hence  
was somewhat lumpy - pugged  
beautifully in water, 0.08. dia.  
work fine in dia, file generally  
round - but the surface is a  
broken shine or rather lumpy shine.  
Microscopic = file tough, the Chloranil  
evidently crystallizes out & act as an  
inert powder to roughen file =  
It certainly acts as a salient  
abrasive & could not have  
pugged it so long or so well,  
didn't stick to glass, saved  
nearly all file - in No 16 of box.  
(stuck together) - after clean

36 - 2 gm Cbt - 4 Div Drug measure  
of saturated solution of  
Sulpho Carbonic acid = not lumpy  
Pugged beautifully = 008  
die @ quilled good at a pic  
only twice = file Round +  
oval,  $\frac{1}{2} + \frac{1}{2}$  = Rough oblong  
In No 15 of 60x = stick in body

45 = 2 gm Cbt. 1 gram  
Betal - 2 Div of Chloro -  
Puge Good - sticks mould  
nearly to last then good  
drye quick - 008 die  
squirts beautifully -  
Theres tendency to stick  
to Rod - run slow then ok  
Could have pugged it a  
little harder I think



filament Round Shiny -  
pretty near as shiny as best  
but not quite - Tough -  
think toughest we have  
had - No 11 in Box,  
didn't lose any file - Stack in Carby  
beautiful shape

57  $\frac{1}{2}$  = No 13 in box = works ok in  
winter fings well - squint  
- easy 008 - file rough  
looks ok under micro in original box  
Carby 006 file 005

53. = ~~it~~ is the same as No 26  
but is squinted with a 005  
die - works beautiful which  
once or twice - file had  
occasionally. Kinks say 10 pct  
stuck little, to rod could easy  
make 100 a minute - perfect shape  
around = No 7 in box -  
Carby 003 file 00525  
orig length 5  $\frac{1}{2}$  - Carby 3 inches - Over

These were well preheated for to  
630 deg-fahr for 8 hours and  
run to maximum heat of gas  
muffle furnace in 7 hours

most all the files were fused  
together except the plugs &  
the 205-fils - get records  
from Joe & also oxide than a  
reporter

56 - 2 gms. C. 2 Dev Chloro -

40 drops pyridin = nice dissolve no  
dumps worked it down by blowing  
afterwards called it a pugger  
10 mins in hand - rather sticky - but  
got ok then pugger long time in  
water say 15 minutes till got  
hand in got it could have got it even  
harder if necessary as it squirted  
beautifully got every one only one  
think noticed, ~~file~~ file didn't  
stick or slightest to rod  
all the fibres look well as real  
and after 10 minutes don't  
curl or appear to shrink in water  
under microscope they appear  
dead round but the shine is,  
not equal to Camphor fits -  
but very good substitute,  
Think these will have more carbon  
left than with Cam -

57 = ~~ARIZONA~~ 57. = MEANT about 2 1/2  
Falberg - but there is some doubt if it  
is known as I have two proportions under  
same no = 10 is 1 1/2 Falberg, other  
20 dps furural, too long time pay down to  
ingot. stir to glass rod, walk fracture.  
and had change from 005 to 005.  
then it kinked, want pbqs I think.  
only got few - or its is the following  
if there are many fibres then it is as  
follows - 57 - 1 Carbrot, 1 Falberg  
200 mg Camphor, 20 drops dipi little  
Wal Chloro - '00'

58. 2 Carbrot, 300 mg Camphor 40  
drops dipi, Chloro 005 - die  
Takes long time to pay works  
Elegant and is fringed quite  
hard but there is a tendency  
to <sup>to wavy but about stick</sup> ~~stick~~ <sup>long glass rod</sup> ~~stick~~ think  
thick and unnecessary amount  
of pyridin used fibres dead  
rounds generally free flame  
surface not a delicate line

59 2 Cbt, 200 mgms. Cam  
40 drops pyridin Chloro soaked  
for 10 minutes + then puffed it.  
took long while acted like 58.  
works well in die stuck 3  
times file round but dullish  
not shiny appearance rather  
free defects

No 60 2 Cbt, 500 mgms  
Campho. 20 drops pyridin  
puff well don't require so  
long a time as 59 - & harder  
in get. 1 squirt fairly die  
stopped 4 times but last run  
got 1/2 of whole output without  
stopping file shiny showing  
Campho lubrication - dead round  
cracks free flaws stick to glass  
rather somewhat. but afterwards ok  
Could have puffed 100 dead round file  
0.05 dia. 9 Hunk -

61 - 2 Cbst 500 mgms  
Chloralhydrate, 20 drops  
Pyridin <sup>Chloro</sup> - ~~just~~ ~~was~~ fairly  
pung well - stopped over in  
dis = fibre round reactor  
free spots fairly shiny  
005 die I think -

Nos 56 to 61 & also Nos of  
the previous file left over -  
We hung in glass reactor in prelin  
furnace and in 5 hours got to  
350 deg. They contracted about  
1/2 inch - then put back  
a run to 630 in 5 hours  
then contracted about another  
1/4" - but the great contraction  
didn't take place hence it  
must be the critical point is  
just below that. We now  
around heat vent, we now  
put them in nickel boxes, then the  
22<sup>nd</sup> I am to run them to fuel in  
gas muffle furnace -

Sept 27/92

Some experiments mixing various things with  
Syring asph hot matter, then pouring in a 3 lb  
+ putting into a steam jacketed press while  
cold + squashed. - then the (oost die) film are  
put in kerosene with 1 2 3 + 4 cc of  
Chloride sulphur respectively + allowed  
to soak. After a time pieces are to be taken  
out + changed to see how far Chlorination  
has gone in.

No 1 Bright fracture Syring melted with  
unboiled linseed oil in considerable quantity  
cleared up, not gassy or puffy very little  
gas came off - poured on acid (vanilla) plate  
cut strips with shears + rolled them up  
+ put in press - film perfect under Micro-  
no lumps or spots - very electrical insul  
broke trouble to handle - tough film used  
1 cc Chloride sulphur in the 3 or 6 inch of  
kerosene -

No 2 do 2 cc Chl S  
No 3 " 3 " "  
No 4 " 4 " "

OVZV

Sept 27/92

No 5 is the same as No 1 but with very considerable amount of asphalt added so as to diminish the quantity of forced lines, makes shiny black if perfect pty could use a cc less oil but as electric as No 1 - tough - 1 cc chl Sul

No 6 do 2 cc chl Sulphur

No 7 " 3 " "

No 8 " 4 " "

No 9 Syrian very little unbound lines & lump of Camphor - rather spongy - don't give good break like 1 + 6 - splits beautifully if perfect apparently - air or rather vapor of Camphor that makes it spongy probably condenses + hot pressure - Considerable it ok -

1 cc chl Sul

No 10 2 " " "

No 11 3 " " "

No 12 = 4 " " "

JVW

No 13 - Syrian & Japan wax - mixes  
elegantly - block fracture, not spongy  
very little seems to do big although  
considerable wax in this - Melt to thin  
liquid - work well in die - Tough,  
fills gap fit.

1 cc Chl Sulphur

No 14 2 cc "

No 15 3 cc "

No 16 4 cc "

Bottles put away



$$\begin{array}{r} 1001 \\ 1001 \\ \hline 1755 \\ 1001 \\ \hline 2756 \\ 1001 \\ \hline 3757 \end{array}$$

$$\begin{array}{r} 1001 \\ 1001 \\ \hline 2002 \\ 2002 \\ \hline 4004 \\ 4004 \\ \hline 8008 \end{array}$$

$$\begin{array}{r} 1001 \\ 1001 \\ \hline 2002 \\ 2002 \\ \hline 4004 \\ 4004 \\ \hline 8008 \\ 8008 \\ \hline 16016 \end{array}$$

$$\begin{array}{r} 1001 \\ 1001 \\ \hline 2002 \\ 2002 \\ \hline 4004 \\ 4004 \\ \hline 8008 \\ 8008 \\ \hline 16016 \\ 16016 \\ \hline 32032 \end{array}$$

$$\begin{array}{r} 1001 \\ 1001 \\ \hline 2002 \\ 2002 \\ \hline 4004 \\ 4004 \\ \hline 8008 \\ 8008 \\ \hline 16016 \\ 16016 \\ \hline 32032 \\ 32032 \\ \hline 64064 \end{array}$$

$$\begin{array}{r} 1001 \\ 1001 \\ \hline 2002 \\ 2002 \\ \hline 4004 \\ 4004 \\ \hline 8008 \\ 8008 \\ \hline 16016 \\ 16016 \\ \hline 32032 \\ 32032 \\ \hline 64064 \\ 64064 \\ \hline 128128 \end{array}$$

$$\begin{array}{r} 1001 \\ 1001 \\ \hline 2002 \\ 2002 \\ \hline 4004 \\ 4004 \\ \hline 8008 \\ 8008 \\ \hline 16016 \\ 16016 \\ \hline 32032 \\ 32032 \\ \hline 64064 \\ 64064 \\ \hline 128128 \\ 128128 \\ \hline 256256 \end{array}$$

$$\begin{array}{r} 1001 \\ 1001 \\ \hline 2002 \\ 2002 \\ \hline 4004 \\ 4004 \\ \hline 8008 \\ 8008 \\ \hline 16016 \\ 16016 \\ \hline 32032 \\ 32032 \\ \hline 64064 \\ 64064 \\ \hline 128128 \\ 128128 \\ \hline 256256 \\ 256256 \\ \hline 512512 \end{array}$$

$$\begin{array}{r} 29 \\ 29 \\ \hline 58 \\ 58 \\ \hline 116 \end{array}$$

$$\begin{array}{r} 453 - \\ 453 \\ \hline 906 \\ 906 \\ \hline 1812 \\ 1812 \\ \hline 3624 \end{array}$$

$$\begin{array}{r} 50. \\ 457 \\ 428 \\ \hline 885 \\ 885 \\ \hline 1770 \end{array}$$

$$\begin{array}{r} 4 \\ 1120 \\ 4300 \\ 3360 \\ \hline 8980 \\ 8980 \\ \hline 17960 \end{array}$$

$$\begin{array}{r} 450 - \\ 450 \\ \hline 900 \\ 900 \\ \hline 1800 \\ 1800 \\ \hline 3600 \end{array}$$

$$\begin{array}{r} 5 \\ 5 \\ \hline 10 \\ 10 \\ \hline 20 \\ 20 \\ \hline 40 \\ 40 \\ \hline 80 \\ 80 \\ \hline 160 \end{array}$$



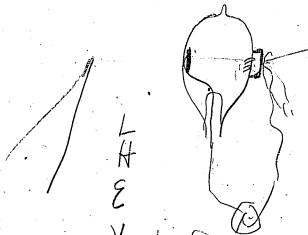
**Notebook, N-91-09-03**

This notebook covers the period September-October 1891. All entries are by Edison. The book contains notes and drawings regarding the "blue bulb" phenomenon and the reduction of mercury vapor in incandescent lamps. There are also notes on various filament experiments, including entries, numbered 1-19, relating to the deposition of carbon and other compounds on the filaments. Edison's notations indicate that he was often assisted by an experimenter named "Joe," possibly John Joseph Force. The spine is labeled "122." The book contains 187 numbered pages.

Blank pages not filmed: 178-185.

x E-172

N-91-09-03



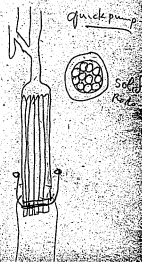
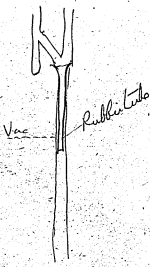
Mandelin

branches



Sept 3/91  
 deposit whole getting  
 vac =  
 deposit glass  
 on waste Exh  
 tube after lamp  
 seals off + this  
 absorb Hg = vac  
 P = wire slum to

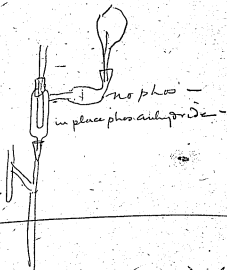
Sept 3/91



Sept 3/91

③

Prokter put in maximum ampere in  
see if max amp when lamp break -



gold & other metallic leaf on  
Carbon when vac for  
fit up valalidye + H<sub>2</sub>  
metal will be thin  
about the H<sub>2</sub>

Sept 3/91

To ascertain if there is a God  
Carrying that ckt after Blue goes  
away also to ascertain if  
lowering of CP after running out  
due to great energy due to the  
that ckt a Red Lamp then  
measure cold — then after it  
fallen to 50 pc of CP measure  
again cold

Sept 3/91 — 4:30 AM —

Put gold leaf  $\frac{1}{2}$  inch sq  
in vac or glass where vapor  
of Hg mandant struck it  
by burning Lamp & looking  
at blue by Spectroscope Hg  
lines gradually disappeared  
was able to bring the lamp higher than ever  
a 18 of 240 was even brot before  
Hurrak!!!

Sept 3/91

(7)

The old way means the old way was a failure was that it was put in side tube with idea it would absorb Hg where it don't absorb cold Hg but being put on side of globe or in the path of the incandescent vapor the instead the Hg strikes the globe it combines because its ~~incandescent~~ incandescent. -

Sept 3/91

putting  $\frac{1}{2}$  inch square of gold or Hg on side don't seem work well =

Sept 3/91

Phenomenon When gold leaf has to fill its repulsed from globe when lamp touched, but when piece touches glass & off filament resting glass attracts glass. -

Sept 4/91

(7)

Hg Expt

- Good Hg in lamp -  
King Hand Sodium amalgam -  
Tin foil -

Sept 4, 91

Keno. 32 = Silver inside - Hg not gone  
Set up Req Curve -

Sept 4/91

33 Keno - miz - side tube Good Hg  
got vac showed Hg - while in vac heated  
side tube not red or to decompose  
gave off air - then heated red  
decompose portion Ag<sub>2</sub> - no air  
turned yellow or reddish - stopped Hg  
little sealed off with tube on  
then heated till turned blackish  
didn't knock vac down - all Hg  
disappear. Glass clear - then seal  
off side tube - sent to get lit



Sept 4/91  
Kens 3 $\frac{1}{2}$  mg Reg for Hecker to  
put on transformer test but approx  
leg =

Sept 4/91  
The Lamp with Cut<sup>2</sup> gave off an  
no 2 = Knobs Vac Down hanging  
blue 11111 spec - globe approx  
clear but by moving spec clear  
to Carbon got this

hanging blue  
billet-type  
same lines

Limit of Spectrograph  
Lines



Sept 4-91

33 No 2 is a duplicate - acts just  
same - (15)

Sept 4/91

Keno 34 - Bromide Aurum

Lots of white lines didn't heat, p.d. by Vac or  
Carbon not (blackened) - light decomp. of globe clean blue  
on clamp - High Vac - no Hg lines

(C) Keno 35 - Chloride Aurum

Lots of white lines on heating  
Carbon (blackened) (blackened) - High Vac + globe blue  
Hq d. line

Keno 36 Chloride Aurum + Iodine

Lots of white lines - Mod Vac of globe blue -  
(Carbon - blackened) (side) - no Hg lines -

Make Dealers Transfarm order

37 -

Sept 4/91

(15)

If getting rid of Hg Vapor prolongs  
life than best scheme is either Iodide  
silver just at tip + heat to red when  
taping - or a supply of Bromide  
between phos + lamp covered from light  
+ when high Vac take off cover +  
expose to light long enough to give  
off free chlorine to combine with  
Whisker of Hg - or a side tube blocked  
sealed off with Lamp - or than  
exposed to light + sealed off -  
or small piece put in Lamp to keep  
Hg constantly combined during  
Exhaustion + piece afterwards  
got ~~down~~ left in or got down  
in Exhaust tube + sealed  
off; there are many schemes  
to work it - practical -

Kens 1 is film heated to 600 & then  
thrown in Naptha -

Kens 2 Req clamp material  
all the clamp painted all over  
& heated.

Kens 3 Aluminum Electrodes

Kens 4 new clamp material,  
chromated Bakala Benzol + plumbago  
painted over clamps clear  
down to glass -

Kens 5 Heated film to 600 then  
thrown in Turpentine - then new  
clamp material all over clamp  
to glass -

Kens 6 - Bulb Coated  
inside with Amyl acetate  
Sol. of Collodion

15  
64

60  
90

Kens 7 - Inside globe coated with Rubber dissolved in Chloroform

8. Balsam Fir (Canada) in Benzol inside globe coated.

Kens 9 - globe cleaned with 25 pct. solution HEP -

Kens 10 cleaned hot 20 pct KO -

Kens 11 - w/3 Carbon in Gum bulb

- 6 X 10 sec
- 7 X 15 "
- 8 X 20 "
- 9 X 25 -

3X 57p  
Ed - by hand 8 1/2

10 X slow chgn -  
11 X 3 inch Cook -

12 Keno Inside bell Control Room  
in Benzol-

13 Keno - Bell cleaned by ammonia

get missing KOs -

Keno 26 Sectional Exhaust  
by punch Cock -

Keno 27 - same as 26 but  
with little sub -

Keno 28 fine Contractor in  
Exhaust Tube -

Keno 29 Ashes looping

30 Keno cobaltos plug -

31 Keno Carbon, heated then  
thrown - hot HCl -

31 Extra same as 31 but bulb  
washed in HCl -

Other nos in 1st part book =

Keno 23 big tube no candle  
no blue on either clamp very  
little air Gore 12/1000

22 Keno No phos used

21 Keno Run with phos  
like 22 -

Keno 20 - jarred clamp  
+ part of glass -

Keno 19 - view clamp material  
in spots over filament.

Keno 18 - long wave wire  
got jammed -

Keno 17 - Cotton spiral on  
N - side -

Keno 16 - jarred long  
wire & glass - baked -



Sept 5/91

(27)

Kens 38 -

fits H.P. 10 hours -  
Culd + not previously heated -

Kens 39 -

~~Tuesday Monday~~ Tuesday

Sept 8 91 -

files soaked 5 hours in test tube  
3 stry KO, then part in 1st distilled  
water then soaked 1½ hours  
in 2nd distilled water

Kens 40 - Glacial acetic  
5 hours - 2½ hours  
in 2nd distilled - hard to wet  
them =

Kens 41 -

SO<sub>4</sub> - soaked 5 hours  
soaked 2nd Dist. 2 1/2 hours

---

42 <sup>Kens</sup> - Soaked Nitric acid 8 hours -  
in 2nd Dist 24 hours -

---

43 Kens - soaked 8 hours  
in about 20 pct Hydroiodic acid  
in 2nd Dist 24 hours

---

44 Kens - 2 sliz KO<sub>2</sub> for 5 hours about  
afterwards tube put in hot KO<sub>2</sub> water  
& boiled about 2 hours until water  
1/2 evap. & bubbles gone, then 10 hrs  
in 2nd Distilled -

---

3 1/2 - 33

10 1/2 - 99

20 pct. 70.

(31)

Since we have good Carbons why not  
burn lamps at 60 candles for 20 min  
& then set them up at get drop Cp at  
normal = Evidently we don't eliminate the  
excessive drop of Cp enough to prevent  
the 1st rapid drop at normal.

Try amalgamated copper clamps -

Try passing following in Emman's bulb  
after lamp worked Reg -

HCP = Hydroiodic acid, acetic acid,

Nitric acid, Bromohydrate, ~~ether~~

Nitro Benzol, Ammonia, Hydrocyanic acid,

Sulphide Carbon, Chloroform,

Ethyl Alcohol, sulphide ammonia,

~~Nitro~~ phenol, trichloroacetic acid

Furfural, pyridin, Bromobenzol

Dip Carbon in following -  
Carbrot in pyridin -  
Uric acid in tetraethylammonium -  
Galberry Nut in Mal,

Kens 45 = 504 in water say 25%  
put inside globe when ready  
to exhaust - Idea being make  
globe acid - no curve ng

Kens 46 = Caslor Oil in  
Lamp - ng no curve

Kens 47 = Balsam Fir with  
Tetraethyl Carbon in bottom  
exhaust tube -

(35)

Keno 48 - file soaked in  
Ammonia 24 hours & several  
hours in 2<sup>nd</sup> distilled -

---

Keno 49 - lead acetate  
24 hours several hours  
2<sup>nd</sup> distilled.

---


Keno 50 Borax 24 hours  
12 hours 2<sup>nd</sup> Distilled.

---

Keno 51 - 1 Lamp side  
tube fused chl Cal big pink  
 $\frac{1}{4}$  this size enough - 1 drop 504  
when got the Lamp Reg. knocked  
Calc into 504 - gave off HCl - no  
blue - lots gas came off for 1/2  
hour when not very bad light

(37)

Lamp this was used gas 3 feet  
fully moist - should pass through  
phos only before going to lamp  
burnt fil though in very bad vac  
yet not oxidized = ~~was~~ finally  
with very large quantity air

 Canning down sealed off  
blue on clamp showed  
H<sub>2</sub> + some strange line  
plainly - also showed  
H<sub>2</sub> in genl. vac although  
no blue in genl vac -

Kens 52 - 24 hours in  
phosphoric acid - 20 hours  
in 2nd dist -

---

53 - files 36 hours in  
strong Chloride Sulphur  
Then soaked in Kerosene  
1 hour —

---

12 X - after final quench, in  
Kerosene - 10 sec of 17 -

---

13 X. Same method not Kerosene

---

14 X Regular which F & H would  
like test in -

---

15 X Are such as they are sent  
to factory 10 sec of 17

---

16 X are discolored carbons which  
they call oxidized -

~~54 Ken~~

53 Keno <sup>Reg Mg</sup> globes dipped in distilled water containing say 1 to 1/2% of HCl.

54 = Reg Mg - globe heated then HCl passed over <sup>flowing</sup> into lamp - then cooled little by blowing on bulb - then reheated - + HCl passed again - blown on - finger held on end exhaust tube then taken off + rubber put on quickly + then on pumps



43

Kens 55 = Healed Bulb  
+ struck Ends in ammonia  
bottle - claid till only  
little warm - put finger  
on + taken to pump  
room -

Kens 56. clamps cleaned  
1st HCl -  $\frac{1}{2}$  min then  $\text{NO}_3$  5 seconds  
then Red Solution then plain  
water then distilled, but  
didn't was whole length so  
glass got somewhat dirty  
by rinsing water running  
down

57 - Mg wash bulbs -

58 Mg washed bulbs  
heated globes after Exhaust  
+ before sealing -

59 - Mg washed - bulb heated  
+ run as high as possible on  
pump -

60 - Mg washed - peroxide  
lead in side tube -

61 - Mg washed - 50 g in side  
tube or 3 pinhead ferrocyanide  
to make CO - 1st lamp  
broke 1/4 inch from P. clamp on pump  
making another -

62 - M<sub>3</sub> washed - get Vac  
plug in 120 volts at once (42)

63 - M<sub>3</sub> washed plug in  
on 120 ~~bottom of~~ 4<sup>th</sup>  
peg down on lamp bank  
run 1/2 hour there then work  
Req -

---

64 - Reno - Req. M<sub>3</sub>  
globe cleaned dried +  
afterwards wet with  
strong Sol KOI

Keno 65 =

49  
boric acid in glaci

Keno 66 = soaked about  
3 days Chloride sulphur  
then 3 days Kerosene

-Keno 67. Soaked about  
4 or 5 days in distilled  
water taken out & boiled  
on hot plate put back  
in water ~~then~~  
~~soak in water~~ then  
out & sent to Joe -

Keno 68 = Uric acid in side <sup>(57)</sup>  
tube -

Keno 69 - 504 + piece  
Zinc knocked over  
into 504 to make H<sub>2</sub>

---

Keno 70 - Mg Reg -

Solid tube, then heat  
then in 15 min heat  
bulb again then after  
25 min from solid tube  
heat again then in 5  
min light up fat + water  
Reg -

Keno 71 = boiled for .1 to  $1\frac{1}{2}$  hours in strong KO; soaked 1 hour in Distilled =

---

Keno 72 = deposited by Aylsworth,

Keno 73 - Oxalate Ferrous in side tube - heated to redness to blue then got solid tubes worked Lamp & sealed off.

= on one I attempted to reheat when it broke vac - destroyed the Lamp

Keno 74 - 10 Mg Reg (55)

Kept fil on minimum  
time just enough to  
time to break tube.

---

Keno 75 = put Lamp  
on & then brought  
it up to <sup>yellow</sup> red heat  
for 2 seconds while there is  
no vacuum so as to  
oxidize the filaments.

---

Keno 76 = oxidized by  
passing through Bunsen  
burner

(57)

77 Keno - abt a week in strong  
HCl = taken out soaked dist  
for  $\frac{1}{2}$  hour then taken out  
dried sent to Joe.

---

Keno 78 - boiled in ordinary  
Nitric several hours -  
(put in Carbon. Calcd) then  
boiled for couple hours in  
fuming Nitric. (may be nitron)

---

Keno 79 = Carbon Calcd put  
in Bone SO<sub>4</sub> - heated over  
~~water~~ sand bath -  
for say 4 hours -  
soaked  $\frac{1}{4}$  hour water



67  
Kens 80 - Cold Carbons in  
hot Cyanide K. for 4 hours

Kens 81 - Cold Carbons in  
hot Iodide K. for 4 hours  
Phenom - don't stick on taking out of water

Kens 82 - Cold Carbons in  
hot Chlorine for 4 hours

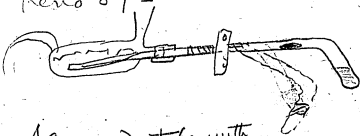
Kens 83 - Cold Carbons in  
Hot glacial phosph acid  
for 4 hours

Kens 84 = Cold Carbons <sup>(6)</sup>  
in hot Ferrous Chloride  
for \_\_\_\_\_

Kens 85 = Cold Carbons in  
hot Hydroiodic acid  
for \_\_\_\_\_

Kens 86 - Cold Carbon  
in hot Aqua Regia  
for \_\_\_\_\_

Reno 87 =



SO<sub>4</sub> inside tube with  
 piece ferrocyanide K<sub>4</sub> by  
 tipping goes into SO<sub>4</sub> to make  
Hcy after lamp worked both  
 clamps + bulb heated Lamp  
 not run very high say 30 @ 40 cp.  
 after gas in = 10 in order.

Reno 88 - same tube arrangement  
 same process only cyanide Ag  
 giving Cy residual. - 10 lamps

first worked Lamp up Rig. then  
 reversed poles - then opt-sealed  
 tube, then put in Cy by heating a/c  
 to 1/2 inch from head - then sealed tube  
 sealed off without lighting lamp  
 after Cy into it

Sept. 14, 1891  
was Monday  
Sept. 16, 1891  
was Wednesday  
T.H.S.

Screen Experiment started  
Monday at midnight 7/4/91  
now Wednesday midnight  
run continuous except  
3 hours —

89. Keno 5 Lamps tube  
got solid. then put 40  
lamps in series and cut  
one at time out every 7  
seconds until 1st peg when  
it was dull red - (Had heated  
bulb by bunsen when solid  
tube was obtained) then  
waked Regular -

90 Keros Fluoride Sodium in 504 <sup>(27)</sup>  
side tube —

No 1 had blue on clamp didn't  
get off

No 2 no blue on clamp —  
Couldnt get clamps off as there  
was action on phos anhyd

Keros 91 = cold Conc 504

1 week Open boil for

3 hours — then taken out

Union in water + immer

taken out dried + put

in box —

Keno 92 =

Drop 504 inside outside  
bulb - m3 Carbon -

Keno 93 - Sodium amalgam  
inside and phos only worked  
took long time -

94 - Sodium - worked  
vac =  $1\frac{1}{2}$  @ 2 hour unperf. vac

95 - 90 deg Frost

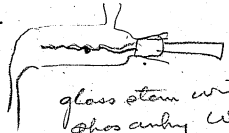
96 - 2nd lamp sodium

only - worked after 100  
than No 1 = 100 - how about  
not perfect vac

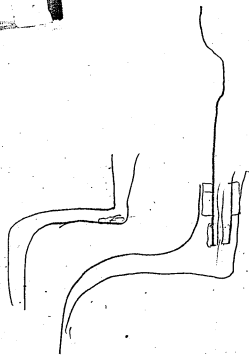
97- 50.3 mm 3rd Lamp after  
 Na had been in phos Cup all  
 night = took about 50 mm  
 got solid tube = didnt break  
 up very much when fed run  
 up - I run it up to finish  
 - 3 minutes. I noticed  
 faint lines between Hg lines

---

98-



glass stem with  
 phos anhy whos  
 good & well -



Kens 99 -

gas jet heater  
990 phos - works fairly  
quack except on tail  
end although it may  
be OK as Reg working  
gets solid tube quack  
at 1st

---

100 - 4<sup>th</sup> Lamp Sodium  
in phos tube -

---



101

*Wanted*

same as 98

102 - ~~102~~ second lamp  
heater -

104 - Worked by Edison Spoon  
phos Cup - ~~1st~~ 2nd run on the  
phos - got solid tube - put in  
plugs a run to full 120 valls  
second by second along box in  
30 seconds or less -  
pulled plug & let run solid -  
then lighted each again on 120  
for 10. seconds <sup>held by</sup> then run ~~again~~

possibly 9 run these lamps too bright -

104 -

boiled, then put on 200 volts  
+ run right up to blue off  
clamps + pulled peg + hot run  
solid, then lighted up at  
5 or 1470 volts for ~~the~~ minute  
then when found tubes solid  
sealed off = noticed ~~some~~

Every one that as soon as blue  
off clamps running little higher  
the blue disappeared in globe  
on all + come back on lower  
volts = Every one of 5 - showed  
this + this is new -

Notice on one leg of fil. near clamp is dark  
we put a paper spot on that side - which is side blue

103 = New small phos Cup with  
Spoon - 5 lamps W3 Run Reg  
1st run on the phos - got reports no  
blue appeared on clamps -

105 Kers -

Spoon phos Cup new  
phos M3 - put head Hg in globe  
worked up quick + minimum  
time lamp on to get air out  
plenty blue in globe - also  
blue didn't go away like 104  
blue on clamp went off not  
quite like 104 but a bit  
like Reg or little harder  
Carbon shiny - economy  
157 against 164 of 104 -  
although both M3 Carbons

~~to~~ Lamps 3 and 2 that  
didn't me lasted 20 hours  
other 90 hours all  
broke in 90 hours  
booy reports lamps kept always  
blue -

106 - worked on heater & Case (8)

107 Rechecked in 50 cp bulbs  
noticed that where bulb was  
annealed one of the lamps  
showed a streak of dew  
inside globe at greatest  
diameter - the moisture was  
so much that dew didn't  
disappear until nearly  
mm vac - then the little  
spot of glass was all  
wet & while the other 4  
got sealed off & worked <sup>reg</sup>  
with no blue on clamp <sup>thin</sup>  
the other lamp is still  
on 5th peg & there is no  
evid to see pbg due to #20  
incamp <sup>new station</sup> <sub>point about #20</sub>

108 - 4 foot tube to diminish  
amount of vapor -

109 - dipd Cbt dipsticks -  
not good dip but gave 40  
Lamp bank didn't give off  
nearly any air pblly dipi  
absbd phos + chl decomp & went  
to Hg & deposited - Arcd -

~~108~~ 110 Dip & bottles  
one of 109 -

K-111 - Run up to about 4  
candles in Mineral oil  
slowly for 10 sec found could not go  
higher 's had run by speaking & part

(25)

112 Reno - Run up on Municipal  
+ another 60x from nothing  
to 50 @ 60 Cp - took about  
3/4 hour no res except at  
about 8 @ 10 C. when very fine  
pin heads after that way up  
suddenly the finest pin heads  
showed - don't think about  
it up so high as they do reg  
as I was afraid would injure  
Carbon on account time it  
took - didn't have Summit  
off but ran for 5 sec at  
8 @ 10 Cp - Hg vapor there  
just same no blue appeared  
at clamps - 10 s at 1st  
then 15 then 1/2 minute  
then went by pin heads  
would if no pin heads in place  
but took 20 sec + more

112 No 2. 3 4 4 run same way <sup>(27)</sup>

notice shade of blackening on blue clamp side near clamp.  $\frac{1}{4}$  inch long also notice glass of inside part browned on blue clamp side a seam to me all others have this brown on N side glass - *Saturday 330 with home disquid with abundant keros*

113 - 5 files depd Cbt dipichlos  
old x xx -

114 = dep 17 - then Chl Sul in Kero  $\frac{1}{2}$  hour then in Kero then to jar - life 5324 hours  
my carbon -

115 = dep 17 then Chl Sul in Kero - then in Kero alone out dried then redip 17 then Chl Sul in Kero then in Kero + out -

Oct 11<sup>th</sup> 91  
116 = Furrice Experiments (5)

Amylen in Kerosene 25 mm. diam.  
whitish yellow - Carbon box  
in tube - very little Lamp  
black - These Carbons are  
NC Carbon on top of Charcoal  
nearest to where gas first enters  
& had some hairs on part  
have been brushed off by  
Camel hair -

---

117 - same as above but  
further on on top Charcoal  
towards exit

---

118 - ~~same~~ same as above  
but immersed in 16 mesh  
Charcoal -




2nd Run Amples but  
which yellow 1 hour and  
non luminous flame - did not  
seem as silvery as other in fact  
did not notice they are any more  
silvery than Regula's full  
hairs which brushed off  
pretty near all - Charcoal  
was not shining very bright  
I put lamp black in each time  
to lay Carbon on may do better  
possibly O<sub>2</sub> got in & lamp bk  
being sensitive soiled Carbon  
but charcoal being more used  
became insensitive

---

Oct 11/91

120 Run through Cely gas  
didn't seem to brighten

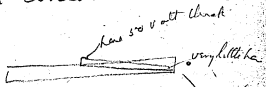
121 - used iron side tube  
heated by burner side  
tube have  $\approx$  or 10 grams  
Coal tar made them slightly  
silvery —

122 - Napthalamine  
this looks ok  
didn't whiten Carbons much but  
had they been at ingress end would  
have been bright.  
propose  silv.

Oct 11 '91

123 Nephthali - ~~no~~ no deposit didn't send muff. through going try send large quantity through next time.

124 - It worked - but cbs stick together. twas 50 vult thick on circle which was on top



gave goe somewhat dry area to test fall Cp but think they will be dull on top

Oct 11, 91  
125 = Tan in <sup>new</sup> retreat.

silence. then lovely didn't  
stick but shale - deposit  
very thin - wanted more time  
say 40 minutes next time -  
5 set up north Oct 17 Lfg 3304

126 - Same as above - but same  
stuff = not so good - 45 minute  
run - 4 set up 26km 3304

127 = BS. On top charcoal  
→ gas → greatest deposit here

good very silvery - fundamentals  
heavily abraded for 1/2 Caban  
at ~~2~~ Loop further from gas  
good but at tall very silvery but  
not so much deposit as had by retreat,  
as was long time before silencing gas  
came back towards end quite thick with 3" long

Oct 11 91 <sup>200</sup>  
Numerous flaws - had this been  
kept up the deposit would have  
been sufficient all over to eliminate  
fracturing - Evidently the  $\frac{1}{2}$   
Cores are either too high or they  
retain gas - Rouse Trails higher  
and that don't do it spread files  
out in various directions ~~etc~~ etc  
find out trouble =

Oct 12 '91 703

Deposit reports =

Experiment 1

(101)

3 inch tube inside measurement.

Turnout for carbon - Big return B.B.

Started 635 - with some little smoke coming out. at 652 the smoke lightest at 655 - turned around gasless or almost flame is now  $3/4$  long  $1/4$  inch of which is luminous at 7 pm  $1 1/4$  long  $1/2$  luminous - Big tube yellow 702 flame 2" long 1 1/2 luminous jerky turned down gasless almost to present to which flame wraps deposit 704 3" flame with all luminous slight appearance lamp like at tip 710 still  $2 1/2$  @ 3" long almost 2 1/2 on tip 715 - considerable lamp black - 719 shut off - found everything covered with lamp black - this was done in the start as no deposit showed on carbon when they were bright - or circulation of gas too slow allowed for too rapid deposit.

Expt 2 Oct 12 91. 7a<sup>(13)</sup>

Same B6 in rebot but used about  $\frac{1}{2}$  oz Benzene to clear tube, there should not have been but a grain or so as it made too much gas for considerable time

Started heat right away & put burner under rebot, flame appeared at 50 at at once

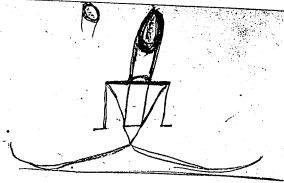
Started 830-

Benzene vapor gave pulsating flame at intervals of 2 sec but stopped at was almost two inches long at 834 tube dull red + benzene flame 3 inches long yellow. 835 very large Benzene flame

larger than gas jet smoky 836 flame  $\frac{1}{2}$  inch luminous is smoky 840 flame smoky  $1\frac{1}{2}$  long + luminous 842 Benzene flame gone B6 flame came 843 2 inch long smoky + the yellow dust in coils

848 big as a gas jet smoky pulsating turned off gas on rebot at 853 when flame died down to pin head & at 858 shut off gas from big burner

Carbons or holders all lamp black but signs of deposit fluffy + hair on about inside filament see report (1)



Experiment No 3 Oct 12, 91. (95)

New dose BL in large retort, 128  
 lighted retort about  $1/2$ " flame at 955,  
 lighted gas in furnace at 957. - ~~128~~  
 958 - non inflammable air blowing  
 through 52 it tube 10.1 tube dull red,  
 raised flame in retort little, 10.4 into  
 Red - non inflam air from 52 it at 10.4  
 raised flame in retort, 10.7 non luminous  
 from flame at 52 it - 10.9 tube yellow  
 flame self supporting  $3/4$  long blue entirely  
 non luminous - 10.11 2 pm flame  $3/4$  blue  
 non luminous - raise out in puff of white - jerky  
 10.14 -  $3/4$  long shows signs of luminosity -  
 tube bright yellow - took top out carefully,  
 10.16 - flame indistinct. semi luminous not blue  
 no signs of volatized particles in center of  
 flame tube very bright yellow 10.20 flame  
 - <sup>rough</sup> not long non luminous not blue, looks like  
 alcohol flame with slight amount sodium - it  
 or say as top of alcohol flame  
 10.23 vs little top of luminosity in  
 center of flame on exit.



Tube whitish yellow <sup>ret 12 91</sup> 1026  $\frac{1}{2}$  pin - 8x11  
flame about same size - luminous top in center  
about  $\frac{1}{2}$  inch long - raised heat on retort. (107)

1030 flame  $1\frac{1}{2}$  long  $\frac{1}{2}$  inch luminous  
no lamp black no solid dust,

1031 flame 8x11 2"  $1\frac{1}{4}$ " luminous

Tube bright yellow - 9 turned off little  
of gas on big furnace - 1032, turned

gas way down on retort as lamp black  
commenced to show - flame now  
dying down on 1 inch - luminous  
no lamp black 1034 turned off

gas on Retort, flame  $\frac{1}{4}$  inch luminous  
1035 8x11 flame gone - 1036  
shut gas off furnace -

No lamp black anywhere carbon  
very silvery - but deposit light  
some fine fluff on them but not  
to harm = Keno 126.

from lighting of furnace  
 6 1/2 minutes blue flame appeared  
 7 minutes 1st appearance tiny  
 nonluminous flame -  
 9 minutes 3/4 inch pure blue  
 15 minutes flame not blue - brown  
 27 tiny tip of burner visible  
 31 minutes 2 long 1/4 of it luminous  
 32 3/4 luminous looking long bright  
 39 horns

Expirt 4 -

Oct 12 91

(109)

~~Stagnation~~ ~~Fire as in No 3~~ Bignolet  
 Put in a fresh dose of BS same amount  
 by guess as before = 9 am going to work it  
 as in experiment 3 as can be continued longer  
 to see if can get heavier deposit also turn  
 furnace more to stop fluffing -  
 lighted rebort 1/2 5130 flame at 11 23 pm -  
 noticed at 11 25 non inflammable air coming  
 from bit lighted furnace at 11 26, spit top on  
 11 27 non inflam gas setting @ 11 30 like dull red  
 scarcely any of any gas from it raised flame on  
 hot dull - 11 31 1/2 spout black flame at 78 ft  
 11 32 1/2 raised flame on rebort, not containing  
 blue flame - 11 33 first appearance tiny  
 non luminous flame - 11 34 tube dull yellow  
 11 35 3/4 inch pure blue non lum flame  
 little spout at time but continues -  
 11 38 - flame 3/4 inch non lum. Pyc. spout -  
 when it spouts it still non luminous - rebort  
 6 inches long 11 40 tube yellow  
 11 41 flame 1/2 inch non luminous not  
 pure blue but like top alcohol flame  
 11 45 spit flame non lum but with blue 1/2 inch  
 11 45 tube pure yellow look small  
 top off -

Oct 12 '91 (11)

1146 flame one inch not blue nor luminous

1150  $\frac{1}{2}$  inch flame not blue nor luminous

1150 raised flame on retort to full - 1153

slight tip of luminosity inside flame which is  $\frac{1}{8}$  long. 1155 luminous point  $\frac{1}{8}$  inch flame  $1\frac{1}{2}$  long. 1156 luminous point  $\frac{1}{4}$  long

1157 flame 2 long  $1\frac{1}{4}$  luminous

1158 turned down retort more than  $\frac{1}{2}$  way  $\frac{2}{3}$  flame nearly  $\frac{3}{4}$  luminous with tendency to lamp black on tip by turning down flame shortens, 9 now raise it again but not full at 12

12 midnight = 12 flame about  $1\frac{1}{2}$  long  $\frac{1}{2}$  luminous  $\frac{1}{2}$  yellowish that being the outside of flame 1203 - flame 2 long little more than  $\frac{1}{2}$  luminous with lamp black tip slight although it don't deposit on tin held over it flame  $1\frac{1}{2}$  at 1204

turn down retort little, tube bright yellow & turn off little air & gas

1205 - flame 2 nearly  $\frac{3}{4}$  luminous yet great deal Hydrogen has 2 lamp black

0.24.22-71  
making horns on tip flame

(118)

1208 - flame  $1\frac{3}{4}$   $\frac{3}{4}$  luminous very small  
black horns - tube bright yellow turned  
off little air + gas - no sign yet off any  
solid dust in center of flame - 1210  
flame  $4\frac{1}{2}$  - luminous faintest trace of  
black horns - about  $\frac{1}{2}$  inch from base flame  
non luminous + only a faint yellowish outer  
edge - 1212 - horns disappeared - flame  
resembles ordinary gas - tube not quite  
so bright a yellow but bright - 1215 - flame  
same - 1216 - flame same but spurt  
little, tube bright yellow 1219 -

1221 - flame same except  $\frac{3}{4}$  long  
+ spurt - 1223 raised flame on  
retort, 1225 - flame  $1\frac{1}{2}$  same channel  
1227 - flame 2" long same channel  
no horns of black - 1228 turned  
flame out on retort 1231 turned  
off gas from furnace - flame at  
best a pin head - JVS

Sept 12 91 (115)

Inrupt filled at bottom with  
 lampblack but Carbons more silvery  
 than Expt 3 with even less hairy  
 The deposit about  $\frac{1}{2}$  had the fine lamellae  
 at band but not on top - none  
 stuck - most of the lampblack was in  
 bottom of semicircle - evidently the  
 luminous last end gave the lampblack  
 still it might have been due to  
 the extra bright yellow verging on  
 white that I had there at one  
 time, this temperature may have  
 decomposed hydrocarbons & gave  
 lampblack which would not have  
 decomposed if temperature had  
 been lower - as the band gets  
 most deposit we are going to  
 use a plate to retard flow of  
 gas just at file

+ also conduct  
 heat to file holder  
 by conduction & radiation



2 set-up number 2802

Revs  
129

11 minutes 1st app tiny non lum  
13 1/2 blue flame 1/2 inch long

Experiment 5-

Oct 7 291

(117)

1 1/2 oz BS



plate turn pit -

13 1/2 inch rest

Lighted retort flame about 1/2 flame  
at 304 am - notice 305 non inflammable  
air coming out slightly Lighted big flame  
at 307 considerable non inflammable coming  
off 309 non inflam still coming 311 ditto  
312 tubered, 312 non inflam still coming  
314 raised retort flame little, 315  
first appearance of tiny non luminous  
flame tube dull yellow at 317. pure  
blue flame 1/2 inch long 317 1/2  
319 - tube pure yellow - I'm not going  
to hit tube get so hot this time  
320 flame pure blue 1/4 @ 1/2 - 322. inch  
long bluish not pure blue 324. mauve  
1/2 inch long. tube bright yellow turned  
off little air regas - small top seen off  
3 or 4 minute forgot to record time  
326 - flame got small turned on more

Oct 12 7 (120)  
flame on retort, 327. flame non luminous  
tip of alcohol lamp color narrow flame  $1\frac{1}{2}$   
@  $1\frac{1}{4}$  long semi trans but no luminous tip in  
center at  $328\frac{1}{2}$  - tube yellow. 330  
semi luminous no luminous tip -  $1\frac{1}{4}$  long  
338 - non lum no lum tip flame  $\frac{1}{2}$  inch  
335 - knocked lamp block off retort flame now  
with non lum @ semi no luminous tip  
no horns luminous tip appear to  
long at 336. at 337. flame  $1\frac{1}{2}$  long tip  
 $\frac{1}{4}$  inch (turns) flame on retort 338  
same - 338 flame  $1\frac{3}{4}$  luminous part  $\frac{3}{4}$   
340 flame 2 inches with non lum border  
one inch luminous no horns  
tube yellow - 341 - same as flame -  
Notice top to the tube flames not apart  
343 same flame - 345 same flame no  
horns of lamp block - sticking at non lum border  
& flame non lum  $\frac{1}{2}$  inch from edge -  
flame  $1\frac{3}{4}$  long 347 same flame but  
less luminous

349 -  $1\frac{3}{4}$  long about  $\frac{1}{2}$  non lum Oct 12 91 (12)  
350 flame  $1\frac{1}{2}$  long more than  $\frac{1}{2}$  non lum  
tube pure yellow - 354 same flame  
so far flame only spouted out once & then  
went bad. 356 - flame inch  $\frac{3}{4}$  non  
lum 4 am - flame  $\frac{3}{4}$  inch  $\frac{3}{4}$  non lum  
404 flame  $\frac{3}{4}$  -  $\frac{1}{2}$  non lum - tube  
yellow - 404 turned off retort  
flame - 405 flame  $\frac{1}{2}$  inch  
for heat diminished heat in tube  
while something is passing from  
retort may give Lampblack  
I keep flame on 2 minutes after  
flame disappeared - flame stopped  
407 - 409 turned off 9 am for  
big retort, = all covered with  
Lampblack - Carbon had been  
deposited on but afterward  
Control with Lamp block



Oct 12 7/2  
The plate is a failure as it gets  
covered with lamp black & pbbly  
dirtyes the Carbons, however  
I am going to leave the  
plate and try it again sloppin  
as soon as luminosity shows  
itself

Expt. 6.

Oct 22 91

(125)

20 oz B.S. Big retort = plate  
transport - The increase from  $12\frac{1}{2}$  oz  
to 20 will give a larger non luminous  
flame as I propose stopping when  
signs of luminosity occur -

505 lighted retort full blast,

508 started furnace

512 time luminous flame appears

513 " " "  $\frac{1}{2}$  inch long

514 notice this blue flame is tipped  
with  $\frac{1}{4}$  inch luminous flame

515 flame  $\frac{3}{4}$  long  $\frac{1}{2}$  off which is  
luminous -

516 tube full red, very yellow

flame  $\frac{1}{2}$  inch long  $\frac{1}{2}$  luminous

517 -  $\frac{1}{4}$  inch "  $\frac{1}{2}$  "

multicolored full flame seen  
on retort.

518 suddenly flame 3 inches  
long for 10 seconds. then goes to  
blue with inner  $\frac{1}{8}$  inch top of.

luminous flame  $1\frac{1}{2}$  <sup>foot</sup> <sup>1291</sup> long thin - at (23)  
519 - 521  $\frac{3}{4}$  nearly all non luminous  
except  $\frac{1}{16}$  inch inner tip - 523 tube full  
yellow - took top off - 525  $\frac{3}{4}$  flame  
reverts - no luminous tip or faint

529 - flame  $1\frac{1}{2}$   $\frac{1}{4}$  inner luminous -

Shut off retort. flame  $1\frac{1}{2}$

$\frac{1}{2}$  with inner luminosity -

530 flame with  $\frac{1}{8}$  inch inner luminous

533 flame dies out, tube whitish  
yellow turned off some gas re-vent -

535 turned gas off by furnace

scarcely any deposit -

hardly any lamp black

though Carbons trace

Some on  
don't make Curve  
no

Expt. 6. -

Oct 12 91

standard flame retort fired 11:58 am -  
at 12:02 started by furnace -

12:10 - 1 1/2 @ 2 inch 1/2 x 1/2 lumina

iron lum - lamp blacky  
spurts smoky center to flame  
white smoke - oscillates

flame 12:11 2 1/2 @ 3 long smoky  
white center - flame size go set

red from iron - smoke tube  
yellowish red at 12:13. at 12:14

flame big thin 7- jet awful hot  
smoke from tips flame at

hole where rod comes through yellow  
smoke - Tar - 5- inch flame

clouds lamp black blown  
straight out - had burn down

retort flame 1/2 5:13 a. at 12:17 -  
Tar dripping from exit tube

blowing strong <sup>Oct 12 91</sup> spilling 1218 (131)

tube full yellow - flame 4 inches  
long round - Redish dbk top -  
fast dying down 1219 turned

up started to full again - now

1221 blowing 4 inch flame -

Tan coming out tends stop hole

So it shut off Balot 1225

flame 1 1/2 inches -

file good deposit but all

lamp black p. bly out of full

disposed

130 Rona. Eshant. D. by jet 12 1/2 (43)  
any other  
clamped

Friday - Oct 12 71 (135)  
Left at 1010 - let in say 300  
drops. B&B - Bayol - at 1016  
big hole that being drilled at 1011  
dull red - I let in up to the  
time at 2 o'clock - 1016  
- shut down 1110  
didn't go above dull cherry  
- 1016 - 1016 - 1016  
cherry - no lamp block showed on  
turn rod when pulled way out  
didn't let more than 100 drop  
in after 1015 =

No deposit bits lamp block  
in bottom on nickel

Experiment 8 Oct 12 91 (137)

Plated 1150 - units whole  
yellow full det in 3 or 4 drops  
4 times - stained with 6  
drops in Gafon highly  
at 1227 put in 4 drops flame.  
tiny - but luminous excise 117  
10 minutes - 10 drops 1230,  
no deposit no sample



131 Keno -

Oct 12 91

(39)

Reqs dep'd 90 then as the sue to put  
Keno soaked several hours &  
then ran through regular  
Carbon Dioxide & washed  
but very shiny & no hairs  
from cloth -

---

Oct 12 91 (141)  
Expt 9 - small relat -  
Coal Tar pitch - heated up <sup>very</sup>  
1 minute before lighting furnace  
started 4:15 - jerky blue  
flame after 5 minutes -  
no luminous flame at all  
blows - jerky through whole  
time stopped 4:47 - tube  
always bright yellow  
No perceptible deposit  
on Carbon some hairs just  
shade lamp block on nickel  
bottom -

Expt. 10 - Oct 12 91

(143)

Anthracene 1 @  $1\frac{1}{2}$  oz Crude

Started 530, -

boiled -

We heated retort after 10  
minutes too strong when  
suddenly lost vent over  
into hot tube + made  
lots gas + deposited  
damp block -

Experiment 12

Oct 13 91

(145)

Temperature in Florence flask on sand  
bath passed City gas through  
tube by lighting - then connected  
to flask ~~gas~~ started furnace

1134 am - <sup>city gas</sup> gas is burning <sup>1/2 inch</sup> ~~at 1140~~  
at 1140 at 1150 there is still gas  
now less coming but this may be the  
impure vapor - 1151 - Temp slight  
boil = flame immaterially temp  
horns lamp like & burner reddish  
1153 - tubes reddish yellow 1210 shut off

No deposit or fuming  
black - scarcely any Temp  
went through on it Condensing  
before it got to tube

Sept 11 =

Oct 13 91

(147)

lighted Retal of gas smoothly

852 pm - put 1/2 oz Hydro-Quinone

in small retort, run for 1/2

hour - Lamp black (dry only)

Carbon smoothly no deposit

ng

Expt 13 - Oct 13 '91

Metachloral - retort started  
retort + furnace 1007

stopped at 1041-

no deposit no lamp black

Exp. 14 =

Oct 13 91

(157)

4 of B8 big rebolt lighted retort  
at 1257. stable furnace at 1,01  $\mu$   
exit blowing non-inflam - tube

<sup>light</sup> Cherry red am going to keep it at

this ~~is~~ slight amount inflam

g. coming 107 pure blue flame

when appeared alcoholic flame

now 173 - alcoholic except tip somewhat

luminous. slight tendency to blacken

114 - now  $2\frac{1}{2}$  inch luminous

flame tube Cherry Red

not light cherry slight dust

in center flame  $2\frac{1}{2}$  or 3 long

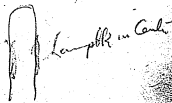
luminous + lambling -

tube bright cherry Hg -

136 started <sup>test</sup> no deto & no  
samples

Sept 15 = Oct 13 91 (153)  
4 oz BB tube to be  
Kept yellow - <sup>orange</sup> stalled retort.  
236 pm = stoped 333

found lamp block at ingress end  
in nickel and semi lamp block in  
from scale  $\frac{3}{4}$  rest way back at  
Exit end where scales was dense  
Carbon blocked.



no deposit -

This blk - Center shows (BB) the  
deposited low temp - the flame  
was always  $1\frac{1}{2}$  @  $1\frac{3}{4}$  long  
 $\frac{1}{2}$  diameter  $\frac{1}{2}$  non liner at base  
There was plenty Carbon here  
if flame  $\frac{1}{2}$  @  $\frac{3}{4}$  inch in



Oct 13 91 (155)  
hours run at deposits right for  
56 minute run there wide as  
plenty Carbon -

Expt 16 - 4oz BS -  
going to run tube yellowish  
white & keep  $\frac{1}{2}$  @  $\frac{3}{4}$  flame  
~~with at 56 minutes -~~

Staled 4:19 - passed City  
gas through - Lighted Retort  
City gas coming out 2. inch  
long many after lighting  
from over sloped 5-10. found  
nickel turner melted to  
iron or rather iron  
melted -

Expmt 17 =

Oct 13 91

(157)

4 of 135 big retort - new tube  
Carbon box filled (10 mesh) carbon fil  
on top. made tube has non scale  
in addition I put about 2  
thumb full ferric oxide in fine  
powder all over inside tube =

Started retort 1222 am -

lighted furnace at 1225 am -

Exit tube blows considerable at 1225 $\frac{1}{2}$   
non inflam - at 1230 still blowing & non inflam  
big tube very dull red at 1232 blowing & non inflam  
big tube yellowish red - at 1235 blowing more but  
not sufficiently inflam to self ignite.

1235 $\frac{1}{2}$  inflam gas appears self lighting -  
Absolutely white faint deep blue base & long  
the Alechilly being  $\frac{5}{8}$  long I notice the usual  
luminous tip Big tube at 1240

light orange yellow took crucible out.

Oct 13 91

#291 base of flame  $\frac{1}{2}$  long very blue flame  
 inch long inner luminous top  $\frac{1}{4}$  inch  
 outside yellow fringe - blue shows CO lines  
 in spectrum - 1246 tube whitish yellow  
 turned over down flame  $\frac{1}{2}$  inch yellow  
 Very little luminous blue base still there  
 raised heat on retort, the flame is yellow  
 + Na line is strong paly b2 blue if turned  
 for Na as CO lines show along even where  
 the flame is yellow with Sodium 1253  
 luminous tip seems to have gone or is  
 masked by Na - flame now  $\frac{7}{8}$  long  
 $\frac{1}{2}$  blue base - tube whitish yellow  
 1255 flame  $1\frac{1}{4}$  long very Na yellow  
 CO line hazy blue base  $\frac{1}{16}$  -  
 1258 flame  $2\frac{1}{4}$  inch blue base very  
 yellow - luminous inside  $\frac{1}{16}$  long  
 1 AM flame  $2\frac{1}{2}$  long lum top  $\frac{1}{8}$   
 Blue  $\frac{1}{4}$  inch - very yellow Na - CO line  
 faint along 104 took flame away  
 from retort

105  $\frac{1}{2}$  in. flame  $\frac{1}{8}$  lay - <sup>not</sup> at 139  $\frac{1}{2}$  shut <sup>(6)</sup>  
gas off furnace. No lampblack  
inside or on box no deposit on  
Carbons except few hairs on Carbon  
nearest  $\frac{1}{8}$  in. -

Experiment 18

4 oz B3 - 4 or 5 -

Unwieldy of Ferric Oxide -  
lighted. Retort at 134  $\frac{1}{2}$  took light  
away at 135 as there is good  
non inflame blow. lighted furnace  
at 135 am - 141 flame not self  
highly notice somewhat inflame  
& am pulling light under & away  
from retort so as not to disill  
off too much before big tube  
gets yellow

Oct 13 91 (63)  
145- tube yellowish Red - w/ flame gas  
but not self sustaining at 145 - at  
149 - flame pulsating so it won't  
remain lighted - tube yellow -  
flame wld be self sustaining but  
pulsating out. 153 about  $\frac{1}{4}$  inch  
but blows at intervals - very strong  
heat as noted yet very poor  
flame at exit - yellow flame  
Even when alined don't notice  
blue base but flame being only  $\frac{1}{4}$   
inch long probably won't show it.  
157 - flame already  $\frac{5}{8}$   $\frac{1}{32}$  blue  
base - yellow - 2 AM flame  $1\frac{1}{2}$   
blue base  $\frac{1}{8}$  inch diameter inside top  $\frac{1}{16}$   
flame yellow for Na CO spectrum

201 flame 2 inches long <sup>oct 3 7/8</sup> 1/4 inch blue  
base. Lum tip 1/4 inch yellow color from  
Na - 202 - flame 2 1/2 long  
3/8 blue base - 5/8 lum tip yellow.  
205 - flame 3" 3/8 @ 1/2 blue base -  
inch lum tip CO spec yellow -  
207 - 3" flame 3/4 lum tip  
base flame. Mauve like a some bluish  
very strong flame on retort,  
increased flame on retort at 210  
flame at 210 3 1/2 long mauve  
base 5/8 - lum tip 5/8 yellow Na  
outside - CO spec 212 suddenly  
tar came through or more pb  
as drops were transparent  
water. flame full simulation

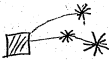
- Oct 13 71 (157)

mess there is no doubt it was  
water from reduction of Feas  
condensed in long tube & then blown  
out by gas at 216 put  
flame back under rehot.  
flame at 216 1 inch - no  
waxier tip no blue base - yellow  
219 - flame 2 1/2 inches narrow  
base 1/2 inch - wax tip 3/4  
think narrow color is pure blue  
masked by the yellow -  
at 224 raised heat on rehot.  
3" flame 5/8 blue of narrow  
color on base + 1/4 wax tip  
flame yellow - little horns appear

230 - flame <sup>yet 13 71</sup> little oscillating  
3 @  $3\frac{1}{2}$  long horns with black -  
with  $\frac{1}{4}$  lamp tip - 5% @  $\frac{7}{8}$  blue  
of mantle but yellow no emission  
Shut off gas from about

230 - tube has been whitish  
yellow right along - stopped at

232 - Carbons stuck  
together somewhat silvery  
but coated with fine hairs  
with balls on ends of hairs  
of still finer hairs



\* this way all over no lampblack



now I think this phenomenon is  
due to electrification from heat  
& decomposition & am going to  
try it over again putting an  
upright in lid of retort & connect  
to city gas, then with rubber  
ball & while retort is going  
gas oscillating waves to  
stuff from retort.

Expmt 19 - Oct 14<sup>th</sup> 91 <sup>(173)</sup> all these things

Big Retort. 4 g B<sub>2</sub>O<sub>3</sub> -

Still plenty Ferric chromate O<sub>2</sub> probably in tube - Have Cock on between big tube & retort so we can pulsate the gas -

Lighted gas on retort. 420 -

Cock open - 520 tube 3 broke off & made hole twice as large probably here to make it larger - sealed furnace -

424 as 520 blowing - we put flame on off Retort so as to get Big tube to yellow - since I knocked

scale off big tube I can guess

temperature better 432 flame inflammable but don't self sustain tube yellowish red - flame self sustaining blue base & non

Lumina stopped at 5 o'clock <sup>Oct 14 91</sup> <sup>(175)</sup> AM

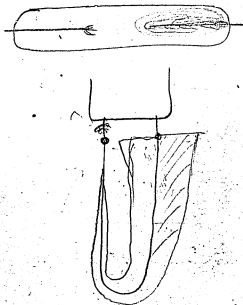
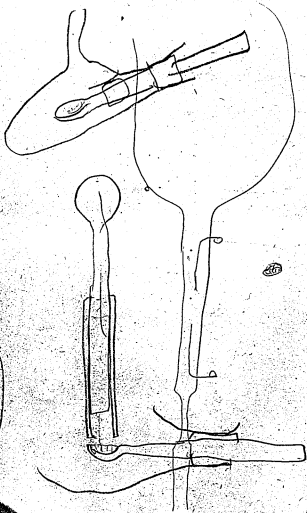
during whale time. I opened  
& closed Cock at intervals  
of 10, 15 & 20 seconds. The  
flame ~~out~~ when retort hot  
shot out 6 @ 10 inches &  
then suddenly died away to  
2 inches then - second to inch  
most time was a Lumina  
inside tip & when flame  
returned from shooting nothing  
large quickly Lumina say  
3 @ 4 inches. Tube was  
nearly all time bright  
yellow somewhat whitish.  
This plan I think must  
be OK = no deposit fine has  
fluff no long fibres  
or bits.

0  
Started another  
Book

Oct 14/91

JNE

(187)



**Notebook, N-91-10-14**

This notebook, which covers the period October 1891-October 1894, is a continuation of N-91-09-03. The book contains notes and drawings by Edison regarding filament deposit experiments, nos. 20-35. There are also notes concerning related experiments, including tests of basic pH solutions for treating filaments. At the end of the book are two loose pages of notes by an unidentified experimenter. The spine is labeled "124." The book contains 197 numbered pages. Two loose pages have been inserted into the book.

Blank pages not filmed: 30-31, 66-67, 118-193.

N-91-10-14

x E-172

Continued

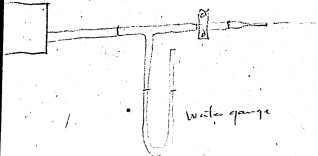
②

Oct 14 1891-

Experiments on deposited  
Carbon on Carbon filaments etc  
in hot tubes from Carbonaceous  
Vapors

Experiment No 20

4 oz Bull's E = Big retort Carboron char -



lighted retort 325 - lighted furnace  
330 pm -

going to deposit under slight pressure

340 6" pressure - shut off

retort 4 pm - very little of a  
hemispherical to flame due

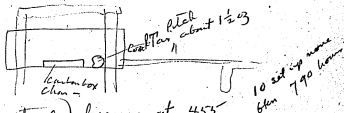
thin film having no deposit

(10) Summary



Experiment No 21

Keio 133



stained furnace at 455

455 - flame blue - then blue &  
dummock tip yellow vapor appear  
in glass - few sec quite smoky  
flame 1 1/2 long at 5 pm  
smoky center appears - flame at 5 -  
flame very luminous.

When I have much pressure  
flame blue with long  
& keep from 5 - till 506  
then increase to 2" pressure  
flame 1 1/4 long 1/4 blue base but  
dummock no yellow

Slight horns 510 took pressure off <sup>(2)</sup>  
flam  $1\frac{1}{2}$  smoky horns  $\frac{3}{16}$  thick  
very luminous  $\frac{1}{8}$  @  $\frac{3}{16}$  main base  
Central direct stream  
tube yellow - 515 I pulsed it  
10 times by holding clamp shut  
until 3 inches pressure then let go  
pulsed it between 515-4520.

Shut off at 5:30 pm -  
Carbons along no block not heavy  $\frac{1}{4}$

Oct 14/91

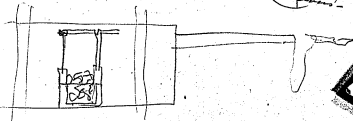
②

Expt No 22

In Center of Tube round iron box  
with say  $1\frac{1}{2}$  of Coal tar pitch.



Carbons smolder  
up but had  
heavy gray deposit  
- plams -

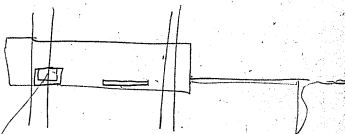


I light furnace run to yellow  
fast as Cow keeping about 2 inch  
pressure + spouting it.  
started furnace at 1012 pm  
stopped at 1037 - Tube  
orange brown last  
flame  $5/8$  - very luminous  
spilled  $2\frac{1}{2}$  lbs

Experiment 23 —

Oct 15 71

(11)



CT Push



open both ends

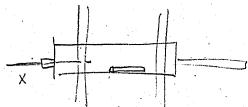
file

Started fire at 1:16 am Oct 15/71

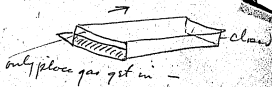
Slepped 145- ~~depend on~~

Too much tea

Expt 24 - Keno 132 (3)



Put small piece <sup>piece</sup> in at X by rod -  
Same box as Expt 23 - Carbons good  
shiny but thin = while I only put  
in altogether say equal to hickory nut  
pieces polystyrene fig - to prevent solder  
rush + carbon deposit - I use ~~same~~  
pitch at same rate as before but fix  
by this



Stallion 605-aw 15/91  
stopped 705

15  
I kept continuously pulling  
in small pieces loosely -  
withdrew rod slowly so as  
not to puff flame at exit  
or disturb inside gas by sudden  
movement - at one time for 5  
minutes we had almost white  
heat it was about 658 to 702  
only 2 top burners were then -  
at no time was the exit  
flame more than  $\frac{1}{4}$  inch  
or generally more -

Calorim not noticeably  
obtained no heavy gases  
no deposit.

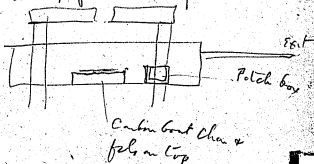
Expt 25 =

15 Oct 91

(17)

Started 4.0 pm

By passing products' dicamp. Plat  
over carbon large amt lampblack  
etc passing over while if you  
put Tin towards Fe + Carbon  
back beyond tin no lampblack  
gives back only compounds hence  
I think better results as in  
Expt 21 = I now use the round  
iron box filled with pitch

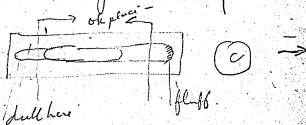


Non inflame smoke comes off  
 first if held right in flame it  
 burns freely shows no gas only  
 solid particles. heat of big tube  
 not being sufficient to ultimate  
 decomposition smoke is mixture of  
 white & yellow - 411 self sustaining  
 flame -  $2\frac{1}{2}$  long very smoky  
 central dust all yellow  
 flame 3 @ 6 inches very irregular  
 - pushed it twice shows considerable  
 pressure - 418 tar cones -  
 4 inch flame very smoky 419,  
 rather steady now primarily  
 pulsating tube dull yellow -  
 422 flame  $3\frac{1}{2}$  very smoky  
 no tar - dusky center 90 pct yellow  
 10 pct white dust, 424 flame 2" long  
 very smoky 68 pct white 40 pct yellow



Tube dull yellow - 428 flame  $1\frac{1}{4}$ .  
very smoky 431 flame with long  
scarcely any smoke but luminous  
90 pct white. I find though  
the flame don't show much heat  
or smoke by holding tip in al  
flame than plenty smoke -  
~~the~~ flame  $\frac{1}{4}$  inch stopped 439.  
a Run of 30 minutes -

Carbons good shiny



We now rearrange them &  
put them back in Carbon Box  
reload patch Cup -

Sept 26 - Oct 15/91

(23)

Experiment 25 - duplicated  
same Carbon just back + pellet  
Cup reloaded -

Stalled 503 pm - 509 -

4 inch flame very smoky  
Central dust .95 pct yellow dust

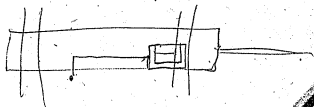
This went little quicker on  
account furnace being hot.

5-11 5 inch large very smoky  
flame Lamp 60K flying all  
over time - 5-16 flame 2 1/4

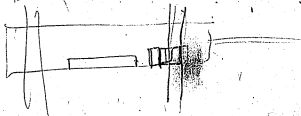
very smoky 50 pct white  
over time 5-20 2" flame smoky

I find that owing to fact  
I forget about thickness  
of tar box the more

have following position in tube



shown to:



525 -  $1\frac{3}{4}$  long flame very  
smoky 530 flame  $\frac{1}{4}$  inch  
little smoky than top bed in  
alcohol flame -

shut off 533

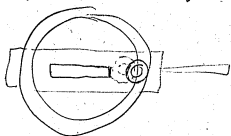
turning to too rapid way of pitch  
back doesn't bring over head Carbon

Expt 27 -

Oct 15/91

(27)

I again duplicate 25" but  
start furnace only about 100 fah -  
and this instead of as before as  
shown by dotted lines with hot furnace



Started 740 pm at 745 smoke  
coming not self lighting but ignites  
in flame - flame 5" @ 6 long  
+ self igniting blows good.  
7 inches long great sample  
at 748 = 749 Tar -

(29)

Tube light yellow - Top off -  
750 - Top - flame 3 @ 4 long vs  
smoky 753 still Harry - flame  
3 @  $3\frac{1}{2}$  long smoky

Top balling hole had Knave  
piece off - hole larger flame.  
sluggish not now the back  
pressure as before flame goes  
up  $1\frac{3}{4}$  long smoky -

---

802 - flame  $1\frac{1}{2}$  smoky tip

---

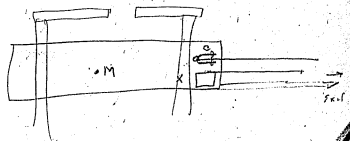
805 - same " "

---

809 - about  $1\frac{1}{4}$  flame smoky  
shut off —

Experiment 28

started 1015 -



Idea is to make Chlorogen by moving the box towards X then twist file at C to Coat, then withdraw the box & thrust file towards M which is to be <sup>red</sup> yellow heat is taken - then go back & do same thing over again -

stopped 1045 Carbon fil rod got sticking shut off -  $\frac{1}{4}$  inch flame composed of smoke came out continually after

The first 6 units or so

Tube full yellow -

notching but even heavy  
deposit all over, not slotted

deposit too rapid

134/Kend

30 hours  
all bucket

Expt 29 - duplicate of 28<sup>(2)</sup>  
but with better working  
rods

Standard 1250

good

Kens 135 =

gray - but showing under microscope on  
black backgrounds and -



Sept 30 16 Oct / 91

(39)

slated 941

stopped 1016 -

Carbons on dark Guelphian Shale  
Enew. Kers 136

Expt 31:

Naphthalene =  one cube

started at 1128. - 1

Stop 1155 - good. better than

Expt 30 more even more shiny

Naphthalene too sensitive but with  
longer tube so it could be kept to  
good low temp. think it would  
be good its better than pitch,

Kens 137 -

Expt 32

43

Stained 12 55-

Harshest perfume -

Carbon returned deposit of  
Lampblack 138 Kew

---

Expt 33 -

(45)

Tetrachloronaphthalene

plated 233 am —

stopped 2:57 -

Carbons I think show that way on air background & are probably nice didn't notice the frosted silver color as in other = There was a pile of smoke came out. Every time pulled out Carbol rod out. The whole inside tube covered with good whitish deposit yet Carbons etc showed no frosted

139 Revs -

Expt 34

(27)

planted 329 am -

Beta-Naphthal =

slipped 4 am - not shiny  
when held up look dark  
lump blocky but both by  
side + end light ~~blocky~~

silvery = heavy frosted  
appearance B-Naphthal too  
sensitive or did not too

quack - Keno 140

Expt 35

49

Meta-Chloral -

Started 4.52 am -

stopped 5.16 -

No 1 dyed seal lines in  
CT Pitch. matter Seal Bagel

No 2 Residue (black) from  
B.S. dyed Bagel dyed  
seal lines.

No 3 Dyed seal lines Bagel  
CTP without heating to  
prevent oxidation.

No 4 dyed Chloro  
CTP seal lines done

zushi lines

Nickel Moved: 20 mesh char  
2:35<sup>pm</sup> - till 12:30 pm -

No 9 - Large bunch of pre-  
prolin Bamboos from Martin  
Force in charcoal

No 10 not prelin Ruv Bamboos  
about 100 in bunch - charcoal  
bed with Three.

No 11 005 - 2 ckt 1 Can...  
dipi & chloro washed spatula  
glass plate = stuck together  
but got a few - Was in small  
box no charcoal

No 12 same in charcoal  
stuck but got 2 or 3 -  
were larger showing very light



weights given to drawing  
up - should be now weight

No 13 Large size 008 I think  
waxed little dip with spade  
& of ~~lumber~~ ~~paper~~ made  
which I believe 11 & 12  
was not =

---

Brown Salolium No 1 -  
Coal tar pitch dissolved  
in chloroform + Lamp Oil  
taken out by filtering  
Brown 1 & 2 are from  
the No 2 Reg length  
showed rise to 128%  
cp & only fell to 98 at  
320 hours

Brown Salinum No 2  
part of Coal Tar pitch  
Aal in Bengal -

Brown Sal No 3  
old cont. little Cam  
& plenty chloro

|        |      |         |              |
|--------|------|---------|--------------|
| No 1 - | 2000 | hours   | broke        |
| 2      | 1500 | "       | "            |
| 3      | 2300 | "       | "            |
| 4      | 2100 | "       | "            |
| 7      | 1700 | burning | gray         |
| 8      | 1700 | "       | "            |
| 9      | 5    | out of  | then 360 hrs |
| 9      | 7    | out of  | then 360 "   |

No 11-003 8 actup 360 cphp  
 No 1 lamp 223 1/2 vatts  
 of 2) Lt No 1 lamp 228 vatts  
 2 " 228  
 3 " 228  
 4 " 228  
 5 " 228  
 6 " 228

Selling Tanday on 1050  
 3-7 4100

## New numbering system

No 300 is Reg bamboo run to yellow  
 in Ni moved Lab - + gave 124  
 vatts & about 300 hours. These  
 taken to Marin & dipped twice  
 in old Reg 17 1st with Camphor  
 only + full yellow 2nd Coal  
 Tar P & some Camphor & red paper  
 + run to full yellow -  
 now gives 108 vatts 262 hp -  
 they are flashed somewhat  
 by Coal tar or Cain

No ~~300~~<sup>301</sup> bunched fibres (100)  
 12 hours run to yellow (low)  
 1st attempt in big furnace  
 Oxidized to blue & yellow -

No 302 = same in another  
 bunch (same run) but not  
 apparently oxidized

303 No 19 aybwith - undip  
 Obs 15-hrs submitted 2-hrs  
 final allgills 1 hour of  
 coherch while heat. heat was  
 dogging fumble v. diff. -  
 now heated, previous life  
 before the 235 hours at least.

304 2 dips cooled in  
 unBox gas - final 3 hrs  
 while hot in old furnace  
 aybwith (21 @) -

~~305 'aybwith' No 20  
 2 dips, unheated final  
 tried this before.~~

305 is Brown 21 <sup>63</sup> ~~quail~~

306 is Brown No 22. Coal T

307 is Brown 23 ~~Campbell~~

Stuff in teal tubes

NO 1 1 gm ~~Can Bicarb.~~ 1 small  
bottle Dipi thin chlova

NO 2 2 grams others same - No 1.

NO 3 3 " " "

NO 4 4 " " "

NO 5 Syrup in chlova with little Dipi

NO 6 - 8 gm <sup>in</sup> C but small bit of Dipi

308. 4 times dipped & pumped  
in Dupi Carbant in Chlors No 4  
4 am

309 - 4 times in 8 gram sal  
as above -

310 - dipped thick sal  
Syrin in Chlors with little  
dippi - then thrown right in  
Rens containing little ChlSal  
remained in about hour then  
out in Benzene all night -

311 - Same <sup>as 310</sup> but allowed to dry  
for 1/2 hour then soaked in  
Rens with ChlSal all night  
out then in Benzene for

Bases May 5/93<sup>(6)</sup>

Carbolic Acid in  $\text{NH}_3$   
Electrolysed then pp by  
acid very black  
stuff considerable  
quantity - fairly  
sol in  $\text{H}_2\text{O}$  and

Piculin - infusible  
Lanellaniline on Carb  
less absorbable  
from the impurities  
in water  $\text{NH}_3$  etc  
good

Another probable base

Oil sassafras in strong H<sub>2</sub>SO<sub>4</sub>  
 little at time until no bubbles -  
 then pour in large quantity  
 water by pip - then filter &  
 wash with water, then when  
 water filtered off add meal  
 dissolves most of it recover  
 the meal sol in strong H<sub>2</sub>O  
 with little H<sub>2</sub>O pip it then  
 filter & dry - test if water  
 can be got out by a vacuum  
 solvent without dissolving  
 it - Don't melt

above without getting out  
 water can probably be got out  
 & probably work ok =



73

Dinitrobenzene is reduced by  
Electrolysis <sup>block</sup> it is ok I think if  
all <sup>very</sup> decomposed Dinitro is  
disolved out.

Nordmann on alcohol al-  
block stuff - ~~is~~ always  
thought to be Carbon  
but found that it is  
Thiomelonic acid - quite  
large product, partially  
sol in H<sub>2</sub>O of NH<sub>3</sub>  
not very much - if covered  
get salm not turned to  
good base

gallic Acid with  $\text{NH}_3$  &  $\text{EtOH}$  is  
 no pp but solution drives to  
 glistening black sub-puffs  
 very little - large product, but  
 there is an enormous ash  
 was this in gallic or  $\text{NH}_3$   
 pbbly good base if ash out

---

Oil Thyme - This 3d -  
 part sal Mat is pp by  
 acid - black only  
 swells little, good -  
 get stuff out that swells  
 it then to elegant  
 15% black - dense - shiny  
 large quantity obtain

Don't think they lower much more if is cheap  
good of H<sub>2</sub>O<sub>2</sub> and be more expensive  
a box of H<sub>2</sub>O<sub>2</sub> alone is 1.00 for the 100 cc. 10% will do more

Good base - no trouble  
Act on Sugar with strong  
H<sub>2</sub>O<sub>2</sub> - by adding just little  
water to make the Carbonaceous  
residue - wash it well by  
grinding fine cobble coat  
in mortar, then filter washing  
with water, be sure & get  
it fine - then oxidize to  
yellowish red by strong HCl  
& KClO<sub>3</sub> - if not red enough  
wash with water thoroughly  
& then alcohol & dry  
repowder & treat with  
fuming nitric, throw in  
water wash & chew Mal  
Sol in H<sub>2</sub>O<sub>2</sub> after heated 200°  
not perfectly sol.

Perfect Base -

Carton from \$1.50 of Bal  
Copalin - added in by HCL +

KClO<sub>3</sub> - very light greenish  
slab of Carbide with no signs  
of smell or smoke - turns

reddish yellow like Reg  
in - Dipe - Feas + gran base

seems just as good - its <sup>strong</sup>

4 Cheap - Make a lot.  
~~leaves no traceable ash.~~

~~Calcium carbonate~~

good

do they  
fairly well

from arabic SO<sub>4</sub> Calm o<sub>2</sub>d<sub>2</sub>  
by firing 24 hours act same  
as above with silvents +  
was absolutely no puff - its not  
so soft in base but very good

glycerum Carbon  
with HCl & KClO<sub>3</sub> -

best solvent none that I can call  
Crucible Essent,

Mat - Chelly or mal ab Ethyl  
Chemical pyridine all shrank  
& crack greatly

glyc Carbon best so far  
don't puff much or any  
more than Reg with Essent  
& dried -

Tried this again but  
not to sol probably not  
acted on suff. Nitric  
Carbon <sup>to then oxidize</sup>  
sol in HCl; water neg

(83)

but think of you - can  
properly which by HETKOs  
might be worked ok -  
I'm trying it again -  
do it double (E) first  
in strong sal then dilute  
& filter & then in a new  
sal -

---

The following is good  
base - no mistake  
No. 1.  $\text{Mg}$  in acetic  
acid & common nitric  
poured in, turns brownish  
its sal, but pp by pouring  
in large quantity water  
over!

Filter this off. what runs  
 through is red & if pp. is  
 white stuff is mg that is  
 pp. as it has. Metro temp  
 & smells also but stuff  
 pp. by 100° in just  
~~10 min~~ & washed well  
 with H<sub>2</sub>O. ~~is~~  
 got about 1/2 of original  
 Sal. P. value. - Seems Sal  
 Chlor Sal. Is but but  
 don't dry them  
 pibly Sal. in Lts  
 things -

Example Carbon 504 <sup>(27)</sup>

Not - good. work this up -

Example Carbon

Is at least ok -

part oxidized by  $HCl + KClO_3$   
yellow dirt puff in slightest  
sol in  $H_2O$  + other good

Shiny red in butter dish

Work it up.

product rather small



Balsam Copra Carbon  
 appears Sol by itself  
 in Br's Picolin -

Have some on hand  
 don't know if I used

that SOX

---

Balsam Cop Carbon

Oxid by Nitric

don't puff ~~up~~ Sol

Picoline Strategy  
 OK

Theriacal wine

Naphthalamine in Hot. Bay

then pour Kilo Throm

drop by water dried

3 days - do to stuff

(a bit hard) dense & high

Specific grav + sal Preser

at 100 + 100 + Preser 1.5

together -

Base fine must work up

(93)

H<sub>2</sub>SO<sub>4</sub> + Camphor - hot  
but don't Carbonyl it (10) go  
so long as it becomes  
suddenly solid. It may  
foam up a great deal  
gas come off but stop  
at right point - filter  
large part left & change  
blocks + if you keep on  
adding water it goes  
through filter etc. so  
finally d.w. - can wash it  
moderately as H<sub>2</sub>SO<sub>4</sub> a  
minute or so won't hurt  
product so base fine showing dry  
but not dense -

Glycerin Carbon very  
finely precipitated or dissolved  
& filtered probably sticky  
stuff to amount of 100  
tries to Extrude very hard  
don't puff & very soft to  
60000 but perhaps  
many cracks.

---

(97)

Can full Carbon  
acted as funny also  
HCl KCl<sub>2</sub> for long time  
became blood red +  
perf Sol - good -  
Good base  
~~but~~

Following are good bases -

Bottle marked No 1 BX

to Cont  $H_2SO_4$  in Evap dish +  
Camphor got it hot lots gas  
come off + I smelt  $SO_2$  but I

carried it only so far as liquid  
got black but not to point  
to make Carbon where it would  
solidifies - then put in Water  
& filtered in its extraordinary

finely divided + must have liquid  
that comes through put back  
several times, then its filter  
paper put out dry, when rather  
dry (TE) selves dry + center still  
moist when touched pour  
Water on + get it into a

(10)  
breaker. then pour into a flat  
enameled dish & evaporate,  
it dries to extremely shiny  
glistening black & cracks &  
contracts to pieces - Not the  
slightest puff or carbonyl  
Even the edges are razor like  
or carbonyl & gives off no smoke  
it shivers even when red hot  
but it cracks & immediately  
a filament ~~is~~ <sup>is</sup> a wire  
crack had hence it probably  
wants something with it  
or a viscous solvent that  
will remain <sup>partially</sup> at high temp  
to prevent cracking  
The product I should

Day is half the Campha<sup>(103)</sup>

Try Solvents

2nd Base - Battli<sup>1</sup>CX

Same as above but heat  
Carried so far as to  
suddenly solidify  
making Carbon - this  
is very strongly worked  
with 3 days ~~of~~  
of HCl + KCl until it  
very red - then air dried  
for 3 days on paper  
it don't stick to



(165)

paper. afterwards it's taken  
from paper & ground in  
mortar then put in again  
flat dish & sun dried all  
day - not the slightest  
puff & apparently no  
gas or smoke comes off  
It is very hard to wash  
it as it's somewhat sticky  
in filter & works slow -  
perhaps best before filter  
to coil -

Try Salverbe  
Jewels very large  
1/2 of Carb at least

2 Bx - 600 cc cp Kalkflor  
H<sub>2</sub>SO<sub>4</sub> - 120 grms Cam. brought  
up slowly. got black. added  
little water but found very  
little sediment so put  
it on again some it till  
there was not very much  
frack But didn't let it  
get solid then turned  
off light & added  
considerable water slowly  
then it stood all night  
& filtered it with only  
about 3 times its bulk  
of water - it didn't  
eat paper - quite large

residue - after 1st filter  
 added but little water to  
 wash out acid - then  
 put the thick black stuff  
 in dishes in drying  
 oven dried several  
 hours stirred &  
 finally powdered it  
 then dried it nearly dry  
 cooled. just tell it wa.  
 damp - got back  
 about 70 grams  
 of the 120 - if dis -  
 ab 2thyl ~~pp~~ in  
 Benzol ok or in Reg  
 Anal ~~pp~~ of chloro  
 OK

also in Mac & pp in string <sup>(11)</sup>  
Hel - ~~and~~ it pp; but  
has not tested resultant  
stuff =

The #2504 Cam hat that  
got soiled, then very  
little washed in  
lamp just as it pp  
put in Kels & Hel -  
24 hours - then followed  
& not washed well  
then dried in drying oven  
212 seems to be just as  
good as BX 1 BX 2.  
but more yellow - some Solans  
& Carbys same Will Call

at. BX-3-

(113)

---

BX 4 is the big lot  
600cc 50% - 150 Cam <sup>50%</sup>  
then washed 24 hours  
then in Kels HCl -  
dilute - then strong  
then filtered took  
5 days + parts still  
washed on

Egg Album H<sub>2</sub>SO<sub>4</sub> -  
afterwards KClO<sub>3</sub> HCl -  
dis NH<sub>4</sub> + Evap -

following have action a  
solvent alone in  
Cannolua Ess -

Methyl iodat

Formal

Aseptol

Methyl formic

pyridin Hydrochlorat

phenylhydrogen

Tetramethylammonhydrox.

a liquid Reoran cond

See front part label

Isopropyl alcohol

Valeraldehyde

Riga Balsam in constant Iso

Fenic Ether

No 12

263

Umochlorhydrin -

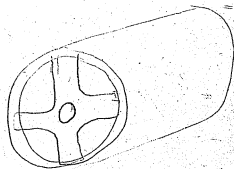
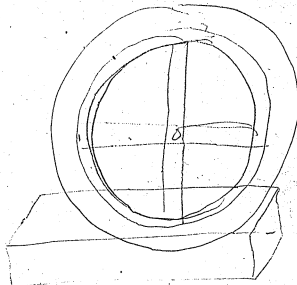
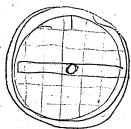
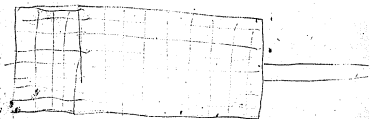
glycum with Iso -

H<sub>2</sub>O -

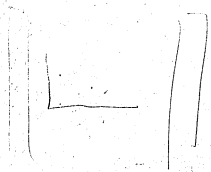
NH<sub>4</sub>

+ 950

195







[ITEM FOUND IN BOOK]

October 13<sup>th</sup> 1944  
No. 28.7, 00, 12, 12, 33, 6.00  
7.00, 10.00  
3<sup>rd</sup> Mack 7.00, 00, 12, 20, 9.30  
44 - 8.00, 7.00, 12.00, 12.30, 9.30  
43.70, 00, 12, 12.80, 8.15

Sunday, October 14<sup>th</sup> 1944  
28.7, 45, 72, 00, 1.00, 6.00  
43.8, 00, 12.00, 1.00, 6.00

Duncan 8.00, 12.20, 1.20  
Mr. Nitty 8.00, 12.20, 1.20

Monday October 15<sup>th</sup> 1944  
28.7, 00  
43.7, 00  
41.7, 00  
37.7, 00

**Notebook, N-92-02-22**

This notebook was begun during February 1892. All entries are by Edison. The book contains notes and drawings relating to various electrical experiments. Included are entries regarding the conductivity of iron, magnetism, Foucault currents, telephone transmission, and the optical behavior of iron subjected to alternating current. There are also theoretical notes about the properties of aluminum, along with drawings, notes, and calculations pertaining to ore milling. Among the latter are several pages of cost analyses, probably based upon operations at the New Jersey and Pennsylvania Concentrating Works plant at Edison, N. J. in September 1892. The pages are unnumbered, and the book has been used in both directions. Several pages have been removed from the book. Approximately 110 pages have been used.

Experiments

Feb 22 1892

Make a cylinder 1 inch dia. of  
75% Zn reduced by Hydrogen +  
melted shellac. It is to be  
6 inches long -

Also air with 100% non wetted  
with minimum shellac in alcohol  
stirred <sup>air</sup> + dried - then put into  
cylinder, heated + pressed slightly  
to obtain cohesion.

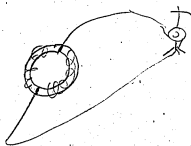
Above to test of continuous electrical  
takes place under high frequency  
Currents. of continuous flow can be  
had -

-00008

-000091

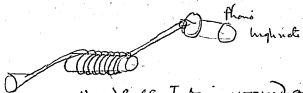
Curve of hysteresis to be run on  
Red hot iron - kept hot by current  
while iron wire in vacuum -

also iron bar surrounded by coils  
of iron wire - whole immersed in  
blacksmith fire -



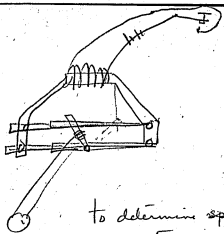
Ring split & 2 diff metals like  
gold & aluminum foil inserted  
deflected effect of heat of hysteresis  
give current - Run Curve



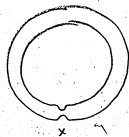


Hand Rubber tubing wound around  
Shell - High note - see if  
amplification in Rubber tube of  
note = wave length must be  
less than  $\lambda$  single turn, or  
say the same -

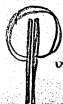
Low at the Ultra light that determines  
the flow of discharge current into  
center wire in glow lamp -  
to prove it shield it with a metal  
not connected - ~~is~~ shield with  
glass - then with quartz -



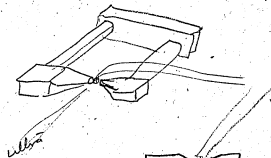
to determine speed of  
 Magnetism in different  
 magnetic bars like  
 Bipolar Spout with flow  
 heat along a bar



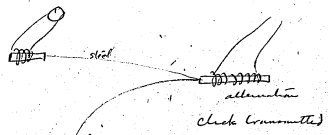
iron ring  
 saturated How  
 would be strength  
 of lines at x &  
 would they be  
 permanent.



would the stay permanent on  
 saturation.

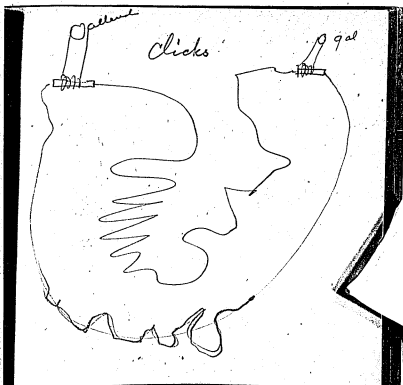


*ultra vibrates to  
get impeller -*



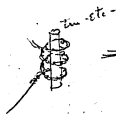
*ultra*



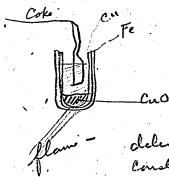


Ballistic

Rings of  
Tin - Lead - Zinc - Brass -  
Copper - Antimony - Bismuth -  
Arsenic, Sulphide Iron  
Phosphide Iron - Aluminum  
Magnesium - Sodium nitrate.



iron stranded wire -  
wound with Copper wire.

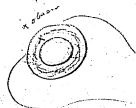


Try Phos acid.  
+ fusible salts  
which may act on  
them  $H_2O$  at red  
heat or above it  
which don't contain  
water.  $CuO$ ,  $Fe_2O_3$   
free  $H_2O$  -

determine R - Voltage &  
conductivity + efficiency

Try thin films of metal on glass  
see if they reverse lines in spectrum  
in same manner as Vapor of  
a metal reverses its own lines

Make films iron in closed  
ring & study its optical  
behavior under alternation of  
Current.



Will there be free lines if very  
thin - will it stay permanently  
changed. will stress polarize  
light or alter amount of color  
reflected or transmitted -

deposit Carbon from Simultaneous  
see if permanent magnetic  
result follows from wave disturbance  
within -

with a cylinder of finely divided  
now mixed with conducting lamp glass  
not any different but reason of  
presence of Carbon.

---

Will a cylinder of iron particles  
compressed together, each iron  
particle which was previously  
deposited with Copper in Vacuum  
act peculiar as to retention of  
magnetism by low resistance  
cells of Copper -

---

Is there such a thing as molecular  
faculties in combination of  
Mass faculties,

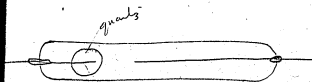
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Is there an amalgam extension  
in Copper wire to that of iron <sup>or magnet</sup>  
by passage of current,  
say Leyden -



block flow  
to ground letter,

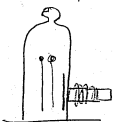
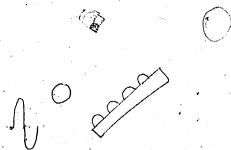
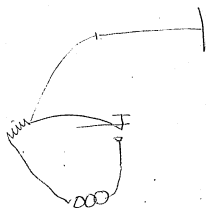
Will there be a Continuum  
of flow -



quads



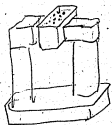
see if this changes  
flow at different  
pressures



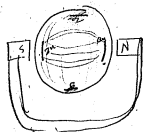
also needs  
coil-

deposit the film of  $\text{Zn}$   $\text{Ni}$  +  $\text{Co}$   
in magnetic field

Lo Carbon (red or white hot)  
Magnetic



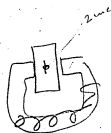
pieces of zinc coated nearly all over with shellac  
which is cork + flat in 20%  $\text{SO}_4 \text{H}_2\text{O}$  at  
the moment of local action current found ought  
to be attracted by magnet.



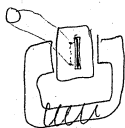
Will the magnet  
Cause the current  
to be deflected?

If so perhaps a current can be got from (or at least)  
action on Zinc in  $HCl$ .

Why is it that Aluminum unlike  
all other metals does not  
deposit metal on the sides of  
a  $sp. r. c.$  tube yet all the  
phenomena takes place as  
with other metals - Can it be  
possible that pure aluminum is  
transparent when deposited  
on the sides + could be flaked  
off or is aluminum decomposed  
into a gas or what?



Zinc in coil in HCl -  
 local action circuit  
 ought to cause  
 motion in Zinc  
 in the field -

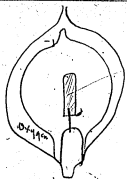


ought not current from  
 local action in HCl  
 with the 2 plate wires  
 by magnet parallel  
 the circuit -

See Knot Phil Mag 1896 246 -

~~Hypothesis~~ Cyclic currents presumably  
 modified by passage current,  
 may not foretell in transformer  
 effect same result.





phosphorus

Vacuum - Cond. between  
Ox. globe & Vac  
glow worm light



closed glass globe

phos -  
free O<sub>2</sub> is  
given off Comb.  
with phos gas  
phos acid which  
dissolves the  
glow carbon  
until all H<sub>2</sub>O  
damp -

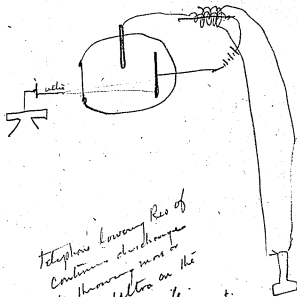


would an intermittent or  
oscillatory spark be recorded  
by luminous paint -

Don't think  $\int$  in glass in  
Leyden jar the electron counting  
can't work at 3 millivolts of a sec

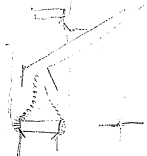
What relations are the Curve  
of loss of energy (watts) in a  
glow lamp to the Curve of  
Magnetic Saturation  
at high Cp. by energy added causes very  
little rise temp enough increased capacity production

Would there be any  
glow in a glass tube  
made of Quartz or Rock Salt



Telephone lowering Res of  
 Cathode discharge  
 by having more or  
 less Ultra on the  
 Right Electrode.  
 use 1200 Volt Cond.

Expansion of Carbon Curve with Volt. Res Cur.  
 How does the Curve of fall  
 of Resistance in glow lamp fil  
 with Reclogy <sup>partly</sup>  
 agree with Satherson Curve  
 of magnet.  
 do the go up & come down the  
 same Carbon is plastic like  
 even at high temp



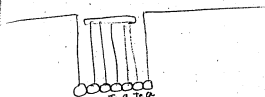
2/1



1/1  
- 072

- 37

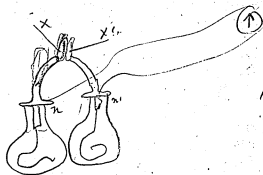
- 56



vacuum emp for Etheric:

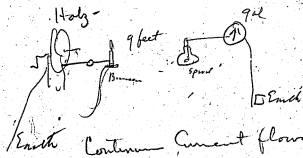
Would glass <sup>or other things</sup> stressed by  
Electrolysis or Magnetism permit  
Ultraviolet to pass like Rock salt,  
will it pass a film of gold and  
Rock salt.

Why is it that Aluminum  
when used as a filament  
melts 75% along axis  
as a crust. it looks like  
75% X42 + 25% aluminum

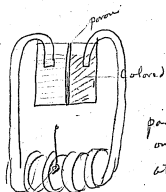


By burning different fluids  
 in the two spirit lamps  
 current should be established  
 between  $n$  &  $n'$  or wires in  
 the flame at different points  
 $X$  -  $X'$  -

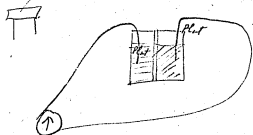
in phlog. 1857 I think is made

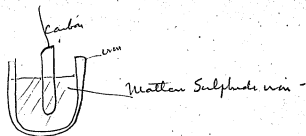


Earth Continuum Current flows



pass light through  
one legend is broken part  
other colors -



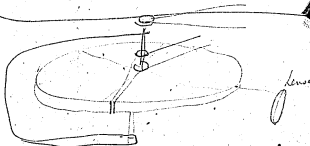


or any other metal which will  
 give a liquid Sulphide at red  
 heat - Reaction,  $CS + Fe$   
 deposited - or the Cell of a  
 metal + lined with a Carbon  
 Sulphide which don't melt at  
 Red - like  $US$   $KS$  or  $NaS$ .  
 Reaction  $CS - K$  reduces of  
 sulphide being to regenerate  $KS$ .

Will a thermo element act different  
in capacity or efficiency when  
heated in diff positions in a  
magnetic field -

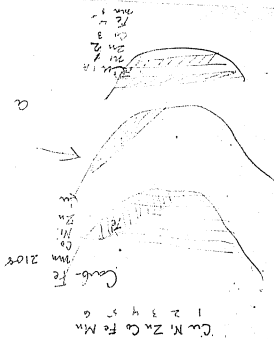
In experiment C + CuO Red hot  
use KO as it dissolves CO<sup>-</sup>  
Reaction is C.KO. CuO

CO - KO - Cu Red use NaO,  
dissolved



Whirling the thin band (transparent film)  
around rapidly with Sun focused on  
it will give a current in some  
direction & between the dance attention  
= Condensation will precede at Sp. of heat.





|      |       |        |
|------|-------|--------|
| 3184 | —     | Co 50% |
| 3023 | —     | Mn 50% |
| 2499 | 5%    | Mn 50% |
| 2356 | 4%    | Fe 50% |
| 2305 | 3%    | Co 50% |
| 2294 | 1%    | Mn 50% |
| 2300 | 2nd   | Zn 50% |
| 1836 | 4 1st | Co 50% |

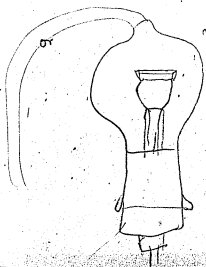
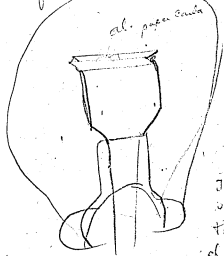
must be transparent yet conducting  
 otherwise would all be short-circuited  
 between surface & opaque interior by  
 interelement heat's same time.

7 miles above W Al in water 1200 feet below  
 Specimen Anal - Watt. p 386 -

Remade Wands - of all the metals  
 Aluminum is found to be the  
 richest in Ultra Violet invisible  
 rays - of 812m extreme refringibility  
 now takes in the fact that Ultra  
 Violet waves reduce the Resonance  
 at spark points & again that  
 in greater tubes nothing is  
 apparently torn off & deposited  
 on the wall it would look as  
 if atoms of all periods with certain  
 waves & hence no race of humpbacked  
 tubes place to melt the surface  
 Digan when Al melts as a loop  
 in the 75% dentmet & 25% iron  
 out in Carbogumma at which  
 is perfectly non oxy furnace if brought  
 up slowly in either pressure its form  
 + one little melt runs out  
 immediately

2499 Mang Sue  
 2108 Cu. Sulf. 1822  
 6197 Gold and 228  
 2356 Cinnabar  
 2305 Ferris sulph  
 2299 Cobalt sulph  
 21504 Ni sulph  
 2200 Ni sulph  
 220 K H + Cl - Ferr  
 592 710 K Sulphur  
 990 m a d - 990  
 1698 Cald - 1698  
 3194 Carbox - 3194  
 2704 Carbox - 2704  
 1510 R All Magnesia  
 3023 Sulphur  
 1227 Aluminium

Melt some Aluminum in vacuum & let it flash on sides of bulb.



This will give  
 sheet CP Alumina  
 to emit delamination  
 of its R + R for Heat  
 + other papers -  
 Might allow ultra pass  
 through even -  
 thick sheet -

The specific heat of Carbon increases as the temperature rises so Silicon - Boron - & Beryllium up to 500 - ~~more~~ and the increase of SH diminishes as the temp rises between any two intervals of 100 - Now the Electrical Res diminishes in same way. There is a close relation between Res & Sp. heat. If the two are proportional the Sp. heat of all metals could be determined by the Curve of increase of Res by rise of temperature -

We might think any substance of ~~low~~ metals having greatest specific heat ought to be best conductor by equal weights - See if this is so -

In very high vac in glow  
Lamp Sulphur & other Vapors  
say <sup>also</sup> iodine - could be driven in  
by heating globe & when the filament  
heated high enough gives a  
true & pure spectrum at Kriem  
temperature -

26 ft 8

|    |     |    |       |
|----|-----|----|-------|
| 84 | und | 3" | 45360 |
| 78 |     | 3  | 42120 |
| 72 |     | 3  | 38880 |
| 66 |     | 3  | 35640 |
| 60 |     | 3  | 32400 |
| 54 |     | 3  | 29160 |
| 48 |     | 3  | 25920 |
| 42 |     | 3  | 22680 |
|    |     | 3  | 2160  |

4 40

54  
 168  
 78  
 29160

189  
 378  
 425580  
 126  
 378  
 22680

48  
 144  
 437160  
 25920

68  
 174  
 597  
 355640

8.1m

72  
 270  
 678  
 8880

24  
 8  
 192

16  
 8  
 128

10 ; 10

1/5

5

100000 -

50 - 55  
 100 - 215  
 200 - 1125  
 400 - 162  
 800 - 31

10 - 10

78  
 54  
 702  
 1212

60)1000-16  
 60  
 400  
 320

10 -

18000 ft/lb

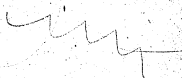
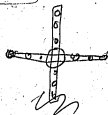
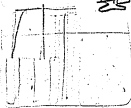
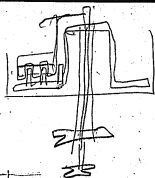
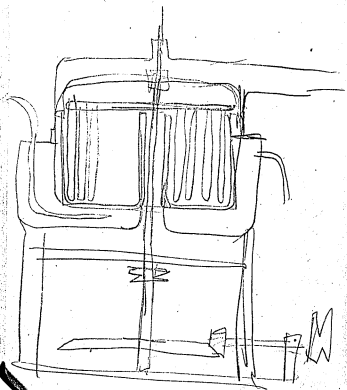
84  
 252  
 756  
 1512

16 ft sec

1/5 sec -

1725) 272400 (157  
 1725  
 9920  
 8640  
 12800  
 12090

157  
 460  
 13560  
 628  
 7530



7' dia 2' thick 75000 lbs km

100000 mm

2300  
2250

43 dia  
42  
41  
40  
39  
38  
37  
36  
35  
34  
33  
32  
31  
30  
29  
28  
27  
26  
25  
24  
23  
22  
21  
20  
19  
18  
17  
16  
15  
14  
13  
12  
11  
10  
9  
8  
7  
6  
5  
4  
3  
2  
1

480  
45  
384  
920  
2310

1396  
160  
83700  
6912  
14644  
13824  
81

48

175  
3000 - 150  
500 500  
2880  
18  
720  
2880

40 mm 200

175  
25 500  
32 500  
100 ton 1 foot in 1/2 of sec

222400 lbs 1 ft in 1/2  
1112000 - in 1 sec  
33000  
6672000  
6672000  
1668000  
60000

16 lbs



15-

$$\begin{array}{r} 18 \text{ dia-} \\ 18 \\ \hline 144 \\ 18 \\ \hline 324 \\ 18 \\ \hline 1620 \\ 324 \\ \hline 4860 = \end{array}$$

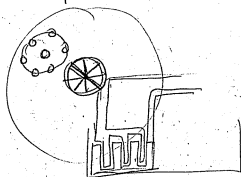
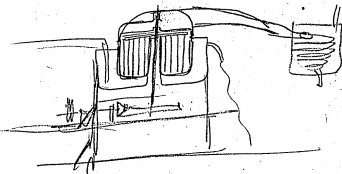
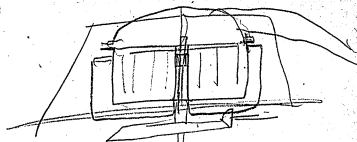
$$\frac{9}{15} = \frac{3}{5}$$

$$\begin{array}{r} 128 \\ 21725 \\ \hline 3456 \\ \hline 12690 \\ \hline 12096 \end{array} \quad \left( \begin{array}{l} 27 \\ 27 \\ 27 \end{array} \right)$$

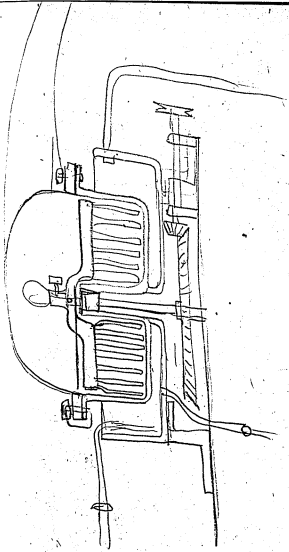
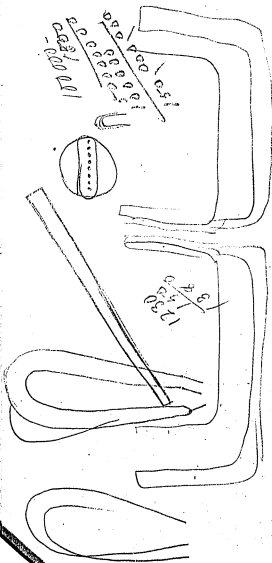
$$\begin{array}{r} 16 \\ 120 \\ 40 \\ \hline 160 \\ 16 \\ \hline 160 \\ 96 \\ \hline 96 \end{array}$$

$$\begin{array}{r} 144 \\ 4 \\ \hline 576 \\ 800 \end{array}$$

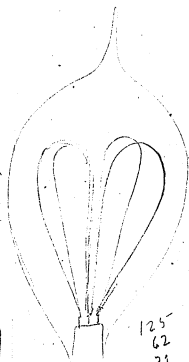
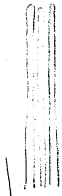
2











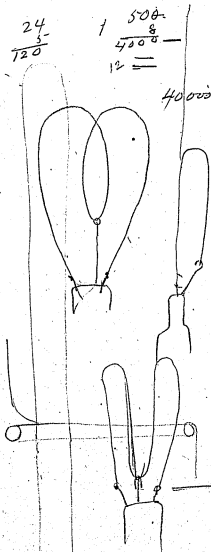
125  
62  
31

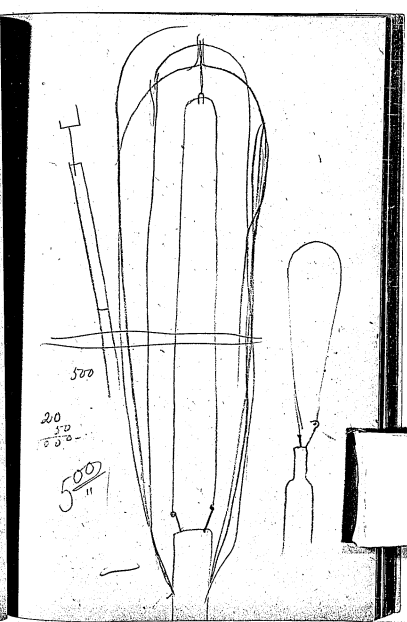
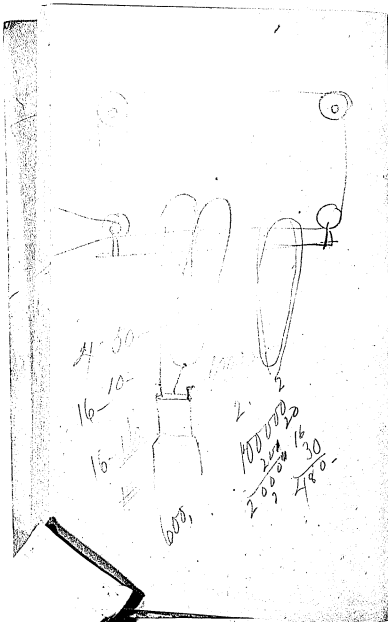
41 / 1000 / 24  
120  
164

24  
120

1 500  
4000  
12

40000 -





E. J. Fredy's John Cande -  
 Siskel  
 E. - at House - Juyato.

Fredy's girls - squirts -  
 female Horse.

Brown Fredy Cande -  
 way up Country.

Store Cande in Jim Cane  
 Horse -

Store ingots - Horse -

Store spirit - safe at

file - store file packed Ciga Mould  
 Lab - long fire - safe  
 deposit - vanilla

$$\begin{array}{r}
 128 \\
 20 \\
 \hline
 480 \overline{) 25600} \quad (53 \\
 \underline{24000} \\
 1600
 \end{array}$$

$$\begin{array}{r}
 16 \\
 30 \\
 \hline
 480
 \end{array}$$

500 lbs

64 -

$$\begin{array}{r}
 480 \\
 68 \\
 \hline
 28800
 \end{array}$$

5 lbs

225 - grams de ozanti -

24 grams 20 grams

$$\begin{array}{r}
 2500 \\
 750 \\
 3 \\
 \hline
 175000 \\
 \underline{125000} \\
 50000
 \end{array}$$

$$\begin{array}{r}
 225 \overline{) 28800} - (128 \\
 \underline{22500} \\
 6300 \\
 \underline{4500} \\
 1800 \\
 \underline{1800} \\
 0
 \end{array}$$

12.

$$\begin{array}{r}
 1728 \\
 20736 \\
 \hline
 3456 \\
 21 \\
 \hline
 42 \\
 25 \overline{) 110} \\
 \underline{100} \\
 10 \\
 \hline
 25 \\
 \underline{25} \\
 0
 \end{array}$$


$$\begin{array}{r}
 485 \\
 11 \\
 \hline
 42 \\
 25 \overline{) 110} \\
 \underline{100} \\
 10 \\
 \hline
 25 \\
 \underline{25} \\
 0
 \end{array}$$

8"

1)

252

$$\begin{array}{r}
 144 \\
 144 \\
 \hline
 576 \\
 144 \overline{) 20736} \\
 \underline{1440} \\
 6336 \\
 \underline{6336} \\
 0
 \end{array}$$

$$\begin{array}{r}
 12 \\
 12 \\
 \hline
 24 \\
 12 \overline{) 288} \\
 \underline{240} \\
 48 \\
 \underline{48} \\
 0
 \end{array}$$

84  
83  
87-  
8-1  
79  
78  
77  
76  
75  
74  
73  
72

laches

$$\begin{array}{r}
 3042 \\
 \hline
 72
 \end{array}$$

$$\begin{array}{r}
 6084 \\
 212944 \\
 \hline
 17852 \\
 17852 \\
 \hline
 0
 \end{array}$$

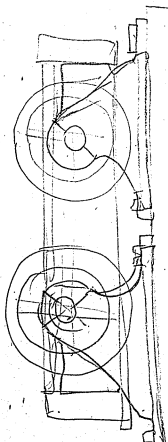
$$\begin{array}{r}
 1728 \\
 \hline
 12
 \end{array}$$

5580 lbs sk of rock

58000 lbs rock

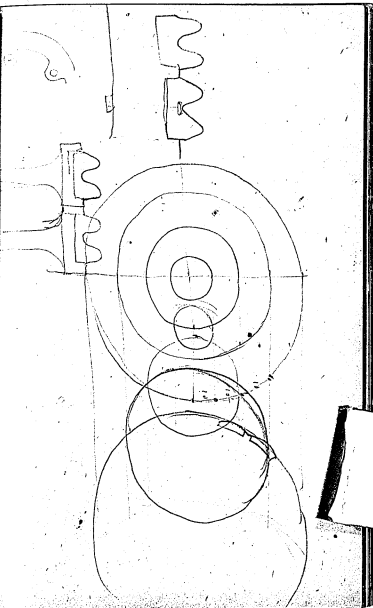
12

Run



other things  
 Run 116,000  
 shaft 11,000  
 pulley 6,000  
 pulley 25,000  
 212,000





1/3  
 21/3  
 30  
 4/3  
 5/3  
 6/3  
 7/3  
 8/3  
 9/3  
 10/3  
 11/3  
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 14/3  
 15/3  
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 94/3  
 95/3  
 96/3  
 97/3  
 98/3  
 99/3  
 100/3

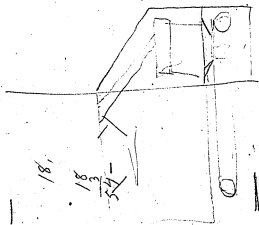


58.8-

18'

58. 13

~~54-~~



272-

1600-ton

272 / 1600-ton 588

2178

2248

212

2176

1360

15-17



84-

$$\begin{array}{r} 34 \\ 24 \\ 62 \\ 2 \end{array}$$

$$\begin{array}{r} 5 \\ 4 \\ 2 \\ 2 \\ 2 \\ 2 \\ 2 \\ 2 \\ 2 \\ 2 \\ 2 \end{array} \left( \begin{array}{r} 2000 \end{array} \right)$$

$$\begin{array}{r} 172 \\ 6000 \\ 5184 \\ 4160 \end{array} \left( \begin{array}{r} 314 \end{array} \right)$$

60

$$\begin{array}{r} 34 \\ 24 \\ 62 \\ 2 \end{array}$$

$$\begin{array}{r} 183 \\ 204 \\ 514 \\ 1052 \end{array}$$

$$\begin{array}{r} 27 \\ 1052 \\ 2487 \end{array}$$

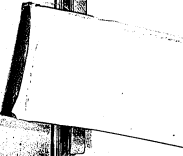
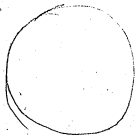
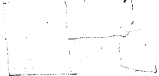
$$\begin{array}{r} 28680 \\ 17340 \\ 56020 \end{array}$$

36-

32

6

$$\begin{array}{r} 275 \\ 32 \\ 550 \\ 825 \\ 8400 \\ 760 \\ 200 \\ 1960 \\ 440 \\ 234 \end{array}$$



285-  
190  
408

Coal 1000-

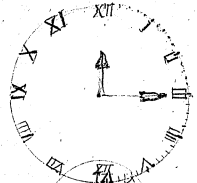
15) 35000 (1933)  
4500  
30500  
750  
29750  
1000  
28750  
10

24 trips  
15 paylip

122.1 off  
3 road  
4 wages  
12 fuel  
4 maintenance  
3 oil  
10 other  
Kernan  
L...  
M...  
3000 lb

175  
135  
310

27500  
70000  
80000  
80000  
80000  
80000  
80000  
80000  
80000  
80000  
2016



1000  
2533  
3533  
666  
4199

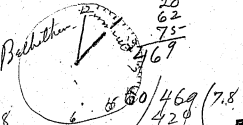
100 at 10  
15) 10000 (666)  
9000  
1000

830-  
35-

175

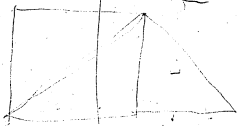
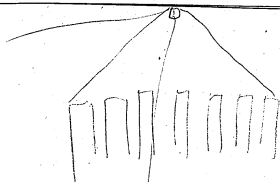
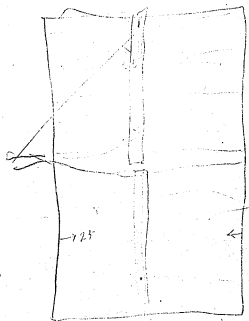
137  
175  
20  
62  
75  
469

100  
175  
275  
357



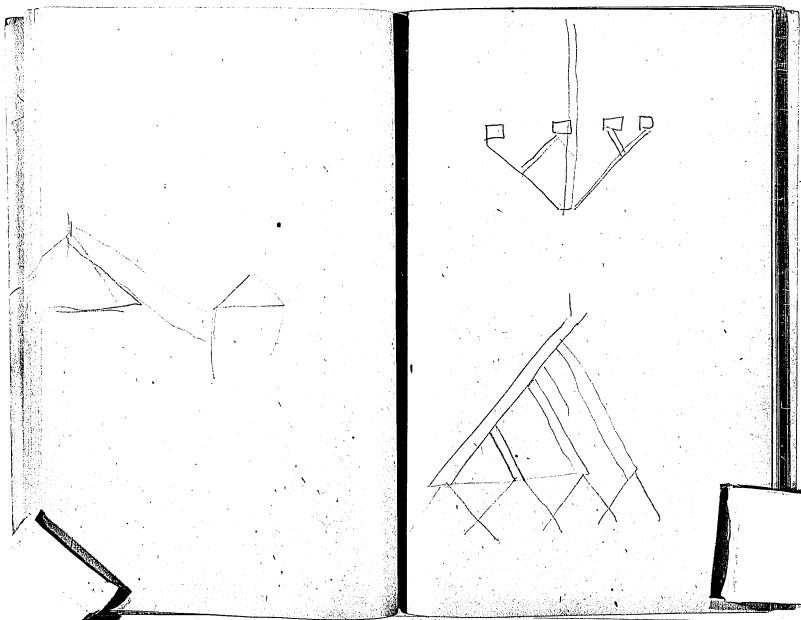
378  
337  
21

469 (7.8)  
428  
490



20/25





7740/15336 (6.84)  
 13440  
 75760  
 13440  
 32.00  
 884  
 2516

50% eff. of screens  
 10% eff. of rolls -

32 tons not for belis. total  
 lifted by elevator & passed  
 by screens -

15336 -  $\frac{2240}{62}$  596  
 $\frac{32}{792}$   
 26852

32 tons come hourly

15336 lbs goes thru screens steadily  
 leaving 25.1 tons to go through  
 rolls. These go through once 10%  
 Cords - give slurrings 1st - 25-82 -  
 2nd 2.25-0 - 3rd 20.25 tons -  
 4th 18.2 5 16.4 6th 14.4  
 7 13.4 8 12.1 9th 10.9  
 8th 9.9 9th 9 10th 8.1 11th 7.3  
 12th 6.6 - 13 5.94 14th 5.35  
 4.62 4.32 3.91 3.52 3.17

391  
 $\frac{39}{552}$

352  
 $\frac{35}{317}$   
 2500 lbs from No 1 stone to  
 screens - 40% sluffing  
 1280 lbs 50% eff. 640 lbs  
 leaving 25.6 tons go through screens -  
 this is raised by elevator.

2500  
 $\frac{250}{225}$   
 $\frac{225}{2025}$   
 $\frac{535}{4182}$   
 $\frac{90}{81}$   
 $\frac{462}{434}$   
 320 434  
 317

2025 2560 whi.  
 $\frac{200}{1025}$  640 fine - 32 3200  
 250 by roll 250 2560

182  
 $\frac{18}{164}$   
 $\frac{164}{748}$   
 25.60 3 134  
 $\frac{14}{3.4}$  840 25.21  
 $\frac{14}{3.4}$  167

if these can not be made more  
 10 tons would remain  
 660  
 $\frac{660}{574}$  240  
 8.1  
 $\frac{8}{7.3}$   
 22.4  
 $\frac{22.4}{87.6}$   
 $\frac{1.7}{616}$  4.3-65

32 - time

30672 lbs in it -  
153361 gals

25.1 ton quarter roll - 50%  
12.5 -

32 1st -  
25 2nd -  
12 3  
6 4  
3 5  
1.5 6  
7 7

$\frac{80.2}{160}$   
 $\frac{40}{200}$

245

$\frac{277}{277}$   
 $\frac{277}{91}$

3  $\frac{225}{112.5}$   
80  $\frac{360}{360}$

32  $\frac{25}{62}$   
2.5 ft  
500

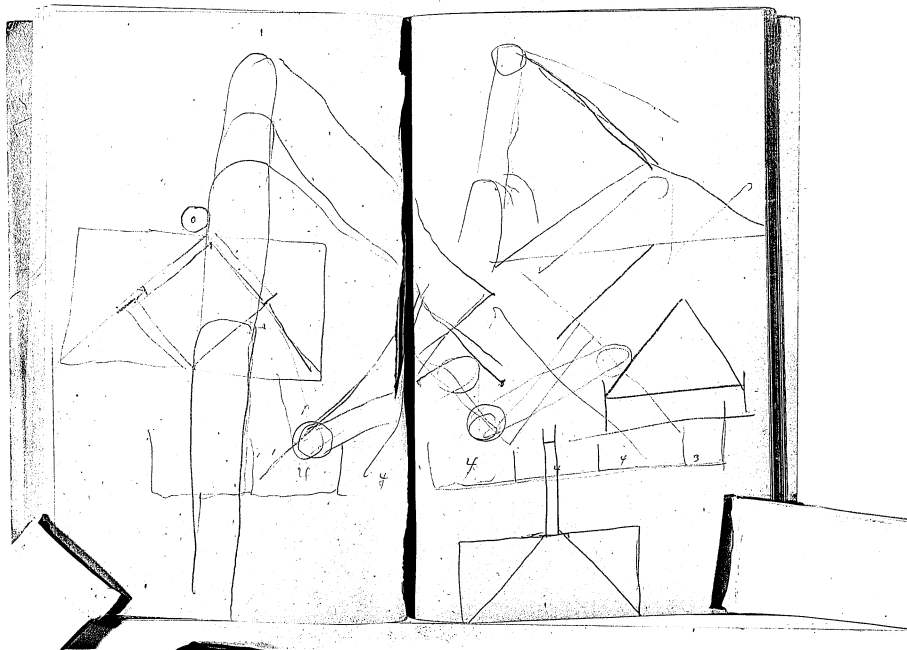
$\frac{2240.00}{1320.00}$   
 $\frac{2240.00}{2240.00}$   
 $\frac{2240.00}{2240.00}$   
 $\frac{2240.00}{2240.00}$

245 - 8.6 times

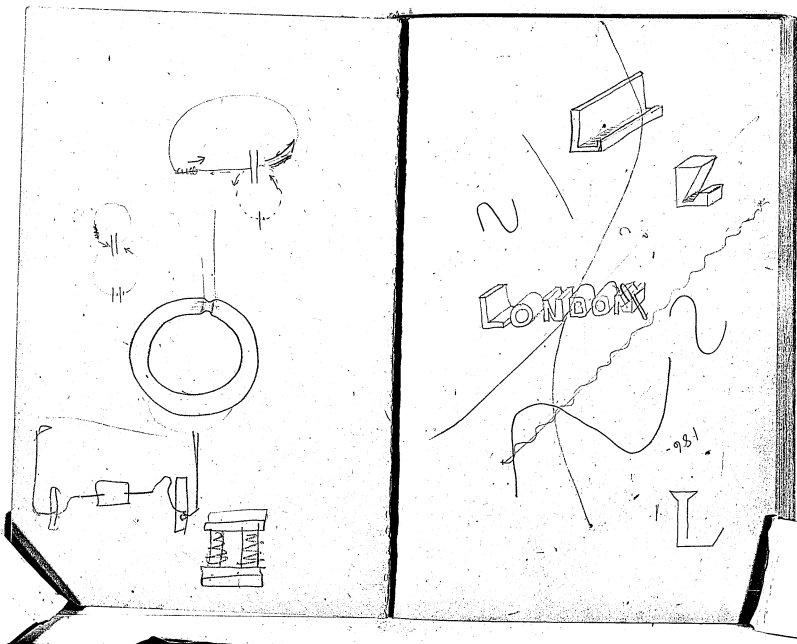
$32 \sqrt{277} \left( \frac{8.65}{2.86} \right)$   
 $\frac{256}{277}$   
 $\frac{210}{277}$   
 $\frac{189}{277}$   
 $\frac{171}{277}$   
 $\frac{154}{277}$   
 $\frac{139}{277}$   
 $\frac{126}{277}$   
 $\frac{114}{277}$   
 $\frac{100}{277}$   
 $\frac{86}{277}$   
 $\frac{72}{277}$   
 $\frac{58}{277}$   
 $\frac{44}{277}$   
 $\frac{30}{277}$   
 $\frac{16}{277}$   
 $\frac{2}{277}$   
2.86  
2.58  
2.33  
2.10  
1.89  
1.71  
1.54  
1.39  
1.26  
1.14  
1.00  
8.6 times  
9  
10  $32 \sqrt{245} \left( \frac{8.16}{72.10} \right)$

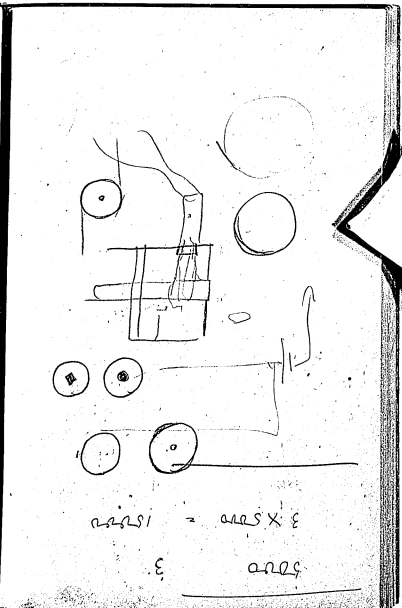
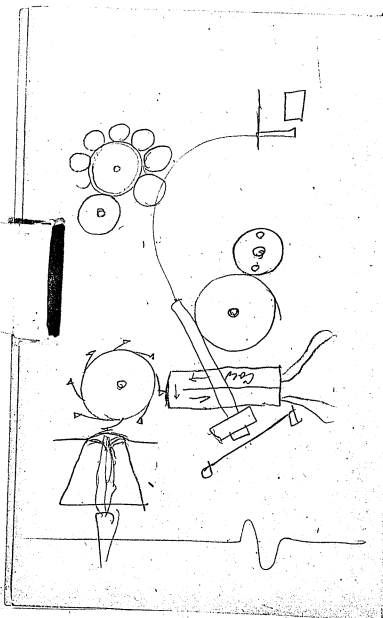


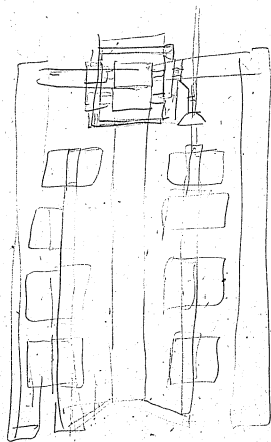




[THIS BOOK WAS USED IN BOTH DIRECTIONS.  
THE FOLLOWING PAGES WERE FILMED FROM  
THE BACK END FORWARD.]





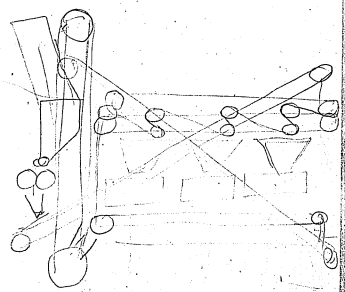
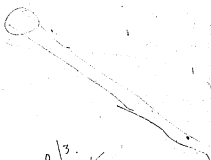


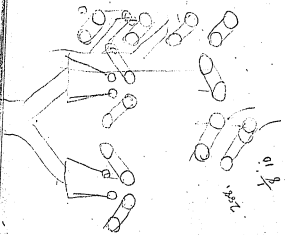
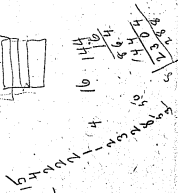
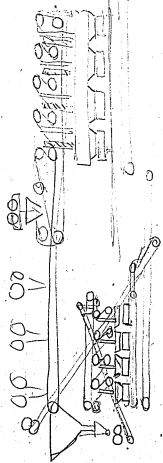
$$30000 \frac{6720}{2240} \div \frac{6720}{2240} \div 3$$

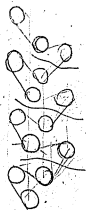
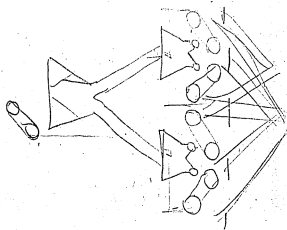
31m.

$$\frac{2240}{6720} \div 3$$

$$2240 \div 5376 = 2.4$$









10- 800 300  
2400 ton 200

100  
2000

16 4/17

2/586  
293

1682  
450  
282  
320  
962

c/384  
96

3/377  
125

Drilling proceeds 80 feet day costing only  
as much as 60, save 4 Cents per ton  
if big rolls made cost drilling per ton say  
25% less, hence save 3.2 Cents ton  
if the 2 broken men are not used

Blockholing now 2.27 cents reduced  
30% — saving 1.12 Cents

Loading 58.6 by rewrapping  
loading should be reduced 50%.  
as there will be no skidding + 1 @ 3  
tons loaded at once, saving 29.3 c

Repairs muncie now  
3.84 c diminished Drilling +  
bulk handling ought to save 25%  
hence .9 of Cent.

Material now 3.77 suppose  
this is skip now Drifts hammer  
checks - bulk handling less Drilling  
by repairs of rep + skip work + more skidding  
should reduce this 33% hence  
saving 1.25 Cents

4/384  
96  
188  
5/377  
125  
188  
213

2.27  
1.50  
1.77

Coal 3110 is cost now - with the Chimes  
 this allowed for less saving .33% - saving  
1.3 Cents

Explosives - if we use rap the explosives  
 should be 33% less hence saving 5.42 Cents  
 this now 16.2/-

Stripping now 9.75 probably less than  
 or about 2 cents - saving 7.75 Cents

Total Cost Sept. \$114.46 per ton Conc

Assum

|            |              |
|------------|--------------|
| Drilling   | 71.20        |
| Blockading | 1.12         |
| Loading    | 29.30        |
| Repairs    | 00.90        |
| Material   | 1.25         |
| Coal       | 1.3          |
| Explosives | 5.42         |
| Shipping   | 7.75         |
| Saving     | <u>54.27</u> |

114.46  
 54.24  
60.22 possible cost.

3/1627  
542  
 1085

310  
 1.33  
 1.8

310  
 1.8

3/1627  
542

12424  
 513  
59137

Cable.

Repairs labor with new cable & proper solution of the bucket. difficulty showed not exceed 2c ton was 7.13 in Sept saving -

5.13 Cent

Total Cost Sept 22.69  
Saving 5.13  
possible cost - 17.56

Mill 1 - All Crushers - Labor Operating in Sept. 5.84 per ton Ore

by Roller feeds on both outside crushers as well as inside, there will be 4 men on each shift saved at 1.25 for 8 men is 10.00 Cost of operating say 2 men each shift at 1.50 for 6 ~~hours~~ men under large crusher 1.20 both shifts 8.00 is a further formula necessary if we is streamlining on both shifts. if we average 1500 tons crush per day, then cost will be about 2c instead of 5.84 saving

3.84 Cent

8.40

16/10/10

5.84  
2.00  
3.84 00

Say repairs labor same + Cent per ton  
Concubats: say all other items same  
by putting small outside crusher on mill  
save 5 hp save 50 Cents per day  
or  $\frac{1}{8}$  of Cent per ton

Crusher Oil now 2.46 per Cmc  
I feel sure I can clean + filter this  
+ save 75 percent say labor would  
save  $\frac{1}{2}$  — Saving 1.23 Cents  
perhaps grease cup principle  
can be used if so arranged  
save nearly all of this item

$\frac{1}{2}$  +  $1\frac{1}{2}$  Screens. as one new  
screen was put in + much repairing  
was done on old ones at the foot we  
are to use steel I think a saving  
of 4.15 Cents so sure it was in  
Sept 6.15 — hence saving 4.15 Cents

$\frac{71330}{47}$   
 $\frac{15}{50}$

25000

147  
460  
99

125  
900

249  
112  
137

Wells Mill 1

Labor now 1.70 - by rods for guide  
stop 1 man only necessary hence  
savings .85 of Cent

Labor repairs outside woodcutters  
which I suppose is putting on & taking  
off shells, now we can certainly design  
rolls which will only require 4 men  
1 hour. I suppose we changed a pair  
per day. Cost would be at most .75  
or for Cent making savings of  
including turning workshop  
= 2.77 Cent

Material Costs Sept 14 1914

now a pair of 300 chrome rolls will  
certainly ~~more~~ pass through 100 rods  
tons & do not sink in Well 1 should  
never be less than 25% of amount  
passed through hence Cost of material  
would be about 7.15 per ton ~~trunk~~  
or 4.60 per ton ~~trunk~~  
to say nothing of fact that 25% of all  
the oil drags through rolls but  
through ~~drags~~ rods ~~comes out~~,  
hence savings on material  
will be 9.94 Cent

1404  
200  
411

Wooden teeth. With labor cost

7.24 cents now this can be  
entirely prevented by a breaking  
block which cost nothing practically  
hence only natural wear of teeth  
& pins which should not exceed  
1 cent per ton Conc hence  
saving 6.24 cents

Driving machinery mill!

Labor .9 material 5.65  
what is this 5. 98

Carrying machinery labor now  
4.39. say this labor is ok  
but investigate - Material  
9.99 - this includes the new 36"  
belt, now the old one has done over  
100000 tons so the cost per ton  
can't exceed 1 cent per ton Conc  
Passing most of this item is for  
belt - do my investigation  
find saving of 7 cents

Costs on Mill 1 now 85.37

Saving

|                              |                 |
|------------------------------|-----------------|
| Labor operating all machines | <del>3.84</del> |
| putting motor on mill        | 3.84            |
| filling Conker oil           | 0.12            |
| 1/2 4 1/2 screen             | 1.23            |
| Labor rolls                  | 4.50            |
| Labor repairing machinery    | 0.85            |
| Material                     | 2.17            |
| Wooden teeth                 | 9.99            |
| Conveying piece              | 6.24            |
|                              | <u>71.00</u>    |
|                              | 35.94           |

Total possible saving

|              |          |
|--------------|----------|
| 85.37        | now      |
| <u>35.94</u> | possible |
| 49.43        | - Cost   |

Bill 2

50 mesh screens - by new method

The saving in power showed more than  
pay for whole cost screen

now cost 14.41 hence 14.41 Cent

Saved

Balls - Experiments with heavy feed  
which new screens will permit gives  
20% net of ore passing through.

hence saving Chrome balls + Saw  
Removal we have 100 000 passed  
20 000 net cost \$300. 1.5 Cents ton  
of 41% ore. or 2.2 Cents per ton  
hence a saving of 4 Cents

Labor Repairs are 10% saving probably  
not less than 66% say .6 Cent

Belt Separator, if they don't split  
assuming 1/32 poor metal on old + 6 times  
that amount on new belt with twice  
wear capacity + non buckling crowning  
+ even old belt did 1500 tons ore only  
then new will do 20 times that or say  
116 Cents saving over 41.4% now  
2.87 Cent

$$\begin{array}{r} \sqrt{167} \\ 22 \end{array}$$

$$\begin{array}{r} 447 \\ 164 \\ \hline 283 \end{array}$$

$$\begin{array}{r} 1500 \\ 12 \\ \hline 30000 \end{array}$$

30.

$$\begin{array}{r} 1500 \\ 18 \\ \hline 83333 \end{array}$$

$$\begin{array}{r} 1500 \\ 30000 \end{array}$$

$$\begin{array}{r} 50 \\ 15 \\ \hline 3333 \end{array}$$

50000



1750  
 900  
 1200  
 270  
 270  
 4125

350 | 42.00 | 10.  
       3 1/2%

350 | 50.00 | 1.4  
       3 1/2%  
       1 1/2%

24 28  
 13 00  
 11 28

If we substitute small bar mag  
 & save 190 horse power & there is  
 practically no repairs but there is  
 some loss of iron in dust figure this  
 & make a better estimate -

The Repair in Sept. 9.02 but this  
 was for staggery, hence in my  
 work probably be not over 2 Cents  
 hence 7 Cents saved

Oiling labor 4.33 think 3 orders  
 not necessary but as mill will  
 do very much more saving  
 should be 1.3 Cents

Power is 76.65-

allowing a couple for labor & any saving  
 350 ton line showed be Nat was  
 show 13 Cents ton line instead of  
 24 28 saving 11.28 Cents

$$\begin{array}{r} 3581 \\ 2380 \\ \hline 1201 \end{array}$$

12

$$\begin{array}{r} 23 \\ 71 \\ 116 \\ \hline 225 \end{array}$$

$$350 \overline{) 8250} \left( 233 \right.$$

$$\begin{array}{r} 1050 \\ 1050 \\ \hline 1500 \end{array}$$

$$3 \overline{) 575} = 191$$

$$2 \overline{) 565} = 282$$

allowing 350 ton average of Gas  
+ 25% than bulk shafts. Costs  
would be 2313 instead of  
3581, saving 1251 Cents

Oil is excessive should be  
saved by increased output  
only to say nothing else of  
33% - Costs now 575  
Saving 191 Cents

Office - all books changed in one  
month instead of over several  
Months for material in Sept 565  
Reduce say 50% Saving 282 Cents

$$\begin{array}{r} 574 \\ 191 \\ \hline 384 \end{array}$$

16-  
2500 2  
2000  
160-

$$\begin{array}{r} 1200 \\ 600 \\ \hline 600 \\ 1200 \\ \hline 2400 \end{array}$$

Saving Mill 2 Cost Sept  
53.61 Cost

Saving +  
Sawing 14.41  
Rolls 4.00  
Repair rolls - 0.60  
Belt Sep 2.50  
Belt Sep Repair 7.00  
Oiling labor - 1.30  

---

30.18

5361  
3018  

---

2343 Cost

Power - now 76.65

labor saving 11.28  
Coal 12.51  
oil 1.91  

---

25.70  
office — 2.82 saving

76.65  
25.70  

---

509.25

25000-

24

$$\begin{array}{r} 250 \\ 105000 \\ \hline 217000 \\ 218000 \\ \hline 435000 \end{array} \quad (24)$$

6000-

432

675

$$\begin{array}{r} 6000 \\ 1200 \\ 1200 \\ 2000 \\ 8000 \\ 2500 \\ 3500 \\ \hline 24400 \end{array}$$

24400

|        |        |
|--------|--------|
| Mining | 60.22  |
| Cable  | 17.56  |
| Mill 1 | 49.43  |
| Mill 2 | 23.43  |
| Power  | 50.95  |
| Office | 2.82   |
| <hr/>  |        |
|        | 205.41 |

205.41 left off  
 $\frac{2}{3}$  or  $\frac{2}{3}$  mile  
 $\frac{2}{3}$  or  $\frac{2}{3}$  mile

If we put increased output to diminish  
 of fuel expense as other items I have  
 not included it will pay the cost  
 of big rolls + new dryer + accidents.

|          |      |
|----------|------|
| Now      | 260- |
| Grinding | 35-  |
| Royalty  | 45-  |
| <hr/>    |      |
|          | 340. |

24. Costs for all other ~~to~~  
 367. is total cost grinded

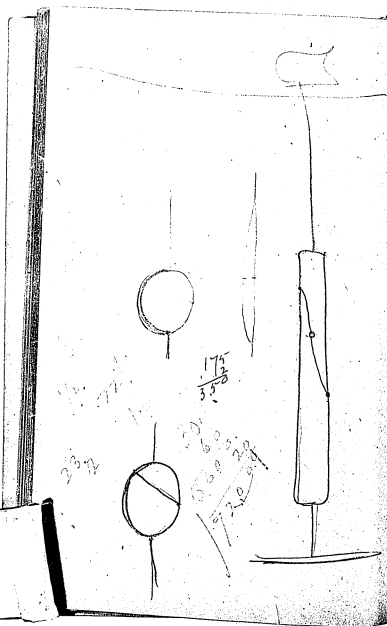
on car + it must be done

Man 114.46  
Cable 22.69  
Sizable 5.57

mill 1 Engine 15.40  
1/2 screen 6.57  
Rolls 18.73  
wood 7.24  
Boring 6.56  
Conveying 14.38  
14"m 1.35  
2.41  
7.28  
4.82

---

Man 143.02 59.37  
mill 1 85.37 35.94  
off mill 2 53.61 50.18  
Electric equip 76.65 25.70  
Blastings etc 29.00  
2.87  
38.99  
mill primer 429.51



350) 13350 (38  
 1050  
 2800  
 2800

4000

Bricklay

Material 22.00  
 1 laborer material 1.00  
 2 masons -

300 ton output

Material 22.00  
 laborer making material - 3  
 2 mason night & day  
 8 loaders day & night  
 1 car man brick and 1 mlt th  
 2 foremen day & night  
 2 firemen boys -  
 4 unloaders night & day  
 4 men in cars loading  
 2 cleanup Eng room  
 Coal for dryer  
 preparation for power cleanup

350) 11300 (32.3  
 8000  
 3000  
 300

340  
 22  
 70  
 77.00

7000  
 3000  
 17000

77.00.

4.00

3.00

10.00.

2.50

3.50

2.50

5.00

5.00.

5.00.

7.00

9.00

133.50

350  
2450  
2800  
18000

100000 / 175000 (175)  
100000  
175000  
425000

200000 / 220000 (22)  
200000  
220000  
420000

3

## Cable -

|               |       |
|---------------|-------|
| 6 Eng's       | 12.00 |
| 4 Skippers    | 6.00  |
| 2 foremen     | 3.50  |
| 2 Repairmen   | 3.50  |
| 4 Signal boys | 4.00  |
| 2 Car boys -  | 2.00  |

## Crusher -

|                         |      |
|-------------------------|------|
| 4 outside men           | 6.00 |
| 2 inside men            | 3.00 |
| <del>2 inside men</del> |      |

Rolls 2 men 3.00

2 rollers 3.50

2 bit men w/ sleds 3.50

4 Drift men 6.00

4 Bar men 7.50

500

27 Laborers 5.00

2 Carpenters 5.50

1 Crusher foreman 5.50

2 foremen 6.00

Mill 2 3.00

2 Rollmen

---

 88.00



|                 |       |
|-----------------|-------|
|                 | 88 00 |
| 6 Oilers        | 9,00  |
| 2 Screen men    | 3 00  |
| 4 Laborers      | 5 00  |
| 2 belt men      | 3 50  |
| 2 Foremen       | 6 00  |
| 2 Carpenter     | 5 00  |
| 2 Watchmen      | 3 50  |
| 2 Truck keepers | 4 00  |

Engines -

|                |       |
|----------------|-------|
| 4 Foremen      | 13,00 |
| 2 Laborers     | 2 50  |
| 2 Fuel boys    | 5 00  |
| 2 Boys         | 2 50  |
| 2 High Mod. in | 3 00  |

Mill 2

|              |      |
|--------------|------|
| 2 Engines    | 5 00 |
| 2 Gofps.     | 2 50 |
| 1 Repair boy | 2 50 |
| 1 Chief      | 5 00 |

50 mech. Boys -

|               |
|---------------|
| 165,00        |
| 3 50          |
| <u>168,50</u> |

1300  
1300  
304  
2900  
1700  
1600  
29600  
15  
12  
238

168,50

Wiley 25,00  
Haul 8,00  
Smith 2,50  
Sledge 1,50  
Sledges 3,00  
Sledges 5,00  
Sledge 4,00  
Workshop 27,00  
Blacksmith 9,00

Chimney 5,00  
2 Loaders 16,00  
2 Crane men 2,50  
2 foreboys 3,00  
2 men 4,00  
2 foremen 5,00  
4 Unloaders 5,00  
4 Gun Crew

Ship repair - 2  
Chemicals -

2 Carpenters

400  
2,50  
296,00  
5500  
30100

301.00.

**Notebook, N-92-03-30**

This notebook covers the period March-May 1892. At the beginning of the book are entries by Edison and Arthur E. Kennelly regarding various electrical experiments, along with theoretical notes relating to dynamos, dielectrics, electric lighting, and magnetism. The remainder of the book contains notes, drawings, and calculations by Kennelly and by unidentified Galvanometer Room experimenters pertaining to differential dynamometers, transformers, batteries, and a 3-wire electrical distribution system. Transcriptions of the Galvanometer Room records can be found in Arthur E. Kennelly Notebook #6. The spine is labeled "135." The book contains 197 numbered pages.

Only the Edison material has been filmed.

X E-172

N-92-03-30



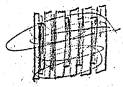
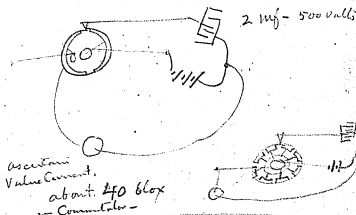
Salinity lines may  
to 4000 lines  
per scm. then wind  
over the salinity  
system a primary  
secondary or  
primary alone  
as well as self  
interaction  
working down up  
between 25 & 4000  
lines.



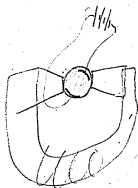
Different vibratory  
periods of  
different magnitudes  
maintain a sea of  
low hydraulic in  
function of slanting

40 500/10

38  
500



So the heat given off in  
alternating through a  
storage battery



20% 50% + 30%  
with the water  
in the  
or what. 5

Wonder how historical  
Cuvr would be with powdered  
now + melted glass mixed  
75% . 25% glass

account for 3/16 blue  
spark on our jumps

②  
• See if there is any change  
of Res or  $C_p$  in glow lamp  
in any position in Magnetic  
field -

Vibrate filament by outside  
magnet that is worked by  
vibrations & strike the vibrat  
period of a glow lamp  
filament at Res & then  
find out the Curves of  
Changes in its vibrating  
period by rise of  $C_p$  -

Statement made that  
an incandescent lamp  
burning near a shield  
will heat up to great  
brightness several times  
& then heat back to normal



(11)

Leyden jars are imperfectly made ~~of glass~~ <sup>of shellac</sup> is used How done.  
It do to use in fact r Hg as with mirrors or by Vacuum process.

Dielectric are the opposite of Magnetic in every? analogy - S.

All Electro-chemical processes follow that of dielectrics.

All sources <sup>electric</sup> of energy tend to take that path in which the greatest <sup>amount of</sup> energy can be transformed into heat, S.

(13)

Radiant heat from a given source recd on one side of a metallic plate gives motion to the atomic systems of the metal & is given out on the other side at different periods or rate than the original depending on each particular metal.

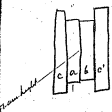
All matter whose atoms <sup>vibrate</sup> are periodic with some pitch in radiant energy - absorb all other periods & emit only the particular vibrations <sup>periodic</sup> with its atomic system - hence thin films of metals show the pitch of the molecular system of the spectrum period.

for Dark box use

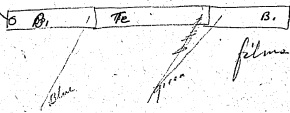
one single turn and put on all the ampere it will stand ~~put~~ so as to get rid of self induction strength of arc - perhaps the single turn could be put in circulating water -

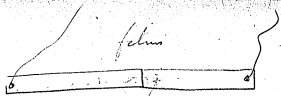


As it cannot be supposed that Te & Arduy actually touch when held tightly together then ultra should permit of a circuit by its action on the surface & by the heat absorbed. The latter being direct source of Engr. + Ultra wave rate the condition to permit of conduction across air space



c c metal a. b. diameter  
 unimpacted dielectric such as glass  
 fused phosphoric acid - idea is  
 beam of light resolved into heat  
 at junction & electrolyte the  
 electrons from c c'





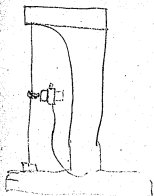
2 metals  
which absorb diff or  
have different colors.

Same metal



blue green etc

absorbs one not the other & will vibrate



With  $St$  <sup>point</sup> and  $T_c$  anvil  
deflection as high as 10 in  
direction indicating cooling

C point } both 5 days indicating heat  
Pt - }

Platinum plate dark purple  
clamped to detector plate

Dec 25, 1932

the back of clamp plate  
wound by spirit lamp. fell  
to 24,000.

**Notebook, N-92-04-16**

The one dated entry in this notebook is from April 1892. All entries are by Edison. The book contains extended theoretical notes relating to gravity and other astronomical forces, the nature and structure of the solar system, and the transmission of light waves and electrical energy. Also included are notes, drawings, and calculations regarding electrical experiments. The pages are unnumbered, and several pages have been removed from the book. Approximately 30 pages have been used. Four loose pages have been inserted into the book.

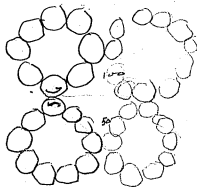


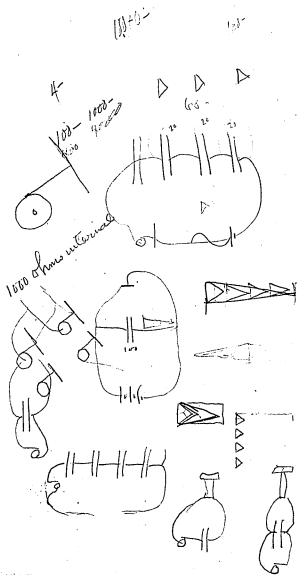
92.04-16

75428

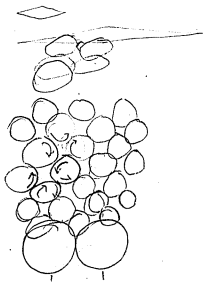
*Acme Co.,*

MFG. STATIONERS,  
96 JOHN ST.  
AND  
19 PLATT ST.  
NEW YORK



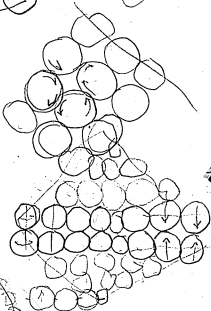
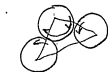
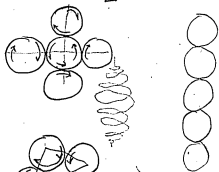


Handwritten text: 4-100 T ac-ca-lee



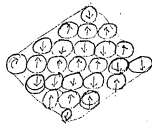
C

2

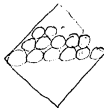


Handwritten text: 2-100

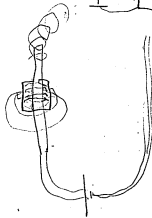


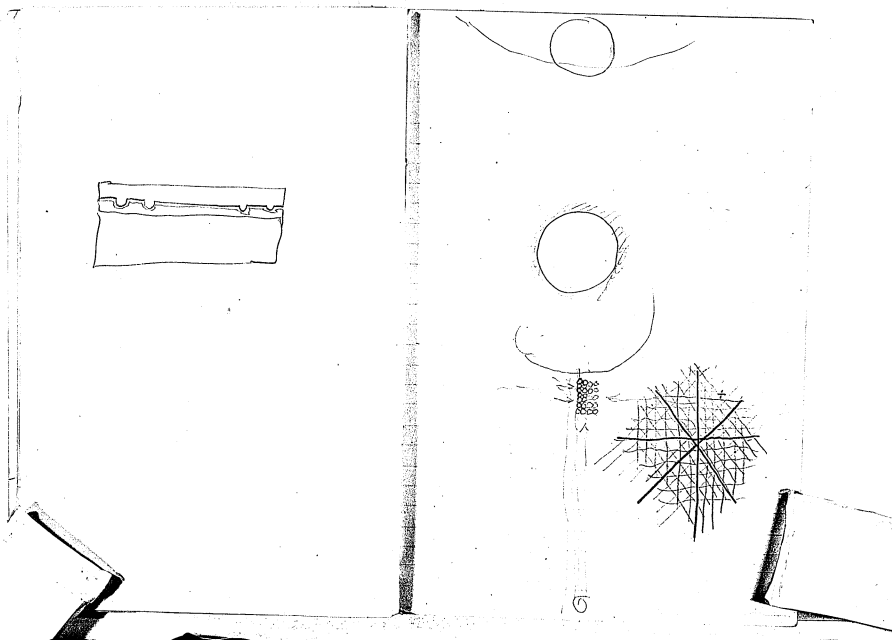


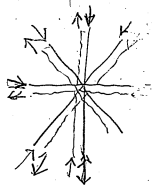
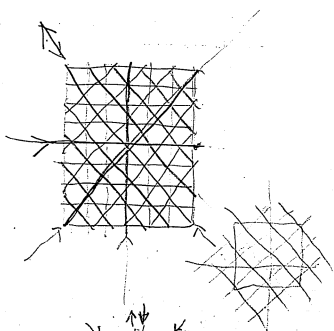
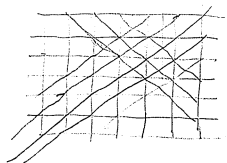
1 3  
5 7  
5 3  
1



Vallage -







Air 1000  
Water 4000  
Steel 16000  
Dirt 18000

1430 ft second

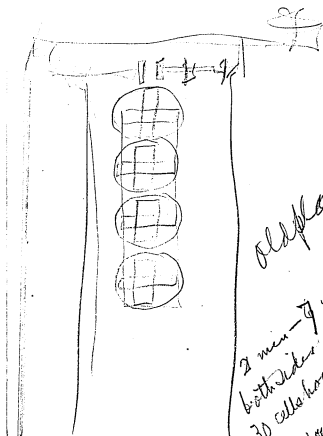
$$\begin{array}{r} 140 \text{ ft} \\ 84 \text{ ft} \\ \hline 56 \text{ ft} \end{array} \quad \begin{array}{r} 1000 \\ 81 \text{ ' } \\ 5400 \text{ ' } \\ \hline 24000 \text{ ' } \\ \text{5000 ' } \\ \hline 19000 \text{ ' } \\ \text{8000 ' } \\ \hline 11000 \text{ ' } \\ \text{3000 ' } \\ \hline 8000 \text{ ' } \\ \text{3000 ' } \\ \hline 5000 \text{ ' } \end{array} \quad (1430)$$

← 18000 ft  
1799

1000 miles  
by time = 1 lb  
pressure

$$\begin{array}{r} 120000 \text{ ' } \\ 184000 \text{ ' } \\ \hline 64000 \text{ ' } \\ \text{30000 ' } \\ \hline 34000 \text{ ' } \\ \text{20000 ' } \\ \hline 14000 \text{ ' } \\ \text{10000 ' } \\ \hline 4000 \text{ ' } \end{array} \quad \begin{array}{r} 1425 \text{ ft } \\ 6 \text{ ' } \\ \hline 85.6 \text{ ' } \\ 30 \text{ ' } \\ \hline 55.6 \text{ ' } \\ 35 \text{ ' } \\ \hline 20.6 \text{ ' } \end{array} \quad \begin{array}{l} 17 \text{ miles a min} \\ 35 \text{ ' } \end{array}$$

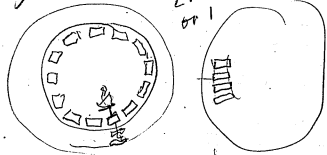
$$\begin{array}{r} 1430 \text{ ' } \\ 35 \text{ ' } \\ \hline 1395 \text{ ' } \\ \text{35 ' } \\ \hline 1430 \text{ ' } \end{array} \quad (17)$$



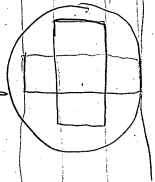
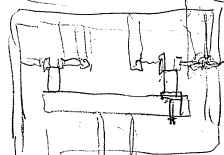
old plan

2 min - 7 wch  
 both sides  
 30 cells hour  
 other side 30  
 makes 15 cells  
 2 hours  
~~1 hour~~

April 16 / 92  
 A 2 / 2 mach  
 or 1



5

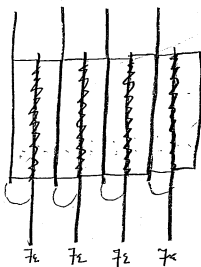
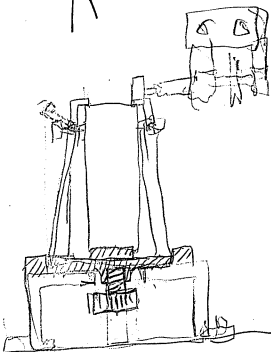


13 cells hour

150 - 100 ft  
 240 2  
 360 3 shafts



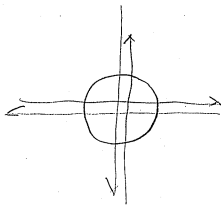
No 1



4 Elements - or pairs  
8 Asbestos  
2 Asbestos to pair -  
80 Asbestos 40 pair  
720

Teddy come to me  
Here is your breakfast  
The bird will want you to  
feed it.

Birds sing on the tree  
Birds feed on the bough  
the storm is rough some times  
We go through the gate to the  
garden



Your world was suspended in space, away from all matter like the sun & planets and neither revolved or moved in space and the primordial bodies moving with a velocity of say 600 times the velocity of light & equally in all directions and these primordial bodies lost ~~the~~ a percent of their velocity in passing individually thru

the earth's matter - it would produce the attraction of one piece of matter for all other pieces of fixed matter and account for the so called attraction of gravitation,

~~If the Earth increased in size or density the velocity of the particles would diminish & the Earth would get hotter deriving the energy from these particles~~

If the earth increased in size or density the relative velocity of the projected particles would diminish & the Earth would get hotter deriving the energy from these particles

hence a large body like  
a sun would be supplied  
perpetually with energy  
& in proportion to the  
density or size -

If these primordial particles  
went in straight lines in  
every direction - & equal in  
quantity, there would be  
no tendency for the  
Earth to advance or turn  
or drift -

Now suppose the earth was  
in motion say at the same speed  
as it now has in its orbit  
around the sun -

It would proceed in a straight  
line -

Supposing 2 or 3 light  
years ahead was our  
present sun several hundred  
millions of miles to the  
right - instead of 92 million  
as now -

If it suddenly appeared  
in this position the earth would  
be affected & could not proceed

in a straight line because the stream of pebbles would be disturbed, a beam of pebbles passing thru the sun would have their velocities changed would be slower and when these struck the earth it would no longer be evenly bombarded by pebbles all at the same speed, those pebbles coming from space on the opposite side to that where the sun is and would have full velocity & that towards the sun lessened velocity hence the earth would be pushed towards the sun, not attracted as by

the present theory  
the result would be  
that ultimately the earth  
would travel in a very  
extended orbit around  
the sun, and this orbit  
must necessarily be  
eccentric - also as  
time goes on will have  
an orbit gradually  
diminished in circumference  
+ ultimately will pass  
into the sun not by  
attraction but pushed  
in =  
Ultimately all over



7  
plants will be forced into  
the sun

All matter is built up  
from these primoids by  
Collisions + Coalescing -  
~~the first~~ The first  
Combination becomes a  
Compound & takes on the  
properties of matter  
light is not carried by  
these Collision Complex  
until their speed falls  
to approximately around  
155,000 miles a second  
when the complex can

is impressed with vibration  
# like a diaphragm  
it can have impressed  
on it the vibrations of  
any other matter like  
Sodium Lithium etc

The stream of these  
complexes passed thru  
Sodium Vapor have  
impressed on them the  
vibrations of the Sodium  
atoms + these complexes  
stream thru space +  
striking the earth matter  
receives these impressions

Vibrations & we see the  
Redium line in the spectrum  
The so called wave length  
has no existence -  
Light may travel  
186,000 miles per second  
but its not necessary  
to suppose that it  
comes on the so called  
ether & be figured in  
wave lengths as these  
periodical Complexes  
travelling at same  
speed may have ~~the~~  
vibrations impressed on  
them which are

a million times slower  
than those figured on  
the Ether & wave length  
system -

and these complex. can  
carry any kind of  
impressed vibrations -  
& we get over space all  
the vibrating rates of  
all the metals etc. as seen  
in the Spectroscopes

(Note, the light carrying complex  
gives light waves by its vibration  
impressed thereon and as the  
amplitude diminishes it ceases  
to give light & then gives radiating  
heat but only because of diminished

### Amplitude

Innocent babies growing light  
Eyes as a child has all wave rates of  
Vibration even when weak, but the  
We know about we cannot appreciate  
being too weak but rise of  
temp increases their amplitude  
and we are then enabled to  
know of their presence.  
Pace of temp simply  
measures amplitude -

---

Light goes in straight  
lines apparently the  
curves of the impressed  
Vibrations actually pass  
through space & impart their

Impressed waves at the  
positive to all fixed or  
amplified bodies -

But Electricity & Magnetism  
apparently always  
move in circuits -

not in a straight  
line - hence the method  
of transmitting light  
& heat energy  
must be quite different

Can Wheatstone's  
measurements of velocity  
of Elec be anywhere  
near correct,

I believe Elec Energy  
has not the speed  
of light Energy  
there must have

been some large  
sources of error in it  
in fact he made it  
faster than light  
which is suspicious

13 perfect

$\frac{18}{200000}$

$\frac{360000}{2000}$

216000

250 40000  
172 20000

250000  
150 000  
100 000

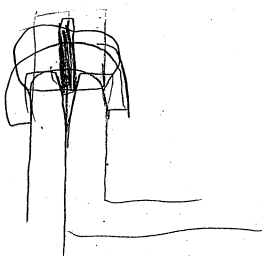
13 500000  
500000  
200000

150  
170  
10500  
150  
2550

120

150000  
30000  
30000  
30000  
120000  
120000

15  
18000  
12000  
350000





[ITEM FOUND IN BOOK]

①  
The Corpuscles which carry light is a compound of the ultimate entities brought about by collisions - probably there are from 2 to 4 particles which have coalesced

These particles are slowed down to 150,000 miles second they have a natural vibrating time near that producing the Sodium line

②  
is a multiple of that time - but of course can have any rate of vibration forced or impressed upon them

The Entities striking matter like a note absorbs the vibrations & transfers it to other light carrying entities which of course pass through space

[ITEM FOUND IN BOOK]

3

in all directions

The notes as compressed  
to the size of the light  
Carrying entire area  
large pieces of matter  
This absorb the vibration  
& dampen it the light  
particle proceeds  
without any or very  
small impressed vibrations  
accomplish of the  
note is spaces with  
considerable of transparency

4

Little case of a Column  
of water. The vibrations  
impressed on the light  
Carrying particle is  
~~entirely~~ absorbed below  
the vision point in perhaps  
50 feet, but the light  
Carrying particle still  
has high velocity &  
penetrates to unknown  
distances into matter

**Notebook, N-94-12-11**

The one dated entry in this notebook is from December 1894. The first page bears the inscription: "Mr. J. Coote Edisons Laboratory Orange, N.J. Dec 11th 94." All remaining entries are by Edison and pertain to squirted filament experiments, with notations for boxes numbered 1-20 and brass boxes numbered 1-8. Near the end of the book is a page entitled "Index," which has no apparent relation to the preceding entries. The front cover is labeled "Various Experiments By Mr J Coote." The pages are unnumbered, and several pages have been removed from the book. Approximately 35 pages have been used. Two loose pages of calculations have been inserted into the book.

X E-172

N-94-12-11

*W. J. Dole*

Edison Laboratory  
Orange, N.J.  
Dec 11 1894

## No. 1. Box

Contains. Ayls file. Cbz<sub>2</sub> in  
Hg above 700. called 2<sup>nd</sup> Cbz<sub>2</sub>  
because file only cracks in  
flame & does not crack after pulling  
out as. first Cbz<sub>2</sub> in Hg 2<sup>nd</sup> is

The above file was heated  
in an oven to 550 Fahr &  
dropped vertically in the oven  
into Carbonate dissolved in  
H<sub>2</sub>O -

Box No 2

2nd Hg Cl<sub>2</sub> then file  
put in oven & when 480  
dropped into Carbonate  
dis in Dipi

Box 3

Fils Carbonized in Pacific  
to 450 deg Fahr -

These fils brought up in  
oven to 450 & dropped  
in Carbmitic disc in Dipi

Box 4

Carlyden paraffine to 450 fah

These files heated in oven

to 450 & dropped in

Syrian asphalt also  
in Bengal.



Box 5-

Carlyle 2<sup>nd</sup> Hq

fit then healed to

450 + draped in

Syrian in Bengal.

Box 6

Carbyd 2<sup>nd</sup> Hg

fil heated in oven to

550 & dropped in Syrian

in Bengal

Box 7

Carbyd 2<sup>nd</sup> Hg -  
fil heated to 550. fols  
+ dropped in Hoffmann  
in Mal

Box 8

Carboxym Paraffine-

450 deg then fil

heated to 450 deg

in oven & dropped

in Hoffman's hot water

Box 9.

Clyde 2<sup>nd</sup> Hg.

fil heated to 550 °

drapped in Hoffman ✓  
in Mal.

Box ten -

2nd Hg Cyl

fil heated to 440 +

drapped in No 33 in mal

Box 11

Clyn 2<sup>nd</sup> Hg

ful heated to 550

+ dropped in 33. w/val.

not cleaned well of 33

w/ none of file -

Box ~~18~~ 20

Carbonized in paraffine at

450 -

fil heated to 450

↓ dropped in X<sup>o</sup> 23 in mal



Box .13

Clym 2<sup>nd</sup> Hg.

fil heated to 450 +  
drapped in Sugar  
in Water, strong Cold  
Solution

Box 14

Cognac Parafine 450

fil heated 450 +

drapped in Sugar



Box 16.

Contains files called  
1st Cbgn by Hq went up  
probably to 700.

They catch fire after  
withdrawn from flames  
which 2nd Hq Cbgn  
don't. - These are  
plain files nothing done  
to them -

Box 17.

These files are 2nd Hg  
Ctgn & went probably  
to 750 @ 800. They  
do not ex after coming  
out flame, buck  
worm hole somewhat  
plain file nothing  
done to them -

Box 18-

Contains film  
Carbonized in paraffine  
up to 450 degrees  
plain film nothing  
done to them -

Box 19

Eight 2 fee filaments  
in parafine that has been  
under heat several times.

run to 600. then run 1 hour  
after thermometer taken out  
probably run to 700 to 750  
feels shiny good - 8x after  
taking from flames -  
5/1000 diameter =

These files are part of a lot  
that was run to 500  
Faking 4 hours, then  
Cooled down & one lot  
taken out, those remaining  
were run up again quickly  
as possible to 500 &  
held 2 hours, then Cooled  
off & run again to 600  
over these No 19 are  
the remaining =

Press Boxes Numbered 1 to 8

Box 1 - Press -

No 1 Contains about 60 files  
small die, 3000 Cbt size Crystals  
in ether, shine good spots good  
no stick or stop, but there is  
tendency to long Crinkles  
~~too~~ there are a number of  $\frac{1}{2}$  size  
spots along file - apparently  
no puff - used steam -  
die body - jet for little while  
to heat Die - These spots  
may be due to not stirring the  
Cry in ether was I failed to  
do it I divided the 3000 Cbt  
in two doses, this we second  
dose had aside for 6 or 8  
minutes covered over with  
bottle dish front glass washed  
as we had too large a die -



Box 2.

3000 Cbt 500 Cuy well mixed  
with ether enough ether to make  
whole thin layer in mortar.  
Worked it till caked + semi dry  
then washed out mortar +  
put it back + washed to  
dark red powder, was  
some slight pieces in it.  
Steam around die a for 1/2 minute  
short flame to heat die.  
Squeets good, shiny but  
tends to crack at times  
small die, put all the file  
from both batches in one  
box = There are pieces  
on surface of file but fine  
many not hurt, file don't  
occur puffs  
file measure  $5 \frac{9}{10} / 1000$   
average

Box 3

Mixed 6000 ckt 1000 Cuy  
+ double quantity<sup>in</sup> ether so as  
to make 4 changes for die-  
cast standard to crinkles,  
only used 2 batches one  
I thrown away this is batch  
used, files have cracks &  
some specs & probably little  
puffs - but not very much  
puff, the puffs are only  
in little specs not general  
puff of whole filament,

Box 4

to 3000 Cst 700 Crosby -  
10<sup>cc</sup> litres - than cakes somewhat  
in morning 2t crinkles &  
regions just as much heat  
as 500 Crosby - We concluded  
that die contraction of press  
was rough so freely replaced  
them all for the next box  
will we hope be better -

## 5 Box

3000 - 500 Cops

Sprouts have no breaks but very  
Cracky - don't have good polish  
think it will stand more  
heat on 2<sup>nd</sup> portion,

2<sup>nd</sup> portion run plunger right  
down before Casey could get away  
& used more steam (no flame) &  
sprouts without a break  
didn't Crumble - file are  
slimy but some are not  
perfect, we put both into  
Crumble & not in this box.

Box 6

3000 + 500 Croy  
not very good -  
mixed with some that  
is good + was cleaned by  
Ethos - We find our Cbt  
is not cleaned with Mal  
We must now clean some  
with Mal + gasoline if  
we are to get good fire

---

Box 7.

3000 Cbt 400 phene.  
in ether - 9 cc = 1st part used immediately  
only boiling water around press no  
flame only partially shiny  
but perfect, no puff - thin & file  
square perfect, file not bigger  
than die - not crumbled

2nd part also in this box  
was dried 10 min in air to  
drive off all volatile stuff &  
solidify silver matrix  
square perfect, but slight  
body puff & crumbled,  
too much heat, if less was  
not shiny & as it was  
file one still but perfect,  
surface = 10<sup>th</sup> (they compare in  
and best of all yet tried)  
shank in air, don't puff  
& are Xtra good over all other  
yet tried =

Box - 8

3000 Cft 300 ft level

Squirts good - watching  
Some cracks - no puff -

## Carbyn

Box No 1 + 2 contain files of  
3000 ckt + every various kind  
Ckts in Caster and at 550.  
Don't think they will amount to  
much; we ckt in all they will  
relegate a break may get  
few out of them - we washed  
them in hot Benzol -

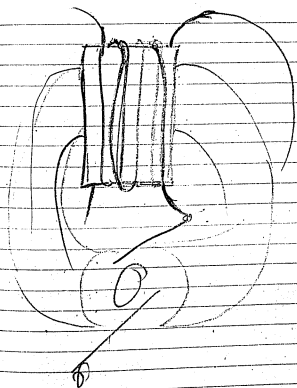


Clyn

Box No 3 Regular LF fets  
brought up to perhaps 450 in  
Castrol oil thickened with  
Spartan asphalt until when  
cold it was thicker than  
Melasses - Then washed the  
fets in Benzol -

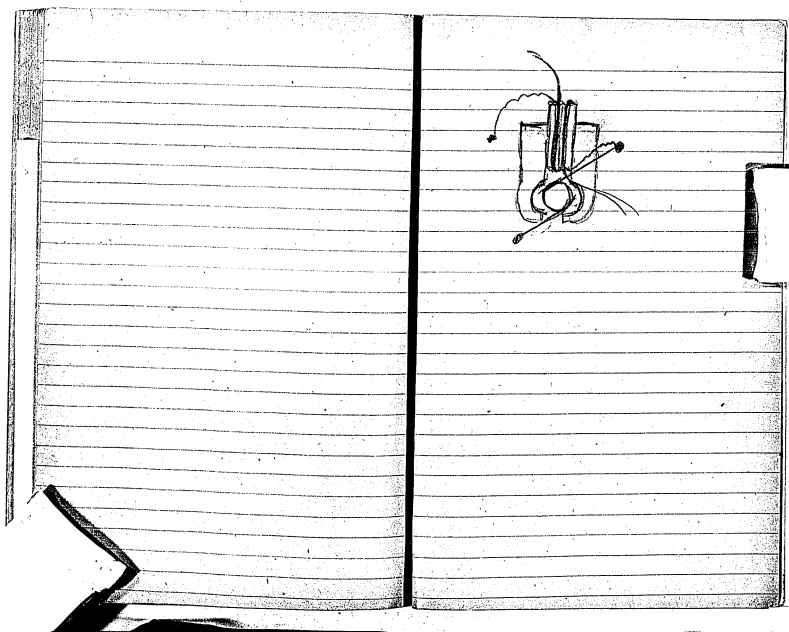
Clay

Box 4 are 3 lots of film I  
made with Cbt purified by Mal,  
afterward purified fairly well  
by ether, then used 300 mg  
phosphor in ether, ground very fine  
& dry (no longer phosne dry, etc.)  
then with slight warmth on  
body disc & warm spinning disc  
spun without any pressure  
quite perfect, we try these  
to see how they come out by  
plain carbon.



# Index.

|      |                     |        |
|------|---------------------|--------|
| No 1 | Sulfate of Magnesia | page 1 |
| " 2  | " " Iron            | 1      |
| " 3  | acetate of Lead.    | 1      |
| " 4  |                     |        |
| " 5  |                     |        |
| " 6  |                     |        |
| " 7  |                     |        |
| " 8  |                     |        |
| " 9  |                     |        |
| " 10 |                     |        |
| " 11 |                     |        |
| " 12 |                     |        |
| " 13 |                     |        |
| " 14 |                     |        |
| " 15 |                     |        |
| " 16 |                     |        |
| " 17 |                     |        |
| " 18 |                     |        |
| " 19 |                     |        |
| " 20 |                     |        |
| " 21 |                     |        |
| " 22 |                     |        |
| " 23 |                     |        |
| " 24 |                     |        |
| " 25 |                     |        |



$$\begin{array}{r} 1800 \\ 2000 \\ 100 \\ \hline 3800 \end{array}$$

$$\begin{array}{r} 52 \overline{) 1700} \quad (290 \\ \underline{104} \phantom{0} \\ 660 \\ \underline{596} \\ 64 \end{array}$$

30 hours / 18 books

[ITEM FOUND IN BOOK]

|    |                       |        |                       |
|----|-----------------------|--------|-----------------------|
| 1  | 400                   | Para   | CB                    |
| 2  | 430                   | "      |                       |
| 3  | <del>440</del><br>440 | Para   |                       |
| 4  | 450                   | "      |                       |
| 5  | 450                   | 2nd Hq |                       |
| 6  | 550                   | "      |                       |
| 7  | 550                   | "      |                       |
| 8  | 4750                  | Para   |                       |
| 9  | 550                   | 2nd    |                       |
| 10 | 440                   | 2nd    |                       |
| 11 | 550                   | 2nd    |                       |
| 20 | 450                   | Para   |                       |
| 13 | 450                   | 2nd    |                       |
| 14 | 460                   | Para   |                       |
| 15 | 550                   | 2nd    |                       |
| 16 | 450                   | 1st Hq |                       |
| 17 | 450                   | 2nd    |                       |
| 18 | 450                   | Para   | Nothing done to this. |

[ITEM FOUND IN BOOK]

|    | gms  |       |     |         |  |
|----|------|-------|-----|---------|--|
| 1  | 70.9 | 83.5  | 369 |         |  |
| 2  | 66.7 | 79.7  | 377 |         |  |
| 3  | 74.0 | 85.18 | 370 |         |  |
| 4  | 74.7 | 82.9  | 375 |         |  |
| 5  | 69.8 | 82.5  | 372 |         |  |
| 6  | 69.3 | 81.6  | 378 |         |  |
| 7  | 68.8 | 83.3  | 378 |         |  |
| 8  | 72.0 | 86.4  | 381 |         |  |
| 9  | 65.7 | 77.3  | 397 |         |  |
| 10 | 67.5 | 81.7  | 383 |         |  |
| 11 | 68.2 | 81.8  | 385 |         |  |
| 13 | 67.4 | 80.0  | 388 |         |  |
| 14 | 69.0 | 81.7  | 389 |         |  |
| 15 | 69.5 | 82.3  | 383 |         |  |
| 16 | 67.7 | 79.8  | 394 | How     |  |
| 17 | 68.2 | 84    | 389 | 123 3/4 |  |
| 18 | 68.1 | 82    | 389 | 50.5    |  |
| 19 | 64.9 | 74.4  | 396 |         |  |
| 20 |      | 76.3  | 386 |         |  |



**Notebook, N-95-01-24**

This notebook contains dated entries from January 1895. All entries are by Edison. At the beginning of the book are notes and drawings regarding the phonograph, including notes and hypotheses relating to phonograph diaphragms and sound reproduction. There are also notes, drawings, and cost calculations pertaining to ore milling. The pages are unnumbered, and the book has been used in both directions. Several pages have been removed from the book. Approximately 130 pages have been used.

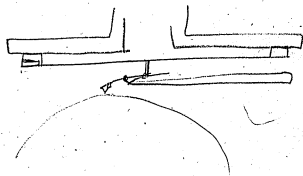
XE172 N950124

Notes

$3\frac{1}{2}$  total -

$3\frac{1}{10}$  amp

$\frac{1}{10}$  amp per inch



$\frac{.16}{.175}$   
0.09

004

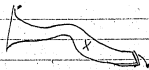
1/2 -

350 )  $\frac{2000}{1750}$  (5.7  
       $\frac{2500}{1750}$  1/4

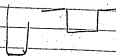
Expt 1

Bore 1/4 hole through floor  
& was stretched like wire  
Cap. # 36 also thread & test  
dia for music & spec  
a la lower length =

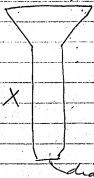
Ear has a resonant tube



dia dia .350/1000  
x 2 inches long or 517 times  
dia of Drum - if ours  
is proportional, it should be  
about 12 long  
X is about 1 1/4 times dia or 1/2  
inch - hence ours should  
be 3 inches dia this seems  
too much by diff dia



perhaps the thin chamber of  
phono which is absent in Ear  
Reverberator, hence it would  
be better not to have the  
chamber on recording but  
a tube 12 long & 3" dia



Reverberators are unlined by  
glass or eborony boards  
hence perhaps a leather  
horn X would be better  
so as not to give reverberations

As in all our Expts  
with new Receiver speaker  
etc & the 10 @ 15 high note  
Keys failed to come out

The friction method may be  
defective in reproducing fine  
vibrations & that improvement  
should be made on it &  
~~with~~ the same records  
showing these defects should  
be tested with other  
reproducers -

In all 'long tubes or funnels  
sounds are muffled &  
fine notes & specific tones  
with repeat Records  
fail although speaker is  
normal through the  
Respeaking tube

Jan 24/95

Evidently there is a great disturbance of the proper vibrations of the diaphragm by the friction of the point in tracking into varying friction during vibration. The more brittle & less tenacious the material is the less this disturbance - as a definite tracking is required hence there will be a definite friction whether the sounds are strong or weak hence the friction might be a small factor with loud sounds is a great factor with weak ~~at~~ sounds hence the very weak sounds must have friction reduced to a minimum which can be done by light tracking & better by using a 5/16" cutter. It is very probable that the diaphragm vibrates freely to the sound & the

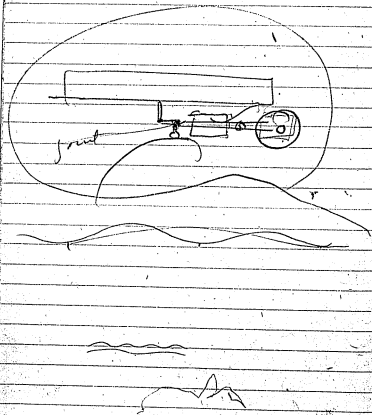
But perhaps the reproducer  
 should be same Curvature  
 as cutter in one way  
 & twice the dia. the  
 other way if double  
 spec.



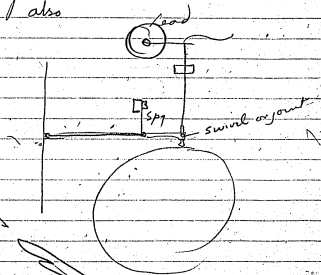
July 24/95

disturbance is purely due to  
 the friction of the point.  
 I have noticed that hissing  
 sounds come out abnormally  
 on very hard material  
 with extremely light tracking  
 Turn for instance the hissing  
 exceeds the vowel sounds  
 for threads to which with  
 only way sounds from  
 distance can be taken &  
 point should be 3 @ 5/1000  
 dia to get depth without  
 width.

As the friction doesn't increase  
 with the speed of cylinder  
 the wearing property of the  
 record can be increased  
 by doubling the speed.  
 Thus we can 1/2 the friction  
 by taking out 1/2 the wax  
 chips & at the same time  
 make total out of wax  
 forming an undercutting.  
 The same hence the wear  
 will be same as double width  
 & 1/2 speed. See page



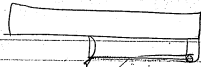
We should try 3 @ 5/1000 glass  
 in regard to the amp but  
 ground metal  
 also



also

July 24 1895

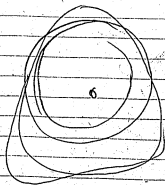
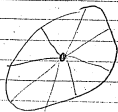




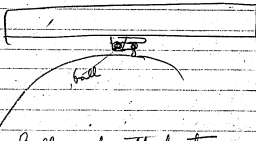
This probably should be  
wood to lighten - perhaps  
whalebone. Sulfurized fibre etc,

The Expts should be made in the glass  
board recording points  
adjustable as to angle easily  
soft shellac to hold the carbon  
sides - some  $\frac{1}{8}$ " @  $\frac{3}{16}$ " long to  
facilitate adjustment,

In recording as well  
as reproducing the  
diaphragms probably  
want to continue vibrating  
after the waves have ceased  
for a fundamental having  
ceased the overtones want to  
continue at lower than they  
should hence a big  
better dampening part



July 24 1895 -  
 might be of use in such



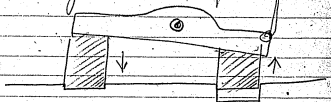
Ball joint with friction  
 Spring strong enough to overcome  
 friction hence eccentricities  
 of cylinder ok but req vibs  
 just to dia. if friction  
 insufficient use belt so  
 lever must move exp. drive  
 to change relation

Note an oval dia  
 so that distance to, ring  
 is everywhere different  
 from the center where  
 point is fastened

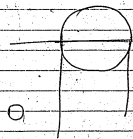
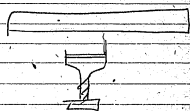
Jan 24 1895

Dr

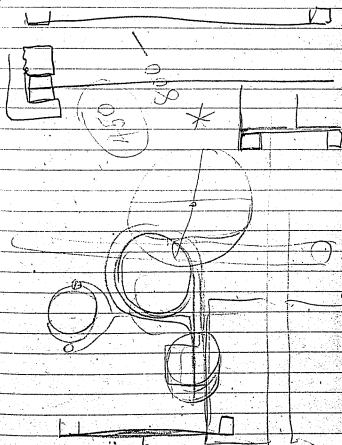
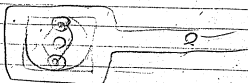
To make a perfectly even  
friction at cutting point  
it might be arranged like

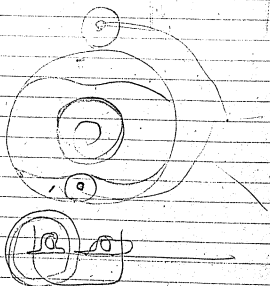


Adjusted so one is in as  
the other comes out,  
This has the advantage also  
that the reproduction is  
a positive movement can't  
jump & it may be important  
as record will not be much  
worn much & very small  
recording points used to diminish  
friction



7/11





300

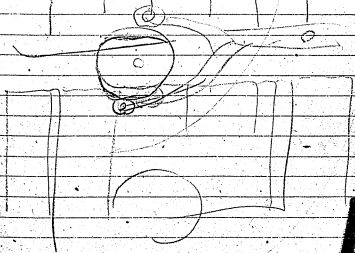
600-

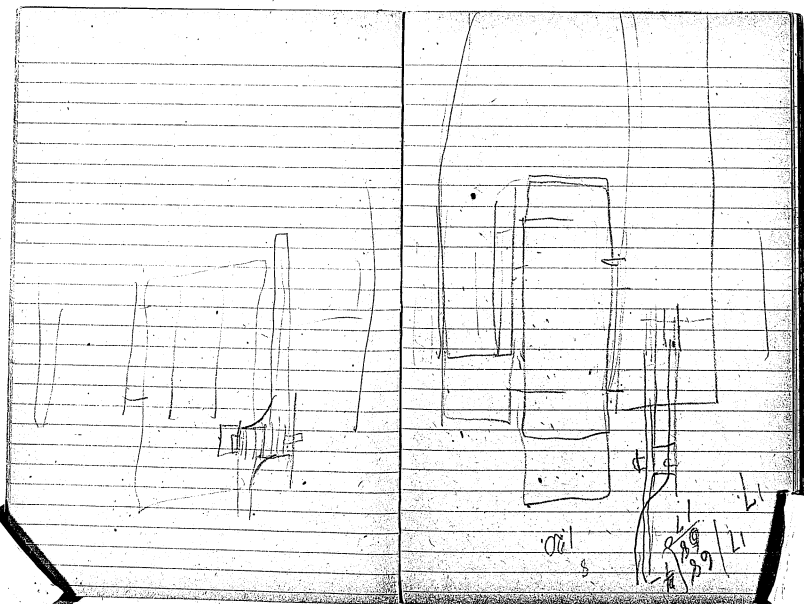
900

150

$$\begin{array}{r} 600 \\ 120 \\ \hline 720 \\ 600 \\ \hline 1320 \\ 9 \overline{) 1320} \\ \underline{900} \\ 420 \\ \underline{360} \\ 60 \end{array}$$

40-





oil  
s

1/2  
2/2  
3/2  
1/1

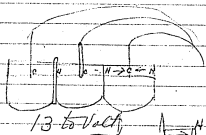
2004  
-20  
180  
2004  
180  
200

5000

112  
5  
20  
7  
8

6  
5  
4  
3  
2  
1





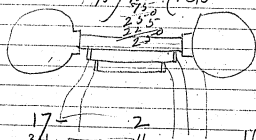
13 to 1/2 inch

225  
450  
900  
1125

15  
30  
60  
75

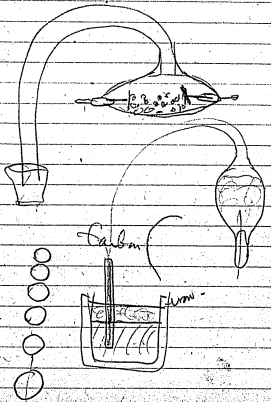
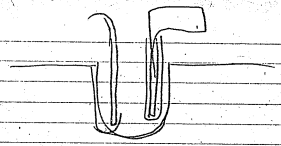


75 / 1000 (13,3)  
235  
225  
225



17 = 2  
3/4 = 4  
68 212  
136 = 180  
16 43  
75  
32  
43  
223

16 1/2 lfy  
& lenses



40  
 27 8 trunks 500  
 175000  
 2000  
 250000  
 500000  
 13  
 400000  
 180000  
 1300  
 4  
 715000  
 680000  
 200000  
 200000  
 170000  
 200000  
 194000  
 130  
 150  
 50  
 300  
 125000  
 200000  
 3  
 600000  
 300000  
 300000  
 4

1 mile runways  
 52000  
 250  
 250  
 250  
 250  
 150  
 300  
 1000

5 2500 tons / 80 tons each, 200000 5000  
 Corrugated sheets 30000 10000  
 Steam shovel 24000 7000  
 Engine - 30 h.p. + pump - 58000 18000  
 Bldg. 8000 3000  
 10 Roads 120000 70000  
 Motors 4000 3000  
 Dynamometer 10000 3000  
 Cincinnati Eng. Dynamometer 7500  
 Machine shop - 25000 - 6000  
 Drays 7000 2000  
 6 36" Rolls 25000 8000  
 1 5 ft Roll 9000 3000  
 6 5 ft Highs - 24000 8000  
 Runways + Belts 52000 12000  
 Other belts 7500 4000  
 Climber scrapers  
 2700 ft scraper etc. fl. } 4200 4000  
 Screens + frame 6000 2000  
 Gears all conveyors - 5000 1500  
 Elevator - 2500  
 Coal grinding 7500 2000  
 Del. House 3000 1000  
 Rope drum pulleys 7000 2000  
 Top crushing plant 10000 2500  
 Hopper 5 ft x 6 ft 7500 2000  
 Hopper 3 ft  
 715700

Profit forward 715 700  
 100 Skp + tax 142 000  
 docos. 25 000  
 erection foundation etc 40 000

Land

822 700

69 000

891 700

196 000

695 700

697

6.26.60

60

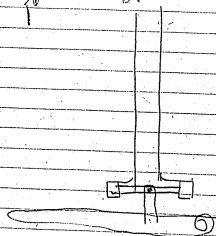
5000

120

24.20

45

2000



- 822 700

822 700

Piping pumps - 12 000  
 Reservoir 4 000  
 Shovel work + frame opening 8 000  
 Extra appliances roads 10 000  
 Lighting 10 000  
 Steam piping Feed heater 10 000  
 Blowers - 5 000  
 Oil system 8 000  
 Steam Heat 12 000  
 Trucks + Ties 18 000  
 office + Storehouse - 6 500  
 Road House - 2 500  
 Air Compressor etc boiler - 7 500  
 Shop, Boiler + Heat 2 500

938 700

168 000

1,006 700

|                                      |                   |            |
|--------------------------------------|-------------------|------------|
| 460 <sup>00</sup>                    | 48                |            |
| 92 <sup>00</sup>                     | 12                |            |
| <hr style="width: 50%; margin: 0;"/> | 96                |            |
|                                      | 48                |            |
|                                      | 572 <sup>00</sup> |            |
| 15000                                | 1152              | 1500-      |
| 1728                                 |                   | 34800 (20) |

|                   |                    |                    |
|-------------------|--------------------|--------------------|
| 24                |                    | 10                 |
| 96                |                    | 8                  |
| 48                | 460                | 460                |
| 548               | 52 <sup>00</sup>   | 368 <sup>00</sup>  |
| 480 <sup>00</sup> | 460                | 1470 <sup>00</sup> |
| 1728              | 33                 |                    |
|                   | 5184               |                    |
|                   | 536 <sup>00</sup>  |                    |
|                   | 460                |                    |
|                   | 33                 |                    |
|                   | 1380 <sup>00</sup> |                    |
|                   | 11184              |                    |
| 35000             |                    |                    |

|     |       |      |
|-----|-------|------|
| 30  | 46    |      |
| 270 | 7500  | 200  |
|     | 10000 | 300  |
|     | 9000  | 360  |
| 30  | 8000  | 2400 |
|     | 10000 | 400  |
|     |       | 500  |
|     |       | 200  |
|     |       | 300  |
|     |       | 50   |

Lead tetraphenyl etc  
 grechem Soc 54 1888 283  
 See Polk's (Ber 20 3331 3336  
 See Chem Soc jnl 1887 572

Dichroins, Compounds having  
 fluorescent & dichroic qualities  
 See H Brunner & P Chinit  
 Ber 21 249 256 Chem Soc  
 jnl 54 1888 p 363 —  
 also Krauser & Brunner  
 Chem Soc jnl 1884 1354  
 ditto de Brunner & Co Soc jnl  
 1874 693 — Ent big book

Phosphorescent Salts  
 A Verneil Compt Rendu  
 106 1104 1107 ditto (107) 106  
 Chem Soc jnl 1888 1248

Actin Line Cyanide or Melles  
 Chloride by R Varet Compt Rendu  
 106 1080 1083 also Chem Soc  
 1868 54 799 780 - 900

For dissolving Cellulose

Acetone Chloroform Chem Soc  
Jnl 1888 810  
J abs of pr Chem [2] 37  
361 374

Action of Pyridine on Metallic  
salts, Chem Soc Jnl 1888  
54 850-77

For giving <sup>gent</sup> ~~gent~~ <sup>gent</sup> ~~gent~~ <sup>gent</sup> ~~gent  
For giving <sup>gent</sup> ~~gent~~ <sup>gent</sup> ~~gent~~ <sup>gent</sup> ~~gent~~  
Felix & Schuler~~

Uranate of Ethyl methyl  
Etc amines C Soc Jnl  
1888 918

Facility best absorbed by  
95% alcoholic sol Ag nitrate  
if in excess dis precipitate,  
~~all~~ all hydrocarbon of acetylene  
Derives from Compound  
mostly Enthalpiae Soc  
S Chem Jnl 1888 930-  
Ent

Da or K Tungstate Equal  
Molecular precipitate from  
salt with Tartaric acid  
C S Jnl 1888 938

Aniline Combines Cuprous chloride  
C Soc Jnl 1888 941

Luminescence of pyrazolite  
C Soc Jnl 1888 1000

Action alk phosphates  
on alk Earth fusion good  
C Soc Jnl 1888 1033

Double phosphate

Magnesium group

C 15 June 1888 1035 9001  
fusum

Benzene derivatives of high

Molecular weight

pentaerythritol para fatty Ketone, etc

Ria aff<sup>nd</sup> Ber 21 2265

2271 also C Soc June

1888 1087 ditto 1887

252

Dichroism Chem Soc June

1888 1182 1183

Chl Zinc Carbonyl with

amines & basic organic

Substances picolin etc -

Methylamine - Crystallized  
from alcohol Chem Soc  
June 1888 1281

Dichroic Crystals Chem Soc June

1880 41

ditto p 105 106

Phosphoric in Vacuum -  
PtCly Kay - etc Chem Soc

1880 ~~48~~ 204

Diphenyl Salt Chem Soc

1880 476 477

Dichroic Crystals

Chem Soc 1880 644

Uranium oxyfluoride Compounds

with fluoric acid Alkali Metals

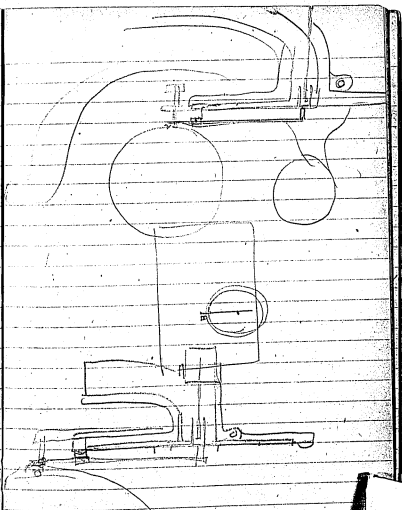
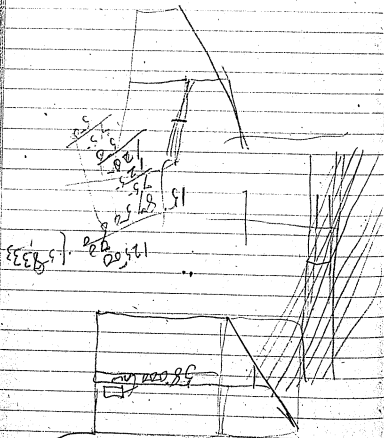
Chem Soc 1880 794 also

853  
A Dittie Compt R 91 166 168

dittie 91 1153 ~~Compounds of Uranium~~  
Compounds of Uranium

Double Salts Propionic acid

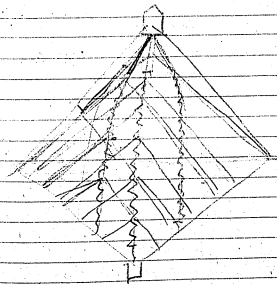
Chem Soc Jul 1880 799

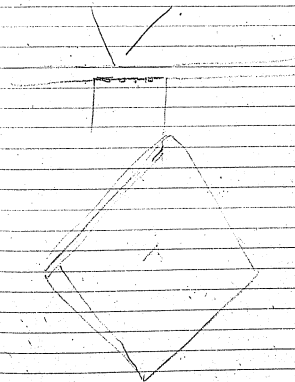
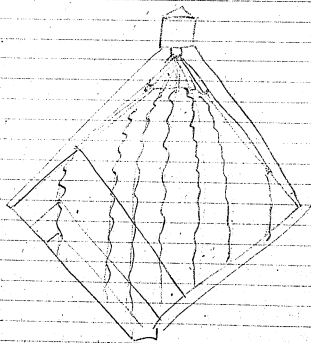


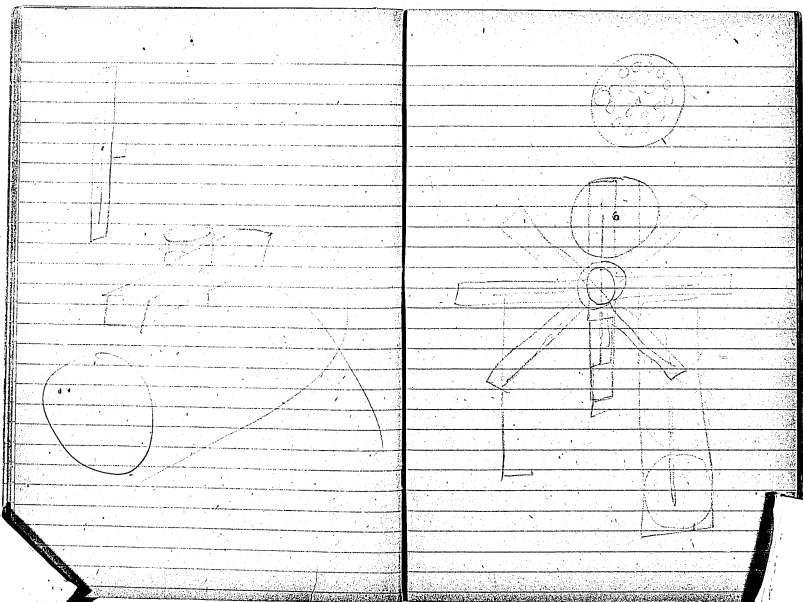


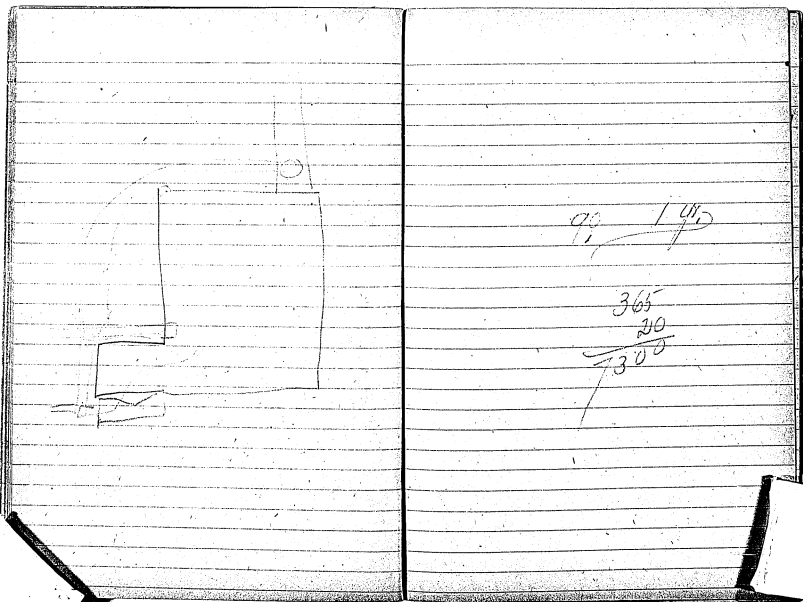
10.0  
4.5  
1.0

10  
4  
1









$$\begin{array}{r} 90 \\ \hline 145 \end{array}$$

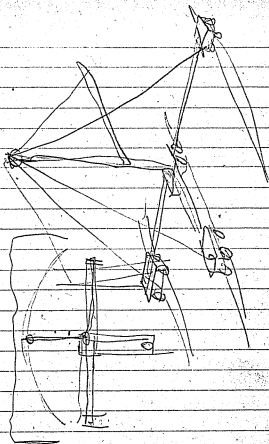
$$\begin{array}{r} 365 \\ 20 \\ \hline 7300 \end{array}$$

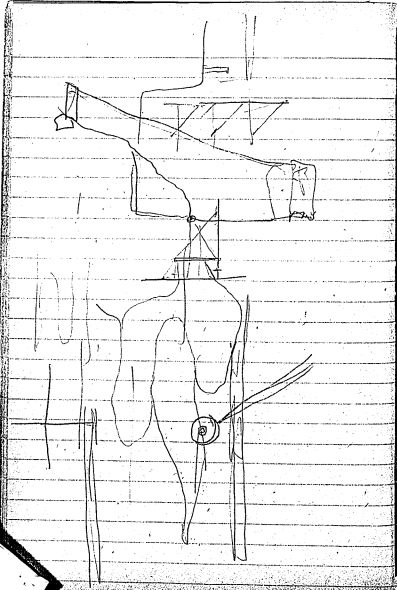
$$\begin{array}{r}
 625 \\
 \underline{16} \\
 875 \\
 \underline{625} \\
 1000
 \end{array}$$

$$16 \overline{) 1000} \begin{array}{l} 625 \\ \underline{96} \\ 40 \\ \underline{32} \\ 80 \end{array}$$

|     |     |                     |
|-----|-----|---------------------|
| 16- | 10. | <del>10.</del> .625 |
| 17  | 10. | .625                |
| 18  | 12. | 250                 |
| 19  | 13. | 875                 |
| 20  | 14. | 750                 |
| 21  | 15. | 275                 |

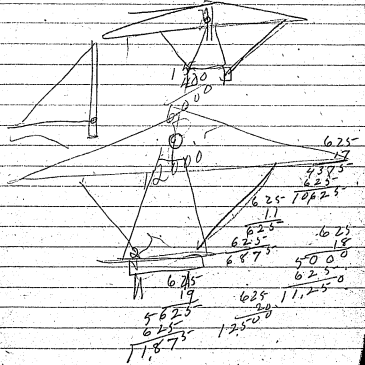
|     |        |
|-----|--------|
| 16- | 10     |
| 17  | 10.625 |
| 18  | 11.250 |
| 19  | 11.875 |

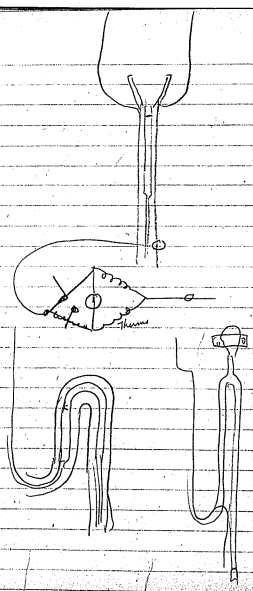




625

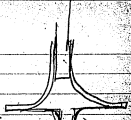
|      |         |
|------|---------|
| 16 - | 10      |
| 17   | 10.625- |
| 18   | 11.250  |
| 19   | 11.875  |
| 20   | 12.500  |





Formic  
acetic  
Propionic  
Butyric  
Capric  
Caprylic  
Pelagonic

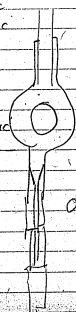
Carbonic  
Glycollic  
Lactic



Malic 8  
Citric 12

Tartaric  
pyrogallic 2500  
Oxalic  
Crotonic 58 1/2

Oxalic  
Malonic  
Succinic  
Lipic  
Adipic  
Pimelic  
Suberic  
Ancheric  
Sebacic



Angelic  
Pyroterbic  
Damalunic  
Camphoric  
Moringic  
Hippogadic  
Oitic  
Parassic

Benzoin  
Toluylid  
Cuminic  
Acrylic  
Adipic

Oxybenzoic  
Oxytoluylid  
Phorbretic  
Oxycuminic  
Aconitic

$\frac{20.64}{2.5}$   
 $\frac{4.07}{1.1}$   
 $\frac{3.58}{1.1}$

#1

$\frac{7.5-7.5}{3}$   
 $\frac{7.5}{3}$   
 $\frac{7.5}{3}$   
 .258

$\frac{2.04}{2.5}$   
 .816



| 0.74 | 4 |
|------|---|
| 2    | 1 |
| 3.75 | 1 |
| 2.58 | 1 |
| 4.07 | 2 |
| 2.00 | 1 |
| 2.00 | 1 |
| 1.52 | 1 |
| 3.00 | 1 |
| 2.70 | 2 |
| 1.00 | 1 |
| 1.41 | 1 |
| 5.00 | 3 |
| 2.33 | 2 |
| 9.00 | 6 |
| 1.50 | 1 |
| 1.50 | 1 |
| 3.00 | 2 |
| 2.00 | 4 |
| 2.5  | 1 |
| 1.71 | 1 |
| 2.5  | 1 |
| 2.08 | 2 |
| 2.07 | 4 |
| 1.57 | 1 |

allophanic  
 Allanturic  
 allitunic  
 alloxanic  
 allthionic

Antimonates  
 Oxymale Antimony  
 Sulphantimonates

Arsenites  
 Arsenates  
 Oxyarsulphide Barium

Double Benzoyles

Bromobenzoyles  
 Chlorobenzoyles  
 Nitrobenzoyles  
 Nitrochlorobenzoyles

Cymide (Benzoyl-Benzoyl)  
 for date & etc



Hydride of Benzoyl Carbonyl  
with Chloral also with  
acid sulphites of alkali  
metals

Hydrides of Nitrobenzoyl

Benzylamine (Toluidine) ( $C_7H_9N$ )  
Combines with acids

Cyanobenzylamine

Brown Nitride (Lohite)

Bromates best made by fusion  
where it is a strong acid

But want all the Bromates  
that can be made in wet way

Bromoacetates  
Dibromoacetates

Bromates

(Pamphonic acid)

Cyanides with Alkaloid bases  
a amines -

Conine Salts of Baccaine, Caffeine

Conine Amine, Styrachinine

Conine Quinine, Conine Alkaloid

Butyrates, Conine

Double chloride Bromides & iodide  
of Cadmium - They form very  
double salts

Chloride Cadmium forms double  
salts with hydrochlorides of  
organic bases Vol 1 704

Oxychloride of Calcium

CaCl<sub>2</sub> CaO with ammonia  
also with Chloric acid or  
acetate or Oxalate Calcium

Copper Nitride will  
nitrogen dissolves in  
Cyanide potassium 5, 5

Cuminic acid - good.

Nitrocuminic acid

Hydride Cummil unites with  
acid sulphites of alkali  
nitrate

Cyanelic acid

Cy made of ethyl formic acid  
Compounds with some metallic  
chlorides

Sulphaldehydes or Mercaptides

Fumaric acid

Malic acid (flavour)

Glyceric acid  
glycollic acid

Diluturic Acid, (good)

7 arm 3' 22' low

$$\begin{array}{r} 13 \\ \underline{32} \\ 67 \\ \underline{130} \\ 720 \end{array}$$

$$\begin{array}{r} 68 \\ \underline{377} \\ 357 \end{array}$$

$$\begin{array}{r} 898 \\ \underline{10} \\ 898 \end{array}$$

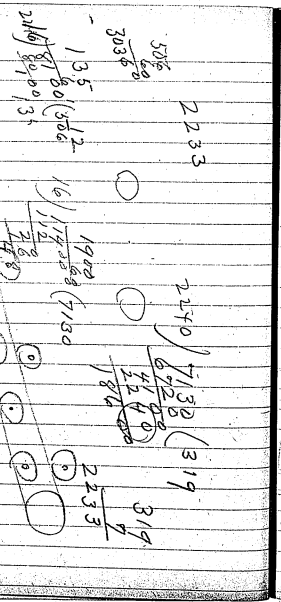
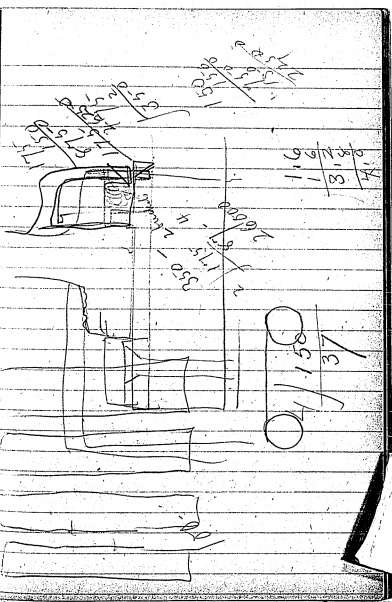
$$\begin{array}{r} 510 \\ \underline{179} \\ 689 \end{array}$$

$$\begin{array}{r} 150 \\ \underline{100} \\ 50 \\ \underline{100} \\ 150 \\ \underline{100} \\ 50 \\ \underline{100} \\ 150 \\ \underline{100} \\ 50 \\ \underline{100} \\ 150 \end{array}$$

$$\begin{array}{r} 272 \\ \underline{170} \\ 102 \\ \underline{340} \\ 340 \end{array}$$

$$\begin{array}{r} 1350 \\ \underline{80} \\ 1270 \\ \underline{80} \\ 1190 \\ \underline{240} \\ 950 \end{array}$$

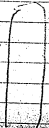
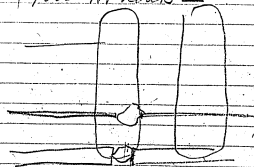
$$\begin{array}{r} 5062 \\ \underline{127} \\ 4935 \\ \underline{1328} \\ 3607 \\ \underline{314} \\ 3293 \end{array}$$



Oxide 35.00  
 Zinc 15.00  
 Frankfort 1.00  
 Furnace 5.00  
 Zinc Retorts  
 Rolls etc 2.00  
9.00

Day 10.000 -

1000 HP hours -



1000 HP - 16650 lbs oxide

\$3333 -

16650 Zinc

\$1500. -

80

Cells & gas - 1000 -

20.

16650 -

9

14985.0

50 - 3 HP hours

1

333

50

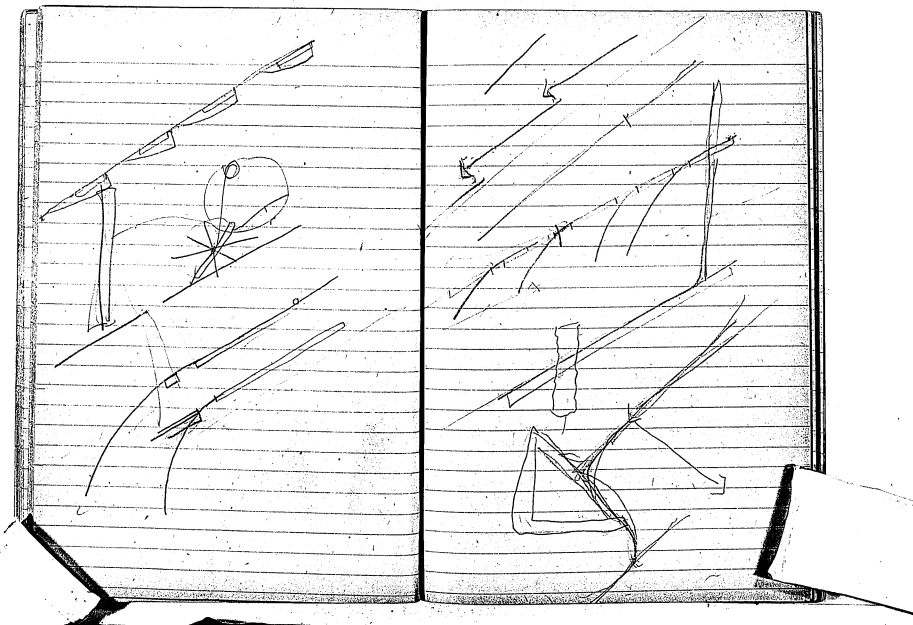
/1000

16650

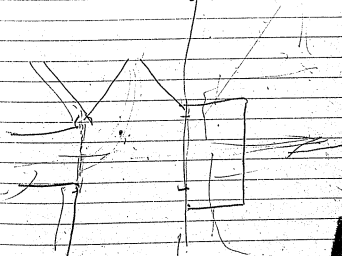
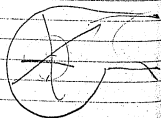
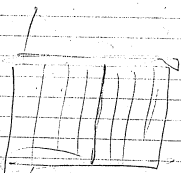
50

3330.00

Bailed







300-

60-

75-

100-

21

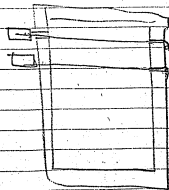
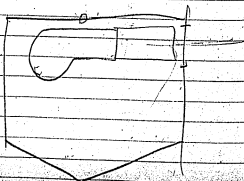
Q

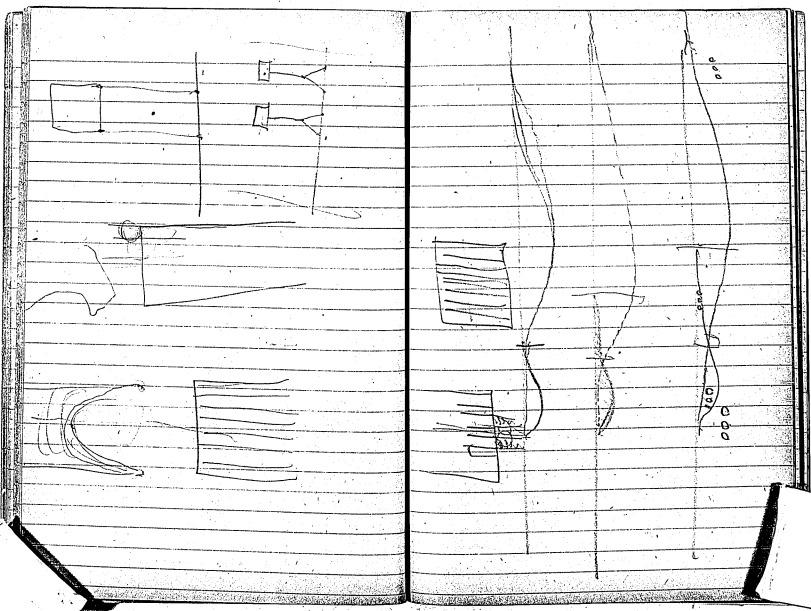
70

20,

60

18-21





1 ft 9 in 8 ft

$$\begin{array}{r} 144 \\ 8 \\ \hline 115 \end{array}$$

8.

690

1 HP hour

$$\begin{array}{r} 690 \\ 27 \\ \hline 25500 \\ 36900 \end{array}$$

$\frac{1}{10}$  ampere per sq in

1 plate 1 ft 9 in corrugated

$\frac{1}{8}$  to  $\frac{1}{4}$  center equals 8 ft

both side or 1152 inches

at 10 amp gives 115 amp

6 plates to give 690 ampere -

4 feet  $2 \frac{1}{2}$  HP



25 50

73

35

35

72



36 new Model

3 sides  
6 Stride

6. 10. 15. 20. 25. 30.

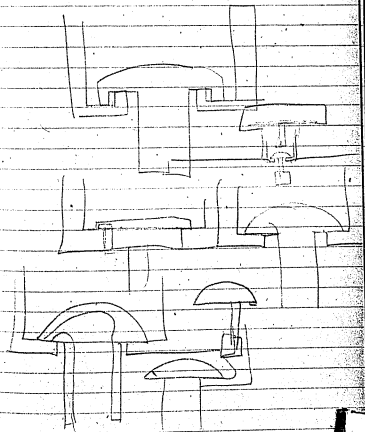
17. 8. 12. 16. 20. 24. 28. 32.

30 33

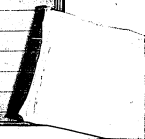
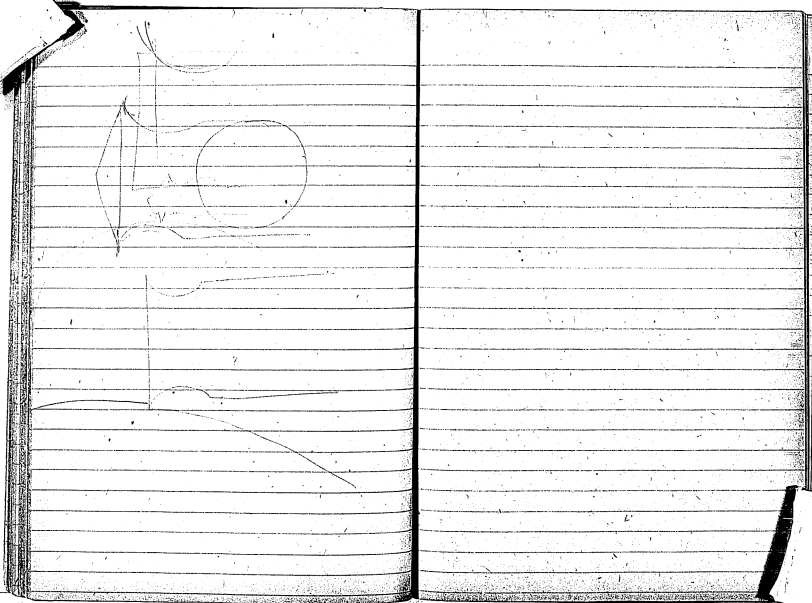
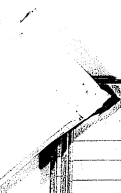
5 54

|    |     |     |     |
|----|-----|-----|-----|
| 70 | 176 | 124 | 62  |
| 30 | 80  | 40  | 20  |
|    | 40  | 20  | 10  |
|    | 20  | 10  | 5   |
|    | 10  | 5   | 2.5 |

|    |    |    |    |
|----|----|----|----|
| 5  | 16 | 16 | 16 |
| 25 | 25 | 25 | 25 |
| 8  | 10 | 10 | 10 |
| 80 |    |    |    |









Hints -

The screens in Rescreening  
building will be 16 high  
arranged same way as over

3 High - The building should  
be 25 ft longer than necessary  
to get in

2000 - 1000 lbs batly  
600 lbs load

3600 lbs -

50 watts per 1000 lb mile.

180 watts per mile -

$$\begin{array}{r} 36 \\ 50 \\ \hline 180 \end{array}$$

$$\begin{array}{r} 100 - 38 \\ 70 = 52 \\ 200 \\ \hline 1000 \end{array}$$

$$\begin{array}{r} 150 \\ 5040000 \\ 21000000 \end{array}$$

$$\begin{array}{r} 19 \overline{) 1000} \\ 95 \\ \hline 50 \\ 30 \\ \hline 800 \end{array}$$

$$\begin{array}{r} 135 \\ 52 \\ 30 \\ 170 \\ \hline 657 \end{array}$$

$$\begin{array}{r} 180 \overline{) 10400} \\ 900 \\ \hline 1400 \\ 1440 \end{array}$$

657 miles

$$\begin{array}{r} 190 \\ 140 \\ \hline 14 \end{array}$$

$$\begin{array}{r} 4500 \\ 25 \end{array}$$

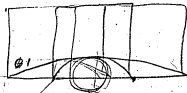
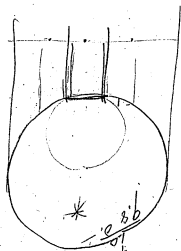
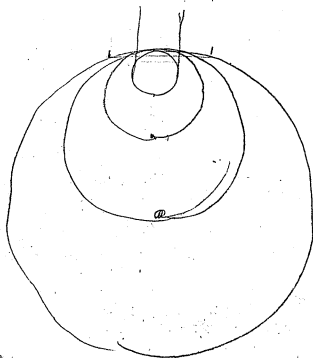
$$\begin{array}{r} 270 \\ 675 \\ \hline 900 \end{array}$$

$$\begin{array}{r} 225 \overline{) 14400} \\ 13500 \\ \hline 900 \end{array}$$

64 miles



[THIS BOOK WAS USED IN BOTH DIRECTIONS.  
THE FOLLOWING PAGES WERE FILMED FROM  
THE BACK END FORWARD.]



$$\begin{array}{r} 1250 \\ 1768 \\ \hline 3384 \\ 1616 \\ \hline 1724 \\ 1724 \\ \hline 3448 \\ 1616 \\ \hline 336 \end{array}$$

$$\begin{array}{r} 1250 \\ 1768 \\ \hline 3384 \\ 1616 \\ \hline 1724 \\ 1724 \\ \hline 3448 \\ 1616 \\ \hline 336 \end{array}$$

$$\begin{array}{r} 1274 \\ 1062 \\ \hline 2110 \\ 1062 \\ \hline 4250 \\ 4250 \\ \hline 8500 \end{array}$$

$$\begin{array}{r} 1250 \\ 1768 \\ \hline 3384 \\ 1616 \\ \hline 1724 \\ 1724 \\ \hline 3448 \\ 1616 \\ \hline 336 \end{array}$$

$$\begin{array}{r} 2618 \\ 202 \\ \hline 1196 \\ 35 \\ \hline 234 \\ 11 \\ \hline 112400 \\ 119300 \\ \hline 56900 \\ 619360 \end{array}$$

$$\begin{array}{r} 14 \\ \hline 36 \\ \hline 117 \\ 113 \\ 119 \\ 36 \\ \hline 679 \\ 73 \\ 870 \\ \hline 20. \\ 234 \\ 85 \\ \hline 149 \\ 234 \end{array}$$

$$\begin{array}{r} 14 \\ \hline 36 \\ \hline 117 \\ 113 \\ 119 \\ 36 \\ \hline 679 \\ 73 \\ 870 \\ \hline 20. \\ 234 \\ 85 \\ \hline 149 \\ 234 \end{array}$$

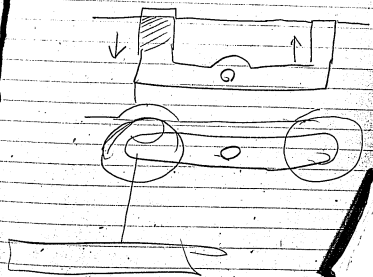
$$\begin{array}{r} 14 \\ \hline 36 \\ \hline 117 \\ 113 \\ 119 \\ 36 \\ \hline 679 \\ 73 \\ 870 \\ \hline 20. \\ 234 \\ 85 \\ \hline 149 \\ 234 \end{array}$$

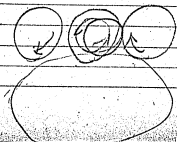
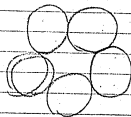
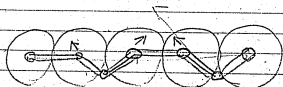
$$\begin{array}{r} 10 \\ 5 \\ 274 \\ 288 \\ 294 \\ \hline 234. \\ 36\% \end{array}$$

Credits - 2 1/2 - 6  
 2nd 36.  
 1st 36.  
 amt  
 amt  
 amt.  
 2 1/2 6  
 288  
 274  
 234. 36%

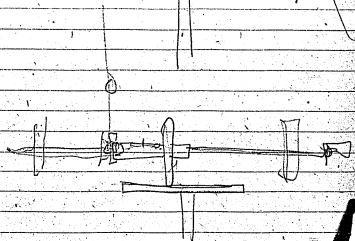
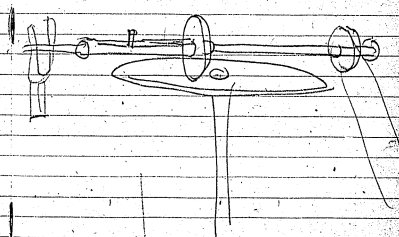
$$\begin{array}{r} 10 \\ 5 \\ 274 \\ 288 \\ 294 \\ \hline 234. \\ 36\% \end{array}$$

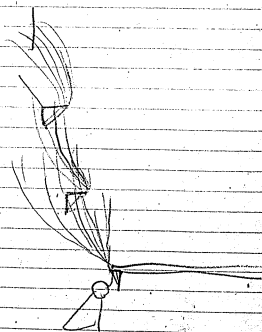
To meet the but t bar  
50 lbs each







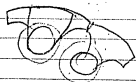




ol

st

ol



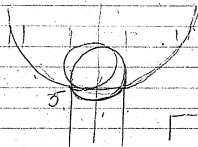
24-

500-

$$\begin{array}{r} 12 \overline{) 250} 20 \\ \underline{240} \phantom{0} \\ 10 \phantom{0} \end{array}$$

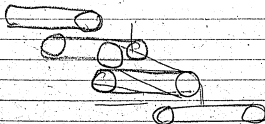
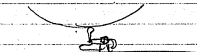
60

$$\begin{array}{r} 110 \\ \underline{20} \\ 220- \end{array}$$



40

5 20



25800  
 500 Supplies etc  
 150  
 26450      360 day

Dec 1st payroll - 10800  
 Supplies etc      26450  
 37250

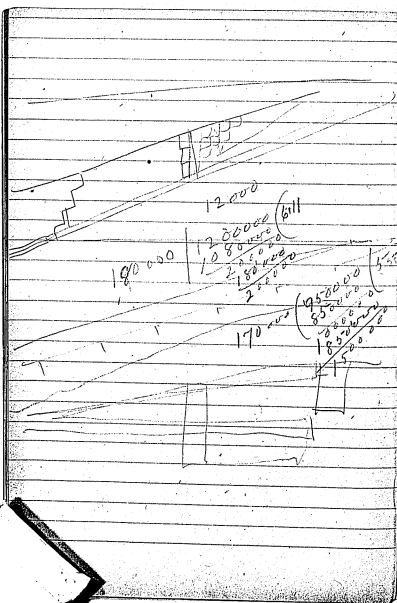
Jan 1st      37250  
 10800  
 48050      48050

Feb      48050      360  
 10800      28  
 58850      298  
                  4200  
                  7008

Mchs      58850  
 10800  
 69650      69630

apl      69650  
 10800  
 80450      80400

50  
 50  
 200  
 60  
 11  
 12000-  
 500-  
 500-  
 400-  
 300-  
 500-  
 2900-  
 400-  
 3500  
 1000  
 1200  
 500  
 500  
 75  
 125  
 100  
 75250  
 400  
 460  
 25  
 40  
 135  
 84  
 90  
 30  
 332  
 25  
 75



$500$   
 $200$   
 $150$   
 $675$   


---

 $525$   
 $2400$   
 $3925$   


---

 $611$   
 $558$

$135$   
 $675$   


---

 $200$   
 $150$   
 $350$

$11500$   
 $350$   
 $1050$   
 $350$   


---

 $5710$

$135$   
 $870$   
 $230$   
 $100$

240

5/20. Sunday

$400$   
 $2000$   
 $16000$

$15c$   
 $15c$

$95$   
 $170000$

$3160$   
 $53$   


---

 $3$

$1020$   
 $170000$   
 $170000$   
 $170000$   


---

 $103870000$

13 Lm 145  
 514 145  
 Top News 160  
 Fun 145  
 Mixer 145  
 Runway 145  
 Billion 290  
 Fun 145  
 Dump 290  
 95 196 145  
 Fun 200  
 Eng 300  
 Ovens 450  
 Fun 400  
 Gent 350  
 Muck 350  
 B 870  
 Mac 550  
 Top 145  
 Coal 270  
 Bunk 150  
 Clean 270  
 Osmf 200  
 Cont 300

Wagon per run  
 \$66-

6555

140 8 670  
 840 30 55 - 57 565  
 200  
 365

375 / 6600 / 176  
 375  
 2850  
 2625  
 2250  
 5000  
 4200  
 176  
 17.6  
 26.0  
 2555  
 365  
 700  
 2533

26  
 104  
 375 / 7000  
 67  
 600  
 375  
 250  
 700 tons / 25  
 20  
 60000  
 83600

35 / 700  
 3300  
 2450  
 240  
 7928

|               |        |              |              |
|---------------|--------|--------------|--------------|
| Drafting off  | 3      | 1200         |              |
| Furniture     | 3 1/2  | 450          |              |
| Power station | 12     | 2050         |              |
| Electric      | 7      | 1150         |              |
| Oil           | 5 1/2  | 700          |              |
| Blacksmith    | 10     | 1750         |              |
| Mechanics     | 40     | 8700         |              |
| Shower room   | 4      | 700          |              |
| Watch         | 4 1/2  | 700          |              |
| Hand          | 15     | 2250         | £            |
| CP            | 31 1/2 | 5705         | 5105         |
| Milk          | 37 1/2 | 6665         | 6665         |
| Rice          | 41 3/4 | 7140         | 7140         |
| Shower 3      | 27 3/4 | 4575         | 4575         |
| Juice         | 8      | 1400         | 1400         |
| Address       | 25     | 4900         |              |
| Missive       | 12 1/2 | 2000         |              |
| Shopper       | 12 1/2 | 2100         |              |
| Paint         | 1      | 200          | 200          |
| Demerol       | 6      | 1500         | 1500         |
| Gen. stuff    | 4      | 2700         | 2700         |
| Books         | 4      | 900          | 900          |
| Gravel        | 6      | 1000         | 1000         |
|               |        | <u>60160</u> | <u>51205</u> |
|               |        | 4            | 1            |
|               |        | 10           | 7            |

|             | man        | cost       |
|-------------|------------|------------|
| Day shift   | 307        | 601        |
| Night shift | 162        | 312        |
|             | <u>469</u> | <u>913</u> |

of which 45 is Italian

1000 67/20000 (298 - 20 net

1134

603

578

53

469

75

397 - net

2000

213

63

160

1300

80

50

13

2500

1000

1000

1000

1000

1000

1000

1000

1000

1000

1000

1000

1000

1000

1000

1000

1000

$$\begin{array}{r} 117 \\ 57 \\ \hline 53 \end{array}$$

$$\begin{array}{r} 800 \\ 59 \\ \hline 94 \\ 596 \\ \hline 535 \end{array}$$

$$\begin{array}{r} 595 \\ 54 \\ \hline 65 \end{array}$$

$$\begin{array}{r} 2950 \\ 6 \\ \hline 2956 \end{array}$$

$$\begin{array}{r} 63 \\ 56 \\ \hline 69 \end{array}$$

$$\begin{array}{r} 3296 \\ 351 \\ \hline 351 \end{array}$$

$$\begin{array}{r} 713 \\ 59 \\ \hline 772 \end{array}$$

$$\begin{array}{r} 4166 \\ 596 \\ \hline 496 \end{array}$$

$$\begin{array}{r} 654 \\ 54 \\ \hline 713 \end{array}$$

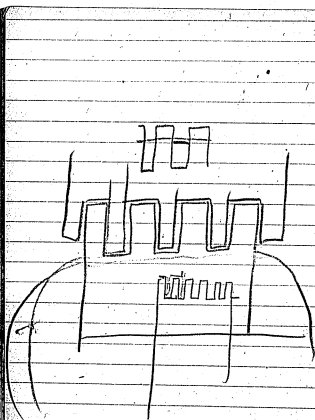
$$\begin{array}{r} 3570 \\ 546 \\ \hline 416 \\ 327 \end{array}$$

$$\begin{array}{r} 150 \\ 400 \\ \hline 550 \end{array}$$

$$\begin{array}{r} 200 \\ 596 \\ \hline 596 \end{array}$$

| 1000 tons | 67 | 20 net | 298 tons |
|-----------|----|--------|----------|
| 1200      |    |        | 357      |
| 1400      |    |        | 416      |
| 1600      |    |        | 476      |
| 1800      |    |        | 535.6    |
| 2000      |    |        | 595      |
| 2200      |    |        | 654      |
| 2400      |    |        | 713      |
| 2600      |    |        | 772      |
| 2800      |    |        | 831      |
| 3000      |    |        | 890      |
| 3200      |    |        | 949      |





100 watts per ton 1 mile  
 40 cells - 80 miles  
 or 1 ton 80 watts in 1% forward in battery -  
 50 cells  
 20 lbs per 1 mile per cent

$\frac{40}{10000}$   
 $\frac{80000}{10000}$

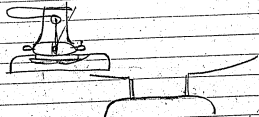
10

2

4

$\frac{75}{30}$

15



18.42

~~22,500~~ ~~300,000~~

~~39,000~~  
~~2,000~~

$\frac{150}{3}$   
 $\frac{45}{3}$   
 $\frac{50}{3}$   
 $\frac{22,500}{3}$

7  
 $\frac{50}{3}$   
 $\frac{35}{3}$   
 $\frac{25}{3}$   
 $\frac{20}{3}$   
 $\frac{15}{3}$   
 $\frac{10}{3}$   
 $\frac{6}{3}$   
 $\frac{4}{3}$   
 $\frac{2}{3}$   
 $\frac{1}{3}$   
129

360 tons

$\frac{1800}{60}$

200

$\frac{1600}{24}$   
 $\frac{200}{24}$   
 $\frac{200}{24}$

300

$\frac{21000}{24000}$   
 $\frac{1200}{28}$

24000

1200

67

$\frac{21000}{201}$   
 $\frac{39}{3}$   
 $\frac{33}{3}$   
 $\frac{33}{3}$   
 $\frac{33}{3}$

900  
2600

1300 sec

8000

10000

35000

6000

25000

12  
 $\frac{40}{20}$   
 $\frac{200}{20}$

85

6500  
650

$$\begin{array}{r}
 67 \overline{) 1200} \begin{array}{l} 18 \\ 21 \end{array} \\
 \underline{670} \phantom{0} \\
 530 \phantom{0} \\
 \underline{536} \phantom{0} \\
 90
 \end{array}$$

12550

$$\begin{array}{r}
 80 \\
 \underline{24} \\
 56 \\
 \underline{92}
 \end{array}$$

24

400-  
H

3000-

2mint

$$\begin{array}{r}
 240 \\
 \underline{8} \\
 1920-68
 \end{array}$$

1800

20  
3400  
3400  
20

530

2850

24

060

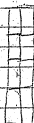
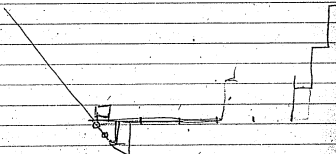
3800

$$\begin{array}{r}
 2 \overline{) 63800} \\
 \underline{31800} \\
 25
 \end{array}$$

250

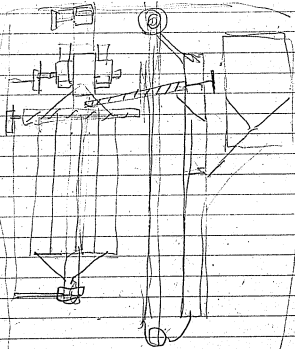
30000-

$$\begin{array}{r}
 100 \\
 15 \overline{) 600} \begin{array}{l} 60 \\ 4000 \\ 20 \\ 2000 \end{array} \\
 \underline{60}
 \end{array}$$





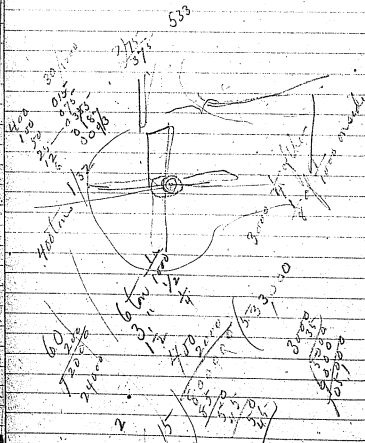
$$\begin{array}{r}
 8959 \\
 2046 \\
 \hline
 6914
 \end{array}$$



$$\begin{array}{r}
 248 \\
 22 \\
 12 \\
 22 \\
 150 \\
 9
 \end{array}$$

100  
 200  
 300  
 400  
 500  
 600  
 700  
 800  
 900  
 1000

100  
 200  
 300  
 400  
 500  
 600  
 700  
 800  
 900  
 1000



|   |     |
|---|-----|
| 2 | 300 |
| 2 | 600 |
| 2 | 300 |
| 6 | 900 |
| 6 | 900 |
| 2 | 300 |
| 1 | 175 |
| 3 | 500 |
| 2 | 270 |
| 1 | 175 |
| 1 | 150 |
| 1 | 135 |
| 1 | 100 |
| 1 | 300 |

5105

|   |     |    |      |
|---|-----|----|------|
| 1 | 150 | 1  | 300  |
| 1 | 150 | 2  | 375  |
| 1 | 175 | 1  | 135  |
| 2 | 350 | 1  | 350  |
| 1 | 180 | 2  | 350  |
| 1 | 600 | 1  | 300  |
| 2 | 350 | 2  | 300  |
| 2 | 350 |    |      |
| 1 | 175 | 37 | 6085 |
| 1 | 340 |    |      |
| 3 | 525 |    |      |
| 1 | 160 |    |      |
| 2 | 320 |    |      |
| 1 | 100 |    |      |
| 1 | 200 |    |      |
| 1 | 300 |    |      |
| 2 | 300 |    |      |

|             |       |
|-------------|-------|
| 4 Engs      | 20.00 |
| 4 film      | 12.00 |
| 2 Antfordi  | 6.00  |
| 4 Dryers    | 12.00 |
| 20 Screens  | 50.00 |
| 16 Blowers  | 48.00 |
| 4 oil       | 12.00 |
| 2 Engs      | 8.00  |
| 2 Halpax    | 6.00  |
| 6 fan       | 15.00 |
| 12 Woodcut  | 36.00 |
| 6 Halpac    | 15.00 |
| 4 Boxes     | 6.00  |
| 2 Linch     | 24.00 |
| 1 Wash mac  | 8.00  |
| 1 Supt      | 15.00 |
| 1 " " mlt   | 8.00  |
| Hand gang   | 20.00 |
| 10 Rizzum   | 48.00 |
| 4 office    | 18.00 |
| 2 Mich Engs | 16.00 |
| Truck gang  | 20.00 |

423.00

|                  |               |
|------------------|---------------|
| shoot - 8 gangs  | 200.00        |
| Loader gang - 12 | 36.00         |
| assys            | 423.00        |
|                  | 8.00          |
|                  | <u>667.00</u> |

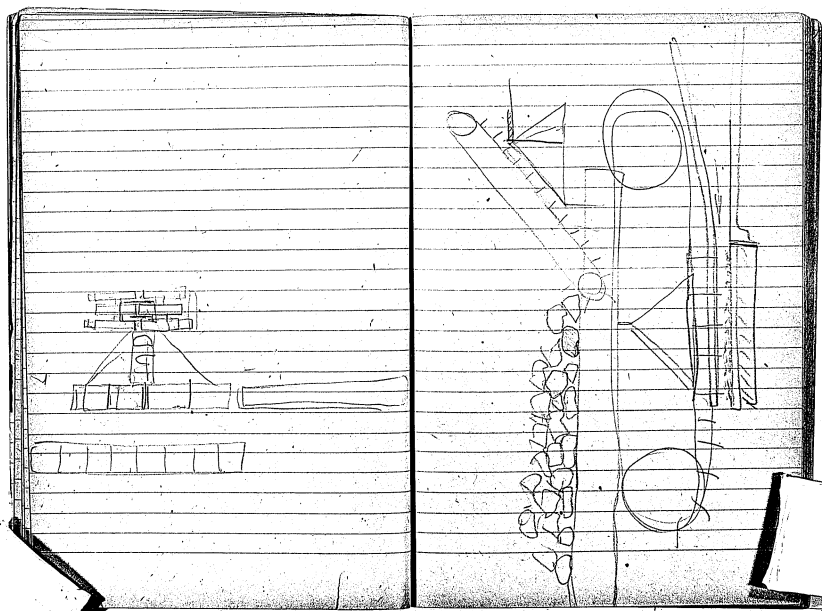
25  
700

|      |                |      |
|------|----------------|------|
| 5000 | 1669.00        | 13.3 |
|      | 3000.00        |      |
|      | 1257.00        |      |
|      | <u>1700.00</u> |      |

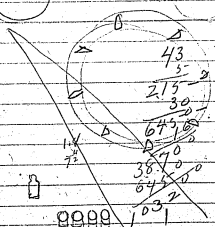
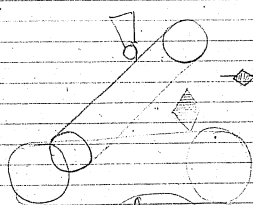
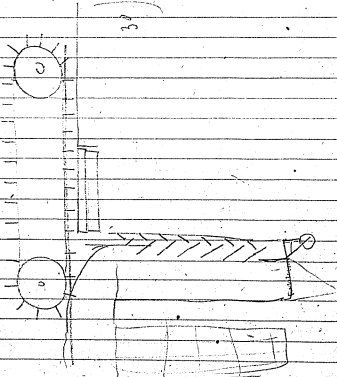
1450  
6

20

64



$\frac{45}{100}$   
 $\frac{100}{100}$   
 37



\$200    2    10    43  
                                  5  
                                  215  
                                  10



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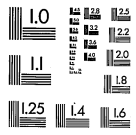
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